BOUNDS ON THE THERMODYNAMIC BEHAVIOR OF SYSTEMS WITH

GENERALIZED COULOMB INTERACTIONS

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ABSTRACT

Various lower and upper bounds are derived for the energy, free energy, and pressure of particles interacting via an r^{-n} pair potential in vdimensions. The results include (i) lower bounds for mixtures with n < 2 of the form const $\rho^{1+2/\nu}$ as ρ increases either isentropically or isothermally, complimenting a recently derived upper bound of Kleban and Puff of the same form in the isentropic case, (ii) upper and lower bounds for the case n > v, n > 2, which for the pressure are both of the form const $\rho^{1+n/\nu}$ as ρ increases isentropically.

We derive various upper and lower bounds for the energy, free energy, and pressure of a system of σ species of particles (including the $\sigma = 1$ case) such that the interaction potential for a pair of particles of species i and j is $e_i e_j / r^n$, where r is their separation and n is a positive constant.

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For n < 2, our lower bounds are complimentary to the recently derived upper bounds of Kleban and Puff,¹ and when combined with their bounds, show that if n < 2, the pressure p behaves under isentropic compression like $\rho^{1+2/\nu}$ for n < 2 and like $\rho^{1+n/\nu}$ for n > 2, where ρ is the density and ν the dimensionality. The case n = 1, ν = 3 describes "real" charged matter (which we shall treat nonrelativistically) and is thus of special interest. In particular, our large- ρ results in this case are relevant to stellar matter. To insure the existence of thermodynamic functions,² a chargeneutrality condition, $\Sigma_i e_i \rho_i = 0$, must be assumed³ in the n = 1, ν = 3, case (and perhaps for all n < ν , although existence proofs are currently lacking except for n = 1, ν = 3), where ρ_i is the density of the ith species; $\Sigma \rho_i = \rho$. For n > ν , on the other hand, one expects the existence of thermodynamic functions only if $e_i \ge 0$ for all i.

Our starting point is a form of the virial theorem 4

$$\frac{p}{\rho} = \frac{2}{\nu} u_{kin} + \frac{n}{\nu} (u - u_{kin})$$
(1)

where u is the total expected energy per particle, and u_{kin} the mean kinetic energy per particle. In classical mechanics, u_{kin} is $\frac{1}{2}vkT$. In quantum mechanics we do not know u_{kin} but we do have the lower bound

$$u_{kin} \ge g(\rho)$$
 (2)

where $g(\rho)$ is the energy per particle for the corresponding ideal gas in its ground state, which for Fermi statistics has the form⁵

$$g(\rho) = A_v \rho^{2/v}$$
(3)

with $A_{\nu} = A_{\nu}$ (ff, m;) independent of ρ .

We distinguish various cases, according to the value of n.

2

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Case I, n = 2.

Here (1) reduces to

$$\frac{p}{\rho} = \rho \left(\frac{\partial u}{\partial \rho}\right)_{S} = \frac{2u}{v}$$
(4)

where s is the entropy per particle. The general solution of this differential equation has the form

$$u = \rho^{2/\nu} h(s) \tag{5}$$

where h is nondecreasing since $(\partial u/\partial s)_{\rho}$ = T is non-negative, giving

s = function of
$$u \rho^{-2/\nu}$$
 (6)

or equivalently

$$f = \rho^{2/\nu} k(T \rho^{-2/\nu})$$
 (7)

where f is the Helmholtz free energy per particle and k must be a nonincreasing function in the quantum-mechanical case since $(\partial f/\partial T)_{\rho} = -s$ is negative. Thus the equation of state has the form

$$p = \rho^{1+2/\nu} \times \text{function of } T\rho^{-2/\nu}.$$
 (8)

In particular, these functional relations apply to the ideal Fermi, Bose, and classical gases, where $e_i \equiv 0$ for all i.

Case II, n < 2. Here Eqs. (1) and (2) imply

$$\frac{p}{\rho} \ge \frac{2-n}{\nu} g(\rho) + \frac{n}{\nu} u, \qquad (9)$$

i.e.

$$\frac{\partial(u\rho^{-n/\nu})}{\partial\rho} \geq \frac{2-n}{\nu} \rho^{-1-n/\nu} g(\rho),$$

so that

$$u = \rho^{n/\nu} \left[\frac{2-n}{\nu} \int_{0}^{\rho} \beta^{-1-n/\nu} g(\beta) d\beta + h(\rho,s) \right]$$
(10)

where h is a nondecreasing function of ρ at fixed s. It is also a nondecreasing function of s at fixed ρ . For Fermi statistics (10) reduces, by (3), to

$$u = A_{v} \rho^{2/v} + \rho^{n/v} h(\rho, s) , \qquad (11)$$

To obtain the corresponding result for f we consider it as a function of ρ and $\tau = T\rho^{-n/\nu}$ and note $(\partial f/\partial \rho)_{\tau} = (\partial f/\partial \rho)_{T} + (\partial f/\partial T)_{\rho} (\partial T/\partial \rho)_{T} = p/\rho^{2} - nsT/\nu\rho$. Thus (9) can be written $(\partial f\rho^{-n/\nu}/\partial \rho)_{\tau} \ge (2-n)\rho^{-1-n/\nu}g(\rho)/\nu$, and it follows that

$$f = \rho^{n/\nu} \left[\frac{2-n}{\nu} \int_{0}^{\rho} \rho^{-1-n/\nu} g(\rho) d\rho + k(\rho, T\rho^{-n/\nu}) \right]$$
(12)

where k is a nondecreasing function of ρ at fixed $T\rho^{-n/\nu}$. In the quantummechanical case k is also a nonincreasing function of $T\rho^{-n/\nu}$ at fixed ρ ; k is then also a nondecreasing function of ρ at fixed T. For Fermi statistics, (12) reduces to

$$f = A_{v} \rho^{2/v} + \rho^{n/v} k(\rho, T \rho^{-n/v}).$$
 (13)

A lower bound for the pressure can be obtained by substituting (10) into (9). By (3), this gives for Fermi systems

$$\frac{p}{\rho} \ge \frac{2}{\nu} A_{\nu} \rho^{2/\nu} + \frac{n}{\nu} \rho^{n/\nu} h(\rho, s).$$
 (14)

For quantum systems a lower bound on p in terms of T rather than s can be obtained by using $u - f = T_S \ge 0$ with (12) in (9). In the Fermi case, from (3), this gives

$$\frac{p}{\rho} > \frac{2}{\nu} A_{\nu} \frac{2/\nu}{\nu} + \frac{n}{\nu} \rho^{n/\nu} k(\rho, T \rho^{-n/\nu}).$$
(15)

Eq. (14) has the consequence that if at least one of the species present obeys Fermi statistics and the system is compressed isentropically, then $p = \frac{2}{\nu} A_{\nu} \rho^{1+2/\nu}$ cannot decrease. Thus p increases at least as fast as const $\rho^{1+2/\nu}$. On the other hand Kleban and Puff¹ show that p has an isentropic upper bound of the form $\rho^{1+2\nu}$ (const + const $\rho^{-(2-n)/\nu}$). Thus p must behave like $\rho^{1+2/\nu}$, in the sense that p $\rho^{-1-2/\nu}$ has positive upper and lower bounds as the system is compressed isentropically.

Case III, n > 2.

Here Eqs. (1) and (2) imply

$$\frac{p}{\rho} \leq \frac{n}{\nu} \quad u - \frac{n-2}{\nu} g(\rho)$$

and the method previously used above now gives upper rather than lower bounds. For example, in place of (14) we now have, for all statistics

$$\frac{p}{\rho} \le \frac{n}{\nu} \rho^{n/\nu} H(\rho, s)$$
(16)

where $H(\rho,s)$ is a nonincreasing function of ρ at fixed s. (We have left out the term that comes from $g(\rho)$ since it no longer dominates for large ρ , the case of greatest interest.) Similarly, for the free energy we have

 $f < \rho^{n/\nu} K (\rho, T \rho^{-n/\nu})$

where K is a nonincreasing function of ρ at fixed $T\rho^{-n/\nu}$.

Case IV, n > v

It is also possible to obtain lower bounds if n > v, where for simplicity we consider only the single-species case, dropping the subscript 1 on e_1 . The method depends on a lower bound for the potential energy per particle $u_{pot} = u - u_{kin}$. To obtain this lower bound, let ω_i denote the volume of the polyhedron comprising all points that are closer to the ith particle than to any other. Then if R_i denotes the distance from the ith particle to its nearest neighbor we have (ignoring surface effects) $\omega_i \geq K_v R_i^v$ where the right-hand side is the volume of a v-dimensional sphere of diameter R_i . The potential energy N u_{pot} therefore satisfies

$$N u_{\text{pot}} \ge \sum_{i} \frac{e^2}{R_i} \ge e^2 \sum_{i} \frac{(\frac{K_v}{\omega_i})^{n/v}}{(\frac{\omega_i}{\omega_i})}.$$
 (17)

From Holder's inequality $\Sigma |x_i y_i| \leq (\Sigma |x_i|^p)^{1/p} (\Sigma |y_i|^q)^{1/q}$, (1/p + 1/q = 1), with $x_i = \omega_i^{n/(n+\nu)}$, $y_i = 1/x_i$, $p = (\nu+n)/n$, we obtain $u_{pot} \geq e^2(K_{\nu}\rho)^{n/\nu}$, and so (1) gives

$$\frac{p}{\rho} = \frac{2}{\nu} u + \frac{n-2}{\nu} u_{\text{pot}} > \frac{2}{\nu} u + \frac{n-2}{\nu} B_{\nu} \rho$$
(18)

where $B_v = e^2 K_v^{n/2}$. Integrating by the same methods used on (10), we obtain

$$u = B_{v} \rho^{n/v} + \rho^{2/v} \dot{h}(\rho, s)$$
 (19)

$$f = E_{\nu} \rho^{n/\nu} + \rho^{2/\nu} \hat{k}(\rho, T \rho^{-2/\nu})$$
(20)

$$\frac{p}{\rho} \ge \frac{n}{\nu} B_{\nu} \rho^{n/\nu} + \frac{2}{\nu} \frac{2/\nu}{h(\rho,s)}$$
(21)

$$\frac{p}{\rho} \ge \frac{n}{\nu} B_{\nu} \rho^{n/\nu} + \frac{2}{\nu} \rho^{2/\nu} \hat{k}(\rho, T \rho^{-2/\nu})$$
(22)

where \tilde{h} and \tilde{k} have the same monotonic properties as the h and k in (10) and (12). From these and upper bounds such as (16) we can estimate the behavior of thermodynamic quantities as ρ is increased when n exceeds both 2 and ν . For example (16) and (21) together imply that under isentropic compression p behaves like $\rho^{1+n/\nu}$ in the sense that $p\rho^{-1-n/\nu}$ has positive upper and lower bounds.

Finally, in the case of classical mechanics with n > v we can obtain similar results for p_{ex} , u_{ex} and f_{ex} , the amounts by which p, u, and f exceed their ideal-gas values at a given density and temperature. The relevant form of (1) is $p_{ex}/\rho = (n/v) u_{ex}$ and by an analysis similar to that of Case I we find, for example,

$$f_{ex} = \rho^{n/\nu} K_{ex} (T \rho^{-n/\nu})$$
(23)

and consequently

$$u_{ex} = \rho^{n/\nu} L (T \rho^{-n/\nu})$$
 (24)

where

$$L(x) = K_{ex}(x) - x dK_{ex}(x)/dx.$$
 (25)

7

Since the excess specific heat is non-negative, L is a nondecreasing function; it follows by (25) that $d^2 L_{ex}(x)/dx^2 \leq 0$, and consequently, since $f_{ex}/T \neq 0$ as $T \neq \infty$, that K_{ex} is also a nondecreasing function. Both L and K_{ex} are positive. Equs. (24) and (25) show, for example, that f_{ex} and u_{ex} increase at most as fast as const $\rho^{n/\nu}$ as ρ increases at fixed T. The functional forms of the results (23) and (24) follow from general scaling considerations, and have been discussed before,⁶⁻⁸ while the nondecreasing character of $K_{ex}(x)$ and L(x) can be obtained from a straightforward use⁹ of the Gibbs-Bogoliubov inequality, instead of the arguments given above. From (23) we have $f \leq \text{const } \rho^{n/\nu}$ as ρ increases isothermally, a result that was also obtained previously¹⁰ by one of us using different methods.

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FOOTNOTES

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 E. has been printed with the wrong sign in Eq. (8) of this reference.
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