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Part 1. Synthesis and Bioassay of SNF Analogues

**Part 2. 1,2-Addition of Organometallic Reagents to Unprotected
Juglones and Progress Toward the Synthesis of Ravidomycins**

**Part 3. Synthesis of Cis-Alkenes via a Novel Copper-Mediated Cross-
Coupling of 1,1-Dibromoalkenes and Halides**

A Dissertation Presented

By

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The Graduate School
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Abstract of the Dissertation

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Part 1. Multidrug resistance (MDR), one of the major clinical problems in cancer chemotherapy, is a condition enabling a disease-causing organism to resist distinct drugs or chemicals of a wide variety of structure and function targeted at eradicating the organism. One of the mechanisms is an overexpression of P-glycoprotein, the product of the *mdr-1* gene in human cells. SNF (the code name of Snow Brand Milk Prod. Co.)

compounds showed a potent immunosuppressive activity in vitro by suppressing B-cell proliferation (induced by lipopolysaccharide) and T-cell proliferation (induced by concanavalin A) in non-cytotoxic concentrations. They also showed a multidrug resistance (MDR) reversal activity in tumor cells by inhibition of P-glycoprotein. In order to obtain MDR reversal agents with better activities, 14 SNF analogs were designed and prepared. During the bioassay of the SNF analogs, some of the compounds exhibited very good multidrug resistance (MDR) reversal effects. In particular, SNF analog **11** was the most active in modulating sensitivity to Paclitaxel (99% reduction in IC₅₀ at 1.0 μM in MCF7-R). SNF analog **12** also showed a good activity by reducing 99% of Paclitaxel IC₅₀. Moreover, a new strategy for the synthesis of the tricyclic core of spatol, a potent inhibitor of cancer cell replication, was developed based on the SNF bicyclo[4,2,0] structure.

Part 2. *p*-Quinols can undergo mild reductions to form *p*-phenols. Oxidations of *p*-phenols can reverse these reductions back to give corresponding *p*-quinols. Some biochemical compounds in nature contain *p*-quinol substructures, such as coenzyme Q. *p*-Quinols have a variety of uses principally associated with their actions as reduction or oxidation agents. We developed a new synthetic methodology for the preparation of quinol-containing nature products, in which organometallic reagents regioselectively added to the non-hydrogen bonding carbonyl groups. Ravidomycin is a potent antitumor and antibiotic, which has attracted the attention of chemists for many years. The new synthetic methodology for the synthesis of quinols was applied to the total synthesis of ravidomycin and a series of models were established. The total synthesis of ravidomycin is still in progress.

Part 3. Organo-copper reagents are very frequently applied in organic syntheses because of this selective reactivity and lower nucleophilicity for carbon. However,

organo-copper compounds are thermally unstable. Thus they can not undergo consecutive cross-coupling processes like some other transition organometals. A new highly stereoselective olefination methodology for preparation of cis-alkenes is developed based on a novel copper-mediated cross-coupling. This is the first example of consecutive cross-couplings of organocopper intermediates and aryl/vinyl halides in a one-pot reaction.

Dedicated to those whom I love and those who love me

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List of abbreviations

α	alpha
$[\alpha]$	specific optical rotation
β	beta
δ	delta, or chemical shift
γ	gamma
Δ	heat
σ	sigma bond
π	pi bond, or orbital
π^*	antibonding orbital
n	nonbonding orbital
\AA	angström
Ac	Acetyl
AcOH	Acetic acid
Ac ₂ O	Acetic anhydride
aq.	Aqueous
Ar	Aryl
Atm	Atmosphere
ATP	Adenosine triphosphate
b	broad
bd	Broad doublet
Bn	Benzyl
bp	Boiling point
bs	Broad singlet
t-Bu	<i>tert</i> -Butyl
n-BuLi	n-Butyllithium
Bz	Benzoyl
Calcd	Calculated
CDI	1,1'-carbonyldiimidazole
CI	Chemical ionization
cm ⁻¹	Reciprocal Centimeter
COSY	Homonuclear (¹ H- ¹ H) correlated spectroscopy
d	Doublet
DBU	1,8-Diazabicyclo[5.4.0]undec-7-ene
dd	Doublet of doublet
DIBALH	Diisobutylaluminum hydride
DIPA	Diisopropylamine
DIPEA	Diisopropylethylamine
DMAP	Dimethylaminopyridine
DMDO	Dimethyldioxirane
DME	Dimethoxyethane

DMF	N,N-Dimethylformamide
<i>ee</i>	Enantiomeric excess
EI	Electron-impact
Eq.	Equivalent(s)
Et	Ethyl
Et ₂ O	Diethyl ether
EtOAc	Ethyl acetate
g	Gram
h	Hour(s)
HMDS	1,1,1,3,3,3-Hexamethyldisilazane
HMPA	Hexamethylphosphamide
HOMO	Highest occupied molecular orbital
Hz	Hertz
IC ₅₀	Concentration for 50% inhibition
i-Pr	Isopropyl
IR	Infrared spectroscopy
<i>in vacuo</i>	Under vacuum
<i>J</i>	First order coupling constant of NMR
KHMDS	Potassium 1,1,1,3,3,3-hexamethyldisilazide
L	liter
LAH	Lithium aluminum hydride
LDA	Lithium diisopropylamide
LiHMDS	Lithium 1,1,1,3,3,3-hexamethyldisilazide
LUMO	Lowest occupied molecular orbital
m	Multiplet
MDR	Multi-drug Resistance
Me	Methyl
mg	Milligram
MHz	Megahertz
Min	Minute(s)
mL	Milliliter
mol	mole
mp	Melting point
MS	Mass spectrometry
Ms	Methanesulfonyl (mesyl)
<i>m/z</i>	Mass-charge ratio
NaHMDS	Sodium 1,1,1,3,3,3-hexamethyldisilazide
NMR	Nuclear magnetic resonance

p	pentet
Ph	Phenyl
Pip	piperidine
p-gp	P-glycoprotein
ppm	Parts per million
Py	Pyridine
q	Quartet
r.t.	Room temperature
s	singlet
t	triplet
TBAF	Tetra-N-butylammonium fluoride
TBS	Tert-Butyldimethylsilyl
Tf	Trifluoromethane sulfonate
THF	Tetrahydrofuran
TLC	Thin layer chromatography
TMS	Trimethylsilyl
Ts	para-Toluenesulfonyl (tosyl)
wt	weight

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Chapter 1. Synthesis and Bioassay of SNF Analogues

1.1 Introduction

1.1.1 Background of SNF 4435C and D

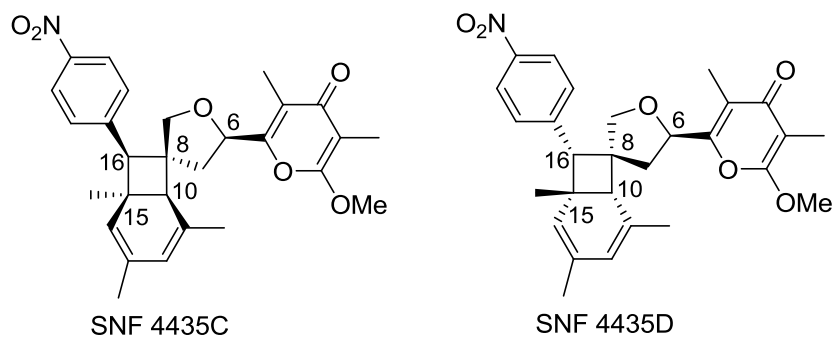


Figure 1. SNF 4435C and SNF 4435D

SNF 4435 C and SNF 4435 D, two diastereomeric natural products in which four of the five chiral centers have the same stereochemistry relative to each other and the fifth chiral center is epimeric, was discovered in the culture broth of a strain of *Streptomyces spectabilis* originally isolated from a soil sample collected in the main island of Okinawa, Japan in 2001 (Figure 1).¹ The structures of SNF 4435 C and D were elucidated to be novel nitrophenyl pyrones containing a tricyclic ring system by NMR. Their novel bicyclic[4,2,0] structures attract attention of many chemists. In 2004, Parker

group finished the first total syntheses of SNF 4435 C and D in which the key step was a tandem $8\pi/6\pi$ electrocyclization.²

1.1.2 Biological Activities of SNF 4435C and D

1.1.2.1 Immunosuppressive Activities

SNF 4435C and D not only own the novel structures which present interesting questions of biosynthetic origins and biomimetic synthetic approaches but also they exhibit immunosuppressive activity *in vitro* at submicromolar concentrations. Kurosawa and coworkers reported that SNF 4435C and D selectively suppressed B-cell proliferation induced by lipopolysaccharide (LPS) and T-cell proliferation induced by concanavalin A (Con A) in non-cytotoxic concentrations.^{3,4}

SNF 4435 C and D exhibited antimicrobial activities against *Candida albicans*, *Mycobacterium phlei*, and *Pyricularia oryzae*. In particular, SNF 4435D was active against *Bacillus subtilis*, *Staphylococcus aureus*, and *Trichophyton rubrum*. However, they did not show any remarkable cytotoxicities against KB, K-562, HL-60, and rat hepatocytes. In comparison with tacrolimus (FK-506)⁵, SNF 4435 C and D did not suppress the production of IL-2 but the productions of IL-6 and IFN- γ . In the experiment of the effects of delayed addition to the culture on the inhibition of mouse mixed lymphocyte reaction (MLR), the potent inhibitory activities of SNF 4435 C and D were still observed when FK-506 exhibited very little inhibitory activity.

The results suggested that the mechanism by which SNF 4435 C and D act is different from those of FK-506 and cyclosporin A (CsA)⁶ which have been developed as

pharmaceuticals. SNF 4435C and D have potential to be effective means for immunosuppressive therapy.⁵

1.1.2.2 Reversal of Multidrug Resistance

Multidrug resistance (MDR), one of the major clinical problems in cancer chemotherapy, is a condition enabling a disease-causing organism to resist distinct drugs or chemicals of a wide variety of structure and function targeted at eradicating the organism. One of the mechanisms is an overexpression of P-glycoprotein, the product of the *mdr-1* gene in human cells.⁷ Kurosawa and coworkers also reported that SNF 4435C and D effectively overcame MDR in an almost nontoxic dose range by direct binding to P-glycoprotein and inhibiting the pump activity.⁸ During their experiments, SNF 4435C and D completely reversed the resistance to vincristine (VCR) *in vitro* in several cell lines, such as mouse leukemia P388 (P388/VCR and P388/ADM, vincristine- and adriamycin-resistant), human myelogenous leukemia K562 (K562/VCR and K562/ADM, vincristine- and adriamycin-resistant) and human ovarian cancer A2780 (adriamycin-resistant AD10), at micromolar concentrations. In addition, SNF 4435C showed positive effects in vincristine-treated mice that bore VCR-resistant P388 leukemia. Furthermore, since the immunosuppressive activity of SNF 4435C and D might limit the clinical use of them for conquering MDR and non-immunosuppressive analogues of CsA exhibited abilities of reversing MDR, Kurosawa and coworkers suggested that non-immunosuppressive SNF4435 derivatives might be potential agents for cancer chemotherapy.

1.1.3 Drug Development for MDR Reversal

1.1.3.1 Multidrug Resistance (MDR)

Multidrug resistance (MDR) is the ability of a disease-causing organism to resist distinct drugs or chemicals of a wide variety of structure and function targeted at eradicating the organism. The organisms can be pathologic cells, including bacterial and neoplastic (tumor, cancer) cells. Thus, multidrug resistance is a serious clinical problem. Because efflux of drugs by the MDR proteins including P-glycoprotein (P-gp), breast cancer resistance protein (BCRP), lung resistance-related protein (LRP) and multidrug resistance protein (MRP-1) is a significant contributor to multidrug resistance in cancer cells, current research is aimed at inhibiting those specific efflux proteins. P-gp, BCRP, LRP, and MRP-1 are membrane-bound ATP-binding cassette transporters which utilize the energy of ATP hydrolysis to transport various substrates across membranes.⁹ These proteins have different selectivities for different drugs.¹⁰ People are interested in both selective inhibition¹¹ and broad spectrum inhibition of these proteins.¹²

1.1.3.2 The Structure of P-gp

P-gp is a 170 kDa transmembrane glycoprotein with a 10-15 kDa of N-terminal glycosylation site. Six transmembrane domains and a large cytoplasmic domain with an ATP-binding site constitute the N-terminal half of P-gp, and the other six transmembrane domains and an ATP-binding site constitute the C-terminal half of P-gp. The two sections share over 65% amino acid similarity.¹³ A number of recent contributions described the general features of the protein. A coherent presentation of the functioning of P-gp was recently published by Clarke and Loo.¹⁴ Seigneuret and Kerr respectively designed the different models for P-gp.^{15,16} On the basis of molecular modeling of known P-gp inhibitors and the competitive binding relationships, Orłowski et al. generated a model for part of the interior binding surface of P-gp.¹⁷ However the functioning of P-gp is

complex and still not well understood. A high resolution x-ray structure had not been produced until a structure of mouse P-gp, which has 87% sequence identity to human P-gp, was solved at 3.8 angstroms in 2009.¹⁸

A nucleotide-free inward-facing conformation arrayed as two “halves” with pseudo two-fold molecular symmetry in the plane of the bilayer was showed in the X-ray structure of P-gp (Figure 2). Two bundles of six transmembrane helices (TMs 1 to 3, 6, 10, 11 and TMs 4, 5, 7 to 9, 12) formed the inward-facing conformation which led to a large internal cavity, so that the nucleotide-binding domains (NBDs) were separated by ~30 Å.

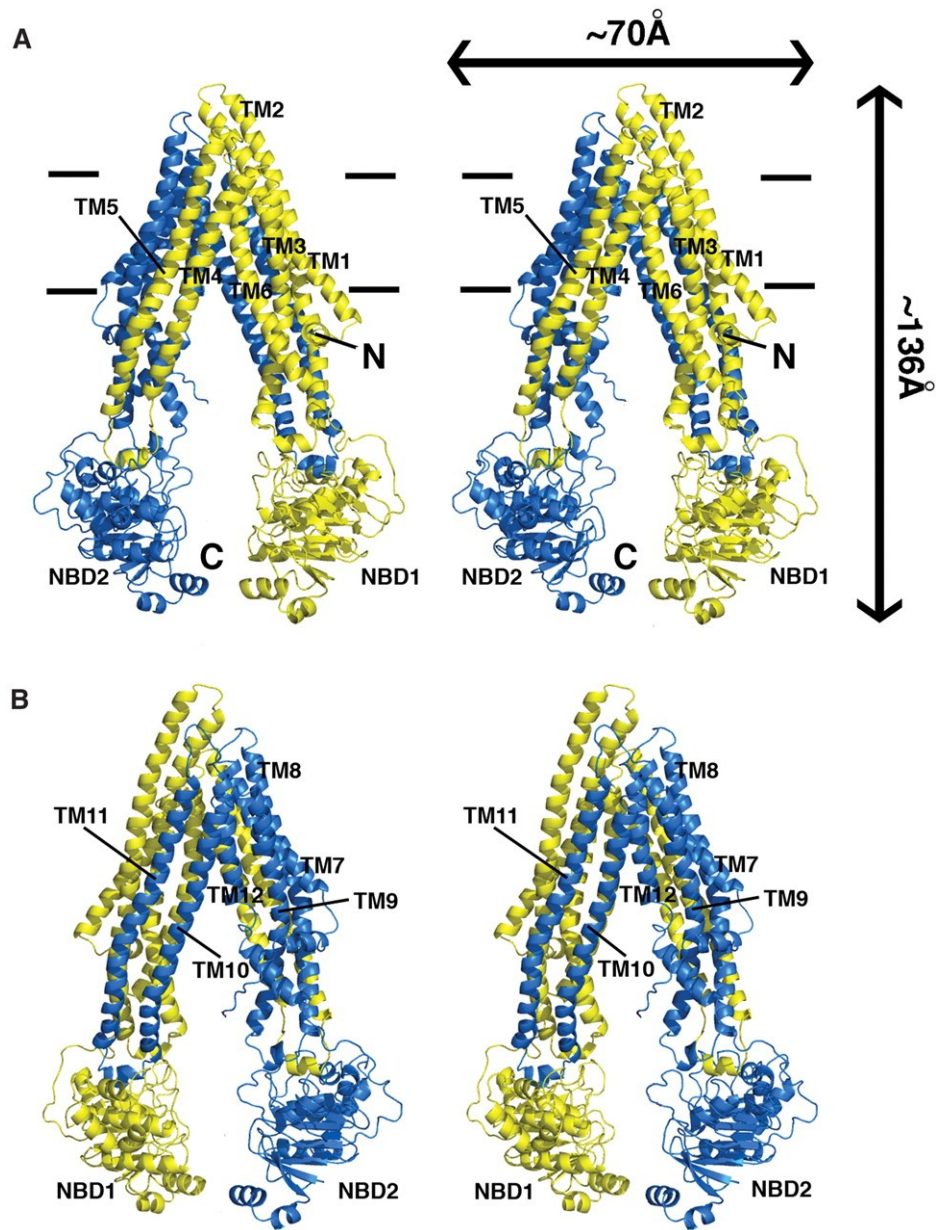


Figure 2. The X-ray Structure of P-gp.¹⁸ (A) Front and (B) back stereo views of P-gp.

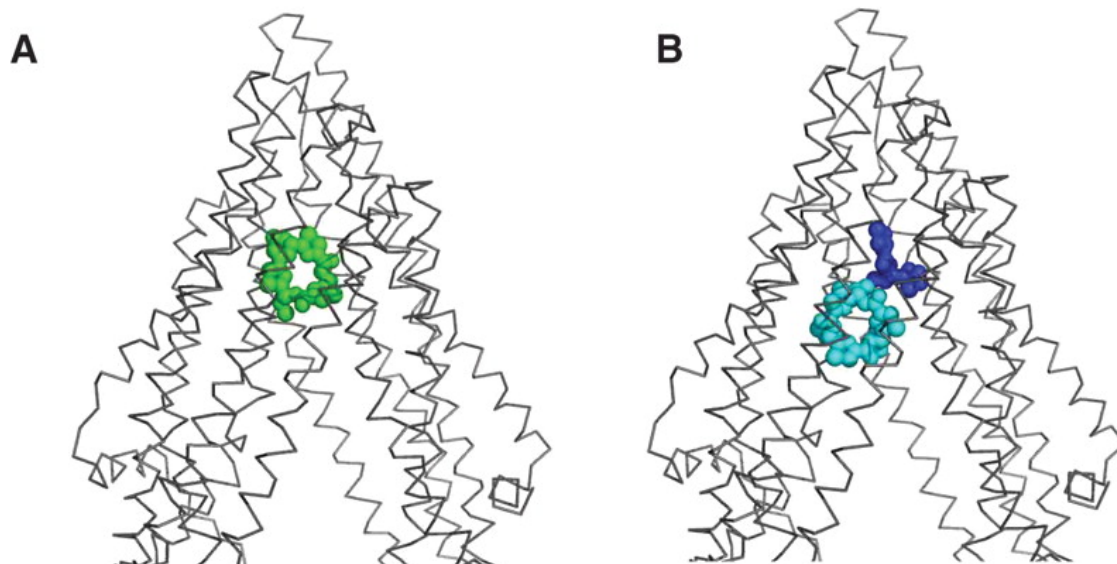


Figure 3. The X-ray Structure of P-gp.¹⁸ (A) Location of one QZ59-RRR (green spheres) and (B) two QZ59-SSS (blue and cyan spheres) molecules in the P-gp internal cavity.

In the cocrystal structures of P-gp with cyclic-valineselena (QZ59), the inward-facing conformation exhibited the capability to bind drugs (Figure 3). P-gp also showed the ability to distinguish between stereoisomers. The appearance of many residues that interacted with drugs in the drug-binding pocket suggests a common mechanism of poly-specific drug recognition (Figure 4). The upper half of the pocket is predominantly hydrophobic and aromatic; the lower half is polar and charged so that hydrophobic polar substrates might bind.

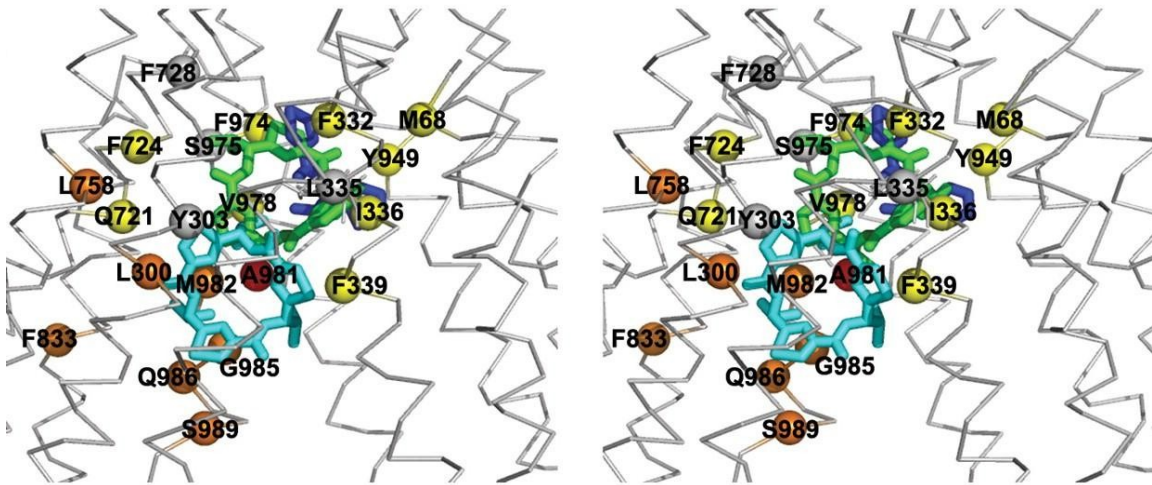


Figure 4. Stereo view of drug-binding residues of P-gp.¹⁸

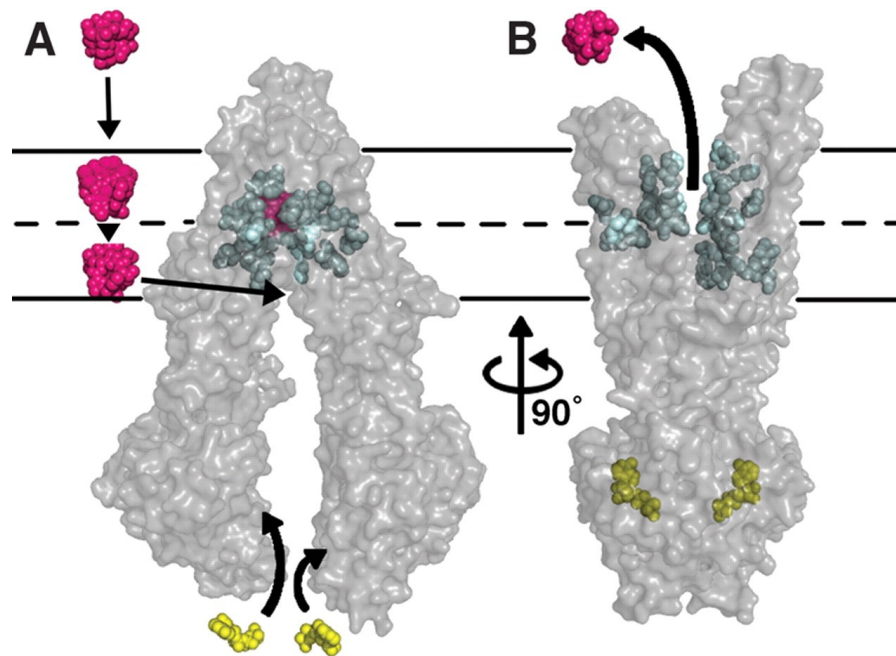


Figure 5. Model of substrate transport by P-gp.¹⁸

The drug-binding pocket of P-gp is much larger than those of some other multidrug transporters.¹⁸ In order to accommodate hydrophobic molecules and phospholipids in the pocket, the inward-facing conformation of P-gp has to open two portals wide. According to the literature,¹⁸ the substrate (magenta) entered the internal

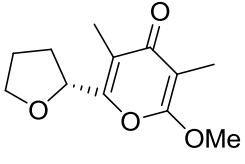
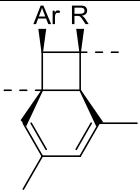
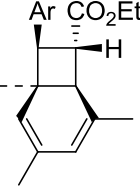
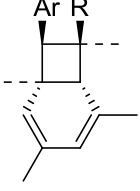
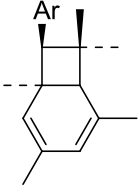
drug-binding pocket through the open portal and then bound with the residues (cyan) in the pocket during transportation (Figure 5). The binding of ATP (yellow) to the NBDs resulted in a big conformational change in which the substrate and drug-binding site(s) exposed to the outer leaflet or extra cellular space.

1.1.3.3 Inhibition of P-gp

Recently, since the importance of P-gp as drug target was realized, some inhibitors against P-gp have been accepted as drugs and a number of specific assays for P-gp inhibition have been developed. Because of SNF compounds' rigid and well understood structures, they provide a very attractive platform for rational drug design.

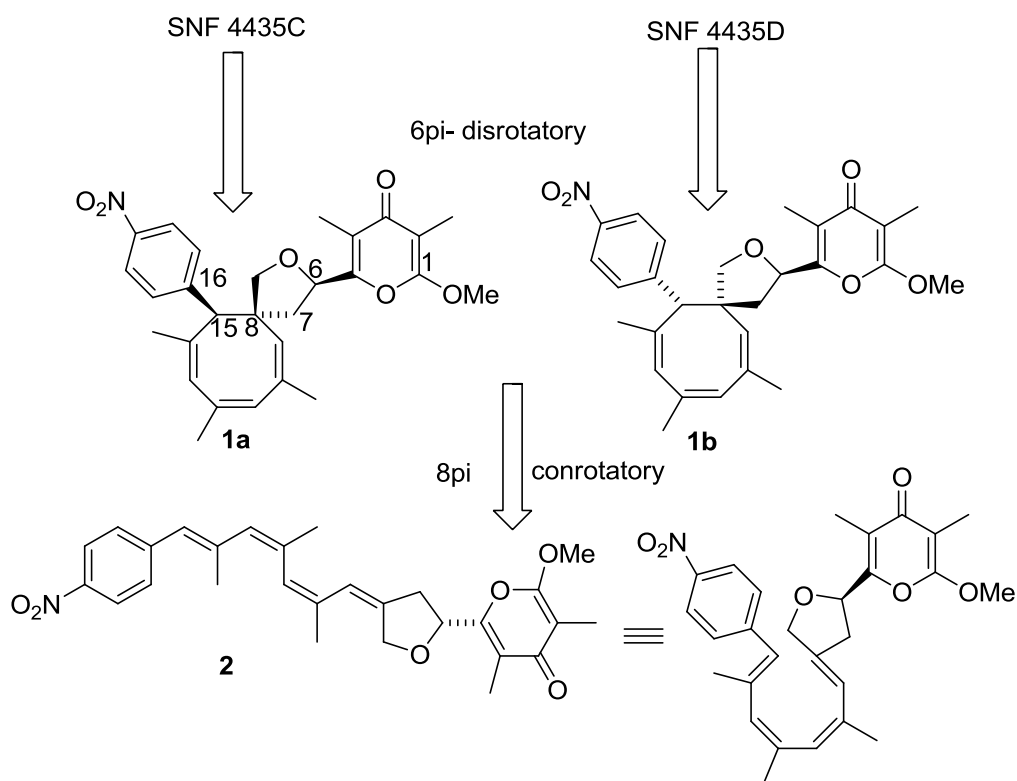
Seelig has approached the problem of a general P-gp pharmacophore by identifying two elements appearing on the surface of P-gp substrates.¹⁹ There were two hydrogen bond acceptor patterns termed type **I** and type **II**. It was claimed that type **I** units with a negative charge, including nitro groups, were unfavorable for the interaction with P-gp. Some structural features appear to be common to many of the substrates and inhibitors that interacted with P-gp. First, a pair of flat hydrophobic surfaces, close in space and with a distance to approximately 5 Å from the center of the hydrophobic region, appear to be important. Again, additional aromatic ring and hydrogen bond donors and acceptors are thought to be advantageous, leading to tighter binding.

Table 1. MDR Test Result.²⁰

Structure	R	IC50 (μM) MCF7-S	MCF7-R	With Paclitaxel (μM) ²	% Paclitaxel IC50 Reduction
		>100	>100	10 1.0	10 10
 Ar = <i>p</i> -nitrophenyl	CH ₂ OH	>50	>50	10 1.0	23 9
	CO ₂ Et	>10	>10	10 1.0	79 22
	CN			1.0	40
 Ar = <i>p</i> -nitrophenyl		>10	>10	10 1.0	90 26
 Ar = <i>p</i> -nitrophenyl	CH ₂ OH	>10	>10	10 1.0	49 5
	CO ₂ Et	>10	>10	10 1.0	46 17
	CN			1.0	12
 endo:exo 50:50 Ar = <i>p</i> -nitrophenyl		>10	>10	10 1.0	54 17

A number of SNF analogs were prepared by Yeon-Hee Lim and tested for MDR reversal (Table 1).²⁰ The IC₅₀ reduction values of some analogs were higher than those of SNF 4435C and D, but the differences are small. The results suggested that SNF compounds' rigid bicyclic structures were very important to inhibit P-gp. Because the action of these SNF analogs has now been validated as inhibition of P-glycoprotein, these compounds may provide an additional tool for the study of the target, for the design of improved inhibitors, and for the design of primary drugs which are not subject to efflux.

1.1.4 The Biomimetic Synthesis of SNF 4435 C and D



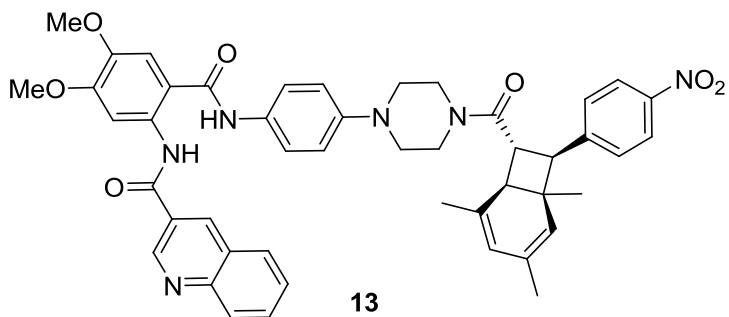
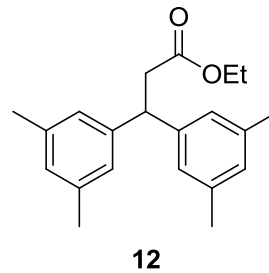
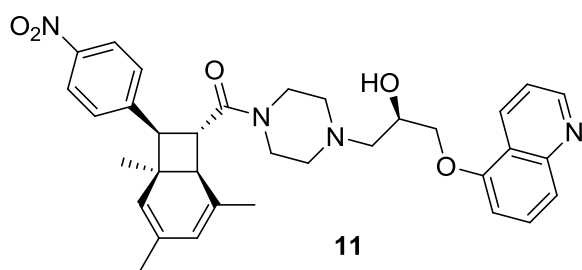
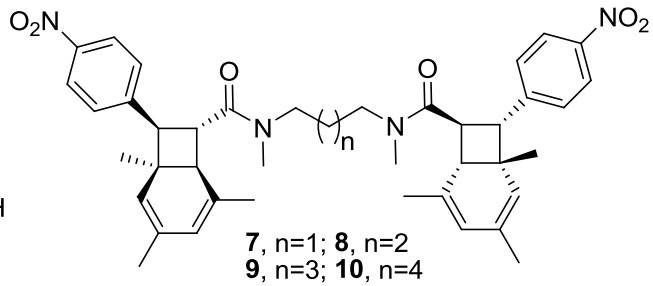
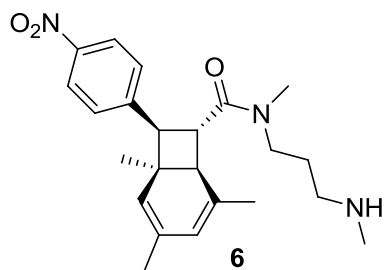
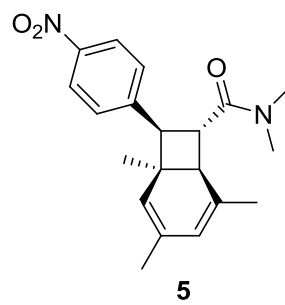
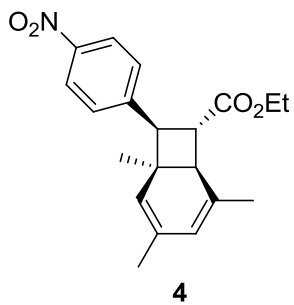
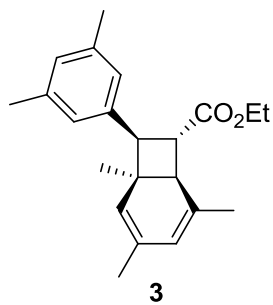
Scheme 1. Suggested biomimetic synthesis of SNF4435C and D.²

The bicyclo[4,2,0]-substructure of SNF4435C and D presents interesting questions of biosynthetic origins and biomimetic synthetic approaches (Scheme 1). The Parker group proposed and finished a biomimetic total synthesis of SNF4435C and D

based on the proposed biosynthesis.² The precursor **2**, a linear (1E,3Z,5Z,7E)-1,3,5,7-tetraene, could bend to a helical transition state. Therefore, it carried out a tandem $8\pi/6\pi$ electrocyclization in which a thermal 8π -conrotatory ring closure occurred first to form 2,4,6,- cyclooctatriene intermediates **1a** and **1b**. Because of the higher Gibbs free energies, these intermediates were individually transformed to SNF 4435C and D with lower Gibbs free energies by a thermal 6π -electrocyclic disrotatory ring closure at room temperature. The linear (1E,3Z,5Z,7E)-1,3,5,7-tetraene **2** was suggested to form by a polypropionate synthase from six units of propionyl coenzyme A, one acetyl coenzyme A, and p-aminobenzoic acid as the source of the nitrophenyl ring.^{2, 21}

1.2 Results and Discussion

Seelig mentioned that negatively charged groups were unfavorable for the interaction with P-gp.¹⁹ Therefore, the analog **3** with a non-polar 3,5-dimethylphenyl group was designed and synthesized for MDR reversal tests (Figure 6). Seelig also maintained that primary and secondary amines and amides didn't serve as recognition elements for P-gp.¹⁹ The previous SNF 4435 analogues made by Parker and Lim don't contain the commonly observed P-gp type **I** and type **II** hydrogen bond acceptors. Furthermore, almost ubiquitous tertiary amine or amide moieties were absent. Thus it seems likely that tighter binders can be produced by adding functionality to the core structure. The SNF 4435 analog **5**, with tertiary amide group, was expected to have much better MDR reversal activity. In order to examine Seelig's hypothesis that primary and secondary amine and amide do not serve as recognition elements for P-gp,¹⁹ SNF analogue **6**, containing (3-aminopropyl)formamide at C8, was designed and synthesized. SNF 4435 analogue **4** was prepared first as the precursor to SNF analogues **5** and **6**.



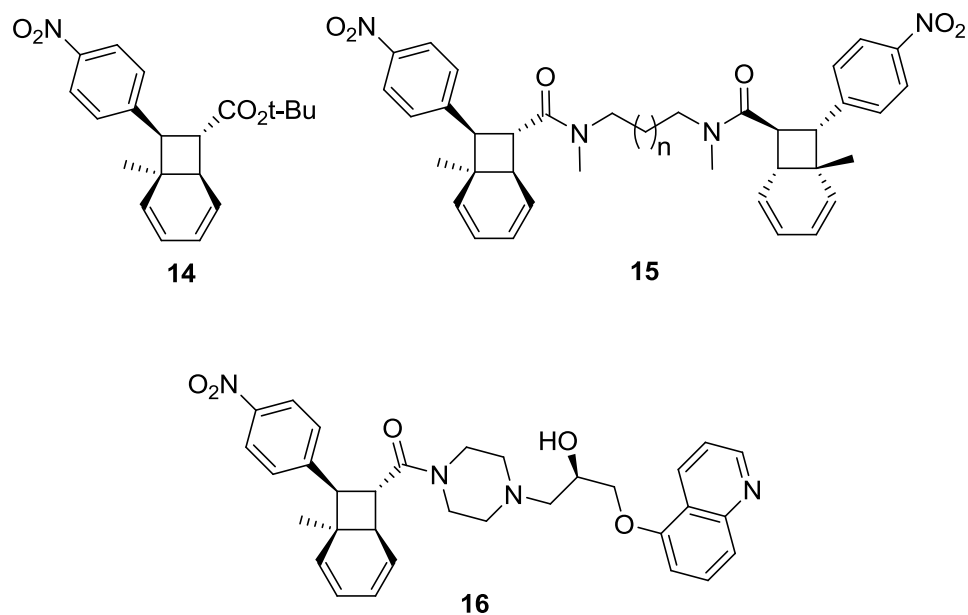


Figure 6. The SNF analogues designed and made.

SNF analogues **7**, **8**, **9**, and **10** contain two endo-bicyclo[4,2,0]-SNF substructures (Figure 6). At the same concentration, these compounds provide more active units than the previous SNF analogues. Thus, SNF analogues **7**, **8**, **9**, and **10** are expected to show outstanding MDR reversal activities.

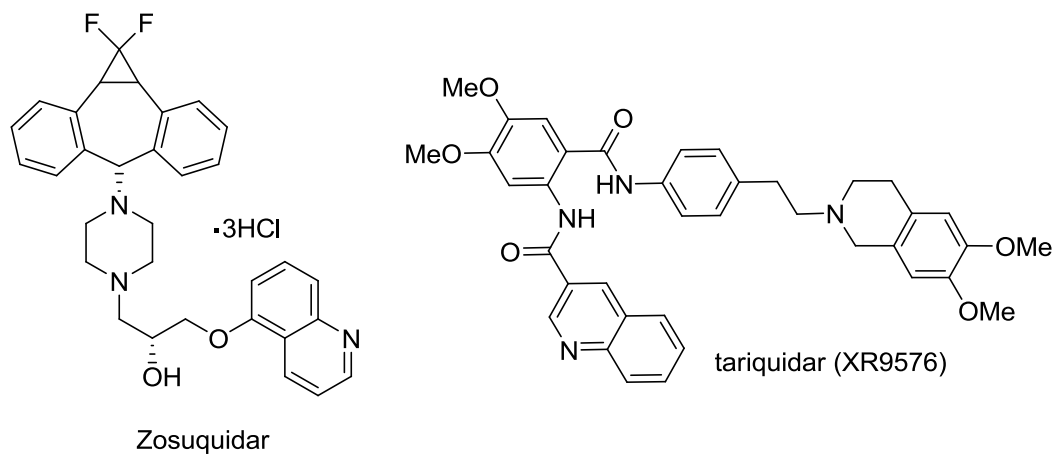


Figure 7. Zosuquidar^{22, 35} and Tariquidar⁴²

Zosuquidar^{22, 35} and Tariquidar⁴² are P-glycoprotein inhibitors undergoing research for treatment of cancer (Figure 7). The endo-bicyclo[4,2,0]-SNF substructures were coupled with the 1-piperazin-3-quinolinylxy-propanol unit of Zosuquidar (LY335979) and the arylpiperazinyl unit of Tariquidar (XR9576) derivatives, individually, to form SNF analogues **11** and **13**.

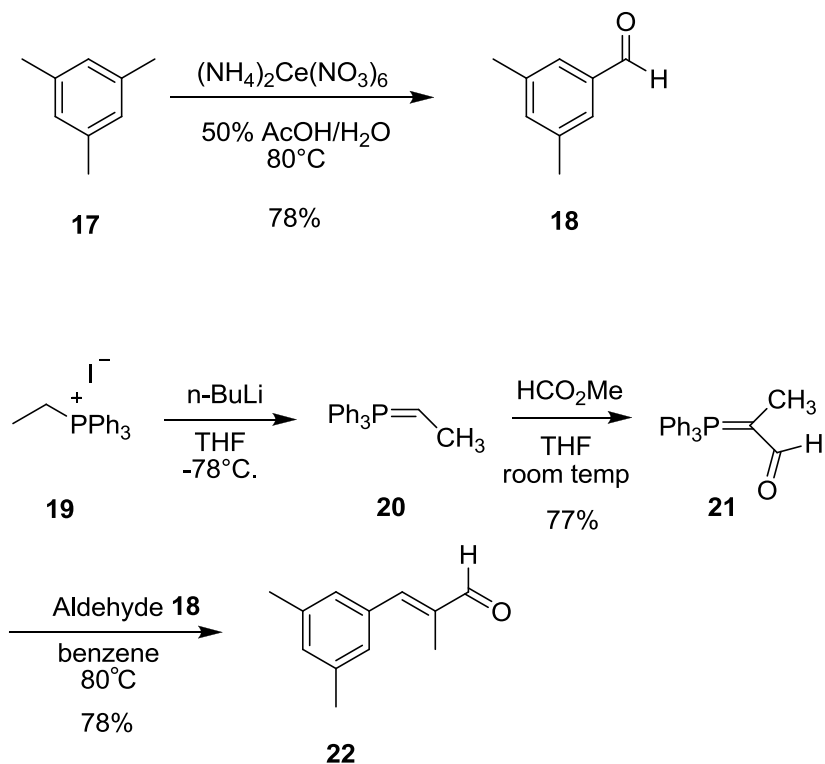
To improve the efficiency of organic synthesis of the SNF analogues, SNF analogue **12**, a mimic compound of SNF analogue **3**, was prepared (Figure 6); a new synthetic methodology for endo-bicyclo[4,2,0] substructure was applied to synthesize SNF analogue **14**. The new endo-bicyclo[4,2,0] substructures of SNF analogue **14** were connected by a diamide linker to form SNF analogue **15**. SNF analogue **16** consisting of a new endo-bicyclo[4,2,0] substructure and a 1-piperazin-3-quinolinylxy-propanol unit of Zosuquidar was prepared for MDR reversal tests.

1.2.1 The Synthesis of the SNF Analogue **3**

The analogue **3** in which the nitrophenyl group is replaced by 3,5-dimethylphenyl was designed and synthesized to examine Seelig's hypothesis of type **I** acceptor (Figure 6).¹⁹

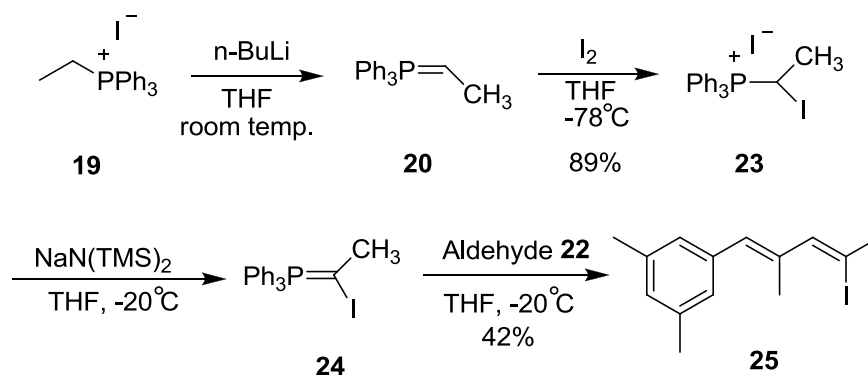
The synthetic strategy for the analogue **3** is coupling of iododiene and vinyl stannane as demonstrated by Parker and Lim.² The iododiene **25** was efficiently prepared by a Stork-Zhao reaction²² which preferred to give the (*Z*)-olefinic product. (*E*)-2-Methyl-3-(3,5-dimethylphenyl)-propenal **22**, the precursor of **25**, had not reported previously. We expected it was prepared by a cross aldol condensation or a Wittig reaction with aldehyde **18**. Gaspar et al. reported that ammonium cerium (IV) nitrate ((NH₄)₂Ce(NO₃)₆) could selectively oxidize one methyl group on an aromatic ring to an aldehyde.²³ Thus, compound **18** was prepared in 78% yield by oxidizing only one methyl group of

mesitylene **17** with $(\text{NH}_4)_2\text{Ce}(\text{NO}_3)_6$. The cross aldol condensation of the aldehyde **18** and propanal did not work. Thus the Wittig reaction preferring (E)-olefinic product became the better choice.



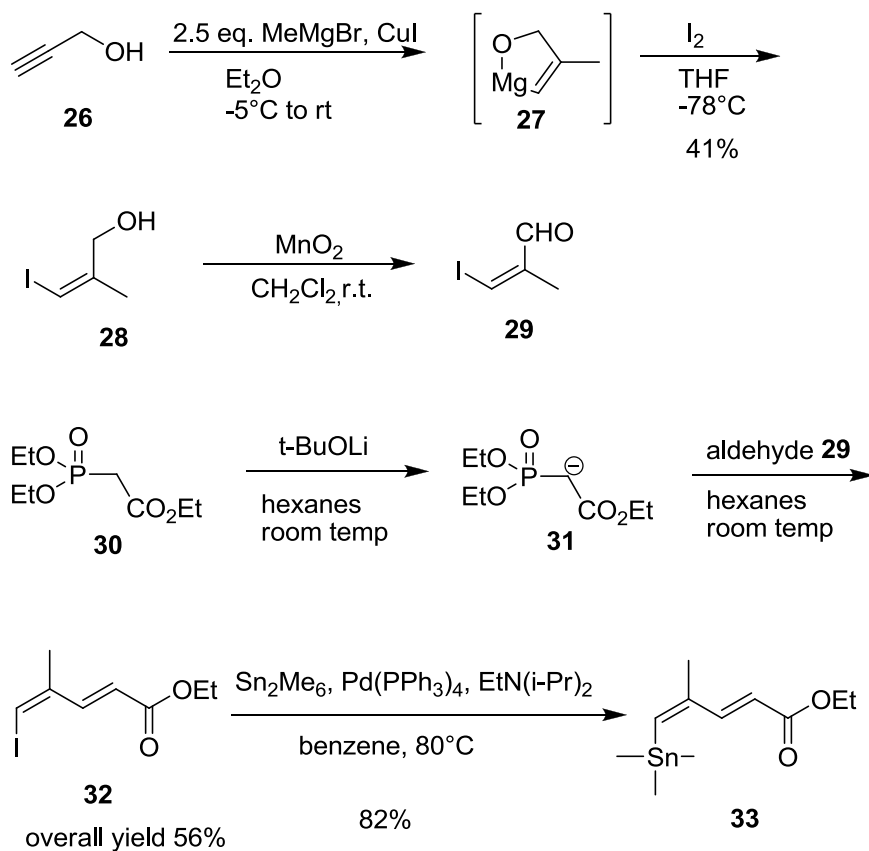
Scheme 2. Preparation of aldehyde **22**

The stable deprotonated Wittig reagent **21** was prepared in 77% yield from deprotonated ethyltriphenylphosphonium iodide **19** and methyl formate; a second Wittig reaction produced (E)- α,β -unsaturated aldehyde **22** in 78% yield (Scheme 2).²⁴ Then the Stork-Zhao reagent, (1-iodoethyl)triphenyl-phosphonium iodide **23**, was synthesized by deprotonation of compound **19** with n-butyllithium and subsequent halogenation of the resulted ylide **20**. Treatment of compound **23** with sodium hexamethyldisilazane at -20°C afforded the ylide **24**. Reaction with aldehyde **22** produced the desired iodoolefin **25** in 42% yield (Scheme 3).



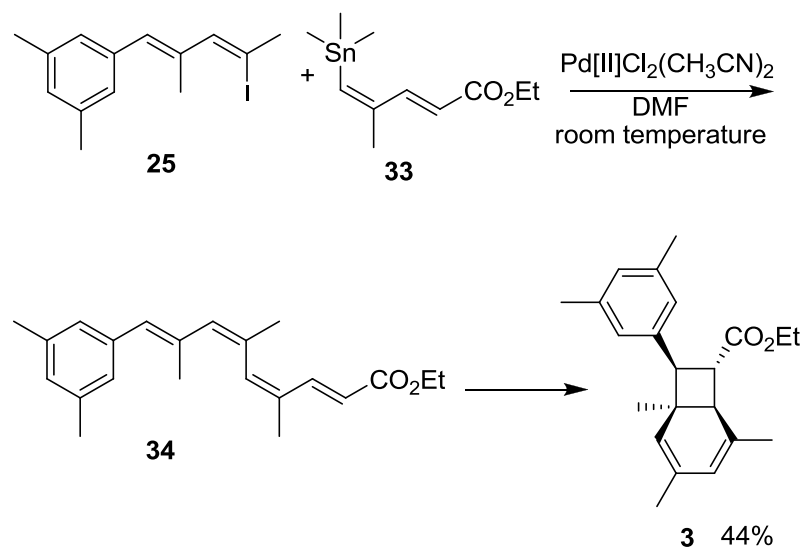
Scheme 3. Preparation of Iododiene **25**

The stannyl ester **33** was prepared as demonstrated by Parker and Lim.¹⁹ First, the (*Z*)-iodoallylic alcohol **28** was obtained in 42% yield from propargyl alcohol **26** through a stereoselective carbocupration/iodination sequence (Scheme 4).²⁵ The intermediate of this reaction was supposed to be a five-member ring **27** which controls the stereoselectivity of the reaction. Oxidation of (*Z*)-iodoallylic alcohol **28** with MnO₂ afforded a very unstable (*Z*)-iodovinyl aldehyde **29** which was transferred into a Horner-Wadsworth-Emmons reaction system as soon as possible after filtration and concentration.²⁶ In the immediate Horner-Wadsworth-Emmons reaction, Wadsworth-Emmons reagent **30** was deprotonated by lithium tert-butoxide (t-BuOLi) to produce an enolate which reacted with aldehyde **29** to give an iododiene **32** in overall 56% yield. This reaction specifically formed the (*E*)-isomer.²⁷ The (1*Z*,3*E*)-iododiene **32** was converted to the corresponding proton-sensitive vinyl stannane **33** in 82% yield by treatment with hexamethylditin under the Pd(0)-catalysis in benzene.



Scheme 4. Preparation of vinyl stannane **33**

A tandem process consisting of a cross coupling and electrocyclizations is shown in Scheme 5. The coupling reaction of vinyl stannane **33** and iododiene **25** in the presence of catalytic amounts of bisacetonitrile-palladium(II) chloride $(\text{CH}_3\text{CN})_2\text{PdCl}_2$ in diisopropylethylamine and DMF at room temperature afforded the intermediate **34** which underwent a $8\pi/6\pi$ electrocyclization to form the endo-bicyclo[4,2,0]-SNF analog **3** in overall yield 44%. This stereoselective tandem reaction only gave the endo product in which the phenyl unit and cyclohexadiene unit were cis on the cyclobutane ring

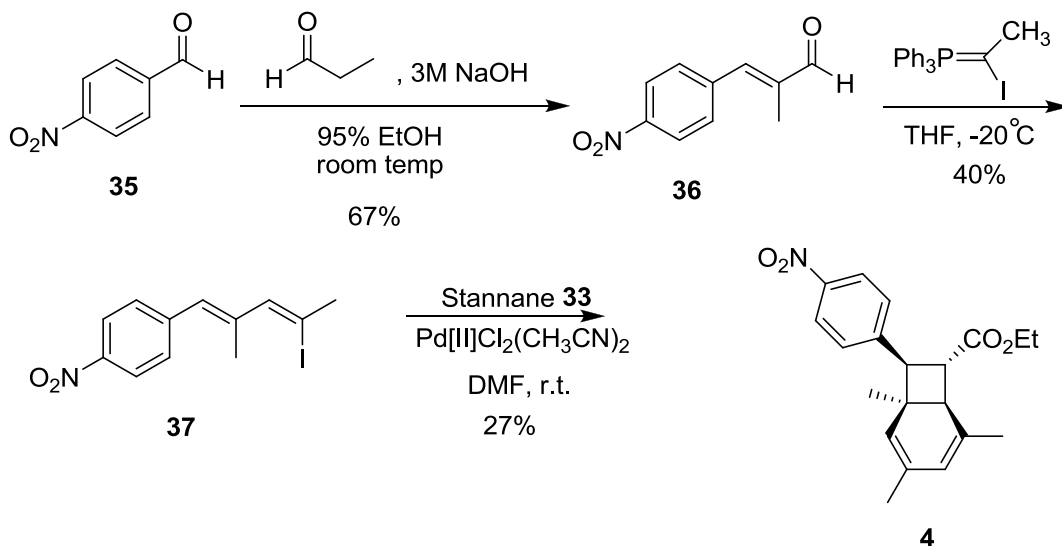


Scheme 5. Preparation of SNF analogue **3**.

1.2.2 The Synthesis of SNF Analogue **4**

According to Kurosawa, modified SNF analogues with lower immunosuppressive activity might be more effective for reversal of multiple drug resistance.⁸ In order to make various SNF analogues, SNF analogue **4** was prepared.

The iododiene **29** was also prepared by a Stork-Zhao reaction (Scheme 6). Under catalysis by sodium hydroxide, the cross aldol condensation of 4-nitro-benzaldehyde **27** and propionaldehyde specifically gave the (E)-1,3-unsaturated aldehyde **28** in 67% yield.²⁸ The ylide **24** reacted with aldehyde **36** to produce the desired iodoolefin **37** in 40% yield.



Scheme 6. Preparation of SNF analogue 4.

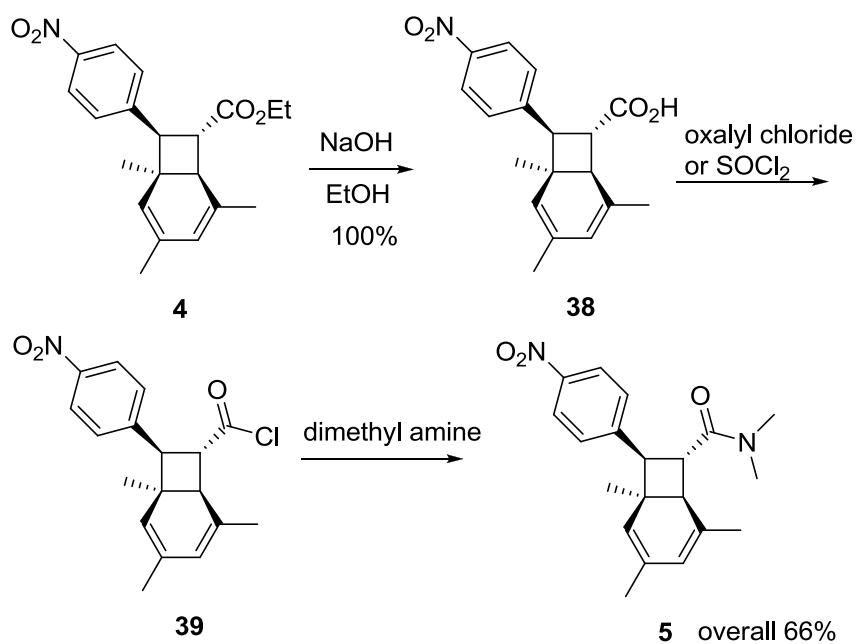
The Stille cross coupling reaction of vinyl stannane **33** and iododiene **37** in the presence of catalytic amounts of (CH₃CN)₂PdCl₂ in diisopropylethylamine and DMF at room temperature stereospecifically afforded the endo-bicyclo[4,2,0] SNF analogue **4** in overall yield 27%.

1.2.3 The Synthesis of SNF Analogue 5

Seelig maintained that primary and secondary amines and amides do not serve as recognition elements for P-gp.¹⁹ Analogues **3** and **4** do not contain the commonly observed P-gp type **I** and type **II** hydrogen bond acceptors. Also common tertiary amine or amide moieties were absent. Thus it seems likely that tighter binders can be produced by adding functionality to the core structure. SNF 4435 analogue **5** with tertiary amide group was expected to have much better MDR reversal activities.

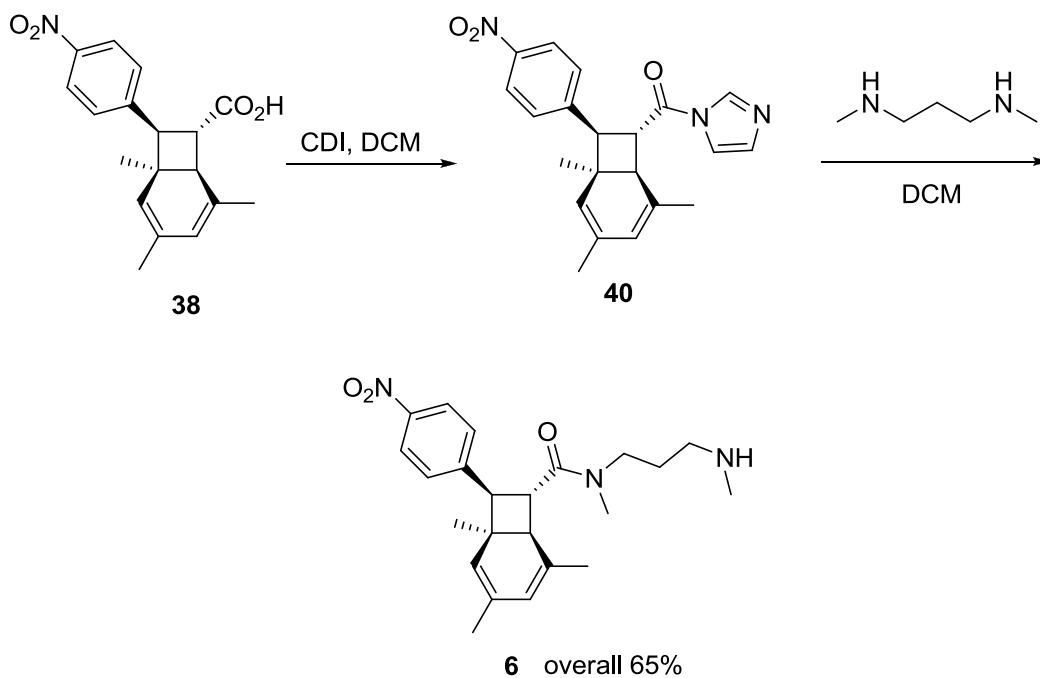
SNF analog **5** was derived from SNF analog **4** (Scheme 7). The ester group of SNF analog **4** was hydrolyzed by sodium hydroxide in ethanol to give the acid **38** in 100% yield. Treatment of the acid **38** with excess oxalyl chloride or thionyl chloride in

dichloromethane (DCM) afforded the acid chloride **39**. An excess 50% solution of dimethyl amine in water and acetone was directly added to the acid chloride **39** to produce SNF analogue **5** in overall yield of 66%.²⁹



Scheme 7. Preparation of SNF analogue **5**.

1.2.4 The Synthesis of SNF Analogue 6



Scheme 8. Preparation of SNF analogue 6

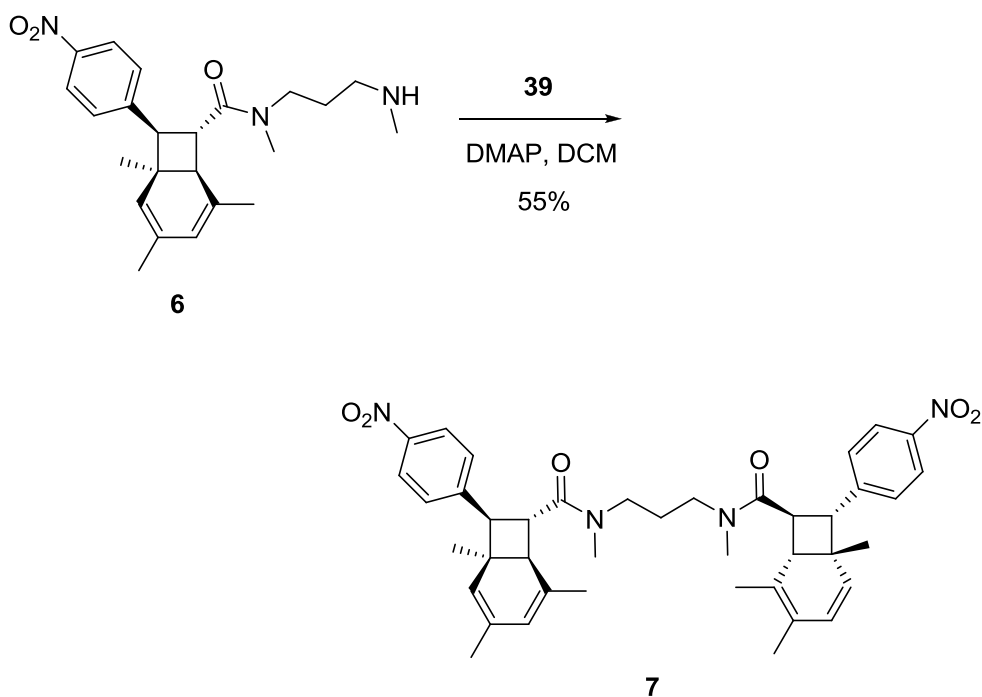
Dr. Eryk Stolarzewicz of the Parker group studied modeling for inhibition of P-gp based on Geoffrey Chang's work.³⁰ The modeling showed that SNF analogues substituted with an aminoamide at the C8 position might have better inhibitive activities against P-gp. The length the chain between the amide functional group and the amine function group could affect the activity. In order to examine Seelig's hypothesis that primary and secondary amines and amides do not serve as recognition elements for P-gp,¹⁹ SNF analogue **6**, substituted by (3-aminopropyl)formamide at C8, was designed and synthesized (Scheme 7).

First, the acid **38** which was produced by hydrolyzing SNF analogue **5** with sodium hydroxide was added by carbonyl diimidazole (CDI) in methylene chloride (DCM) to lead to the carbonyl imidazole intermediate **40**. Then the solution of carbonyl

imidazole **40** was added quickly to an excess solution of *N,N'*-dimethylpropyldiamine in methylene chloride to give a monoacylation diamine **6** in overall yield 65%.³¹

1.2.5 The Synthesis of SNF Analogue 7

SNF analogue **7** consisting of two endo-bicyclo[4,2,0]-SNF substructures can increase the effective concentration of the inhibitive units against P-gp. Moreover, according to the X-ray structure of P-gp, the upper half of the pocket is predominantly hydrophobic and aromatic; the lower half is polar and charged. Thus SNF analogue **7** contains more hydrophobic cyclohexadiene units and more polar nitrophenyl units than the previous SNF analogs. Thus, SNF analog **7** is expected to show an outstanding MDR reversal activity.



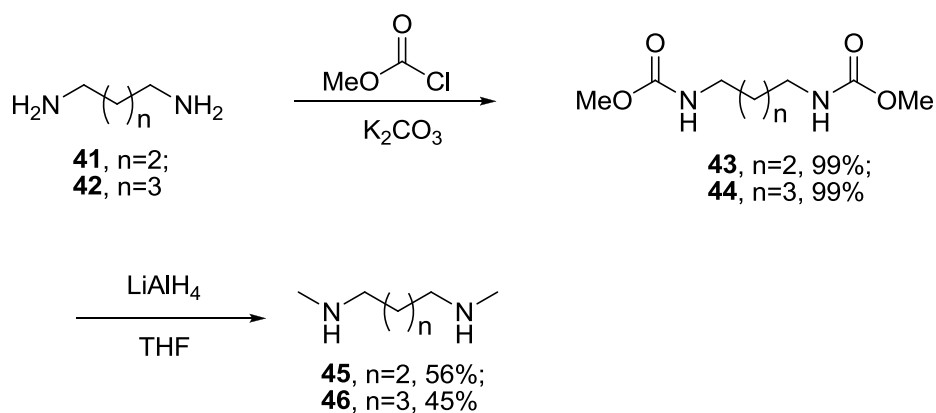
Scheme 9. Preparation of SNF analogue **7**

The amine group of SNF analogue **6** was acylated by the second portion of the acid chloride **39** under the presence of 4-(dimethylamino)pyridine (DMAP) in methylene

chloride to form SNF analog **7**, a mixture of two enantiomers and one diastereomer, in 55% yield (Scheme 9).³²

1.2.6 The Synthesis of SNF Analogues **8**, **9**, and **10**

Considering that the length of the diamide chain linking two endo-bicyclo[4,2,0]-SNF substructures of analogue **7** could effect the inhibitive activities against P-gp, SNF analogues **8**, **9**, **10** with diamide linker of different lengths were designed and synthesized.

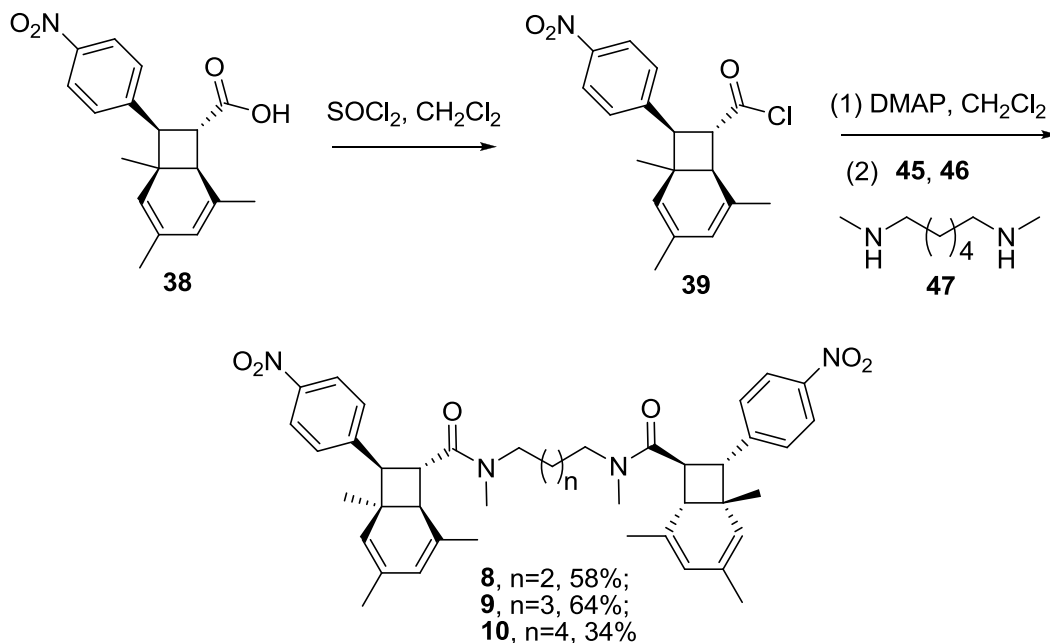


Scheme 10. Preparation of N,N'-dimethyldiamines **45** and **46**

Unlike N,N'-dimethylpropyldiamine, N,N'-dimethyldiamines **45** and **46** are not commercially available. Treatment with methyl chloroformate and potassium carbonate converted diamines **41** and **42** to dicarbamates **43** and **44** in very high yields.³³ Then lithium aluminum hydride (LiAlH₄) in tetrahydrofuran (THF) reduced the dicarbamates **43** and **44** to the desired N,N'-dimethyldiamines **45** and **46** in about 50% yields.³⁴

Treatment of the acid **38** with excess thionyl chloride in dichloromethane (DCM) afforded the acid chloride **39** (Scheme 11). In the presence of 4-(dimethylamino)pyridine (DMAP), a solution of 0.5 equivalent of N,N'-dimethyldiamine **45** in dichloromethane (DCM) was added to the acid chloride **39** to produce SNF analogue **8** in overall 58%

yields. Analogs **9** and **10** were prepared by the same method in overall 64% and 34% yields, respectively.



Scheme 11. Preparation of SNF analogues **8**, **9**, and **10**.

1.2.7 The Synthesis of SNF Analogue 11

1.2.7.1 Another Modulator of Multiple Drug Resistance, LY335979.

Compound LY335979 3HCl has been found to be a clinically useful modulator of *P*-glycoprotein-mediated multiple drug resistance in the research by Syntex Corporation and Lilly Research Laboratories (Figure 7).^{22, 35} LY335979 3HCl contains a rigid bicycle[5,1,0] substructure in the upper part of the molecule. Now, Zosuquidar, the commercial name of LY335979 3HCl, is under development in "Phase 3" of clinical trial in the United States.

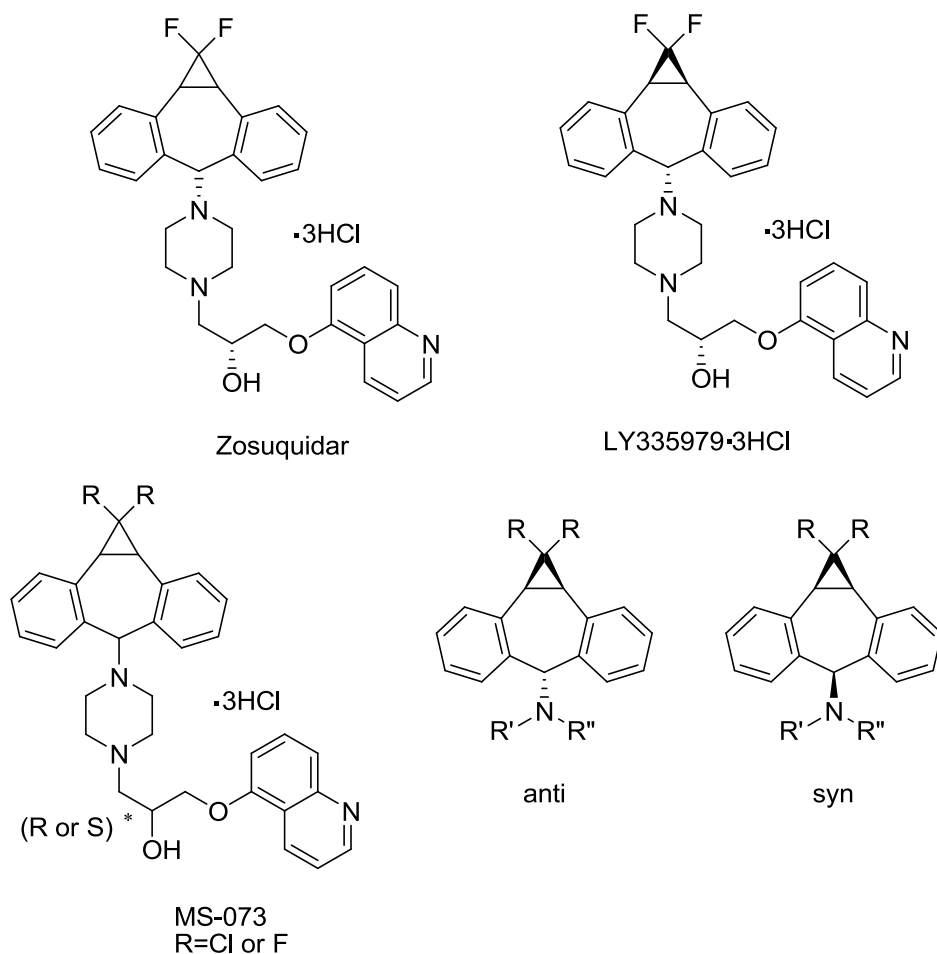


Figure 8. Zosuquidar, LY335979 3HCl and MS-073

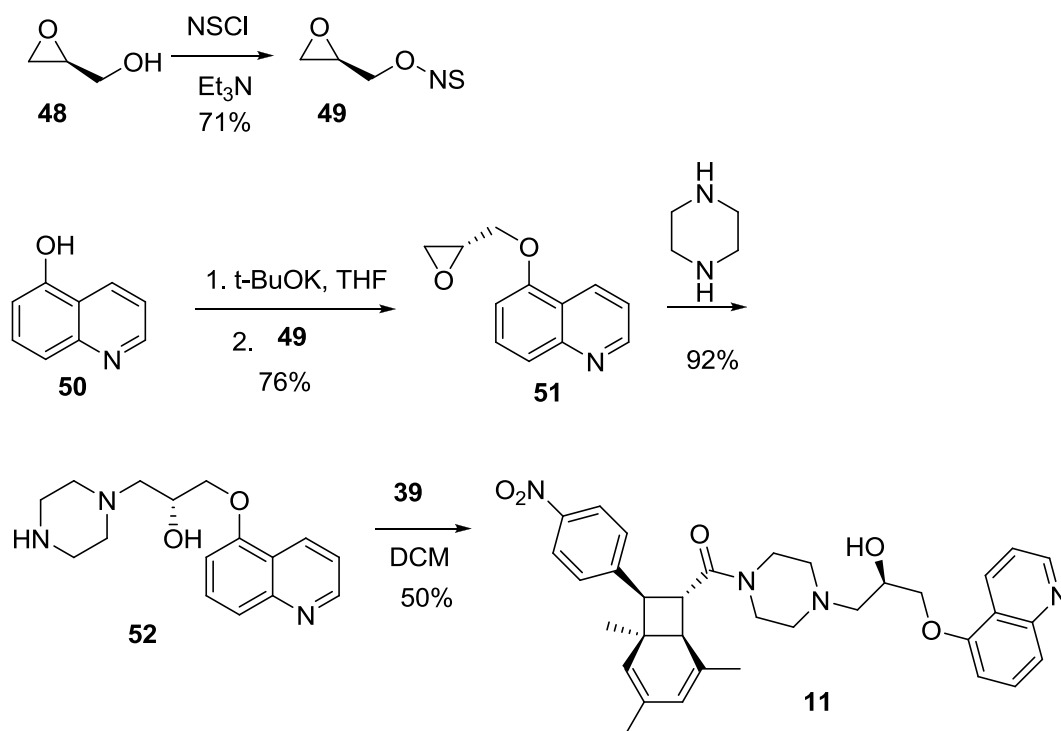
Syntex Corporation tested various MS-073 compounds (Figure 7) with extensive structural modifications on MDR (Table 2).³⁵ These data in Table 2 show that the anti-compounds (piperazine and cyclopropane axial) are more potent than their syn-counterparts.³⁵ Furthermore, the compounds with the (R)-configuration in the hydroxylpropoxy spacer unit are more potent than the (S)-enantiomer.³⁵ LY335979, an anti-(R)-MS073 compound, holds the best MDR activities. According to the literature,³⁵ the drug molecule would partition to a well-defined, energetically favorable location, orientation and conformation in the membrane bilayer pathway for binding of MDR reversal agents to P-gp.

Table 2. MDR reversal EC₅₀ values *in vitro*³⁵

Ex.	R	Stereochemistry	EC ₅₀ (nM)
MS-073	F	anti-(R,S)	50
		anti-(R)	20.7±3.8
		anti-(S)	60
		syn-(R)	200
		syn-(S)	180
		syn-(R,S)	190
	Cl	anti-(R,S)	70
		anti-(R)	55
		anti-(S)	100
	H	anti-(R,S)	40
S 9788			290
Ro 11-2933			540
Verapamil			1200

Since both LY335979 and the SNF analogues contain rigid bicyclic substructures and show good inhibitive activities, we wondered whether the combination of these two kinds of *P*-glycoprotein inhibitors could afford a more biologically active compound. Verapamil analogs and 1,4-dihydropyridines which differ in their potencies as calcium entry blockers are effective in modulating drug transport by P-gp, and SNF's arylbicyclo[4,2,0]-substructure may replace LY335979's bicyclo[5,1,0]-substructure. Thus the 1-piperazin-3-quinolinyl-oxy-propanol unit of LY335979, which contains several hydrogen bond acceptors and donors, was chosen for a structural component of the new designed SNF analogue **11**.

1.2.7.2 Preparation of SNF Analogue 11.

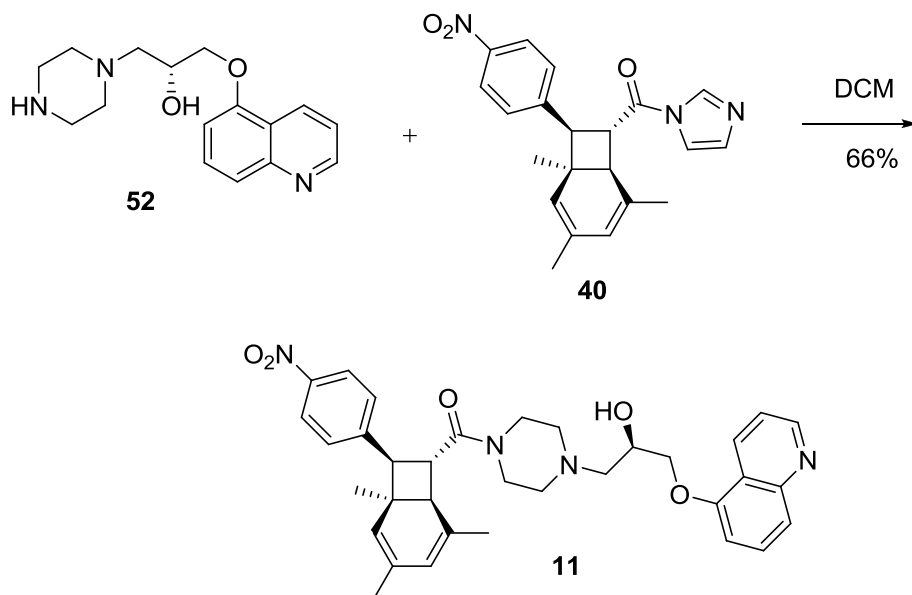


Scheme 12. Preparation of SNF analogues 11.

Following Dr. Charles J. Barnett's work, piperazine moiety **52** was synthesized selectively (Scheme 12).²² (-)-D-Glycidol **48** was protected by 4-nitrobenzenesulfonyl chloride in the presence of trimethylamine in methylene chloride to form epoxide **49** in 71 % yield. 5-Quinolol **50** was deprotonated by potassium t-butoxide in THF. The resulting nucleophilic quinoliny-5-oxyl replaced the 4-nitrobenzenesulfonate group of epoxide **49** by a S_N2 reaction to produce quinolinyglycidol **51** in 76% yield. Excess piperazine opened the epoxide ring of quinolinyglycidol **51** in ethanol to give N-monosubstituted piperazine **52** in 92% yield.³⁶

Piperazine **52** was added to acid chloride **39** in methylene chloride to afford SNF analogue **11** in 50% yield (Scheme 12). Alternatively, the coupling of acid and amine by

carbonyldiimidazole (CDI) was applied. In methylene chloride, the piperazine **52** coupled with SNF imidazole **40** to give SNF analog **11** in 66% yield (Scheme 13). The analogue **11** is a mixture of two diastereomers.



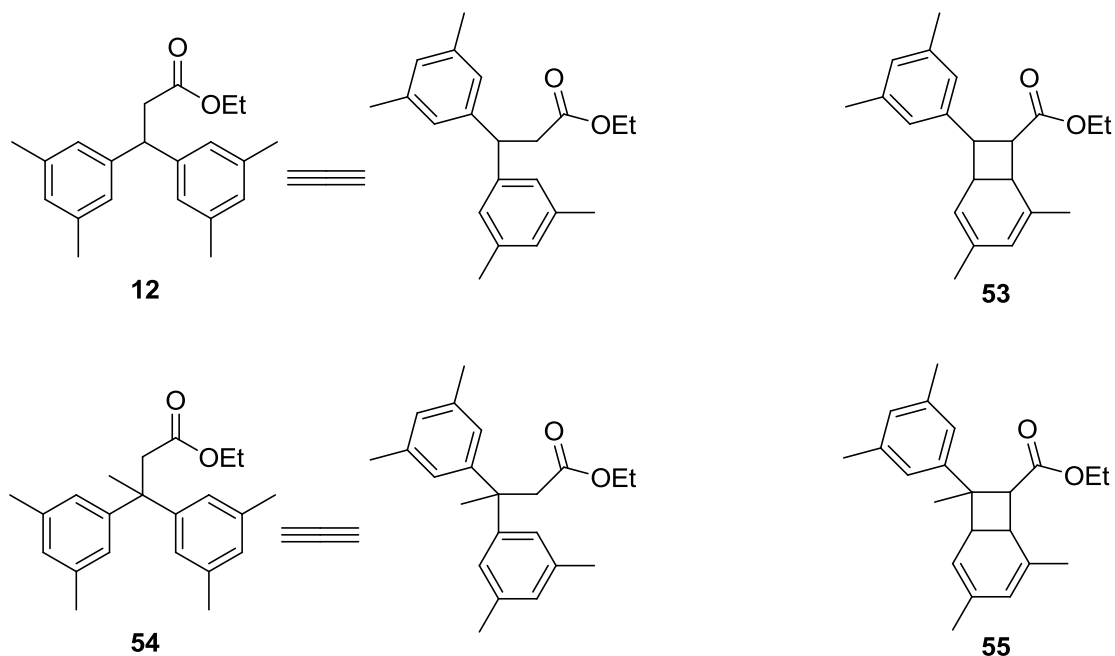
Scheme 13. Plan B for preparation of SNF analogues **11**.

1.2.8 The Synthesis of 3,3'-Dixylylpropionic Ester **12**

In the syntheses of SNF analogues, there are several disadvantages. First, the Stork-Zhao reaction and Stille cross coupling gave the products in low yield. Second, the key Stille cross coupling utilized tin reagents which are toxic and are not acceptable for pharmaceutical intermediates. Third, to make stereospecific olefins, the reaction conditions are limited. Thus the bicyclic structure of SNF compounds was redesigned.

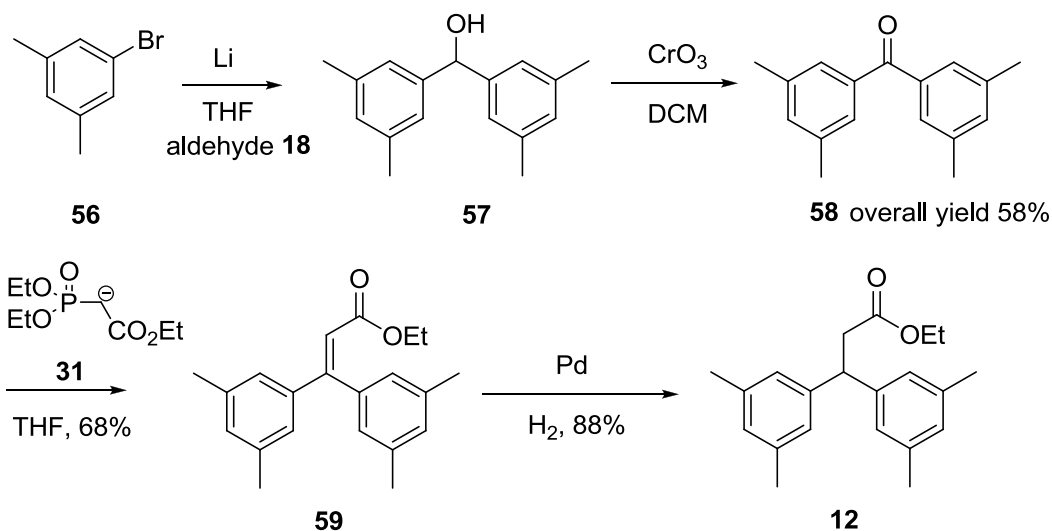
The ethyl 3,3'-dixylylpropionates **12** and **54** might form mimic conformations of compounds **53** and **55**, respectively. Therefore, we expected that compounds **12** and **54**

might occupy the same volume as SNF analogue **3** does in P-glycoprotein. Compounds **12** and **54** would be much easier to make and they do not have any stereoisomer pattern.

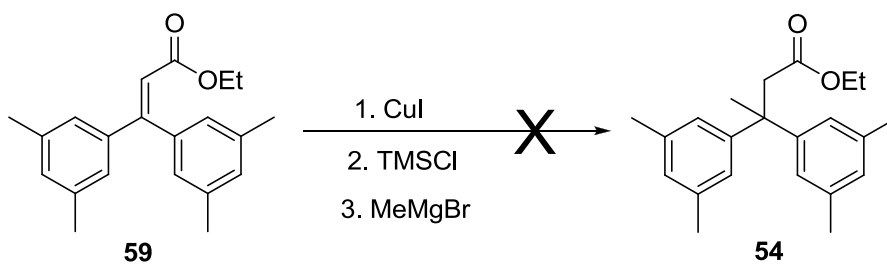


Scheme 14. A proposed cyclization in biological systems.

A mixture of 5-bromo-*m*-xylene **56** and aldehyde **18** was stirred with lithium wire in THF to give the adduct **57** (Scheme 15).³⁷ The resulting alcohol **57** was oxidized by chromium(VI) oxide in a mixture of methylene chloride and acetic acid to give ketone **58** in 58% overall yield.³⁸ Horner-Wadsworth-Emmons reagent **31** was added to the ketone **58** in THF to afford α,β -unsaturated ester **59** in 68% yield.³⁹ The α,β -double bond of ester **59** was reduced by Pd-mediated catalytic hydrogenation in methanol to give the desired dixylylpropionic ester **12** in 88% yield.³⁹



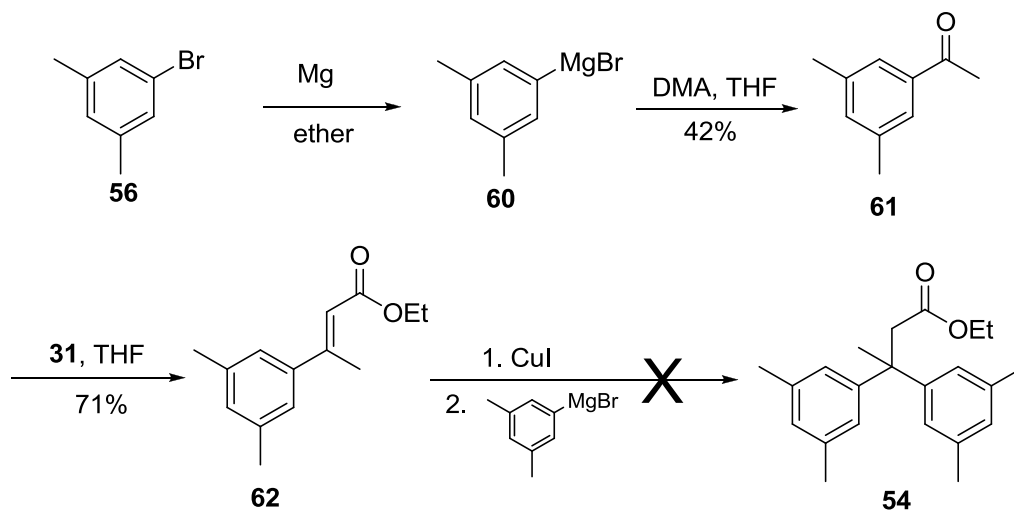
Scheme 15. Preparation of 3,3'-dixylylpropionic ester **12**.



Scheme 16. Attempt at preparation of 3,3'-dixylylpropionic ester **56**.

We attempted to synthesize the dixylylpropionic ester **54** (Scheme 16). First, copper(I) iodide (CuI) and chlorotrimethylsilane (TMSCl) were added to a solution of α,β -unsaturated ester **59** in THF. Then methylmagnesium bromide (MeMgBr) was added to the suspension.⁴⁰ However, compound **54** was not produced, and the reaction only gave back the starting material **59**. A possible reason that the addition did not work is the big steric effect of 3,5-dimethylphenyl function groups. Another synthetic route to ester **54** was tried (Scheme 17): 5-Bromo-*m*-xylene **56** was treated with magnesium to prepare 3,5-dimethylphenylmagnesium bromide **60**, which was added to *N,N'*-dimethylacetamide to form the ketone **61** in 42% yield.⁴¹ Horner-Wadsworth-Emmons reagent **31** converted

the ketone **61** to a trans- α,β -unsaturated ester **62** in 71% yield.³⁹ The suspension of α,β -unsaturated ester **62** and copper (I) iodide in THF was added 3,5-dimethylphenylmagnesium bromide **60**.⁴⁰ Still because of the big steric effects, the desired compound **54** was not obtained and the reaction only gave the starting material **62** back.



Scheme 17. The second attempt at preparation of 3,3'-dixylylpropionic ester **54**

1.2.9 The Synthesis of SNF Analogue 13

1.2.9.1 Background: Anthranilamide Derivatives as Modulators of Multiple Drug Resistance

Tariquidar (XR9576, Figure 8) is one of the third generation of the inhibitors of P-gp. Recently, because of a significant proportion of adverse events, the progress of clinical trials of Tariquidar slowed down.⁴² Therefore, a new generation of inhibitors of P-gp based on anthranilamide derivatives was developed (Figure 9).⁴³

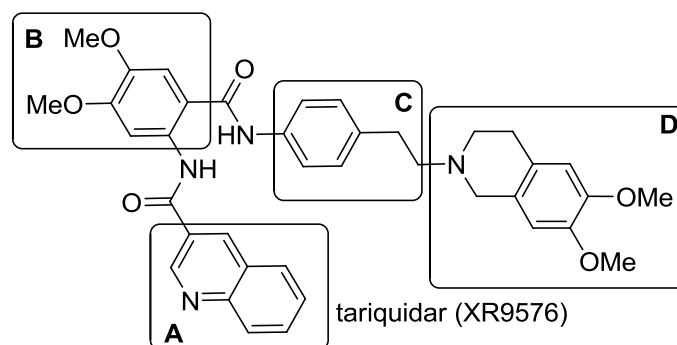


Figure 9. Tariquidar (XR9576). **A**, **B**, **C**, and **D** define the four regions of XR9576

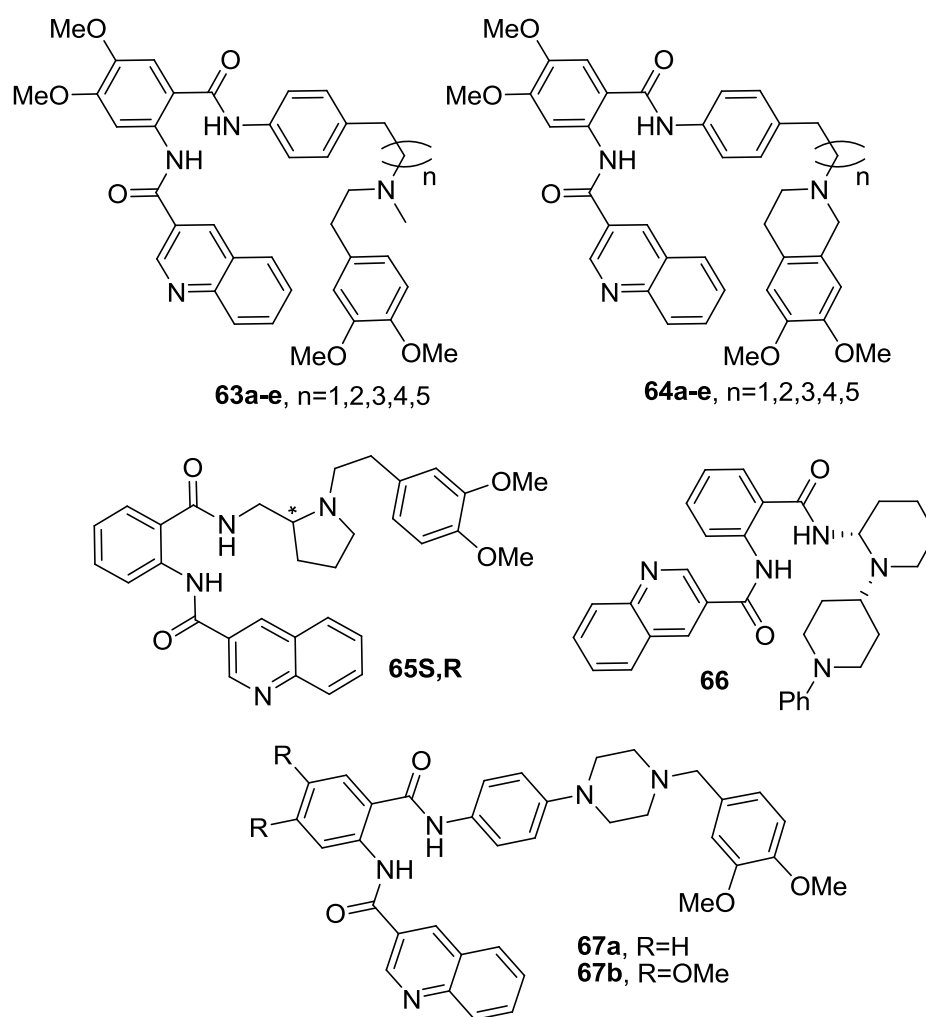


Figure 10. Anthranilamide derivatives

The P-gp inhibition potencies of compounds **63a–e**, **64a–e**, **65S/R**, **66**, and **67a–b** in combinations respectively with daunorubicin (DNR) and vinblastine (VBL), were assessed using CEM/VLB500 human leukemia cells. These compounds exhibited more potent MDR reversal activities than verapamil did (Table 3). In particular, compounds **67a** and **b** were respectively 9- and 20-fold more potent than verapamil (VRP); they were almost equipotent to XR9576. These compounds also exhibited minimal cytotoxicity at concentrations up to 10 μ M while XR9576 was significantly toxic at this concentration

The metabolism of drugs by cytochrome P-450 (CYP) initiates many drug–drug interactions.⁴³ In addition, P-gp can act synergistically with cytochrome P450 (CYP) 3A4 to enhance drug metabolism and elimination.⁴⁴ CYP1A2, CYP2C8, CYP2C9, CYP2C19, CYP2D6, and CYP3A4 are involved in the metabolism and the drug–drug interactions. Compounds **63a**, **67a–b**, and XR9576 were examined in a CYP450 inhibition assay to evaluate their ability to inhibit catalytic activities (Table 4). The results showed that compound **67b** exhibited the best inhibition of CYPs.

According to the effect of the modifications on the regions B-C-D versus the P-gp/CYPs inhibition and the cytotoxicity, an arylpiperazinyl group instead of an alkyl chain in region C may led to significant inhibitive activities against P-gp and CYPs; and the addition of two methoxy groups to the anthralinidyl moiety (region B compound **67a**) increased by 200% the P-gp inhibition activity and completely inhibited CYP-450. In comparison with other compounds, compound **67b** was considered as the most active P-gp inhibitor at nM.⁴³

Table 3. In vitro inhibition activities of new anthranilamide derivatives on CEM/VBL500 cells incubated in the presence of vinblastine and their cytotoxic activity.⁴³

Compound	Inhibitory activity on CEM/VLB₅₀₀ EC₅₀ (nM)	Cytotoxicity activity on CEM/VLB₅₀₀ GI₅₀ (nM)
63a	270 ± 84	27707 ± 15909
63b	490 ± 90	29177 ± 13401
63c	781 ± 177	19063 ± 7858
63d	1345 ± 113	17845 ± 10300
63e	904 ± 384	10201 ± 2147
64a	899 ± 127	42298 ± 4588
64b	500 ± 147	29339 ± 11787
64c	945 ± 346	26353 ± 4588
66d	1274 ± 27	17210 ± 4952
64e	1084 ± 255	17060 ± 9197
65R	329 ± 138	>15000
65S	482 ± 226	>15000
66	648 ± 208	11500 ± 1400
67a	124 ± 76	>15000
67b	59 ± 35	≥15000
XR9576	68 ± 40	13500 ± 3100
VRP	1041 ± 443	>15000

Table 4. Cytochrome-P450 inhibition assays of compounds **63a**, **67a-b**, and XR9576.⁴³

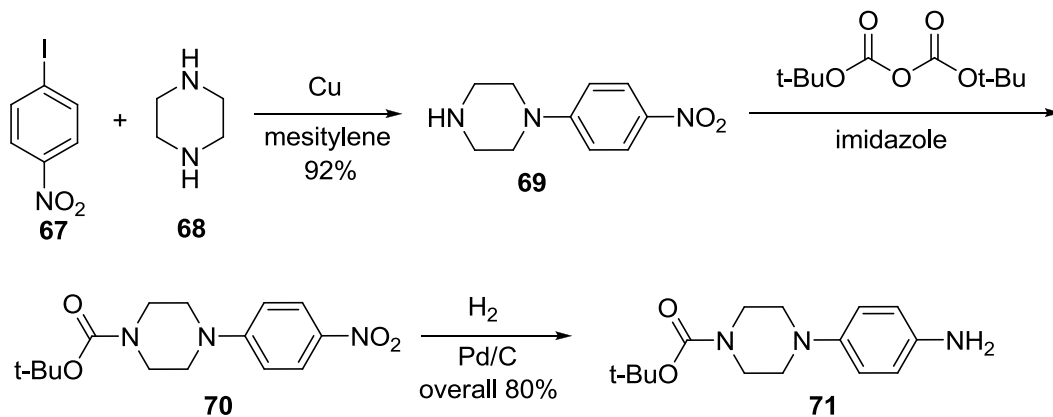
P-450 subunit	Inhibitor Ref.	Ref. IC ₅₀ (μM)	XR9576 IC ₅₀ (μM)	63a IC ₅₀ (μM)	67a IC ₅₀ (μM)	67b IC ₅₀ (μM)
CYP1A2	Furafylline	4.98	27.2	>100	>100	6.68
CYP2A6	Tranilcypromine	0.98	>100	>100	>100	n/a
CYP2B6	Tranilcypromine	10.3	>100	>100	>100	100
CYP2C8	Quercetin	1.56	45.3	5.82	13.3	1.08
CYP2C9	Sulfaphenazole	0.56	7.1	2.08	>10	0.0968
CYP2C19	Tranilcypromine	8.74	>10	14.7	23	0.393
CYP2D6	Quinidine	0.0074	100	>100	>100	17.3
CYP2E1	DDTC	17.1	>100	>100	>100	n/c
CYP3A4/BFC	Ketoconazole	0.096	>100	1.62	20.2	0.195
CYP3A4/BQ	Ketoconazole	0.26	>100	7.75	3.58	2.48

Compound **69b** inhibited both CYP3A4 and P-gp, while XR9576 and LY335979 selectively inhibited P-gp. Therefore, the new SNF analogue based on compound **67b** was designed. Since regions B and C significantly affect the activities, we decided to replace region D with SNF substructure. SNF analogue **13** with a SNF unit and an arylpiperazinyl unit of Tariquidar was expected to inhibit not only P-gp, but also CYP450.

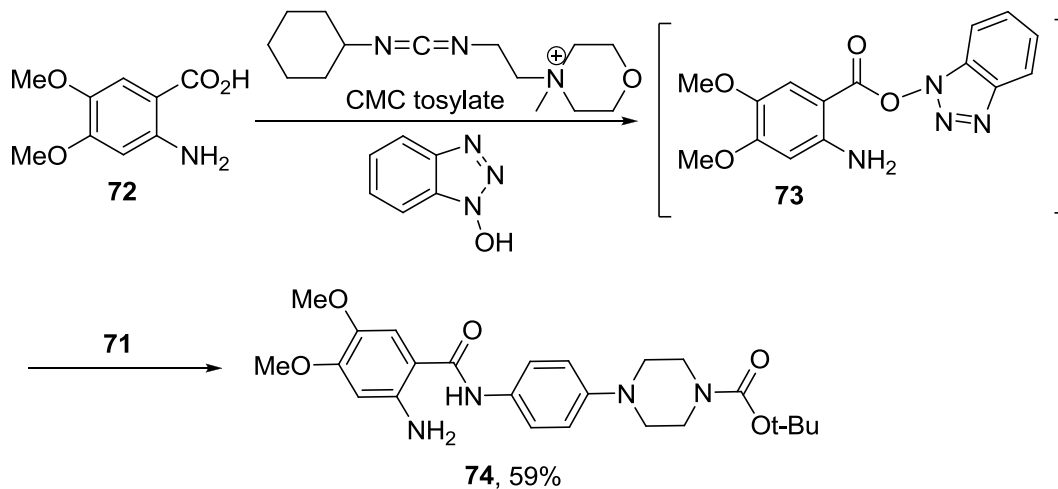
1.2.9.2 Preparation of SNF Analogue 13.

N-tert-Butoxycarbonyl (t-Boc) protected arylpiperazine **71** was prepared (Scheme 18). Under the catalysis of copper, p-nitroiodobenzene coupled with piperazine in mesitylene to afford compound **69** in 92% yield.⁴⁵ The protection of compound **69** with di-tert-butyl dicarbonate led to compound **70**.⁴⁶ The following selective hydrogenation

under the catalysis of palladium reduced the nitrophenyl group to give the desired compound **71** in overall 80% yield.⁴⁷



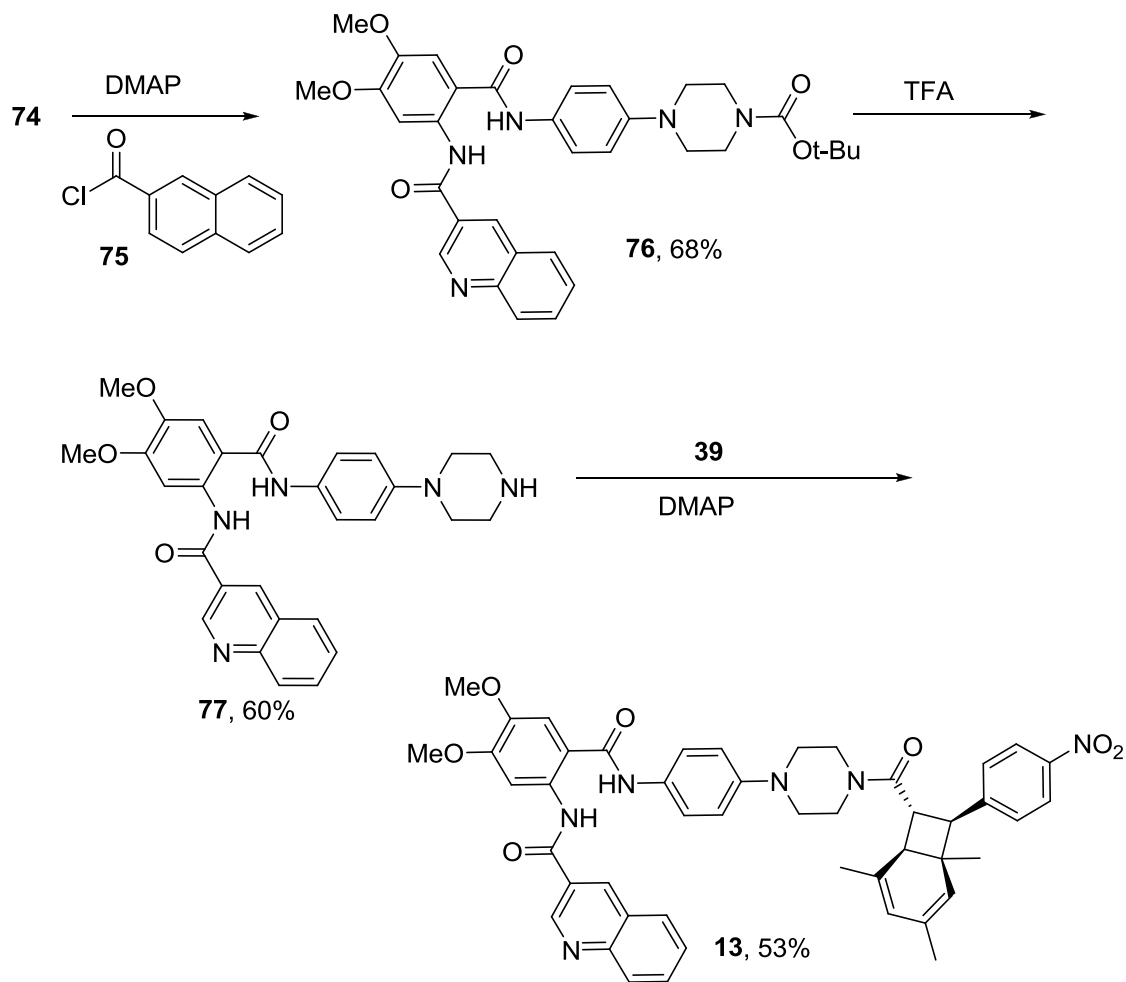
Scheme 18. Preparation of arylpiperazine **71**.



Scheme 19. Preparation of compound **74**

A special procedure was employed to couple compound **71** and 2-amino-4,5-dimethoxybenzoic acid **72** together (Scheme 19). First, 1-hydroxybenzotriazole (HOBt) and CMC tosylate activated the amino acid **72** to form an intermediate **73**. Then

compound **71** substituted the benzotriazole of the intermediate to afford the product **74** in 59% yield.⁴³

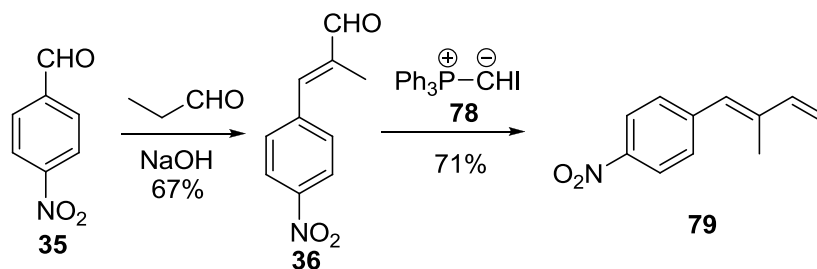


Scheme 20. Preparation of SNF analogue **13**

The acylation of compound **74** with acid chloride **75** in the presence of 4-dimethylaminopyridine (DMAP) resulted in compound **76** in 68% yield. Treatment with trifluoroacetic acid (TFA) selectively removed the t-Boc protection group to give the piperazine **77** in 60% yield. In the presence of DMAP, SNF acid chloride **39** was added piperazine **77** to give the desired SNF analogue **13**, which contains the active regions of **67b** and a rigid SNF substructure, in 53% yield (Scheme 20).

1.2.10 The Synthesis of SNF Analogue 14

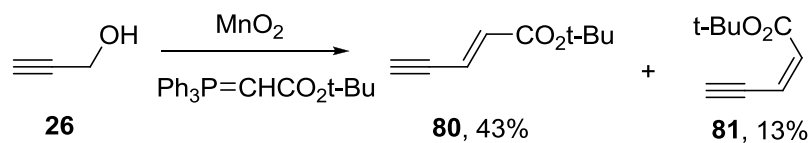
The synthetic route to SNF's bicyclo[4,2,0] system is not efficient. It takes seven steps to prepare the basic SNF analogue **4**, and the overall yield is 5% based on propargyl alcohol. The reason is that several low-yield reactions such as the Stork-Zhao reaction and the addition of methylmagnesium bromide to propargyl alcohol were used to make stereospecific olefins. Furthermore, the key Stille cross-coupling reaction utilizes toxic tin starting materials and also gives the products in low yield. Considering that the rigid bicyclic structure of SNF analogues plays an important role in multi-drug resistance reversal, we proposed that SNF analogues without two methyl groups respectively on C-11 and C13 might also exhibit good activities against MDR. Therefore we developed a new strategy for the syntheses of SNF analogues, which does not require the vinyltin reagents.



Scheme 21. Preparation of iododiene **79**.

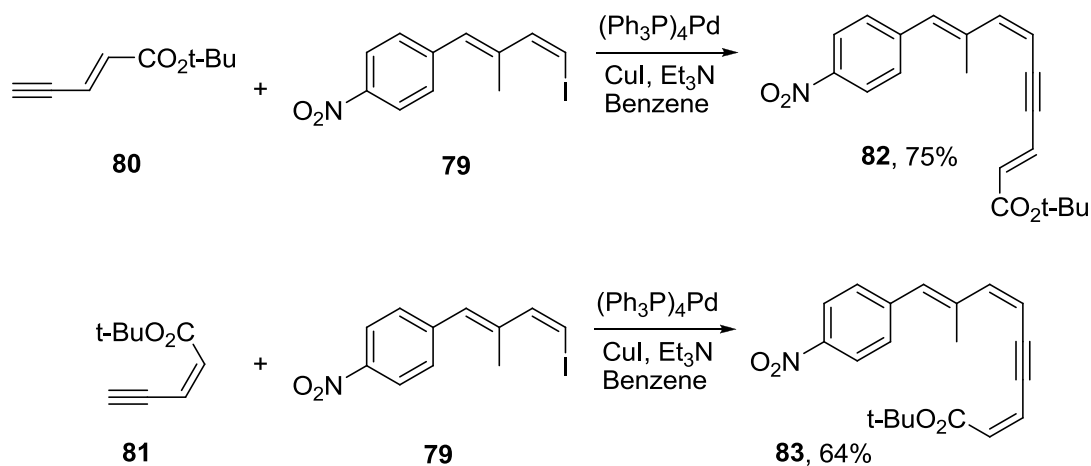
The cross-Aldol condensation of 4-nitrobenzaldehyde **35** and propionaldehyde afforded the aldehyde **36** in 67% yield (Scheme 21). Treatment with iodo-Wittig reagent **78**⁴⁸ converted the aldehyde **36** to (1Z,3E)-iododiene **79** in 71% yield.⁴⁹

In a one-pot oxidation-olefination reaction, propargyl alcohol **26** was oxidized and then the resulting aldehyde was treated with a Wittig reagent to produce two isomers **80** and **81** in 43% and 13% yield, respectively (Scheme 22).⁵⁰



Scheme 22. Preparation of esters **80** and **81**.

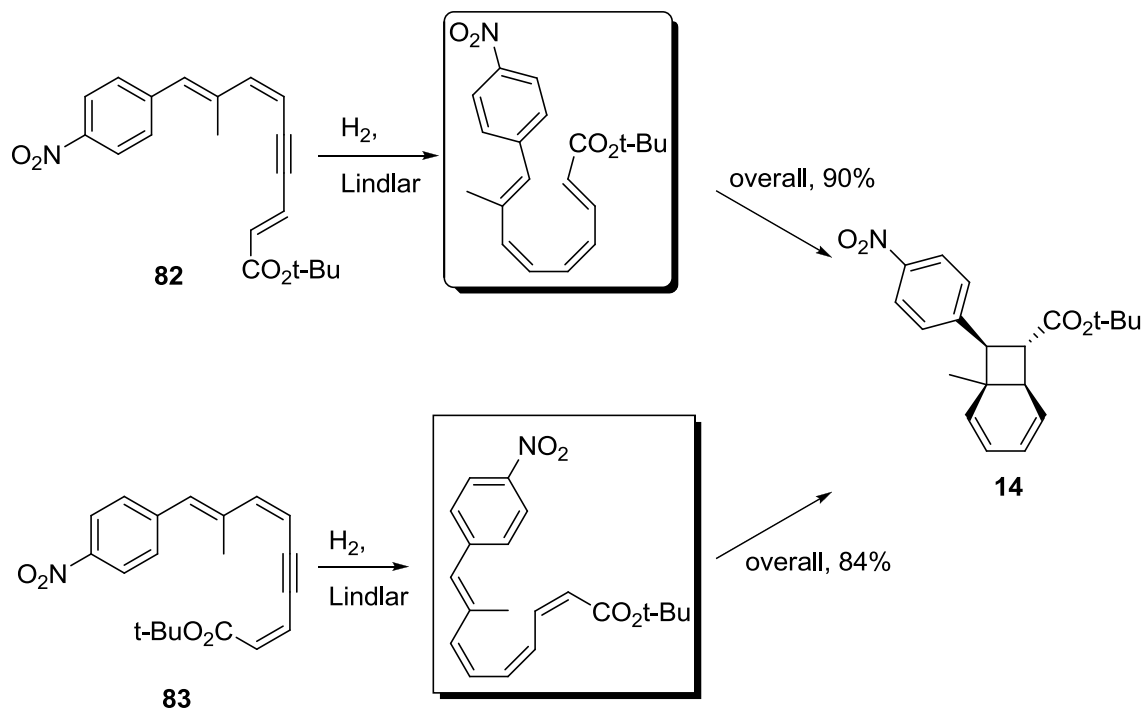
Sonogashira coupling of the trans-isomer **80** and iododiene **79** in the presence of tetrakis(triphenylphosphine)palladium ((Ph₃P)₄Pd) and copper(I) iodide (CuI) led to compound **82** in 75% yield (Scheme 23).⁵¹ The cis-isomer **81** and iododiene **79** coupled to afford compound **83** in 64% yield. In the transition state of the reaction, since the cis-starting material **81** might cause a big steric effect, the yield of the corresponding product is lower.



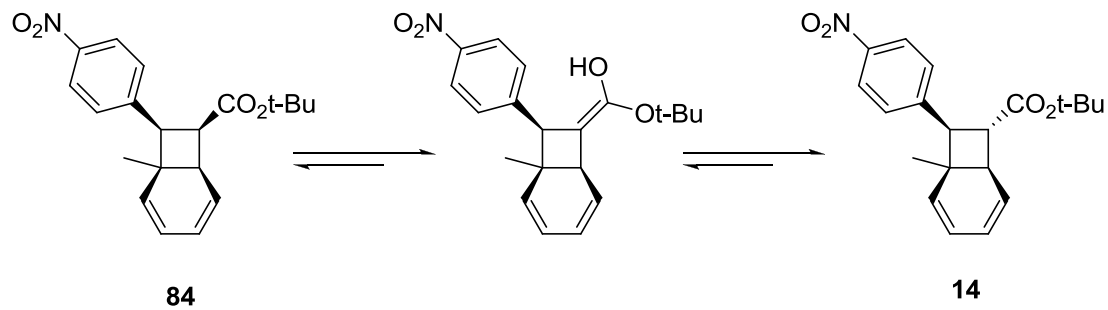
Scheme 23. Preparation of esters **82** and **83**.

The hydrogenation of the triple bonds of isomer **82** in the presence of Lindlar catalyst gave the trans-endo product, SNF analogue **14**, in 90% yield (Scheme 24)⁵². The reduction of isomer **83** afforded the same product in 84% yield. The results suggest that the cis-endo product **84**, which resulted from epimerization of cis-isomer **83** to the

thermodynamically stable trans-endo product **14**, which was also made directly from trans-isomer **82** (Scheme 25).

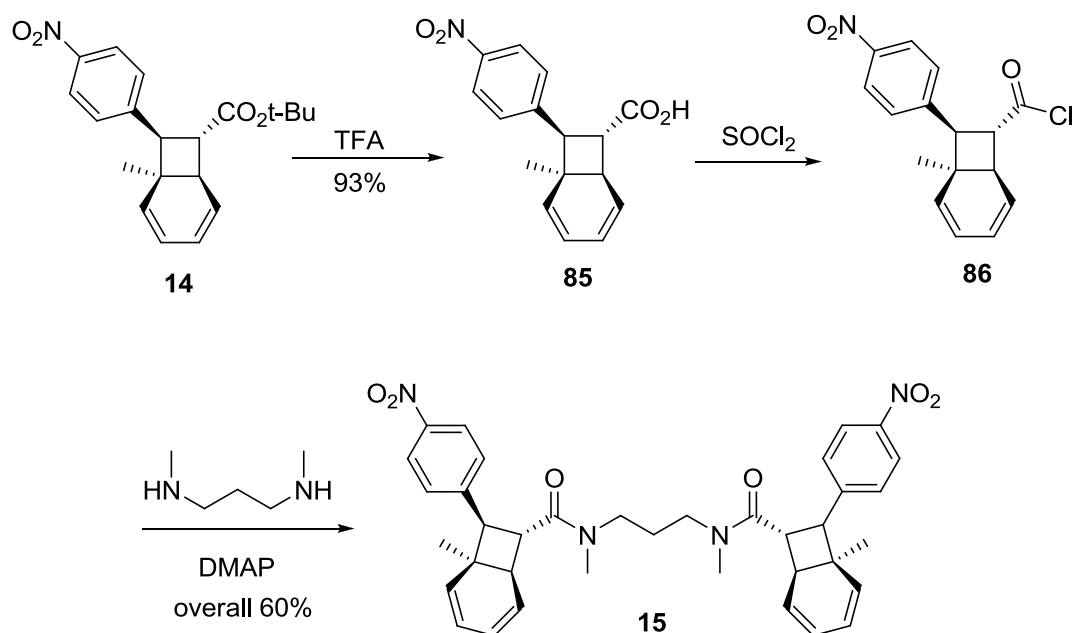


Scheme 24. Preparation of SNF analogue **14**.



Scheme 25. Epimerization of **84** to **14**.

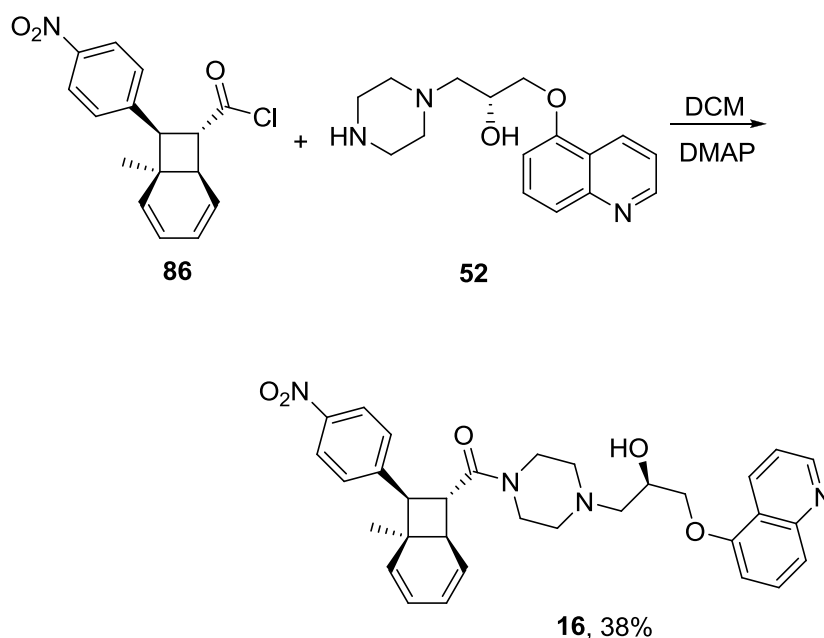
1.2.11 the Synthesis of SNF Analogue 15



Scheme 26. Preparation of SNF analogue 15.

SNF analogue 7 exhibited a good MDR reversal activity. In order to thoroughly investigate the P-gp inhibition of the new bicyclic compound 14, analogue 15 with two new bicyclic(4,2,0) units was prepared. Treatment with trifluoroacetic acid (TFA) removed the t-butyl group of SNF analogue 14 to afford acid 85 in 93% yield (Scheme 26). Excess thionyl chloride (SOCl₂) converted acid 85 to acid chloride 86. In the presence of excess 4-(dimethylamino)pyridine (DMAP), 0.5 equivalent of N,N'-dimethylpropylamine was added to the acid chloride 86 to give SNF analogue 15 in 60% yield. Like SNF analogues 7-10, analogue 15 was obtained as a pair of enantiomers and a meso-diastereomer.

1.2.12 the Synthesis of SNF Analogue 16



Scheme 27. Preparation of SNF analogue 16.

In biological tests, SNF analog **11** consisting of a functional unit of LY335979 and a SNF bicyclic unit showed an outstanding MDR reversal ability. We wondered whether analog **16** with a LY335979 unit and the new bicyclic substructure of **14** could possess a good activity as **11** does (Scheme 27). In the presence of excess 4-(dimethylamino)pyridine (DMAP), the piperazine **52** was acetylated by the acid chloride **86** to give SNF analogue **16** in 38% yield (Scheme 27).

1.2.13 Biological Activity of SNF Analogues

Dr. Ralph J. Bernacki and Dr. Paula Pera of Experimental Therapeutics Department at Roswell Park Cancer Institute tested some of the SNF analogues for reversal of multidrug resistance in the paired drug-sensitive and drug-resistant cell lines

MCF7 and MCF7 Adr(R), a drug-selected NCI cell line that has high levels of P-glycoprotein.¹⁰ The protocol is that described in the literature.¹²

Table 5. Effect of Modulators on Sensitivity to Taxol in MCF7-R Human Mammary Cells

Compound	IC ₅₀ (μ M) MCF7-S ¹	IC ₅₀ (μ M) MCF7-R ¹	Compound Tested in Combination With Paclitaxel (μ M)	Paclitaxel IC ₅₀ (nM) MCF7-R	% Paclitaxel IC ₅₀ Reduction ⁴
paclitaxel			none	1000	
3			10	90	91
4	>10	>10	10 and 1.0	100 and 740	90 and 26
5			10	270	73
6	>10 (0) ²	>10 (0) ²	1.0 ³	861	14
7	>10 (0) ²	>10 (0) ²	1.0 ³	329	67
11	>10 (42) ²	>10 (42) ²	1.0 ³	5.1	99
12	>10 (0) ²	>10 (0) ²	1.0 ³	825	18

¹Human mammary carcinoma – MCF7-S – sensitive cell line, MCF7-R – multidrug resistant cell line.

²The IC₅₀ value was greater than 10 μ M for all reversal agents tested, however the % growth inhibition seen at 10 μ M is indicated in parentheses.

³Reversal agent alone (at 1.0 μ M) resulted in 0% growth inhibition

⁴[100 - (IC₅₀ reversal agent + paclitaxel/IC₅₀ paclitaxel x100)]

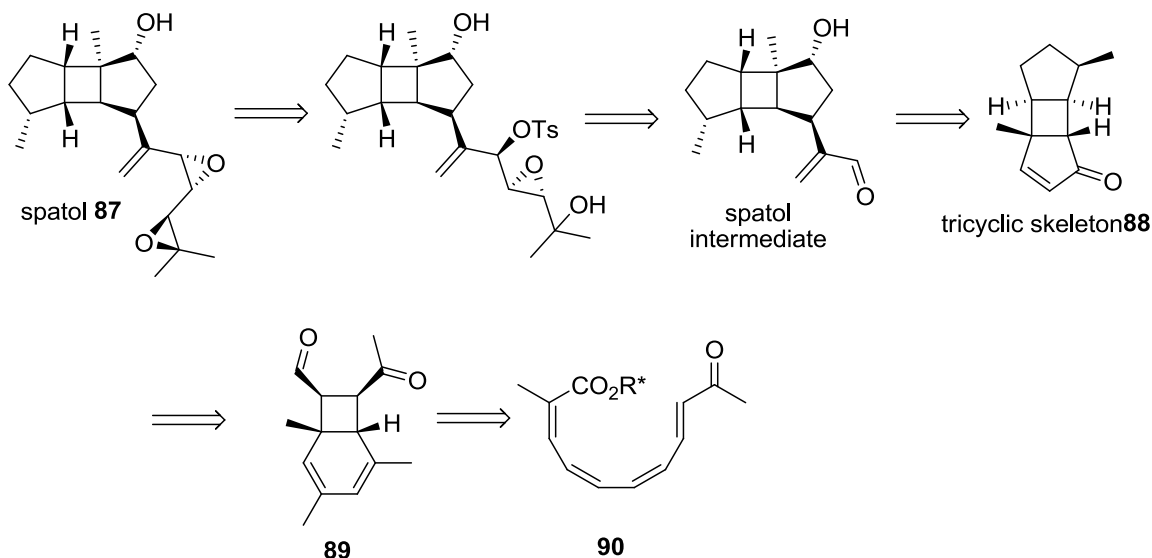
The biological results are showed in Table 5. The growth inhibitory effect of each compound alone is indicated in columns 2 and 3. Analog **11** inhibited cell growth by 42% at 10 μ M in both cell lines and 0% at 1.0 μ M in MCF7-R. For combination studies (performed in MCF7-R cells) with taxol, a concentration of compound which had no effect on cell growth (1 μ M for analogs **4**, **6**, **7**, **11**, and **12**; 10 μ M for analogs **3-5**) was combined with varying concentrations of taxol. Columns 5 and 6 indicate the results of

this combination. Analog **11** was the most active in modulating sensitivity to taxol (99% reduction in IC₅₀), followed by analog **12** (67% reduction in IC₅₀). Analogs **5**, **6**, and **7** showed very little modulating activity. Ester analog **3**, in which the p-nitrophenyl substituent has been replaced by a 3,5-dimethylphenyl group is almost as active as ester **4**, indicating that the nitro group is not a significant contributor to activity.

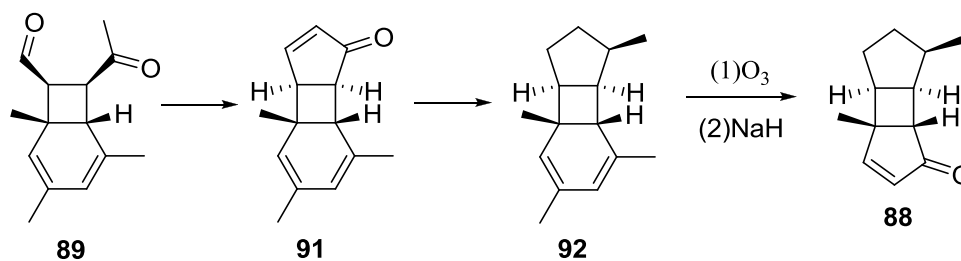
1.2.14 A Model for Synthesis of Spatol

Spatol **87**, a diterpene isolated from the brown algae *Spatoglossum howleii* (Scheme 28),⁵³ attracted much attention from chemists not only because of its relatively unique structure but also because of its biological activity. Spatol was considered as a potent inhibitor of cancer cell replication.⁵³ It was reported that spatol inhibited synchronous cell division in the fertilized sea urchin egg (*Lytechinus pictus*) at 1.2 $\mu\text{g/mL}$. Spatol was also found to inhibit human T242 melanoma and 224C astrocytoma cell lines in vitro with activity ranging from 1 to 5 $\mu\text{g/mL}$. A number of synthetic studies directed toward spatol and other spatane diterpenes have appeared.⁵⁴ The synthetic challenges include the stereoselective preparation of the cis,anti,cis-tricyclo[5.3.0.0]decane skeleton, as well as the labile diepoxide functionality.⁵⁵

We proposed a new strategy for the synthesis of spatol tricyclic core **88**, which starts from the SNF bicyclic skeleton (Scheme 28). The spatol intermediate could be made from ketone **88** sequentially by reduction, Sharpless epoxidation, and further transformations. The synthetic route from the spatol intermediate to spatol has been reported.⁵⁶ The process consists of 1,2-addition, epoxidation, and cyclization.⁵⁷



Scheme 28. A proposal of the synthesis of spatol from a SNF bicyclic skeleton.

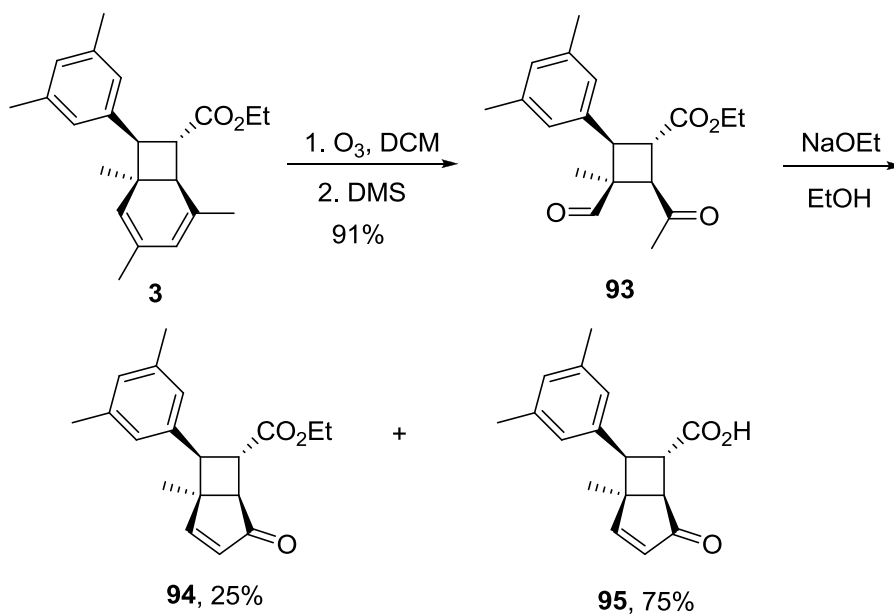


Scheme 29. A proposed transformation from the compound 89 to ketone 88.

Ketone **89** might cyclize by an intramolecular aldol condensation to the α,β -unsaturated ketone **91** (Scheme 29). The ketone **91** could be reduced to tricyclo[4,3,0,0] system **92**. Ozonolysis might open the six-membered ring of sesquiterpene **92** to form a cis-acetylcyclobutylcarbaldehyde intermediate which could cyclize by a following intramolecular aldol condensation to the desired spatol precursor **88**.⁵⁸

In order to examine the proposal, a model based on SNF analog **3** was built (Scheme 30). The ozonolysis in the presence of dimethyl sulfide (DMS) cleaved the cyclohexadiene substructure of SNF analogue **3** to give cis-acetylcyclobutylcarbaldehyde

93 in yield 91%. Base-catalyzed intramolecular aldol condensation led to bicyclo[3.2.0]heptones **94** and **95** in yields 25% and 75%, respectively. Ester **94** underwent a partial hydrolysis to the acid **95** during the aldol condensation.



Scheme 30. Model reactions for spatol synthesis

1.2.15 Application of $8\pi/6\pi$ –Electrocyclization to the Synthesis of Welwitindolinone A Isonitrile

1.2.15.1 Background: Welwitindolinone A Isonitrile

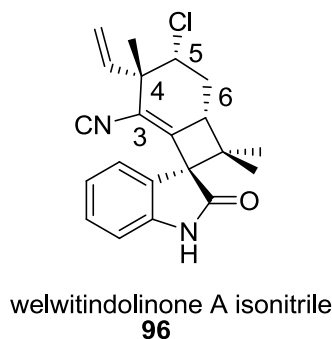


Figure 11. Welwitindolinone A isonitrile.

Several related oxindole-containing alkaloids, including hapalindole, fischerindole, ambiguine, and welwitindolinone, were isolated from marine blue-green algae *Hapalosiphon welwitschii* and *Westiella intricata* Borzi by Moore and co-workers.⁵⁹ Among the known welwitindolinones, the only one comprised of a highly functionalized spirocyclobutane oxindole carbon skeleton is welwitindolinone A isonitrile **96**. Consequently, this compound has been postulated to serve as the biosynthetic precursor to the remaining congeners.^{59,60} Welwitindolinone A isonitrile **96** is a densely functionalized oxindole harboring three all-carbon quaternary centers, a neopentyl chlorine atom, and a striking spiro-fused cyclobutane that appears to be strained to exist upon cursory inspection.⁶¹

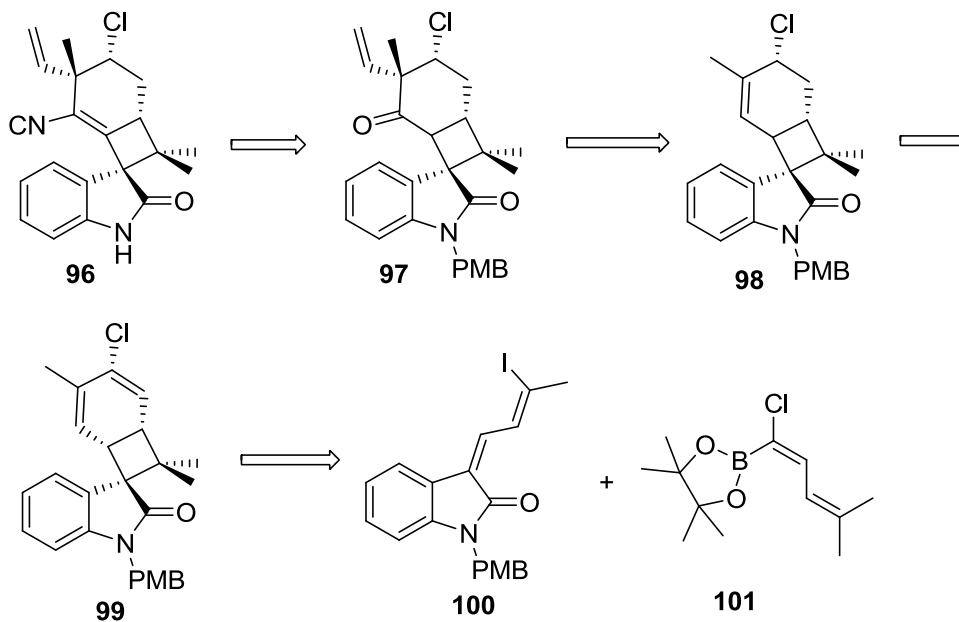
1.2.15.2 Biological Activity

Welwitindolinone A displays diverse and valuable biological activity, including reversal of P-glycoprotein-mediated multiple drug resistance (MDR), antifungal activity,

larvacidal activities, and inhibition of cancer cell proliferation. Similar to the effects of verapamil, welwitindolinone might attenuate the resistance of MCF-7/ADR cells to anticancer drugs, such as vinblastine, taxol, actinomycin D, daunomycin, and colchicine. These effects are apparent at doses as low as 0.1 μM , without affecting the cytotoxicity of cisplatin. Therefore it was considerably more potent than verapamil for reversal of MDR.⁶²

1.2.15.3 Retrosynthetic Analysis of Welwitindolinone A Isonitrile

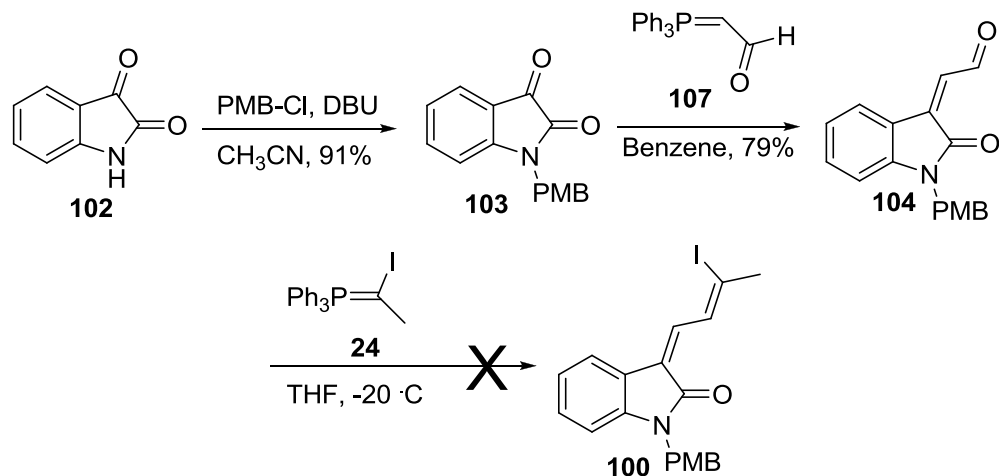
The synthetic method of the vinyl isonitrile region in welwitindolinone A isonitrile **96** from a ketone **97** has been reported (Scheme 31).⁶³ The carbonyl group of compound **97** could be converted to a formamide. The dehydration of the formamide by trifluoroacetic anhydride could form the target **96**. Bicyclo[4,2,0]octane **98** could be converted to the key intermediate **97** by a formation of cyclobutanone on C3,C4 double bond and a following ring opening.⁶⁴ Hydrazine might selectively hydrogenate the double bond of the vinyl chloride region (C5,C6) of cyclohexadiene **99** to give bicyclo[4,2,0]octane **98**.⁶⁵ The stability of the C3,C4 double bond and lower electron density of the C5,C6 double bond might provide the selectivity in the reduction. Pd-catalyzed Suzuki cross coupling of iododiene **100** and dienyborate **101** could lead to a tetraene.⁶⁶ The consequent tandem 8π , 6π -electronic cyclization could result in bicycle[4,2,0]octadiene **99**.



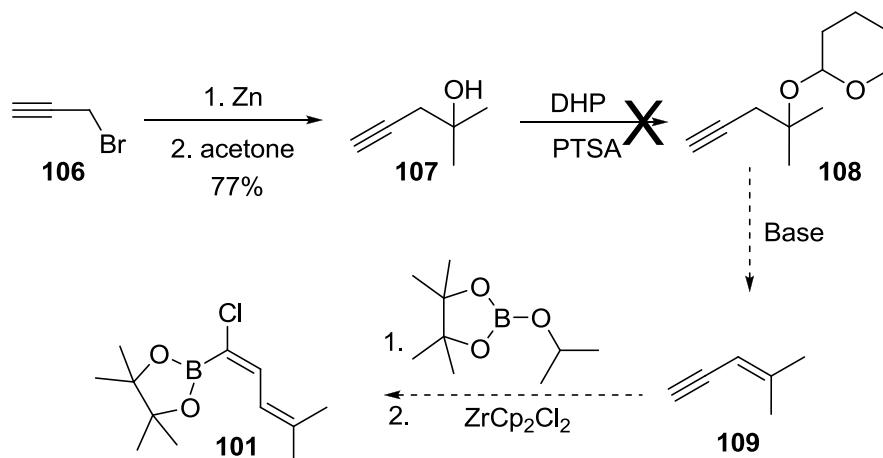
Scheme 31. Retrosynthetic analysis of welwitindolinone A isonitrile.

1.2.15.4 Toward the Synthesis of Welwitindolinone A Isonitrile

The amide group of isatin **102** was protected by p-methoxybenzylchloride (PMB-Cl) to afford PMB-isatin **103** in 91% yield (Scheme 32).⁶⁷ Wittig reagent **105** transformed PMB-protected isatin to the aldehyde **104** in 79% yield.⁶⁸ However, Stork-Zhao reaction of the aldehyde **104** with ylide **24** did not produce iododiene **100** and the starting material **104** decomposed.

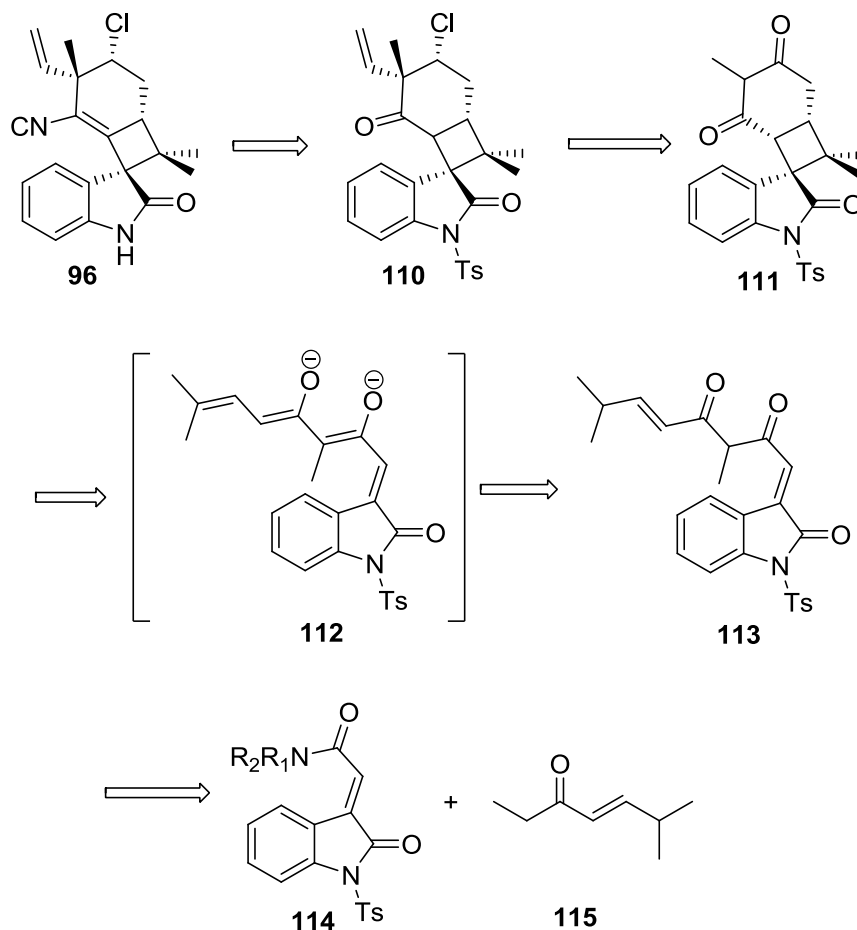


Scheme 32. Attempt at preparation of iododiene **100**.



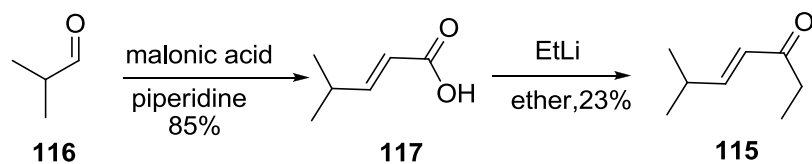
Scheme 33. Attempt at preparation of dienyborate **101**.

Zinc-mediated addition of propargyl bromide **106** to acetone gave the tertiary alcohol **107** in 77% yield (Scheme 33).⁶⁹ Acid-catalyzed addition of **107** to 2,3-dihydropyran (DHP) did not afford desired 2-alkoxytetrahydrofuran **108** which was supposed to be transformed to a conjugated 4-methylpentenyne **109** by base-catalyzed elimination.⁷⁰



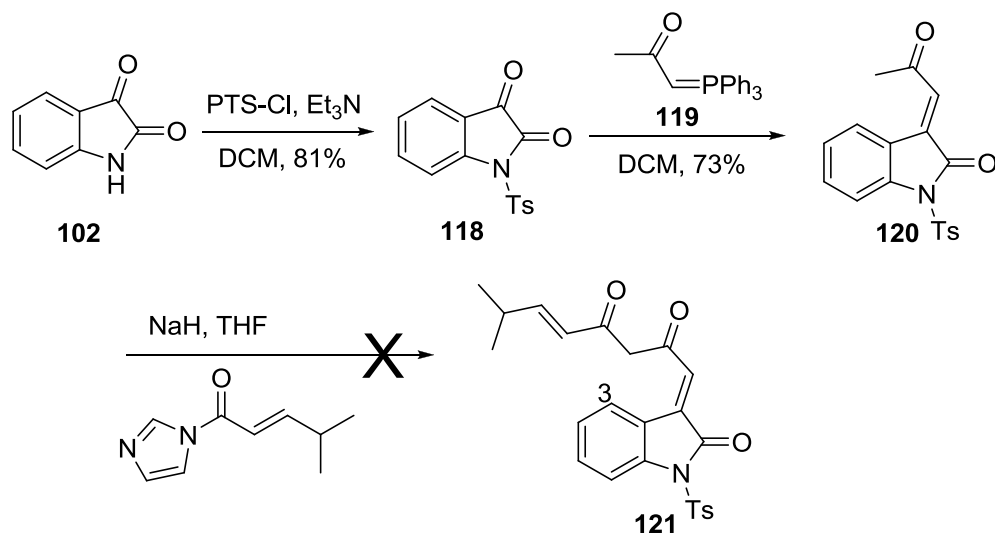
Scheme 34. Revised synthetic route for welwitindolinone A isonitrile **96**.

Due to the absence of iododiene **100**, the proposed route (Scheme 31) for the synthesis of welwitindolinone A isonitrile **96** was revised (Scheme 34). The system of triphenylphosphine and tetrachloromethane might transform β -diketone **111** to β -chloro- α,β -unsaturated ketone.⁷¹ Reduction and further transformations of the β -chloroketone might produce the welwitindolinone intermediate **110**. Base catalysis might move the γ,δ -double bond of β -diketone **113** to δ,ϵ -position and result in a dienolate intermediate **112**.⁷² The tandem $8\pi,6\pi$ -electronic cyclization of **112** could lead to bicycle[4,2,0]octadione **111**.⁷³ The cross condensation of ketone **115** and amide **114** might afford β -diketone **113**.⁷⁴



Scheme 35. Preparation of ketone **115**.

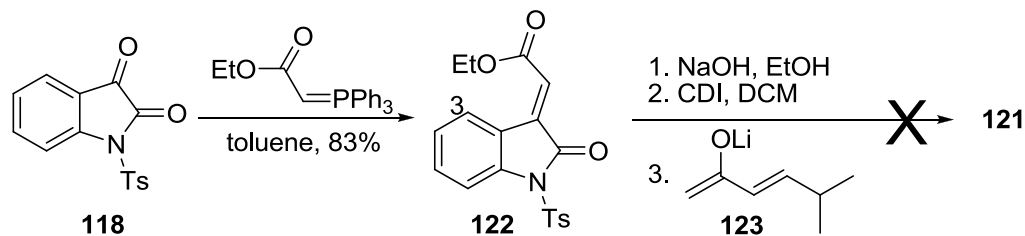
Isobutanal **116** condensed with malonic acid in the presence of piperidine to produce (E)- α,β -unsaturated acid **117** in 85% yield (Scheme 35). The addition of two equivalents of ethyl lithium (EtLi) to the acid **117** resulted in α,β -unsaturated ketone **115** in 23% yield.⁷⁵



Scheme 36. Attempt at preparation of β -diketone **121**.

Isatin **102** was protected by p-toluenesulfonyl chloride (PTS-Cl) to form N-tosylisatin **118** in 81% yield (Scheme 36).⁷⁶ Wittig reaction **119** transformed the tosylisatin **118** to α,β -unsaturated ketone **120** in 73% yield.⁷⁷ The NMR spectrum of ketone **120** showed that the proton on C3 of the benzene ring is deshielded by the C=O double bond and has a larger down-field chemical shift than that of compound **118** do. Thus, compound **120** is the (E)-isomer. Condensation of the ketone **120** was subjected

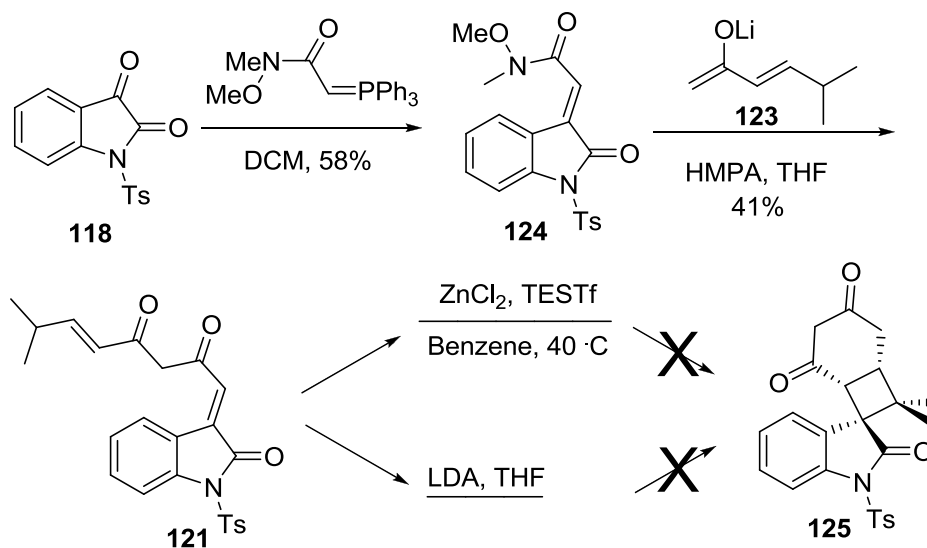
to the acyl imidazole derived from the acid **117**.⁷⁸ However, the desired compound **121** was not produced.



Scheme 37. The second attempt at preparation of β -diketone **121**.

(E) - α,β -Unsaturated ester **122** was derived from tosylisatin **118** by a Wittig reaction in 83% yield (Scheme 37). The NMR spectrum of the product also showed that the proton on C3 of benzene ring has a larger down-field chemical shift, which means the compound has the (E) -configuration.⁷⁷ The ester **122** was hydrolyzed and then the acid was transformed to an acyl imidazole. The addition of an enolate of (E) -5-methylhex-3-en-2-one **123** to the acyl imidazole failed to lead to the diketone product **121** and gave the starting material back.

Weinreb amide **124** was provided by a Wittig reaction of tosylisatin **118** in 58% yield (Scheme 38). The NMR spectrum indicated only the (E) -isomer was present.⁷⁷ The cross condensation of Weinreb amide **124** and (E) -5-methylhex-3-en-2-one **123** afforded the target diketone **121** in 41% yield.⁷⁴ Two attempts failed to transform the diketone **121** to bicyclo[4,2,0]octadione **125**. First, treatment of the diketone **121** with zinc chloride and trimethylsilyl triflate in benzene was tried, but the target product **125** was not obtained and the starting material was recovered. Second, the diketone **121** was deprotonated by excess lithium diisopropylamide (LDA); however, the resulting dienolate did not cyclize to bicyclo[4,2,0]octadione **125** and the starting material was recovered.



Scheme 38. Attempt at electrocyclization of β -diketone 125.

1.3 Experimental Section

General

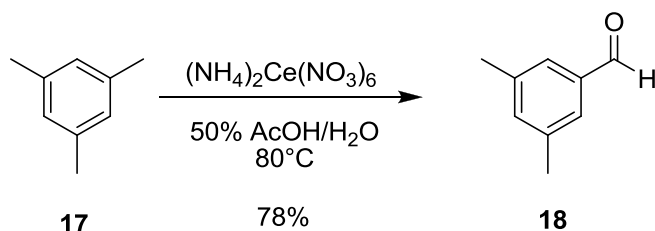
For all compounds, ^1H , ^{13}C and NOE NMR spectra were recorded on Varian Gemini 2300 (300 MHz), Inova 400 (400 MHz), Inova 500 (500 MHz), Inova 600 (600 MHz) spectrometer. Chemical shifts were measured relative to the residual solvent resonance for ^1H and internal reference. IR spectra were recorded on a Mattson Galaxy Series FTIR 3000 or 2020 spectrophotometer.

Reagents and solvents were purchased from Aldrich Chemical Company, Fisher Scientific or Acros Organics. Liquid reagents were purified by distillation prior to use. Unless otherwise noted, solid reagents were used without further purification. Methylene chloride, acetonitrile, cyclohexane, and toluene were distilled from CaH_2 under Ar atmosphere. Diethyl ether and THF were distilled from Na and benzophenone under Ar atmosphere.

All reactions unless otherwise noted were performed in oven-dried glassware under positive pressure of Ar. All yields are isolated yields.

Chromatographic separations were performed using Fisher grade 1740 type 60 Å silica gel (170-400 mesh). Analytical thin layer chromatography was performed using Analtech Uniplate pre-coated glass 250 micron silica gel GHLF plates with 254 nm fluorescent indicator. Preparative thin layer chromatography was performed using Analtech Uniplate pre-coated glass 1000 or 2000 micron silica gel GHLF plates with 254 nm fluorescent indicator.

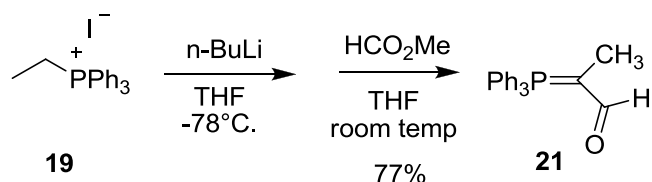
3,5-Dimethylbenzaldehyde **18**.²³



In a two-necked round bottom flask, 2 mL (14 mmol) of mesitylene **17** was mixed with 70 mL of 50% acetic acid in H_2O . The reaction mixture was heated to 80°C . 31.37 g (57 mmol) ammonium cerium(IV) nitrate in 140 mL of 50% acetic acid/ H_2O was added dropwise into the reaction mixture. The color of the reaction system was always yellow. After the addition of the oxidant, the mixture was refluxed for 120 minutes till the reaction mixture turned colorless. The reaction system was cooled down and diluted by water. The solution was extracted three times with CH_2Cl_2 . The organic layers were combined and dried by Na_2SO_4 . After concentration by rotovap, the organic products were separated by a silica gel column with 30:1 hexanes and AcOEt to give 1.493 g (11 mmol) of colorless liquid **18** in 78% yield.

$^1\text{H NMR}$ (CDCl_3): δ 9.93 (d, 1H, $^4J = 0.6$ Hz), 7.48 (dd, 2H, $^4J = 0.6$ Hz, $^4J = 0.3$ Hz), 6.96 (m, 1H), and 2.39 (dd, 6H, $^4J = 2.4$ Hz, $^4J = 0.3$ Hz). IR (KBr): ν_{max} 3031, 2969, 1738, 1698, 1598, and 1463 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.²³

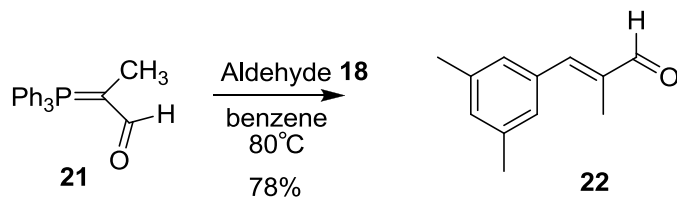
2-(Triphenylphosphoranylidene)propionaldehyde **21.**²⁴



In 250 mL round bottom flask, 22 g (53 mmol) of ethyltriphenylphosphonium iodide **19** was mixed with 125 mL of THF. The 20 mL of 2.5 M n-BuLi was added dropwise to the suspension. The reaction turned into a red transparent solution. After the addition of n-BuLi, the reaction solution was stirred for 20 minutes. Then the red solution was transferred to a suspension of 6.5 g (68 mmol) of t-BuONa and 30 mL of THF. The reaction mixture was allowed to cool to 0 °C. After stirring for 10 minutes, 5 mL (83mmol) of methyl formate was added to this suspension rapidly. The reaction mixture turned white from red quickly. After stirring for 7 hours, when this mixture was completely white, the reaction was quenched by 1M HCl. The pH value was adjusted to 8.0. Then the solution was extracted by CH_2Cl_2 . The organic layers were combined and dried over Na_2SO_4 . After solvent was removed by rotovap, 19 g of pale yellow solid was gotten. It was recrystallized in CH_2Cl_2 and Et_2O to give 17.1 g (42 mmol) of white solid **21** in 77% yield.

$^1\text{H NMR}$ (CDCl_3): δ 8.06 (d, 1H, $^4J = 4.8$ Hz), 7.6 (m, 15H), and 1.87 (d, 3H, $^4J = 13.5$ Hz). The spectroscopic and other physical data reported are consistent with values reported in the literature.²⁴

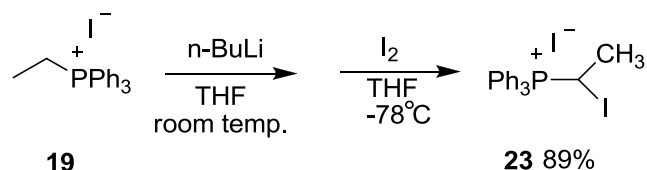
(E)-2-Methyl-3-(3,5-dimethylphenyl)-propenal 22.



249 mg (0.94 mmol) of 2-(triphenylphosphoranylidene)propionaldehyde **21** and 105 mg (0.783 mmol) of aldehyde **18** were mixed in 20 mL of dry benzene.²⁴ The reaction was refluxed at 80 °C. The NMR of the reaction mixture was checked every five hours. After two days, the reaction was quenched by water. The organic layer was separated and the water layer was extracted three times with CH₂Cl₂. The organic layers were combined and dried over Na₂SO₄. After solvents were removed by rotovap, the organic products were separated by a silica gel column with 30:1 hexanes and AcOEt to give 90 mg (0.61 mmol) of yellow liquid **22** in yield 78%.

¹H NMR (CDCl₃): δ 9.57 (s, 1H), 7.16 (s, 2H), 7.05 (s, 1H), 6.98 (s, 1H), 2.37 (s, 6H), and 2.09 (s, 3H). ¹³C NMR (CDCl₃): δ 195.9, 150.6, 138.5, 135.3, 131.6, 130.1, 128.1, 126.3, 21.5, and 11.2. IR (KBr): ν_{max} 3010, 2930, 2875, 2710, 1745, 1710, and 1610 cm⁻¹

(1-Iodoethyl)triphenyl-phosphonium iodide 24.²⁰

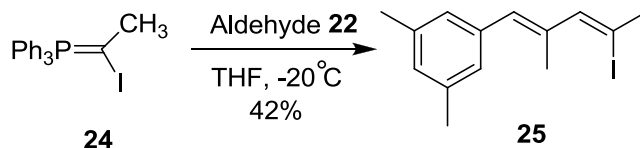


In 250 mL round bottom flask, 11 g (26 mmol) of ethyltriphenylphosphonium iodide **19** was mixed with 125 mL of THF. 10 mL of 2.5 M n-BuLi was added dropwise to the suspension. The reaction turned into a red transparent solution. After the addition of n-BuLi, the reaction solution was stirred for 20 minutes. The red solution was transferred to 21 mL of 1.1 M I₂/THF solutions at -78 °C by a cannula. The reaction

turned white from red. After addition of the red ylide, the reaction was stirred at room temperature for 5 hours. Then this mixture was quenched by 1 M HCl and a yellow solid was filtered off. This solid was washed by Et₂O till it turned white. 12 g (23 mmol) of white solid **23** was given (89%).

¹HNMR (CDCl₃): δ 7.8 (m, 15H), 6.93 (m, 1H), and 2.2 (d, 2H, ³J = 7.2 Hz). The spectroscopic and other physical data reported are consistent with values reported in the literature.²⁰

(E,Z)-Iodo-1,3-pentadiene 25.

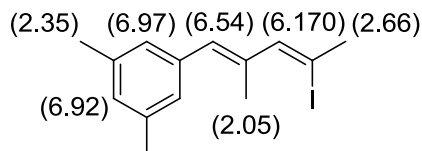


In 100 mL round bottom flask, 602 mg (1.1 mmol) of (1-iodoethyl)triphenylphosphonium iodide **24** was mixed with 20 mL of THF. The reaction was cooled down to -20 °C. The 1.8 mL of 1 M NaHMDS was added dropwise to the reaction. The yellow mixture turned into dark red and transparent. After stirring for 15 minutes, 150 mg (0.86 mmol) of aldehyde **22** in 5 mL of THF was added dropwise to the red solution. The dark red reaction solution turned into a brown black mixture. After stirring for 4 hours, this reaction was quenched by saturated NH₄Cl solution. The organic layer was separated and the water layer was extract three times with CH₂Cl₂. The organic layers were combined and dried over Na₂SO₄. After solvents were removed by rotovap, the organic products were separated by a silica gel column with hexanes to give 64 mg (0.36 mmol) of yellow liquid **25** in 42% yield.

¹HNMR (CDCl₃): δ 6.94 (s, 2H), 6.89 (s, 1H), 6.51 (s, 1H), 6.14 (s, 1H), 2.63 (s, 3H), 2.33 (s, 6H), and 2.02 (s, 3H). ¹³CNMR (CDCl₃): δ 138.8, 137.8, 137.5, 136.0,

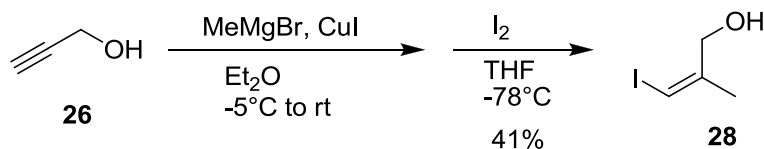
131.6, 128.7, 127.2, 127.1, 97.8, 35.4, 29.9, 21.6, and 18.2. IR (KBr): ν_{max} 3010, 2980, 2885, 2810, 1620, and 1420 cm^{-1}

NOE:



Irradiated (saturated) peak	Enhanced peaks
2.66ppm	6.17ppm
6.17ppm	6.54, 2.66ppm
6.54ppm	6.97, 6.17ppm

(2Z)-3-Iodo-2-methyl-2-propen-1-ol **28.**²⁵

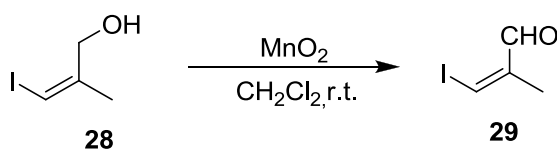


In 1 L round bottom flask, 1.5 mL (26.8 mmol) of propargyl alcohol **26** and 2 g (10.4 mmol) of dry CuI were mixed in 200 mL of Et₂O. 14 mL of 2.5 M methyl magnesium bromide (MeMgBr) was added dropwise to the reaction at 0 °C. The first 6 mL was added very carefully. After the CH₄ bubbling stopped, the left 8 mL was added a little bit fast. After addition of MeMgBr, the reaction became very viscous. After 20 hour continuous stirring, the reaction mixture turned dark green. 15 mL of 1.1 M I₂ solution in THF was added dropwise to the reaction mixture at -78 °C. The color of the reaction mixture first turned red from dark green, and then became yellow. The reaction mixture was stirred over night and quenched by the saturated NH₄Cl solution. The organic layer was separated and the water layer was extracted three times with CH₂Cl₂. The organic layers were combined and dried over Na₂SO₄. After solvents were removed by rotovap,

the organic products were separated by a silica gel column with 5:1 hexanes and AcOEt to give 2.1714 g (11 mmol) of yellow liquid **28** in 41% yield.

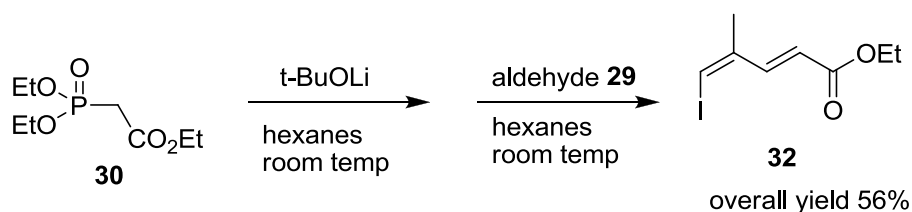
$^1\text{H NMR}$ (CDCl_3): δ 5.98 (s, 1H), 4.25 (s, 2H), and 1.98 (s, 3H). IR (KBr): ν_{max} 3455, 2920, 2950, 1721, 1460, 1379, 1138, and 1085 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.²⁵

(2Z)-3-Iodo-2-methyl-2-propenal 29.²⁹



In 50 mL round bottom flask, 200 mg (1 mmol) of alcohol **28** was mixed with 30 mL of dry CH_2Cl_2 . 2 g (23 mmol) of MnO_2 was added into the reaction solution rapidly. The black reaction suspension was stirred for 3 hours. The extra MnO_2 was filtered off. The solvents were removed by rotovap. This aldehyde is not stable, so it was used in the next step as soon as possible.

(2E,4Z)-5-Iodo-diene ester 32.²⁷



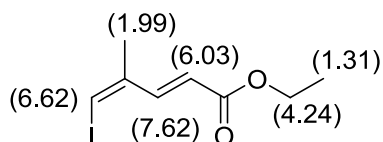
In 50 mL round bottom flask, 0.16 mL (0.71 mmol) of triethyl phosphonoacetate **30** was mixed with 20 mL of hexanes. 1 mL of 1 M lithium tert-butoxide was added dropwise to the solution. The reaction solution turned pale yellow from colorless. After 15 minutes stirring, 172 mg (0.65 mmol) of 2-(Z)-3-iodo-2-methyl-2-propenal **29** produced in the last step in 10 mL of hexanes was added dropwise to the reaction solution by syringe. The reaction solution turned orange. After 8 hours stirring, some small

insoluble oil drops appeared. The reaction was quenched by water. The organic layer was separated and the water layer was extract three times with CH₂Cl₂. The organic layers were combined and dried over Na₂SO₄. After solvents were removed by rotovap, the organic products were separated by a silica gel column with 20:1 hexanes and AcOEt to give 141 mg (0.36 mmol) of yellow liquid **32** in 56% yield.

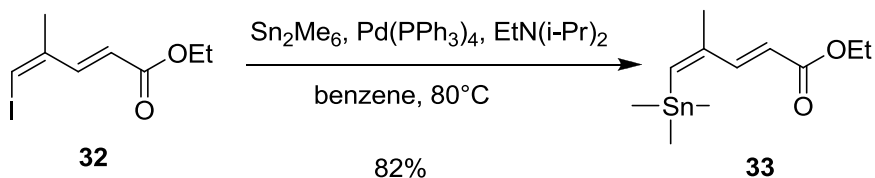
¹HNMR (CDCl₃): δ 7.62 (d, 1H, ³J = 15.6 Hz), 6.62 (s, 1H), 6.03 (d, 1H, ³J = 15.6 Hz), 4.23 (q, 2H, ³J = 7.2 Hz), 1.99 (s, 3H), and 1.31(t, 3H, ³J = 7.2 Hz). The spectroscopic and other physical data reported are consistent with values reported in the literature.²⁷

NOE:

Irradiated (saturated) peak	Enhanced peaks
6.62ppm	1.99ppm



(2E,4Z)-5-(Trimethylstannyl)-diene ester 33.²⁰

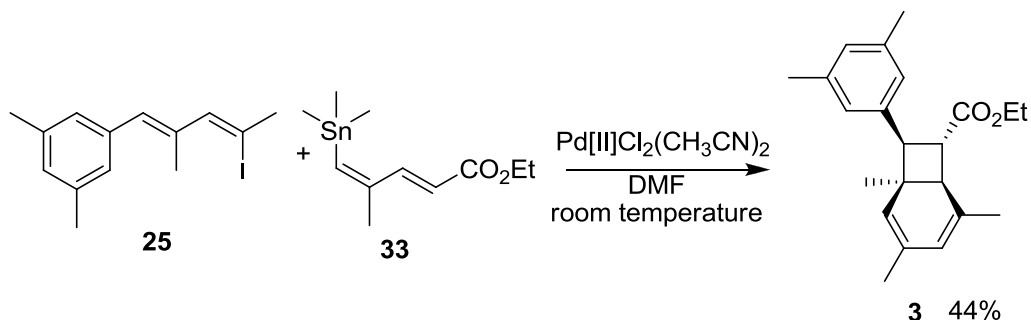


In 50 mL two-necked RBF, 130 mg (0.49 mmol) of (2E,4Z)-5-iodo-diene ester **32** was mixed with 30 mL of benzene. Then, 30 mg of catalyst tetrakis(triphenylphosphine)palladium(0) and 30 μL of N,N-diisopropylethylamine were added to the reaction solution. Finally, 200 μL of hexamethylstannane was added to the reaction solution. The reaction solution was refluxed at 80°C overnight. The pale yellow

reaction solution became a black mixture. After 15 hours refluxing, the reaction was cooled down to room temperature and stirred with KF and celite. The resulted insoluble matters were filtered off by celite. The residues were purified by a basic alumina column with benzene to give 120 mg (0.41 mmol) of pale yellow liquid **33** (82%).

$^1\text{H NMR}$ (C_6D_6): δ 7.69 (d, 1H, $^3J = 15.6$ Hz), 6.23 (s, 1H), 5.96 (d, 1H, $^3J = 15.6$ Hz), 4.03 (q, 2H, $^3J = 7.2$ Hz), 1.77 (s, 3H), 0.98 (t, 3H, $^3J = 7.2$ Hz), and 0.19 (s, 9H). The spectroscopic and other physical data reported are consistent with values reported in the literature.²⁰

SNF analog **3**.



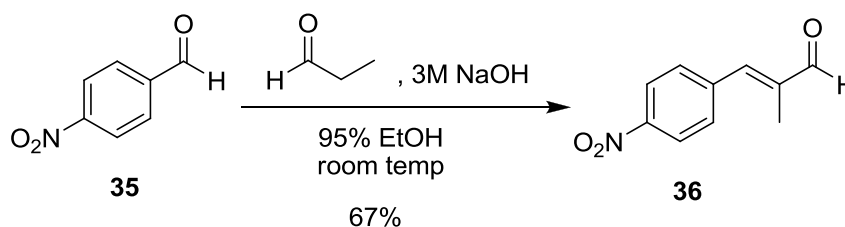
In a 50 mL round bottom flask, 40 mg (0.132 mmol) of (2E,4Z)-5-(trimethylstannyl)-diene ester **33** was mixed with 30 mL of dry DMF. Then 40 mg (128 mmol) of (E,Z)-iodo-1,3-pentadiene **25** and 3 mg of bis(acetonitrile)dichloropalladium(II) were added into the reaction solution. The reaction mixture was stirred in dark for 24 hours. 1 mg of catalyst $\text{PdCl}_2(\text{CH}_3\text{CN})_2$ was added. After 5 hours stirring, the reaction was quenched by potassium fluoride (KF) and Celite. The resulted insoluble matters were filtered off by Celite. The residues were washed by saturated NaHCO_3 . The solution was extract three times with CH_2Cl_2 . The organic layers were combined and dried over Na_2SO_4 . After solvents were removed by rotovap, the organic products were separated by a silica gel column with 200:1 hexanes and AcOEt to give 21 mg (0.058 mmol) of yellow liquid **3** (44%).

^1H NMR (CDCl_3): δ 6.86 (s, 1H), 6.81 (s, 2H), 5.44 (s, 1H), 4.58 (s, 1H), 4.13 (m, 2H), 3.59 (d, 1H, $^3J = 10.8$ Hz), 3.42 (dd, 1H, $^3J = 10.8$ Hz, $^3J = 9.0$ Hz), 2.63 (d, 1H, $^3J = 9.0$ Hz), 2.30 (s, 6H), 1.78 (s, 3H), 1.64 (s, 3H), 1.23 (t, 3H, $^3J = 6.6$ Hz), and 1.22 (s, 3H). ^{13}C NMR (CDCl_3): δ 174.0, 137.7, 134.1, 129.7, 128.1, 122.2, 60.3, 56.0, 46.2, 46.2, 43.4, 29.7, 28.4, 21.9, 21.6, and 14.2. IR (KBr): ν_{max} 2950, 2855, 1740, 1600, 1440, 1380, and 1005 cm^{-1}

NOE:

Irradiated (saturated) peak	Enhanced peaks
5.455 ppm	1.793, 1.645 ppm
4.602 ppm	1.645 ppm
3.42 ppm	6.822 ppm
3.608 ppm	6.813 ppm
2.64 ppm	3.617 ppm

(E)-2-Methyl-3-(4-nitro-phenyl)-propenal 36.²⁸

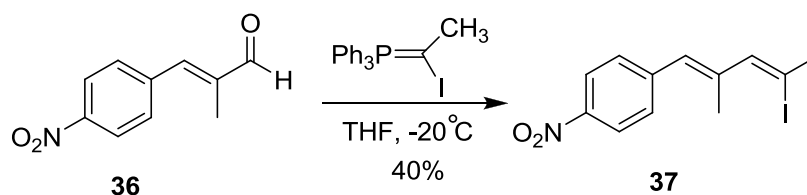


In a 50 mL round bottom flask, 2 g (13.2 mmol) of 4-nitrobenzaldehyde **35** and propionaldehyde were mixed in 95% ethyl alcohol. 2 mL of 3 M NaOH was added into the reaction mixture at room temperature. The reaction mixture became dark red. After 4 hours stirring, the reaction was quenched by 1 M HCl. The pH value was adjusted to 8.0. The reaction mixture was extracted three times with Et_2O . The organic layers were combined and dried over Na_2SO_4 . After the solvents were removed in vacuo, the organic products were separated by a silica gel column with 2:1 hexanes and AcOEt. The product

was recrystallized in a mixture of EtOH and hexanes to give 1.7 g (8.8 mmol) of yellow solid **36** (67%).

$^1\text{H NMR}$ (CDCl_3): δ 9.65 (s, 1H), 8.30 (d, 2H, $^3J = 8.8$ Hz), 7.66 (d, 2H, $^3J = 8.8$ Hz), 7.32 (s, 1H), and 2.09 (s, 3H). The spectroscopic and other physical data reported are consistent with values reported in the literature.²⁸

(E,Z)-1-(4-Nitro-phenyl)-2-methyl-4-iodo-1,3-pentadiene 37.²⁰

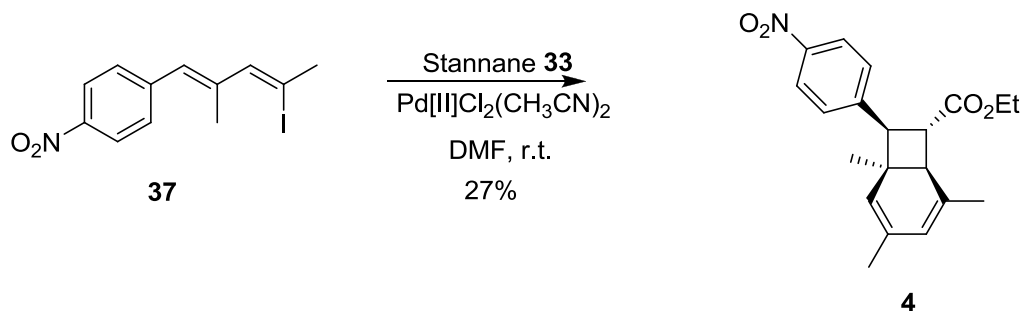


In 50 mL round bottom flask, 3 g (5.5 mmol) of (1-iodoethyl)triphenylphosphonium iodide was added to 25 mL of THF. The reaction suspension was cooled down to -20 °C. Then 4 mL of 1 M NaHMDS was added dropwise to the reaction suspension. The yellow mixture became dark red and transparent. After 15 minutes stirring, 1 g (5.3 mmol) of (*E*)-2-methyl-3-(4-nitro-phenyl)-propenal **36** in 5 mL of THF was added dropwise to the red solution. The dark red reaction solution turned into a brown black mixture. After 4 hours stirring, this reaction was quenched by the saturated NH_4Cl solution. The organic layer was separated and the water layer was extract three times with CH_2Cl_2 . The organic layers were combined and dried over Na_2SO_4 . After solvents were removed in vacuo, the organic products were separated by a silica gel column with 50:1 hexanes and AcOEt to give 703 mg (2.2 mmol) of yellow liquid **37** (40%).

$^1\text{H NMR}$ (CDCl_3): δ 8.20 (d, 2H, $^3J = 8.0$ Hz), 7.44 (d, 2H, $^3J = 8.0$ Hz), 6.59 (s, 1H), 6.17 (s, 1H), 2.65 (s, 3H), and 2.01 (s, 3H). IR (KBr): ν_{max} 3071, 3000, 2961, 2913,

1592, 1514, 1491, and 1434 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.²⁰

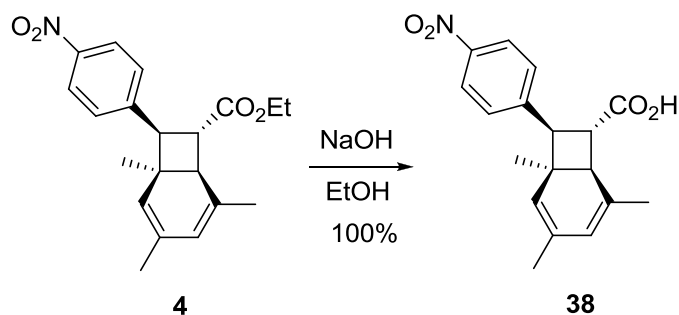
SNF analog 4.²⁰



In a 50 mL round bottom flask, 35 mg (0.127 mmol) of (2E,4Z)-5-(trimethylstannyl)-pentadiene ester **33** was dissolved in 30 mL of dry DMF. Then 42 mg (0.138 mmol) of compound **37** and 3 mg of bis(acetonitrile) dichloropalladium(II) were added into the reaction solution. The reaction mixture was stirred in dark for 24 hours. 1 mg of catalyst $\text{PdCl}_2(\text{CH}_3\text{CN})_2$ was added. After extra 5 hours stirring, the reaction was quenched by KF and Celite. The resulted insoluble matters were filtered off by Celite. The residues were washed by saturated NaHCO_3 . The solution was extract three times with CH_2Cl_2 . The organic layers were combined and dried over Na_2SO_4 . After solvents were removed in vacuo, the organic products were separated by a silica gel column with 40:1 hexanes and AcOEt to give 10 mg (0.034 mmol) of yellow liquid **4** (27%).

^1H NMR (CDCl_3): δ 8.17 (d, 2H, $^3J = 8.0$ Hz), 7.23 (d, 2H, $^3J = 8.0$ Hz), 5.47 (s, 1H), 4.43 (s, 1H), 4.14 (m, 2H), 3.75 (d, 1H, $^3J = 10.8$ Hz), 3.44 (dd, 1H, $^3J = 10.8$ Hz, $^3J = 9.0$ Hz), 2.74 (d, 1H, $^3J = 9.0$ Hz), 1.80 (s, 3H), 1.62 (s, 3H), 1.26 (s, 3H), and 1.24 (t, 3H, $^3J = 6.6$ Hz). ^{13}C NMR (CDCl_3): δ 173.5, 146.9, 145.9, 134.3, 131.1, 128.3, 123.6, 122.4, 121.372, 60.8, 56.0, 46.1, 44.3, 28.5, 22.1, 21.7, and 14.3. IR (KBr): ν_{max} 2966, 2941, 2913, 1727, 1620, 1518, 1445, and 1178 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.²⁰

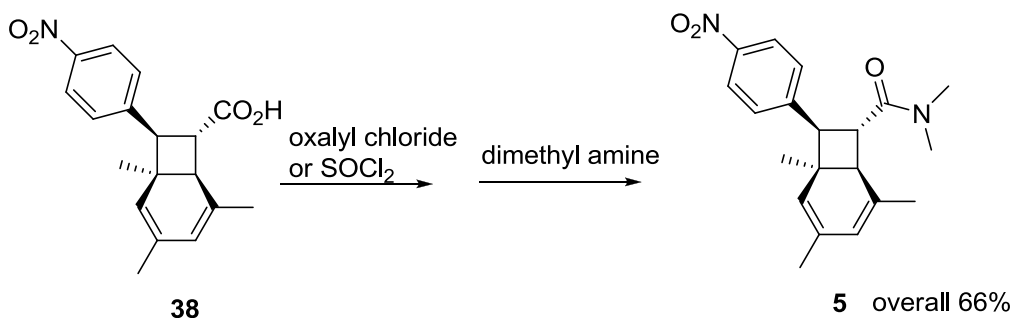
SNF acid 38.



The ester (**4**, 6 mg, 0.0175 mmol) was dissolved in ethanol (95%, 1 mL). KOH/EtOH (0.05 M, 1 mL) solution was added to the mixture dropwise. The yellow solution was stirred at r.t. over night. The reaction was quenched by HCl/H₂O (1 M, 5 mL). The water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na₂SO₄. The organic solution was concentrated in vacuo to give 6mg crude product. The NMR showed that the crude product was pure enough.

¹HNMR (CDCl₃): δ 8.17 (d, 2H, ³J = 8.5 Hz), 7.34 (d, 2H, ³J = 8.5 Hz), 5.47 (s, 1H), 4.42 (s, 1H), 3.74 (d, 1H, ³J = 10 Hz), 3.49 (t, 1H, ³J = 10 Hz, ³J = 9.5 Hz), 2.76 (d, 1H, ³J = 9.5 Hz), 1.80 (s, 3H), 1.62 (s, 3H), and 1.26 (s, 3H). ¹³CNMR (CDCl₃): δ 179.1, 147.0, 145.6, 134.2, 131.4, 128.4, 123.7, 122.7, 121.2, 56.1, 46.0, 44.4, 29.9, 28.7, 22.2, and 21.7. IR (KBr): ν_{max} 3500, 2945, 2825, 1662, 1619, 1474, and 1258cm⁻¹.

SNF dimethyl amide 5.

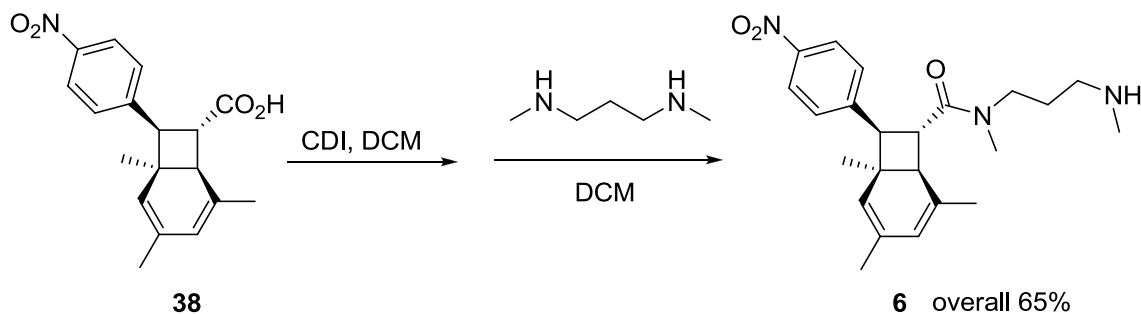


The acid (**38**, 6 mg, 0.019 mmol) was dissolved in dry benzene (2 mL). Dimethylformamide (1 drop) was added to the solution. Oxalyl chloride (0.05 mL) was added to the mixture at 0 °C.³¹ The reaction was stirred at 0 °C for 2 hours and at r.t. for 1 hour. The mixture was concentrated in vacuo. Dimethyl amine (40% in water, 0.5 mL) in acetone (0.5 mL) was added to the acid chloride at 0 °C. The reaction was stirred at r.t. overnight and quenched by HCl (1 M in water, 2 mL). The water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na₂SO₄. The organic solution was concentrated in vacuo. The residue was purified by a TLC plate with ethyl acetate and hexanes (1:2) to give the amide **5** (4 mg, 0.012 mmol, 66%).

¹HNMR (CDCl₃): δ 8.16 (d, 2H, ³J = 9.0 Hz), 7.33 (d, 2H, ³J = 9.0 Hz), 5.49 (s, 1H), 4.50 (s, 1H), 3.87 (d, 1H, ³J = 9.6 Hz), 3.71 (dd, 1H, ³J = 9.6 Hz, ³J = 9.0 Hz), 2.94 (d, 6H, ³J = 7.2 Hz), 2.81 (d, 1H, ³J = 9.0 Hz), 1.74 (s, 3H), 1.68 (s, 3H), and 1.24 (s, 3H).
¹³CNMR (CDCl₃): δ 161.9, 134.5, 132.8, 130.0, 128.7, 123.6, 56.0, 54.1, 46.5, 44.4, 36.2, 29.9, 28.3, and 22.2. IR (KBr): ν_{max} 3390, 3290, 2985, 2810, 1715, 1382, and 1149 cm⁻¹.

NOE:

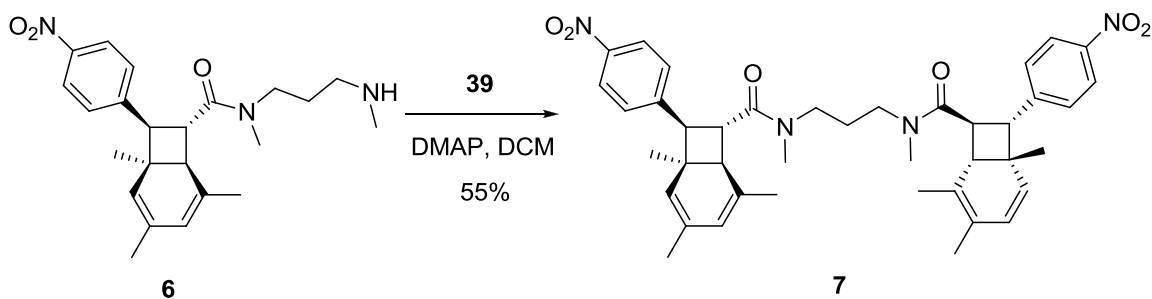
Irradiated (saturated) peak	Enhanced peaks
5.492 ppm	1.744, 1.675 ppm
4.502 ppm	1.677 ppm
3.8 ppm	3.81, 1.237, 7.33 ppm
3.69 ppm	2.936, 7.33 ppm
2.81 ppm	3.87, 1.746, 1.238 ppm

SNF analog 6.

The acid (**38**, 50 mg, 0.16 mmol) and carbonyl diimidazole (40 mg, 0.25 mmol) were dissolved in dry methylene chloride (2 mL).³¹ The mixture was stirred at r.t. for one hour. A little gas was released. The N,N'-dimethyl-1,3-propanediamine (0.2 mL) in dry methylene chloride (1 mL) was added to the mixture. The reaction was stirred at r.t. for 18 hours. The solution was diluted with 10ml methylene chloride. The mixture was washed by water. The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na₂SO₄. The organic solution was concentrated in vacuo. The residue was separated by a TLC plate with methanol and methylene chloride (1:10) to give yellow oil **6** (42 mg, 10 mmol, 65%)

¹HNMR (CDCl₃): δ 8.15 (d, 2H, ³J = 8.0 Hz), 7.31 (d, 2H, ³J = 8.0 Hz), 5.48 (s, 1H), 4.74 (s, 1H), 3.77 (d, 1H, ³J = 9.6 Hz), 3.70 (t, 1H, ³J = 9.6 Hz, ³J = 8.4 Hz), 3.49 (m, 2H), 2.94 (s, 3H), 2.81 (m, 2H), 2.72 (d, 1H, ³J = 8.4 Hz), 2.60 (s, 3H), 2.09 (m, 3H), 1.70 (s, 3H), 1.65 (s, 3H), and 1.21 (s, 3H). ¹³CNMR (CDCl₃): δ 174.2, 147.0, 145.7, 133.5, 131.1, 128.5, 123.7, 123.3, 122.2, 70.6, 55.8, 46.8, 46.6, 45.1, 44.3, 44.2, 35.9, 33.3, 28.3, 24.0, 22.1, and 22.1. IR (KBr): ν_{max} 3390, 2985, 1740, 1630, 1515, 1345, 1275, and 1115 cm⁻¹. ESI-MS (M-H⁺): 398.14, 399.19, and 400.22 g/mol.

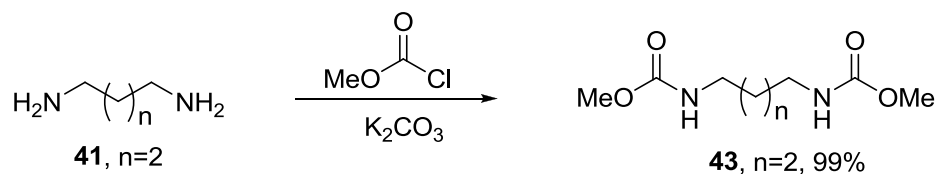
SNF analog 7.



The acid (**38**, 29 mg, 0.09 mmol) was dissolved in dry methylene chloride. Thionyl chloride (0.1 mL) was added dropwise to the mixture at 0 °C.³² The reaction was stirred for half an hour and then concentrated in vacuo. The residues were placed in high vacuo for 2 hours to remove extra thionyl chloride. 4-Dimethylaminopyridine (12 mg, 0.1 mmol) and the amine (**6**, 15 mg, 0.1 mmol) were mixed in methylene chloride (2 mL). The residues were added dropwise to the mixture by a pipet at 0 °C. The reaction was stirred at r.t. for 3 hours and quenched by *N,N'*-dimethyl-1,3-propanediamine (0.1 mL). The mixture was washed by sat. ammonium chloride solution. The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na₂SO₄. The organic solution was concentrated in vacuo. The residue was separated by a TLC plate with ethyl acetate and hexanes (1:1) to give yellow oil **7** (15 mg, 0.05 mmol, 55%).

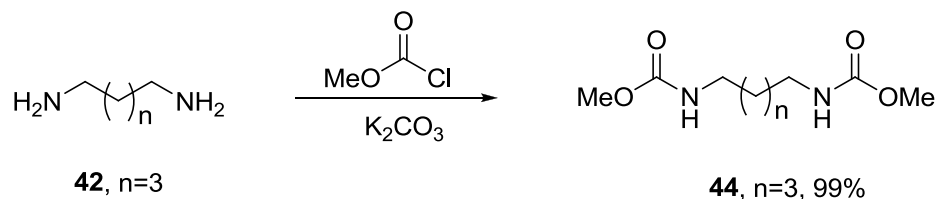
¹HNMR (CDCl₃, mixture): δ 8.14 (m, 4H), 7.33 (m, 4H), 5.47 (d, 2H), 4.50 (d, 2H), 3.84 (m, 2H), 3.66 (m, 2H), 2.91 (m, 6H), 2.74 (m, 2H), 1.68 (m, 15H), and 1.22 (m, 6H). ¹³CNMR (CDCl₃): δ 172.5, 146.9, 146.3, 134.1, 131.0, 128.8, 128.5, 127.9, 123.6, 123.0, 122.4, 55.8, 47.6, 46.7, 46.6, 45.9, 44.4, 35.7, 34.0, 28.4, 26.7, 24.9, 22.2, and 22.0. IR (KBr): ν_{max} 2945, 1770, 1640, 1600, 1520, and 1135 cm⁻¹. ESI-MS (M-H⁺): 693.27, and 694.37 g/mol.

Dicarbamates **43**.³³



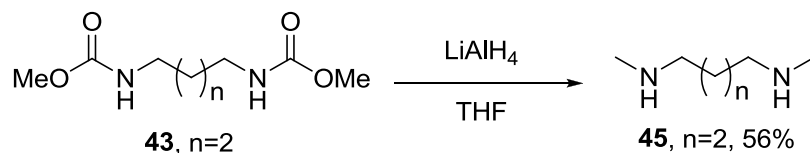
1,4-Butanediamine (2 g, 22.7 mmol) in water (30 mL) and methylene chloride (30 mL) was added potassium carbonate (3.2 g). Ethyl chloroformate (6.5 mL, 66 mmol) was added to the reaction at 0 °C. The reaction was stirred for 4 hours at room temperature. The solution was poured in 250 mL of 1 M ice cold hydrochloric acid solution. The solution was extracted three times with methylene chloride. The organic solution was concentrated to give a pure white solid **43** (4.54 g, 22.5 mmol, 99%) which was used directly in reduction.

Dicarbamates **44**.³³



1,5-Pentanediamine (2 mL, 20 mmol) in water (40 mL) and methylene chloride (40 mL) was added potassium carbonate (3.0 g). Methyl chloroformate (5.5 mL, 60 mmol) was added to the reaction at 0 °C. The reaction was stirred for 4 hours at room temperature. The solution was poured in 250 mL of 1 M ice cold hydrochloric acid solution. The solution was extracted three times with methylene chloride. The organic solution was concentrated to give a pure white solid **44** (5.92 g, 20 mmol, 99%) which was used directly in reduction.

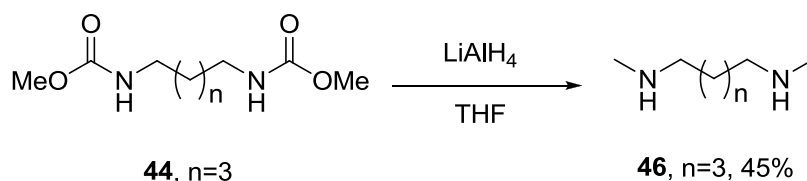
N,N'-Dimethyldiamines **45**.³⁴



4.54 g (28.4 mmol) of **43** in 120 mL of THF was added 2.0 g of LiAlH₄ very slowly at 0 °C. The reaction was warmed up to room temperature and refluxed for 3 hours. Celite was added to the reaction and the suspension was stirred for 15 minutes. The solids were filtered off. The filtrate was quenched by methanol. The solution was concentrated. The resulting liquid was distilled to give 1.143 g (15.9 mmol) of **45** (56%).

¹HNMR (CDCl₃): δ 2.47 (t, 4H, ³J = 6.0 Hz), 2.34 (s, 6H), 1.41 (m, 2H), and 1.36 (b, 2H). ¹³CNMR (CDCl₃): δ 52.036, 36.480, and 27.677. The spectroscopic and other physical data reported are consistent with values reported in the literature.³⁴

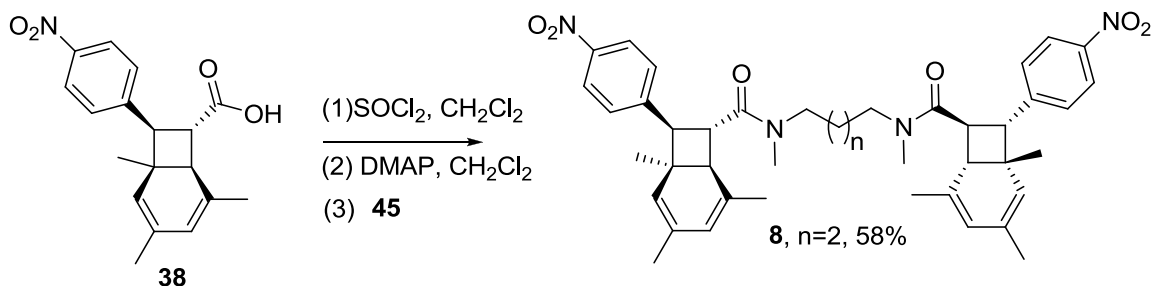
N,N'-Dimethyldiamines **46**.



5.92 g (31.1 mmol) of **44** in 120 mL of THF was added 2.0 g (52.6 mmol) of LiAlH₄ very slowly at 0 °C. The reaction was warmed up to room temperature and refluxed for 3 hours. Celite was added to the reaction and the suspension was stirred for 15 minutes. The solids were filtered off. The filtrate was quenched by methanol. The solution was concentrated. The resulting liquid was distilled to give 1.21 g (14 mmol) of **46** (45%).

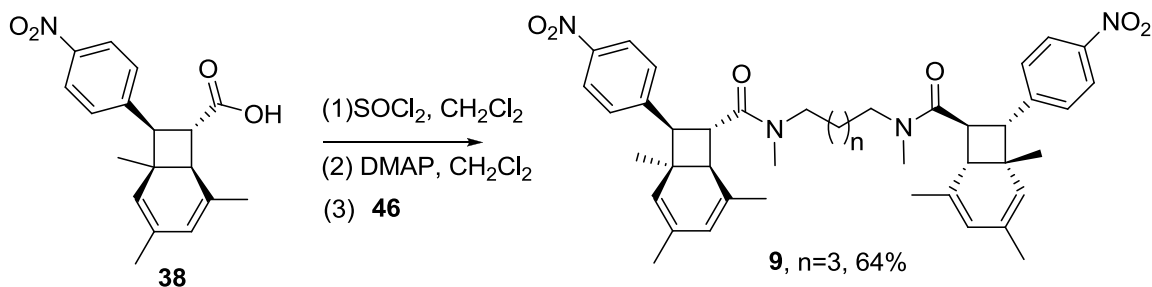
^1H NMR (CDCl_3): δ 2.49 (t, 4H, $^3J = 7.8$ Hz), 2.34 (s, 6H), 1.91 (b, 2H), 1.42 (tt, 4H, $^3J = 7.8$ Hz, $^3J = 6.6$ Hz), and 1.28 (m, 2H). ^{13}C NMR (CDCl_3): δ 52.0, 36.5, 29.8, and 25.1. IR (KBr): ν_{max} 3296, 2931, 2855, 2796, 1660, 1547, 1474, 1380, and 1116 cm^{-1} .

SNF analogue **8**.



SNF acid (19 mg, 0.06 mmol) in 3 mL of methylene chloride was added 0.1 mL of thionyl chloride at 0 °C. The mixture was stirred for half an hour at room temperature. The solution was concentrated in vacuo. 1 mL of methylene chloride was added to the resulting acid chloride and then of 7 mg of 4-dimethylaminopyridine was added. 3.3 mg (0.03 mmol) of diamine **45** was added to the solution. The reaction was stirred for 24 hours and then quenched by water. The water solution was extracted three times with methylene chloride. The organic solutions were combined, dried over MgSO_4 , and concentrated in vacuo. The residues were separated by a TLC plate with 1:1 hexanes and ethyl acetate to offer 12 mg (0.035 mmol) of compound **8** (58%).

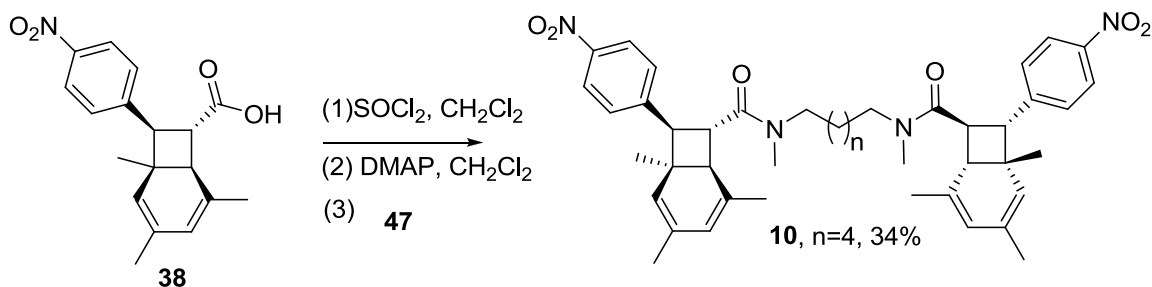
Mixture of diastereomers ^1H NMR (CDCl_3 , mixture): δ 8.16 (m, 4H), 7.34 (m, 4H), 5.48 (m, 2H), 4.50 (m, 2H), 3.84 (m, 2H), 3.65 (m, 4H), 3.45 (m, 2H), 2.86 (m, 6H), 2.75 (m, 2H), 1.68 (m, 12H), 1.44 (m, 4H), and 1.23 (m, 6H). ^{13}C NMR (CDCl_3): δ 172.4, 146.991, 134.0, 131.0, 129.9, 128.8, 128.6, 123.6, 122.7, 62.9, 56.6, 49.3, 47.7, 46.6, 44.7, 35.5, 28.3, 22.2, and 22.2. IR (KBr): ν_{max} 2963, 2929, 2875, 1640, 1603, 1514, 1462, 1379, and 1108 cm^{-1} . ESI-MS ($\text{M}-\text{H}^+$): 707.3, and 708.3 g/mol.

SNF analogue 9.

SNF acid (20 mg, 0.06 mmol) in 3 mL of methylene chloride was added 0.1 mL of thionyl chloride at 0 °C. The mixture was stirred for half an hour at room temperature. The solution was concentrated in vacuo. 1 mL of methylene chloride followed by 7 mg of 4-dimethylaminopyridine was added to the acid chloride. 4.6 mg (0.03 mmol) of diamine **46** was added to the solution. The reaction was stirred for 24 hours and quenched by water. The water solution was extracted three times with methylene chloride. The organic solutions were combined, dried over MgSO₄, and concentrated in vacuo. The residues were separated by a TLC plate with 1:1 hexanes and ethyl acetate to offer 14 mg (0.038 mmol) of compound **9** (64%).

Mixture of diastereomers ¹HNMR (CDCl₃, mixture): δ 8.16 (m, 4H), 7.34 (m, 4H), 5.48 (m, 2H), 4.50 (m, 2H), 3.84 (m, 2H), and 3.66 (m, 4H), 3.40 (m, 2H), 2.87 (m, 6H), 2.77 (m, 2H), 1.69 (m, 12H), 1.50 (m, 6H), and 1.23 (m, 6H). ¹³CNMR (CDCl₃): δ 172.3, 147.0, 134.3, 131.0, 129.9, 128.8, 128., 123.7, 122.9, 56.3, 49.8, 47.9, 46.5, 44.3, 35.5, 34.0, 28.3, 26.8, 26.9, and 22.2. IR (KBr): ν_{max} 2963, 2877, 1631, 1514, 1484, 1346, 1267, and 1108 cm⁻¹. ESI-MS (M-H⁺): 721.4 g/mol.

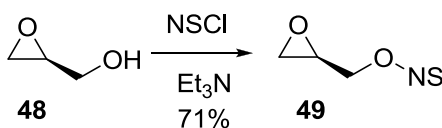
SNF analogue 10.



SNF acid (30 mg, 0.09 mmol) in 3 mL of methylene chloride was added 0.1 mL of thionyl chloride at 0 °C. The mixture was stirred for half an hour at room temperature. The solution was concentrated under vacuum. 1 mL of methylene chloride followed by 7 mg of 4-dimethylaminopyridine was added to the acid chloride. 0.010 mL (0.045 mmol) of *N,N'*-dimethyl-1,6-hexanediamine was added to the solution. The reaction was stirred for 24 hours and quenched by water. The water solution was extracted three times with methylene chloride. The organic solution was combined, dried over MgSO_4 , and concentrated in vacuo. The residues were separated by a TLC plate with 1:1 hexanes and ethyl acetate to offer 12 mg (0.03 mmol) of compound **10** (34%).

Mixture of diastereomers ^1H NMR (CDCl_3 , mixture): δ 8.16 (m, 4H), 7.33 (m, 4H), 5.48 (m, 2H), 4.48 (m, 2H), 3.83 (m, 2H), 3.62 (m, 4H), 3.42 (m, 2H), 2.88 (m, 6H), 2.78 (m, 2H), 1.69 (m, 12H), 1.50 (m, 8H), and 1.23 (m, 6H). ^{13}C NMR (CDCl_3): δ 172.2, 146.437, 134.7, 131.0, 129.9, 128.6, 123.6, 122.9, 122.5, 172.3, 147.0, 134.3, 131.0, 129.9, 128.8, 128.6, 123.7, 122.9, 62.9, 55.9, 49.7, 48.0, 46.5, 44.4, 35.5, 28.4, 27.2, 26.7, and 22.2. IR (KBr): ν_{max} 2963, 2876, 1640, 1603, 1514, 1484, 1451, 1414, 1381, and 1346 cm^{-1} . ESI-MS ($\text{M}-\text{H}^+$): 735.5 g/mol.

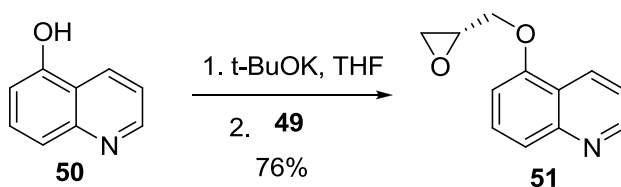
Protected glycidol **49**.²²



Triethyl amine (0.4 mL) was added to the mixture of (-)-D-glycidol (**48**, 200 mg, 2.7 mmol) and dry methylene chloride (20 mL) at 0 °C under nitrogen. 4-nitrobenzenesulfonyl chloride (700 mg, 3 mmol) in methylene chloride (10 ml) was added dropwise to the mixture at 0 °C. The reaction was stirred for 2 hours and quenched by dilute HCl/H₂O. The organic phase was separated; the water phase was extracted three times by ether. The organic phases were combined and dried over Na₂SO₄. The organic solution was concentrated in vacuo. The residue was separated by a silica gel column with ethyl acetate and hexanes (1:10) to give the protected (-)-D-glycidol **49** (607 mg, 1.9 mmol, 71%).

¹HNMR (CDCl₃): δ 8.39 (d, 2H, ³J = 8.5 Hz), 8.11 (d, 2H, ³J = 8.5 Hz), 4.46 (dd, 1H, ²J = 11 Hz, ³J = 2.5 Hz), 4.01 (dd, 1H, ²J = 12 Hz, ³J = 6.0 Hz), 3.20 (d, 1H, ³J = 2.5 Hz), 2.83 (dd, 1H, ³J = 4.0 Hz, ³J = 4.0 Hz), and 2.60 (dd, 1H, ³J = 2.0 Hz, ³J = 2.0 Hz). ¹³CNMR (CDCl₃): δ 151.1, 141.8, 129.5, 124.7, 71.9, 48.9, and 44.7. IR (KBr): ν_{max} 3516, 3108, 2921, 1607, 1532, 1404, 1351, 1186, and 1095 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.²²

5-Quinolinol **51**.²²

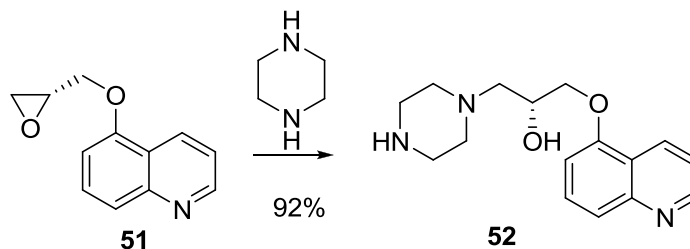


Potassium t-butoxide (1 M in ether, 0.78 mL, 0.78 mmol) was added dropwise to the mixture of 5-quinolinol (**50**, 100 mg, 0.69 mmol) and THF (10 mL) at 0 °C. The reaction was stirred for an hour at r.t. to give a yellow suspension. Then the protected (-)-D-glycidol (**31**, 170 mg, 1.42 mmol) in THF (5 mL) was added dropwise to the suspension. The reaction was stirred for two hours at r.t. and quenched by sat. ammonium

chloride solution (1 mL). The organic phase was separated; the water phase was extracted three times by ether. The organic phases were combined and dried over Na₂SO₄. The organic solution was concentrated in vacuo. The residue was separated by a silica gel column with methanol and methylene chloride (1:150) to give the ether **51** (99 mg, 0.53 mmol, 76%).

¹HNMR (CDCl₃): δ 8.84 (dd, 1H, ³J = 4.8 Hz, ⁴J = 2.4 Hz), 8.54 (dd, 1H, ³J = 7.2 Hz, ⁴J = 0.6 Hz), 7.65 (d, 1H, ³J = 8.4 Hz), 7.51 (dd, 1H, ³J = 7.8 Hz, ³J = 7.8 Hz), 7.31 (dd, 1H, ³J = 8.4 Hz, ³J = 4.2 Hz), 4.35 (d, 1H, ³J = 7.8 Hz), 4.01 (dd, 1H, ²J = 12 Hz, ³J = 6.0 Hz), 3.40 (dd, 1H, ²J = 10.8 Hz, ³J = 6.0 Hz), 2.89 (dd, 1H, ³J = 9.6 Hz, ³J = 4.8 Hz), and 2.76 (dd, 1H, ³J = 10.8 Hz, ³J = 2.4 Hz). ¹³CNMR (CDCl₃): δ 153.6, 150.5, 148.8, 130.6, 129.0, 121.9, 120.5, 120.1, 105.1, 69.0, 49.8, and 44.3. The spectroscopic and other physical data reported are consistent with values reported in the literature.²²

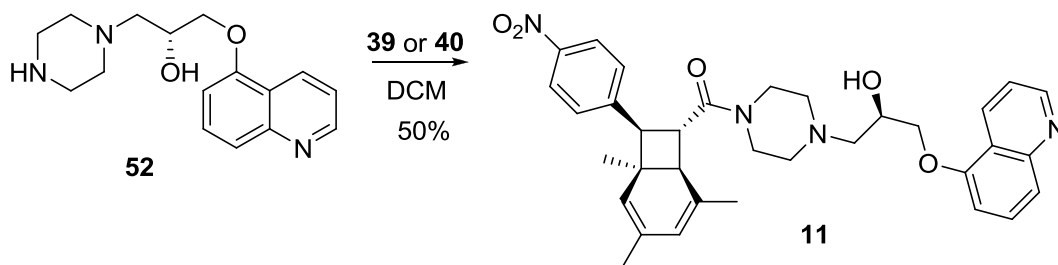
Piperazine **52**.³⁶



The quinolinol (**51**, 188 mg, 0.94 mmol) was mixed with piperazine (3.7 g, 43 mmol) and ethanol (20 ml). The mixture was refluxed at 85 °C for 24 hours. Then potassium carbonate (200 mg) was added to the mixture. The reaction was refluxed at 85 °C for 2 more days. The solid was filtered off. The filtrate was concentrated in vacuo. The residues were separated by a silica gel column with methanol and methylene chloride (1:5) to give yellow oil **52** (264 mg, 0.86 mmol, 92%).

^1H NMR (CDCl_3): δ 8.80 (d, 1H, $^3J = 4.0$ Hz), 8.48 (d, 1H, $^3J = 9.5$ Hz), 7.62 (d, 1H, $^3J = 9.5$ Hz), 7.50 (dd, 1H, $^3J = 8.0$ Hz, $^3J = 8.0$ Hz), 7.27 (dd, 1H, $^3J = 8.0$ Hz, $^3J = 4.0$ Hz), 4.18 (t, 1H, $^3J = 4.0$ Hz, $^3J = 4.0$ Hz), 4.08 (t, 2H, $^3J = 4.0$ Hz, $^3J = 4.0$ Hz), 3.74 (s, 2H), 2.87 (m, 4H), 2.61 (d, 2H, $^3J = 5.0$ Hz), 2.55 (d, 2H, $^3J = 7.0$ Hz), and 2.44 (d, 2H, $^3J = 5.5$ Hz). ^{13}C NMR (CDCl_3): δ 153.7, 149.9, 148.2, 130.5, 129.0, 120.8, 120.3, 119.7, 104.9, 70.7, 65.4, 61.1, 53.8, 49.2, 45.4, 45.2, and 44.9. IR (KBr): ν_{max} 3383, 3298, 1268, 1170, 1097, 1048, and 887 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.³⁶

SNF analog 11

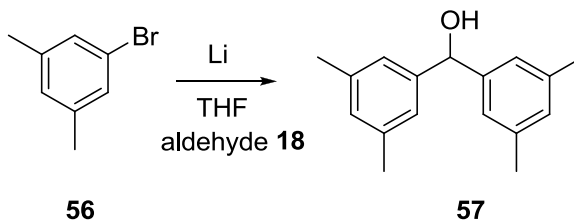


Method one: The acid (**38**, 40 mg, 0.13 mmol) was dissolved in dry methylene chloride. Thionyl chloride (0.2 mL) was added dropwise to the mixture at 0 °C. The reaction was stirred for an hour and then concentrated in vacuo. The residues were placed in high vacuo for 2 hours to remove extra thionyl chloride. 4-Dimethylaminopyridine (20 mg, 0.3 mmol) and the piperazine (**52**, 40mg, 0.14 mmol) were dissolved in methylene chloride (2 mL). The residues were added dropwise to the mixture by a pipet at 0 °C. The reaction was stirred for 3 hours at r.t. and quenched by water. The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na_2SO_4 . The organic solution was concentrated in vacuo. The residue was separated by a TLC plate with methanol and methylene chloride (1:10) to give yellow oil **11** (40 mg, 0.065 mmol, 50%)

Method two: the acid (**38**, 50 mg, 0.16 mmol) and carbonyl diimidazole (30 mg, 0.19 mmol) were dissolved in dry methylene chloride (2 mL). The mixture was stirred at r.t. for one hour. A little gas was release. The piperazine (**52**, 40 mg, 0.014 mmol) in dry methylene chloride (1 mL) was added to the mixture. The reaction was stirred at r.t. for 18 hours. The solution was diluted with 10 mL of methylene chloride. The mixture was washed by water. The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined, dried over Na₂SO₄, and concentrated in vacuo. The residue was separated by a TLC plate with methanol and methylene chloride (1:10) to give yellow oil **11** (56 mg, 0.096 mmol, 66%)

¹HNMR (CDCl₃): δ 8.80 (s, 1H), 8.51 (d, 1H, ³J = 8.5 Hz), 8.13 (d, 2H, ³J = 7.5 Hz), 7.69 (d, 1H, ³J = 7.0 Hz), 7.56 (t, 1H, ³J = 7.5 Hz), 7.32 (d, 2H, ³J = 8.0 Hz), 6.83 (d, 1H, ³J = 7.5 Hz), 5.49 (s, 1H), 4.49 (s, 1H), 4.272 (m, 1H), 4.15 (s, 2H), 3.83 (d, 2H, ³J = 8.0 Hz), 3.64 (t, 2H, ³J = 9.0 Hz), 3.45 (s, 2H), 2.85 (d, 1H, ³J = 6.5 Hz), 2.65 (m, 4H), 1.74 (s, 3H), 1.67 (s, 3H), and 1.23 (s, 3H). ¹³CNMR (CDCl₃): δ 154.1, 147.0, 145.9, 134.2, 131.0, 130.9, 129.5, 128.6, 123.6, 122.9, 122.4, 122.0, 120.4, 105.5, 70.7, 65.8, 60.9, 60.8, 56.1, 56.2, 46.1, 46.0, 45.5, 44.3, 44.2, 42.1, 28.3, 22.2, and 22.0. IR (KBr): ν_{max} 3480, 3080, 2920, 2780, 1630, 1585, 1475, 1345, and 1265 cm⁻¹. ESI-MS (M-H⁺): 583.31, 584.42, and 585.46 g/mol.

Alcohol **57**.

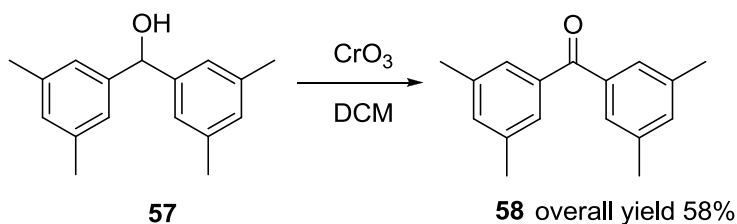


In a two necked RBF, 1-bromo-3, 5-xylene (**56**, 120 mg, 0.65 mmol) and mesitylaldehyde (**18**, 50 mg, 0.37 mmol) were added to lithium (200 mg) in THF (25

mL) at 0 °C.³⁷ A mixture of bromide (1.08 g, 9.7 mmol) and aldehyde (470 mg, 3.5 mmol) in THF (25 mL) were added dropwise to the reaction system during 2 hours. After the addition, the reaction was stirred for one hour and then quenched by sat. NH₄Cl solution (10 mL) at 0 °C. The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined, dried over Na₂SO₄, and concentrated in vacuo to give a crude product **57** (810 mg, 3.37 mmol, 87%).

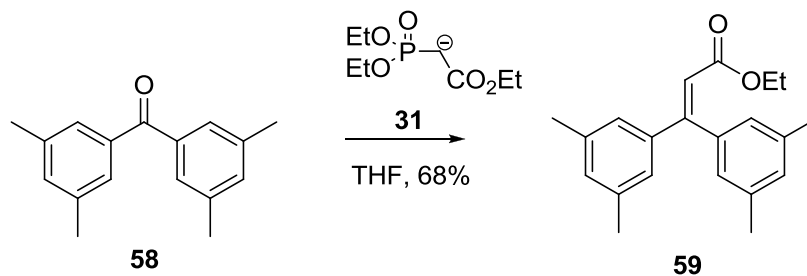
¹HNMR (CDCl₃): δ 7.000 (s, 4H), 6.902 (s, 2H), 5.700 (s, 1H), and 2.302 (s, 12H).
¹³CNMR (CDCl₃): δ 144.1, 138.2, 129.3, 124.4, 76.5, and 21.5. IR (KBr): ν_{max} 3380, 3005, 2910, 1601, 1460, 1151, 1052, and 852 cm⁻¹.

Ketone **58**.



The crude product **57** (810 mg, 3.37 mmol) was dissolved in acetic acid (25 mL). Chromium oxide (2.5 g, 16.4 mmol) was added to the mixture at 0 °C.³⁸ The reaction was stirred at r.t. for 2 hours and quenched by ice cold water. The water phase was extracted three times by ether. The organic phases were combined and concentrated. The residue was separated by a TLC plate with ethyl acetate and hexanes (1:15) to give a white solid **58** (523 mg, 1.95 mmol, 58%).

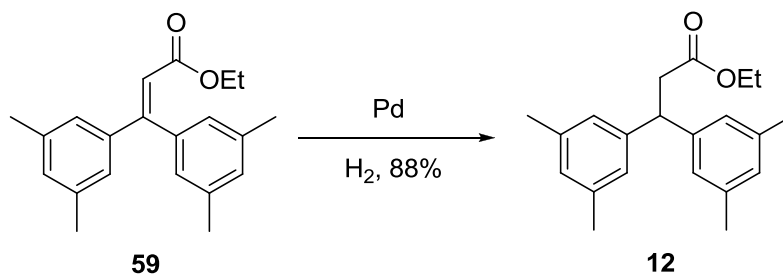
¹HNMR (CDCl₃): δ 7.393 (s, 4H), 7.217 (s, 2H), 2.378 (s, 12H). ¹³CNMR (CDCl₃): δ 183.4, 138.2, 138.0, 134.1, 127.9, and 21.4. IR (KBr): 2990, 2810, 1590, and 1205 cm⁻¹.

Ester 59.

Lithium t-butoxide (1 M in ether, 1.4 mL, 1.4 mmol) was added dropwise to a mixture of triethyl phosphoacetate (200 mg, 0.89 mmol) and benzene (10 mL) at r.t.³⁹ The mixture was stirred for half an hour at r.t.. The ketone (**58**, 104 mg, 0.44 mmol) in benzene (2 mL) was added dropwise to the reaction. The reaction was refluxed for 18 hours and quenched by sat. NH₄Cl solution (10 mL). The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na₂SO₄. The organic solution was concentrated in vacuo. The residue was separated by a silica gel column with ethyl acetate and hexanes (1:200) to give a white solid **59** (89 mg, 0.30 mmol, 68%).

¹HNMR (CDCl₃): δ 7.00 (s, 2H), 6.91 (s, 2H), 6.81 (s, 2H), 6.29 (s, 1H), 3.49 (q, 2H, ³J = 7.0 Hz), 2.31 (s, 6H), 2.28 (s, 6H), and 1.16 (t, 3H, ³J = 7.0 Hz). ¹³CNMR (CDCl₃): δ 166.5, 157.3, 141.4, 139.2, 138.0, 137.3, 131.2, 129.9, 127.0, 126.4, 117.3, 117.3, 60.1, 21.5, 21.5, and 14.2. IR (KBr): ν_{max} 2985, 2915, 1740, 1605, 1495, 1395, 1190, 1075, and 880. GC-MS (M⁺): 308, 293, 263, 262, 235, 221, 202, 178, 165, 133, 115, 105, 77, 63, and 46 g/mol.

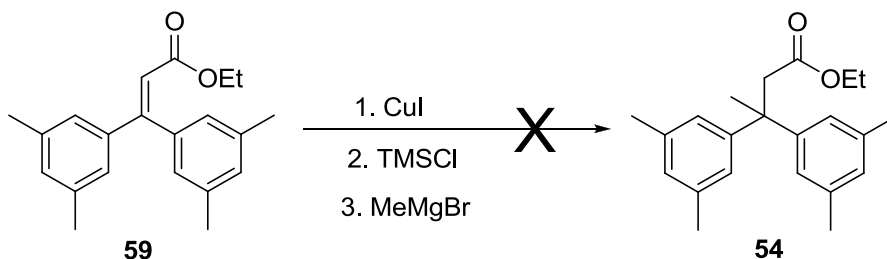
Ester 12.



The 25 mL RBF containing palladium (10% 10mg) was connected with vacuo to remove air.³⁶ And then hydrogen was injected. The operation was repeated five times. Hydrogen in a balloon was pumped to the RBF; and then ethanol (95%, 10 mL) was added. The ester (**59**, 48 mg, 0.16 mmol) in ethanol (95%, 2 mL) was added to the RBF. The reaction was stirred at r.t. overnight. The solid was filtered off and the solution was concentrated in vacuo to give a pure solid **12** (42 mg, 14 mmol, 88%).

¹HNMR (CDCl₃): δ 6.76 (s, 4H), 6.73 (s, 2H, ³J = 7.0 Hz), 4.31 (t, 1H, ³J = 7.0 Hz), 3.95 (t, 2H, ³J = 7.0 Hz), 2.91 (d, 2H, ³J = 8.5 Hz), 2.18 (s, 12H), and 1.03 (t, 3H, ³J = 7.0 Hz). ¹³CNMR (CDCl₃): δ 172.2, 143.8, 138.0, 128.3, 125.7, 60.5, 47.1, 41.1, 21.6, and 14.3. IR (KBr): ν_{max} 2995, 2940, 1730, 1590, 1455, 1380, 1260, and 1160 cm⁻¹. GC-MS (M⁺): 310, 236, 223, 193, 165, 131, 115, 91, 77, and 53 g/mol.

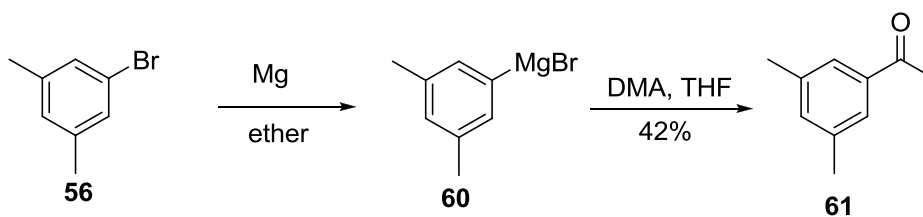
Attempt at ester 54 .



The ester (**59**, 41 mg, 0.13 mmol) was mixed with copper(I) iodide (10 mg, 0.05 mmol) and THF (10 mL). Chlorotrimethylsilane (21 mg, 0.2 mmol) was added to the suspension at r.t.; and then the suspension was stirred for half an hour. It was allowed to

cool down to $-15\text{ }^{\circ}\text{C}$. Methylmagnesium bromide (3M in THF, 0.070 mL, 0.21 mmol) was added dropwise to the suspension at $-15\text{ }^{\circ}\text{C}$. The reaction was stirred for 8 hours and quenched by sat. NH_4Cl solution (10 mL). The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na_2SO_4 . The organic solution was concentrated in vacuo. An NMR of the residues showed that no reaction occurred and the starting material was obtained.

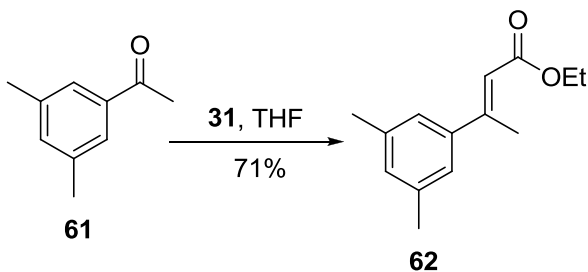
Ketone **61**.



1-Bromo-3, 5-xylene (2 mL, 14.7 mmol) in THF (15 mL) was added dropwise to the suspension of magnesium and THF (10 mL) during an hour to give a dark brown solution.⁴¹ The solution was transferred to the other RBF. Dimethylacetamide in THF (5 mL) was added dropwise to the dark solution at $0\text{ }^{\circ}\text{C}$. The solution was stirred at r.t. overnight and then quenched by sat. NH_4Cl solution (10 mL) at $0\text{ }^{\circ}\text{C}$. The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na_2SO_4 . The organic solution was concentrated in vacuo. The residue was separated by a silica gel column with ethyl acetate and hexanes (1:50) to give oil **61** (1.09 g, 6.0 mmol, 42%).

^1H NMR (CDCl_3): δ 7.57 (s, 2H), 7.20 (s, 1H), 2.57 (s, 3H), and 2.37 (s, 6H). The spectroscopic and other physical data reported are consistent with values reported in the literature.⁴¹

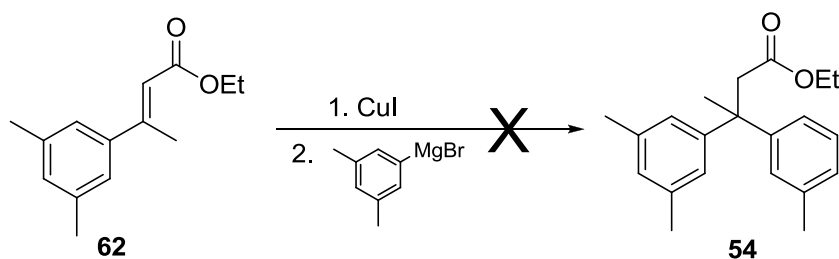
Ester 62.



Lithium t-butoxide (1 M in ether, 2.5 mL, 2.5 mmol) was added dropwise to the mixture of triethyl phosphoacetate (400 mg, 1.79 mmol) and THF (10 mL) at r.t.³⁷ The mixture was stirred at r.t. for half an hour. The ketone (**61**, 207 mg, 1.4 mmol) in THF (5 mL) was added dropwise to the reaction system. The reaction was refluxed overnight and quenched by sat. NH_4Cl solution (10 mL). The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na_2SO_4 . The organic solution was concentrated in vacuo. The residue was separated by a silica gel column with ethyl acetate and hexanes (1:40) to give a white solid **62** (197 mg, 1.0 mmol, 71%).

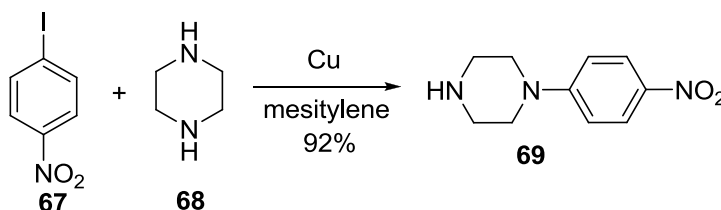
^1H NMR (CDCl_3): δ 7.11 (s, 2H), 7.02 (s, 1H), 6.13 (s, 1H), 4.23 (q, 2H, $^3J = 7.0$ Hz), 2.57 (s, 3H), 2.36 (s, 6H), and 1.34 (t, 3H, $^3J = 7.0$ Hz). ^{13}C NMR (CDCl_3): δ 167.2, 156.2, 142.6, 138.2, 134.9, 130.8, 126.4, 124.8, 124.4, 117.0, 60.0, 21.5, 18.2, and 14.6. IR (KBr): ν_{max} 2980, 2837, 1702, 1630, 1572, 1511, 1463, 1366, 1303, and 1252 cm^{-1} .

The second attempt at ester 54.



1-Bromo-3, 5-xylene (250 mg, 1.35 mmol) in THF (10 mL) was added dropwise to the suspension of magnesium (1.3 g, 54 mmol) and THF (25 mL) during an hour to give a dark brown solution.³⁶ The solution was transferred to the suspension of copper iodide (30 mg, 0.15 mmol), the ester (**62**, 250 mg, 1.15 mmol) and THF (10 mL) at 0 °C. The reaction was stirred at r.t. overnight and quenched by sat. NH₄Cl solution (10 mL). The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na₂SO₄. The organic solution was concentrated in vacuo. The crude NMR showed that no reaction occurred and the starting material was obtained.

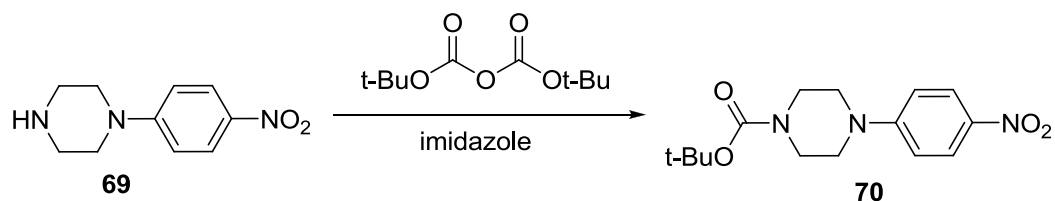
Arylpiperazine **69**.



6.5 g (76 mmol) of piperazine and 1.2 g (4.8 mmol) of 1-iodo-4-nitrobenzene were mixed in 30 mL of mesitylene. 3.7 g (56 mmol) of copper, 3 g (21 mmol) of potassium carbonate, and 1.4 g (5.3 mmol) of 18-crown-6 were added to the mixture. The reaction was refluxed at 185 °C for 24 hours. Then the reaction was cooled down to room temperature and was added celite. The solid was filtered off, the filtrate was concentrated. The residue was separated by a silica gel column with 1:1 hexanes and ethyl acetate to give 965 mg (4.4 mmol) of product (92%). The product was recrystallized from acetone to give 358 mg (1.6 mmol) of yellow solid **69**.

¹HNMR (CDCl₃): δ 8.00 (d, 2H, ³J = 9.6 Hz), 6.72 (d, 2H, ³J = 9.6 Hz), 3.30 (t, 4H, ³J = 5.2 Hz), and 2.93 (t, 4H, ³J = 5.2 Hz). The spectroscopic and other physical data reported are consistent with values reported in the literature.⁴⁵

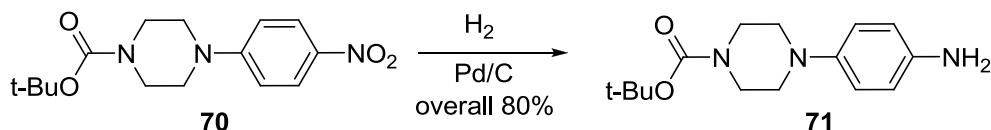
t-Boc protected arylpiperazine 70.⁴⁶



358 mg (1.73 mmol) of **69** was dissolved in THF. 600 mg (2.75 mmol) of di-tert-butyl dicarbonate and 600 mg of triethylamine were added to the solution at 0 °C. The reaction was refluxed overnight. The reaction mixture was washed by 1 M sodium hydroxide solution. The aqueous solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. The extraction was concentrated to give 596 mg (1.94 mmol) of crude product **70** which was used directly in the next step.

¹HNMR (CDCl₃): δ 8.15 (d, 2H, ³J = 9.6 Hz), 6.81 (d, 2H, ³J = 9.6 Hz), 3.60 (t, 4H, ³J = 5.2 Hz), 3.42 (t, 4H, ³J = 5.2 Hz), and 1.49 (s, 9H). ¹³CNMR (CDCl₃): δ 155.0, 138.4, 126.0, 112.7, 80.5, 70.7, 50.9, 48.1, and 47.1. The spectroscopic and other physical data reported are consistent with values reported in the literature.⁴⁶

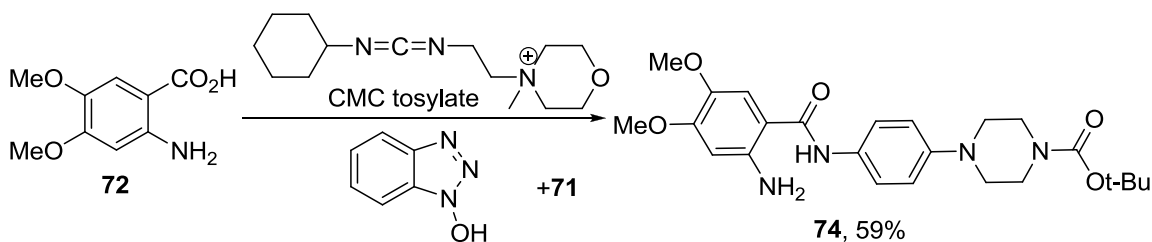
t-Boc protected arylpiperazine 71.⁴⁷



In a 50 mL RBF, 10% Pd/C was connected to vacuum for 5 minutes. And then hydrogen was flushed to the RBF. 595 mg (1.94 mmol) of compound **70** in 15 mL ethanol was added to the reaction under hydrogen. The reaction was stirred overnight under hydrogen. The solid was filtered off. The filtrate was concentrated to give 442 mg (1.59 mmol) of product **71** (80%).

^1H NMR (CDCl_3): δ 6.79 (d, 2H, $^3J = 6.4$ Hz), 6.63 (d, 2H, $^3J = 6.4$ Hz), 3.54 (t, 4H, $^3J = 4.8$ Hz), 2.94 (t, 4H, $^3J = 4.8$ Hz), and 1.46 (s, 9H). ^{13}C NMR (CDCl_3): δ 154.9, 144.5, 140.7, 119.3, 116.2, 79.9, 58.4, 51.5, and 28.5. IR (KBr): ν_{max} 3329, 2976, 2817, 1697, 1594, 1518, 1478, 1416, and 1317 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.⁴⁷

Amine 74.

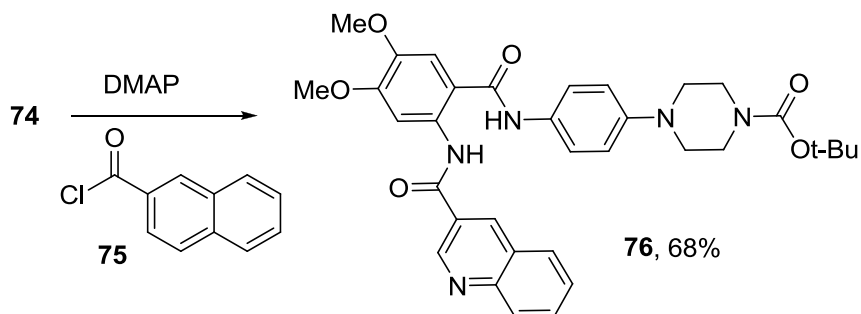


234 mg (1.18 mmol) of 2-amino-4,5-dimethoxybenzoic acid and 297 mg (1.07 mmol) of **71** were mixed in dry methylene chloride. 423 mg (1.2 mmol) of 4-methylmorpholine N-oxide monohydrate followed by 135 mg (1.0 mmol) of 1-hydroxybenzotriazole was added to the reaction. The black solution was stirred for 24 hours and then washed by 1 M sodium hydroxide solution. The water solution was extracted three times with methylene chloride. The organic solution was dried over magnesium sulfate and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 methylene chloride and methanol to give 386 mg of product **74** (0.61 mmol, 59%).

^1H NMR (CDCl_3): δ 7.91 (s, 1H), 7.41 (d, 2H, $^3J = 8.0$ Hz), 6.98 (s, 1H), 6.88 (d, 2H, $^3J = 8.0$ Hz), 6.19 (s, 1H), 3.83 (s, 3H), 3.79 (s, 3H), 3.55 (t, 4H, $^3J = 4.0$ Hz), 3.05 (t, 4H, $^3J = 4.0$ Hz), and 1.47 (s, 9H). ^{13}C NMR (CDCl_3): δ 167.3, 154.8, 153.6, 148.4, 145.0, 141.1, 131.1, 122.4, 117.3, 111.4, 107.8, 101.1, 80.0, 57.2, 55.8, 49.9, and 28.5. IR (KBr):

ν_{max} 3351, 2974, 1753, 1692, 1642, 1572, 1515, 1422, and 1286 cm^{-1} . ESI-MS ($\text{M}-\text{H}^+$): 457.4 g/mol.

Amide 76

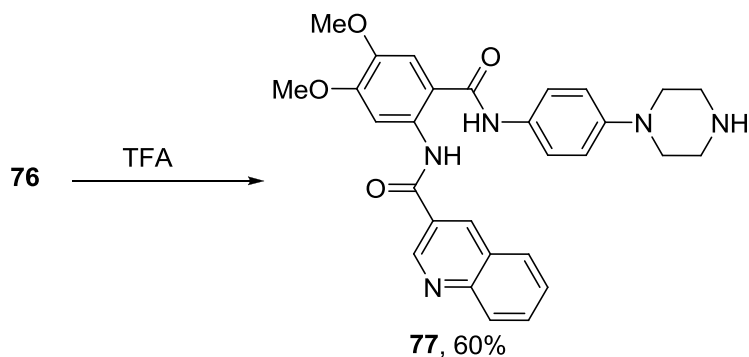


60 mg (0.32 mmol) of 3-quinolinecarboxylic acid in 10 mL of methylene chloride was added 1 mL of thionyl chloride at room temperature. The reaction was refluxed overnight. The solvent and extra thionyl chloride were removed in vacuo. 43 mg (0.35 mmol) of 4-dimethylaminopyridine in 10 mL of methylene chloride was added to the reaction. Then 100 mg (0.158 mmol) of 74 in 1 mL of methylene chloride was added to the reaction. The reaction was stirred overnight and quenched by water. The water solution was extracted three times with methylene chloride. The organic solution was dried over magnesium sulfate and concentrated in vacuo. The residues were crystallized in 95% ethanol to give 97 mg of product **76** (0.137 mmol, 68%).

^1H NMR (CDCl_3): δ 12.51 (s, 1H), 9.51 (s, 1H), 8.73 (s, 1H), 8.54 (s, 1H), 8.30 (s, 1H), 8.15 (d, 1H, $^3J = 8.8$ Hz), 7.97 (d, 1H, $^3J = 8.0$ Hz), 7.81 (dd, 1H, $^3J = 8.8$ Hz, $^3J = 7.2$ Hz), 7.62 (dd, 1H, $^3J = 8.0$ Hz, $^3J = 7.2$ Hz), 7.53 (d, 2H, $^3J = 8.8$ Hz), 7.10 (s, 1H), 6.93 (d, 2H, $^3J = 8.8$ Hz), 5.29 (s, 1H), 3.93 (s, 3H), 3.77 (s, 3H), 3.58 (t, 4H, $^3J = 4.2$ Hz), 3.10 (t, 4H, $^3J = 4.2$ Hz), and 1.48 (s, 9H). ^{13}C NMR (CDCl_3): δ 167.5, 164.0, 154.9, 152.8, 149.5, 149.0, 149.0, 144.7, 135.9, 135.7, 131.6, 130.2, 129.5, 129.3, 127.6, 127.4, 127.0, 122.8, 117.2, 112.4, 109.8, 105.1, 80.2, 56.5, 56.3, 49.7, and 28.6. IR (KBr): ν_{max}

3365, 2975, 2916, 2849, 1749, 1694, 1622, 1558, 1475, 1418, 1366, and 1264 cm^{-1} . ESI-MS (M-H^+): 612.4, and 613.4 g/mol.

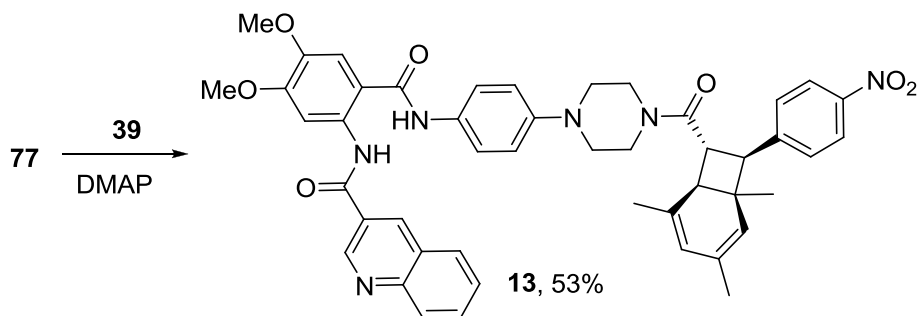
Piperazine 77.



90 mg (0.128 mmol) of **76** was dissolved in 10 mL of methylene chloride. 2 mL of trifluoroacetic acid was added to the solution at 0 °C. The reaction was stirred for 5 hours at room temperature and then quenched with 1 M sodium hydroxide solution. The water solution was extracted three times with methylene chloride. The organic solutions were combined, dried over magnesium sulfate, and concentrated in vacuo. The residues were crystallized in 95% ethanol to give 45 mg of yellow solid **77** (0.077 mmol, 60%).

^1H NMR (CDCl_3): δ 12.51 (s, 1H), 9.53 (s, 1H), 8.75 (s, 1H), 8.59 (s, 1H), 8.15 (d, 1H, $^3J = 8.0$ Hz), 7.98 (d, 1H, $^3J = 8.0$ Hz), 7.81 (dd, 1H, $^3J = 8.0$ Hz, $^3J = 7.2$ Hz), 7.63 (dd, 1H, $^3J = 8.0$ Hz, $^3J = 7.2$ Hz), 7.51 (d, 2H, $^3J = 8.8$ Hz), 7.08 (s, 1H), 6.95 (d, 2H, $^3J = 8.8$ Hz), 5.29 (s, 1H), 3.98 (s, 3H), 3.84 (s, 3H), 3.15 (t, 4H, $^3J = 4.2$ Hz), and 3.05 (t, 4H, $^3J = 4.2$ Hz). ^{13}C NMR (CDCl_3): δ 167.4, 164.1, 152.9, 149.7, 149.5, 149.0, 144.8, 135.9, 131.0, 129.5, 129.5, 129.4, 127.6, 127.5, 127.1, 122.7, 116.7, 112.4, 109.7, 105.2, 56.6, 56.3, 50.7, and 46.3. IR (KBr): ν_{max} 3455, 2975, 2827, 1678, 1589, 1514, 1414, 1366, 1319, 1251, and 1165 cm^{-1} . ESI-MS (M-H^+): 512.4, 513.4, and 514.4 g/mol.

SNF analogue 13.



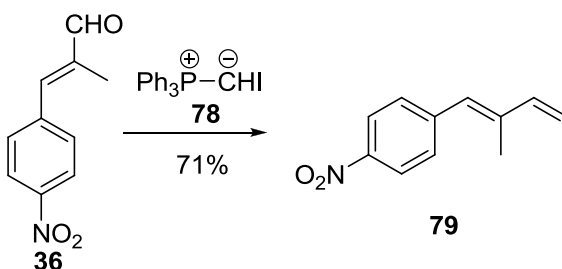
40 mg (0.132 mmol) of SNF acid in 5 mL of methylene chloride was added 0.5 mL of thionyl chloride. The mixture was stirred for half an hour. Then the solvent and extra thionyl chloride were removed in vacuo. 30 mg (0.059 mmol) of compound **77** and 10 mg (0.082 mmol) of 4-dimethylaminopyridine in 5 mL of methylene chloride was added to the reaction. The reaction was stirred for 24 hours and then quenched by 1 M sodium hydroxide solution. The water solution was extracted three times with methylene chloride. The organic solutions were combined, dried over magnesium sulfate, and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 methylene chloride and methanol to give 25 mg of product **13** (0.031 mmol, 53%).

$^1\text{H NMR}$ (CDCl_3): δ 12.52 (s, 1H), 9.52 (s, 1H), 8.75 (s, 1H), 8.58 (s, 1H), 8.17 (m, 3H), 7.97 (d, 1H, $^3J = 8.4$ Hz), 7.82 (dd, 1H, $^3J = 8.0$ Hz, $^3J = 6.8$ Hz), 7.63 (dd, 1H, $^3J = 8.0$ Hz, $^3J = 6.8$ Hz), 7.51 (d, 2H, $^3J = 8.4$ Hz), 7.33 (d, 2H, $^3J = 8.4$ Hz), 7.12 (s, 1H), 6.86 (d, 2H, $^3J = 8.8$ Hz), 5.51 (s, 1H), 4.51 (s, 1H), 3.96 (s, 3H), 3.82 (s, 3H), 3.85 (d, 1H, $^3J = 9.6$ Hz), 3.74 (dd, 1H, $^3J = 9.6$ Hz, $^3J = 8.8$ Hz), 3.55 (m, 4H), 3.08 (m, 4H), 2.87 (d, 1H, $^3J = 8.8$ Hz), 1.76 (s, 1H), 1.69 (s, 1H), and 1.24 (s, 1H). $^{13}\text{C NMR}$ (CDCl_3): δ 171.2, 167.5, 164.0, 153.0, 149.5, 148.9, 148.4, 147.0, 146.0, 144.8, 136.0, 134.3, 132.2, 131.6, 131.1, 130.5, 129.5, 129.3, 128.6, 127.6, 127.5, 127.1, 123.8, 123.0, 122.8, 122.4, 117.2, 112.2, 109.9, 105.2, 56.7, 56.3, 56.2, 50.4, 49.7, 46.1, 45.7, 44.3, 44.2, 28.4, 22.2, and 22.1. IR (KBr): ν_{max} 3305, 2970, 2857, 1667, 1589, 1519, 1412, 1346, and 1277 cm^{-1} . $^1\text{ESI-MS}$ (M-H^+): 807.3, and 808.3 g/mol.

Iodomethyltriphenylphosphonium iodide **78**.⁴⁸

5 g (18.6 mmol) of diiodomethane was added to a solution of 5 g (19 mmol) of triphenylphosphine in 50 mL of benzene at room temperature. The reaction was stirred overnight. The solid was filtered off and then was washed three times with ether to yield 9.21 g of compound **78** (17.1 mmol, 92%).

Iododiene **79**.

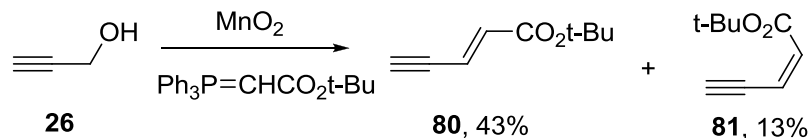


In a 100 mL round bottom flask, 5 g (11.6 mmol) of iodomethyltriphenylphosphonium iodide **78** was mixed with 50 mL of THF. The reaction mixture was cooled down to -20 °C. Then 12 mL of 1 M NaHMDS (12 mmol) was added dropwise to the reaction mixture. The yellow suspension became red. After 15 minutes stirring, 2.0 g (10.4 mmol) of (*E*)-2-methyl-3-(4-nitro-phenyl)-propenal **36** in 10 mL of THF was added dropwise to the reaction. The reaction gave a brown suspension. After 4 hours stirring at -20 °C, this reaction was quenched by a saturated NH₄Cl solution. The organic layer was separated and the water layer was extract three times with CH₂Cl₂. The organic layers were combined and dried over MgSO₄. After solvents were removed in vacuo, the organic products were separated by a silica gel column with 50:1 hexanes and AcOEt to give 2.03 g of yellow liquid **79** (7.4 mmol, 71%).

¹HNMR (CDCl₃): δ 8.21 (d, 1H, ³J = 9.0 Hz), 7.46 (d, 1H, ³J = 9.0 Hz), 6.92 (d, 1H, ³J = 8.5 Hz), 6.74 (s, 1H), 6.44 (d, 1H, ³J = 8.5 Hz), and 2.16 (s, 3H). ¹³CNMR

(CDCl₃): δ 146.2, 143.7, 141.8, 138.6, 130.3, 129.6, 123.4, 79.2, and 17.8. IR (KBr): ν_{max} 3419, 2929, 2855, 1706, 1652, 1594, 1558, 1456, 1393, 1256, and 1156 cm⁻¹.

Ester **80** and **81**.

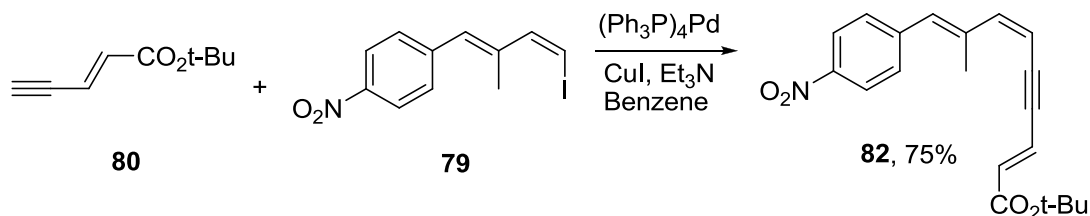


In a 250 mL round bottom flask, 3 g (54 mmol) of propargyl alcohol was dissolved in 170 mL of dry CH₂Cl₂. 45 g (517 mmol) of MnO₂ was added to the reaction solution rapidly. After 10 minutes stirring, 15 g (40 mmol) of (tert-butoxycarbonylmethylene)triphenylphosphorane was added to the reaction. The black reaction suspension was stirred for 3 hours. The solid was filtered off. After solvents were removed in vacuo, the organic products were separated by a silica gel column with 20:1 hexanes and AcOEt to give 2.64g of trans-product **80** (17.2 mmol, 43%) and 0.79 g of cis-product **81** (6.2 mmol, 13%).

Cis **80** ¹HNMR (CDCl₃): δ 6.09 (d, 1H, ³J = 12.0 Hz), 6.04 (d, 1H, ³J = 12.0 Hz), 3.55 (s, 1H), and 1.50 (s, 9H). ¹³CNMR (CDCl₃): δ 163.8, 132.9, 120.8, 88.7, 81.5, 79.9, and 28.2. IR (KBr): ν_{max} 3273, 2979, 2934, 1716, 1476, 1394, 1313, 1256, 1154, and 979 cm⁻¹

Trans **81** ¹HNMR (CDCl₃): δ 6.62 (d, 1H, ³J = 16.0 Hz), 6.23 (d, 1H, ³J = 16.0 Hz), 3.28 (s, 1H), and 1.47 (s, 9H). ¹³CNMR (CDCl₃): δ 164.8, 134.6, 123.0, 85.3, 81.4, 80.5, and 28.2. IR (KBr): ν_{max} 3273, 2979, 2934, 1716, 1476, 1394, 1313, 1256, 1154, and 979 cm⁻¹

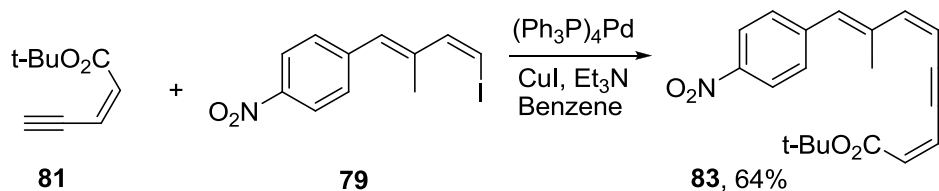
(E)-ester 82.



In a 25 mL round bottom flask, 300 mg (0.95 mmol) of iododiene **79** was added to a solution of 174 mg (1.14 mmol) of ester **80** in 10 mL of benzene at room temperature. Then 40 mg of tetrakis(triphenylphosphine)palladium and 10 mg of triethylamine were added individually under argon. After 14 mg of copper iodide was added to the reaction, the yellow solution became dark brown immediately. The reaction was stirred for 12 hours at room temperature. The solid was filtered off. The filtrate was concentrated in vacuo. The residues were separated by a silica gel column with 10:1 hexane/ethyl acetate to afford 274 mg of compound **82** (0.80 mmol, 75%).

^1H NMR (CDCl_3): δ 8.21 (d, 1H, $^3J = 8.8$ Hz), 7.45 (d, 1H, $^3J = 8.8$ Hz), 6.84 (dd, 1H, $^3J = 15.6$ Hz, $^4J = 2.8$ Hz), 6.74 (s, 1H), 6.47 (d, 1H, $^3J = 12.0$ Hz), 6.14 (d, 1H, $^3J = 15.6$ Hz), 5.763 (dd, 1H, $^3J = 12.0$ Hz, $^4J = 2.8$ Hz), 2.34 (s, 3H), and 1.49 (s, 9H). ^{13}C NMR (CDCl_3): δ 165.1, 147.6, 146.5, 144.8, 143.9, 139.9, 132.9, 132.0, 130.1, 123.7, 108.1, 96.4, 94.2, 81.3, 28.2, and 17.0. IR (KBr): ν_{max} 3401, 2976, 2856, 2175, 1712, 1593, 1519, 1392, and 1259 cm^{-1}

(Z)-ester 83.

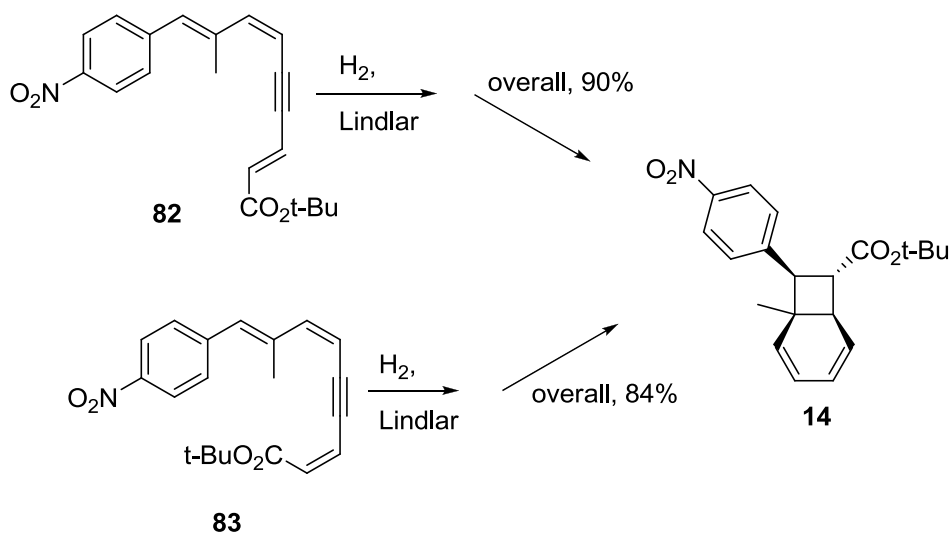


In a 25 mL round bottom flask, 150 mg (0.48 mmol) of iododiene **79** was added to a solution of 134 mg (0.88 mmol) of ester **81** in 10 mL of benzene at room temperature.

Then 30 mg of tetrakis(triphenylphosphine)palladium and 10 mg of triethylamine were added individually under argon. After 8 mg of copper iodide was added to the reaction, the yellow solution became dark brown immediately. The reaction was stirred for 12 hours at room temperature. The solid was filtered off. The filtrate was concentrated in vacuo. The residues were separated by a silica gel column with 10:1 hexane/ethyl acetate to afford 107 mg of compound **83** (0.31 mmol, 64%).

$^1\text{H NMR}$ (CDCl_3): δ 8.20 (d, 1H, $^3J = 7.2$ Hz), 7.45 (d, 1H, $^3J = 7.2$ Hz), 6.83 (s, 1H), 6.44 (d, 1H, $^3J = 12.0$ Hz), 6.23 (dd, 1H, $^3J = 11.2$ Hz, $^4J = 2.8$ Hz), 6.01 (d, 1H, $^3J = 12.0$ Hz), 5.86 (dd, 1H, $^3J = 11.2$ Hz, $^4J = 2.8$ Hz), 2.34 (s, 3H), and 1.49 (s, 9H). $^{13}\text{C NMR}$ (CDCl_3): δ 165.2, 147.6, 146.6, 144.8, 143.9, 139.9, 132.9, 132.0, 130.2, 124.0, 108.2, 96.4, 94.3, 81.3, 28.2, 163.8, 132.9, 120.8, 88.7, 81.5, 79.9, and 28.2. IR (KBr): ν_{max} 3405, 2977, 2932, 2179, 1707, 1594, 1519, 1455, 1368, 1258, and 1150 cm^{-1} .

SNF analogue 14.



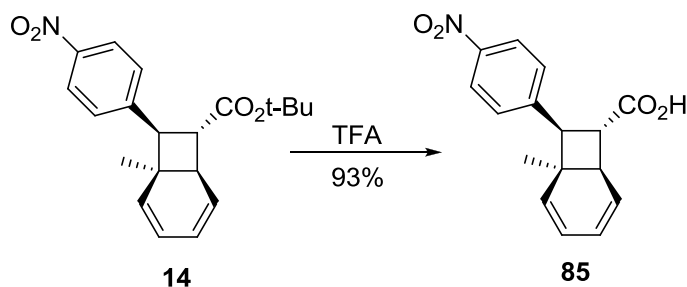
(a) From **82**: Under hydrogen, a solution of 340 mg (1 mmol) of ester **82** in 5 mL of ethyl acetate was added to 100 mg of Lindlar catalyst in a 25 mL round bottom flask. The reaction was stirred for half an hour under H_2 in the darkness at room temperature. Then the reaction was flushed with Argon. The reaction was stirred for a day at room

temperature. The solid was filtered off. The filtrate was concentrated in vacuo. The residues were separated by a TLC plate with 20:1 hexanes/ ethyl acetate to afford 305 mg of compound **14** (0.9 mmol, 90%)

(b) From **83**: Under hydrogen, a solution of 80 mg (24 mmol) of ester **83** in 5 mL of ethyl acetate was added to 34 mg of Lindlar catalyst in a 25 mL round bottom flask. The reaction was stirred for half an hour under H₂ in the darkness at room temperature. Then the reaction was flushed with Argon. The reaction was stirred for a day at room temperature. The solid was filtered off. The filtrate was concentrated in vacuo. The residues were separated by a TLC plate with 20:1 hexanes/ ethyl acetate to afford 67 mg of compound **14** (20 mmol, 84%)

¹HNMR (CDCl₃): δ 8.17 (d, 2H, ³J = 9.2 Hz), 7.37 (d, 2H, ³J = 9.2 Hz), 5.79 (m, 3H), 4.89 (d, 1H, ³J = 10.0 Hz), 3.76 (d, 2H, ³J = 10.5 Hz), 3.53 (dd, 2H, ³J = 10.5 Hz, ³J = 10.0 Hz), 2.91 (dd, 2H, ³J = 9.6 Hz, ³J = 6.0 Hz), 1.41 (s, 9H), and 1.30 (s, 3H).
¹³CNMR (CDCl₃): δ 172.7, 147.0, 145.7, 130.2, 128.3, 124.8, 123.7, 123.6, 123.5, 81.0, 57.8, 47.1, 42.8, 41.3, 28.2, and 28.1. IR (KBr): ν_{max} 2916, 1724, 1592, 1515, 1444, 1341, 1181, and 962 cm⁻¹.

Acid **85**.

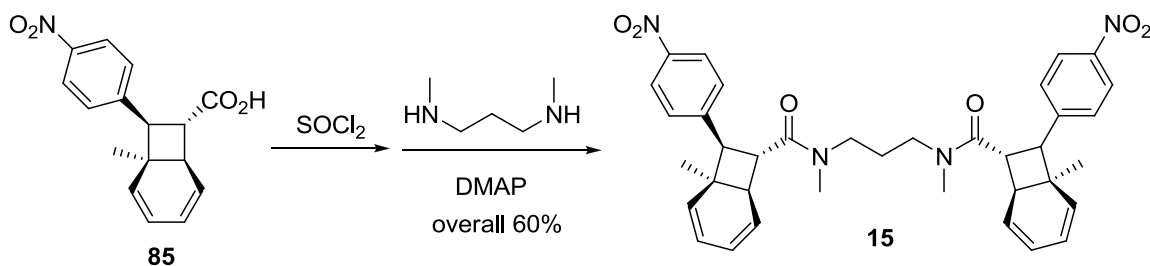


80 mg (24 mmol) of ester in 10 mL of THF was treated with 1 mL of TFA at 0 °C. The reaction was warmed up to room temperature and stirred overnight. The reaction was quenched with 10 mL of water. The solution was extracted three times with methylene

chloride. The organic layers were combined and dried over MgSO_4 , concentrated in vacuo to afford 62mg of crude acid **85** (22 mmol, 93%).

^1H NMR (CDCl_3): δ 8.18 (d, 2H, $^3J = 8.8$ Hz), 7.38 (d, 2H, $^3J = 8.8$ Hz), 5.84 (m, 2H), 5.75 (dd, 1H, $^3J = 8.8$ Hz, $^3J = 5.6$ Hz), 4.91 (d, 1H, $^3J = 9.6$ Hz), 3.85 (d, 2H, $^3J = 10.4$ Hz), 3.65 (dd, 2H, $^3J = 10.4$ Hz, $^3J = 9.2$ Hz), 3.00 (dd, 2H, $^3J = 9.2$ Hz, $^3J = 6.0$ Hz), and 1.31 (s, 3H). ^{13}C NMR (CDCl_3): δ 179.0, 147.1, 144.9, 130.1, 128.2, 124.1, 123.9, 123.8, 123.7, 57.4, 45.8, 43.2, 41.4, and 29.8. IR (KBr): ν_{max} 3401, 2922, 2851, 1702, 1598, 1518, 1455, and 1345 cm^{-1} .

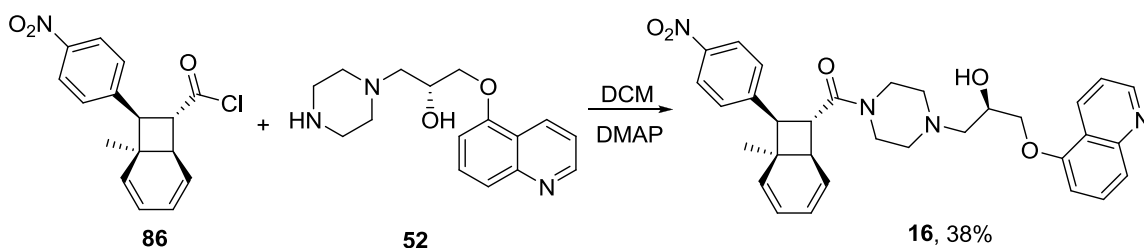
SNF analogue **15**.



At 0 °C, 0.5 mL of thionyl chloride was added to a solution of 62 mg (0.22 mmol) of acid **85** in 5 mL of methylene chloride. The reaction was stirred for half an hour. The solvent and extra thionyl chloride was removed in vacuo. The crude product was used directly. 11 mg (0.11 mmol) of $\text{N,N}'$ -dimethylpropane-1,3-diamine and 25 mg (0.2 mmol) of DMAP in 2 mL of methylene chloride were added to the acid chloride at 0 °C. The reaction was warmed up to room temperature and stirred overnight. The reaction was quenched with saturated NH_4Cl . The solution was extracted three times with methylene chloride. The organic layers were combined and dried over MgSO_4 . After solvents were removed in vacuo, the residues were separated by a TLC plate with 2:1 hexanes and ethyl acetate to afford 42 mg of compound **15** (0.13 mmol, 60%).

Mixture ^1H NMR (CDCl_3): δ 8.15 (m, 2H), 7.35 (m, 2H), 5.86 (m, 2H), 5.75 (m, 1H), 4.98 (d, 1H, $^3J = 10.6$ Hz), 4.04 (m, 1H), 3.83 (dd, 1H, $^3J = 10.0$ Hz, $^3J = 8.8$ Hz), 2.98 (m, 6H), 2.89 (m, 5H), 1.71 (m, 2H), and 1.29 (m, 6H). ^{13}C NMR (CDCl_3): δ 172.2, 147.0, 145.9, 131.4, 128.6, 124.6, 124.1, 123.7, 123.0, 56.3, 47.8, 45.7, 44.9, 42.6, 42.6, 36.0, and 27.8. IR (KBr): ν_{max} 3032, 2946, 1633, 1603, 1517, 1493, 1452, 1346, and 1109 cm^{-1} .

SNF analogue 16.

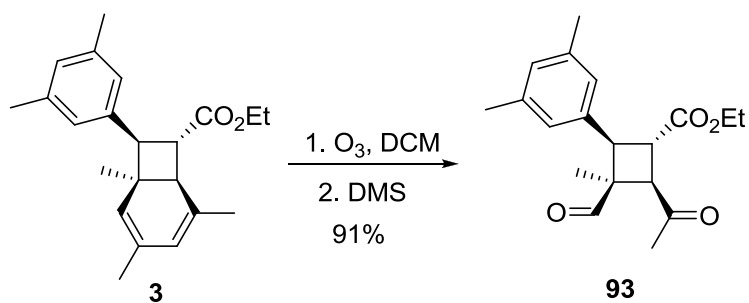


At $0\text{ }^\circ\text{C}$, 0.5 mL of thionyl chloride was added to a solution of 60 mg (0.21 mmol) of acid **85** in 5 mL of methylene chloride. The reaction was stirred for half an hour. The solvent and extra thionyl chloride was removed in vacuo. The crude product was used directly. 31 mg (0.11 mmol) of amine **52** and 12 mg (0.1 mmol) of DMAP in 2 mL of methylene chloride were added to the residues at $0\text{ }^\circ\text{C}$. The reaction was warmed up to room temperature and stirred overnight. The reaction was quenched with saturated NH_4Cl . The solution was extracted three times with methylene chloride. The organic layers were combined and dried over MgSO_4 . After solvents were removed in vacuo, the residues were separated by a TLC plate with 2:1 hexanes and ethyl acetate to afford 44 mg of compound **16** (0.079 mmol, 71%).

Mixture of diastereomers ^1H NMR (CDCl_3): δ 8.90 (d, 1H, $^3J = 3.2$ Hz), 8.55 (d, 1H, $^3J = 8.0$ Hz), 8.16 (d, 2H, $^3J = 8.8$ Hz), 7.71 (d, 1H, $^3J = 8.4$ Hz), 7.59 (dd, 1H, $^3J = 8.4$ Hz, $^3J = 7.6$ Hz), 7.36 (m, 3H), 6.87 (d, 1H, $^3J = 7.6$ Hz), 5.89 (m, 2H), 5.72 (m, 1H),

5.02 (dd, 1H, $^3J = 9.6$ Hz), 4.28 (m, 1H), 4.19 (m, 2H), 4.10 (d, 1H, $^3J = 10.0$ Hz), 3.78 (dd, 1H, $^3J = 10.0$ Hz, $^3J = 9.6$ Hz), 3.60 (m, 4H), 2.91 (dd, 1H, $^3J = 7.6$ Hz, $^3J = 7.2$ Hz), 2.68 (m, 4H), 2.50 (m, 2H), and 1.27 (d, 3H). ^{13}C NMR (CDCl_3): δ 170.7, 154.1, 150.9, 149.2, 147.0, 145.6, 131.5, 130.8, 129.5, 128.5, 124.5, 124.1, 123.7, 123.0, 122.3, 120.9, 120.4, 105.5, 70.8, 65.9, 60.9, 56.0, 46.0, 44.6, 42.8, 42.5, 42.1, and 27.9. IR (KBr): ν_{max} 3397, 3035, 2923, 2854, 1634, 1589, 1516, 1456, 1407, 1345, and 1266 cm^{-1} .

Cyclobutane **93**.

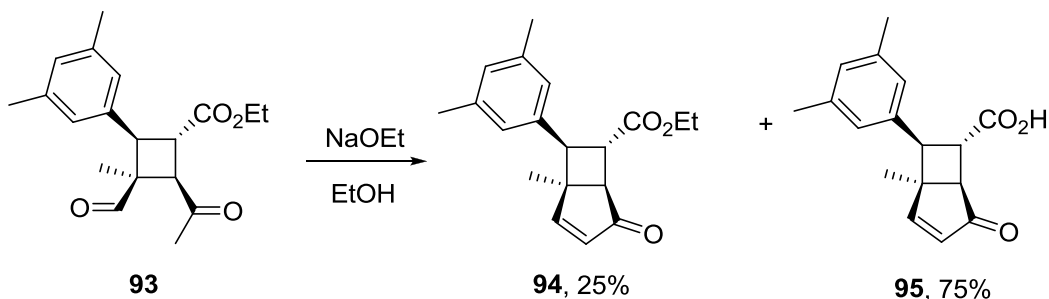


A solution of the ester (**3**, 9.1 mg, 0.028 mmol) in methylene chloride (15 mL) was cooled down to -78 °C under nitrogen. Oxygen was flushed through the mixture at -78 °C for 5 minutes. Then ozone was pumped to the mixture for 3 minutes to give a blue solution. After the ozone stream was continued for 3 more minutes, an oxygen stream was flushed through the mixture for 10 minutes. Nitrogen was passed through the mixture for half an hour to remove oxygen and ozone. Dimethyl sulfide was added to the mixture. Then the mixture was stirred overnight at room temperature. The mixture was concentrated in vacuo. The residues were separated by a TLC plate with ethyl acetate and hexanes (1:10) to give pure oil **93** (8.1 mg, 0.026 mmol, 91%).

^1H NMR (CDCl_3): δ 9.24 (s, 1H), 6.87 (s, 1H), 6.79 (s, 2H), 4.19 (q, 2H, $^3J = 7.5$ Hz), 3.98 (t, 1H, $^3J = 10.5$ Hz), 3.57 (d, 1H, $^3J = 10.5$ Hz), 3.27 (d, 1H, $^3J = 10.5$ Hz), 2.25 (s, 6H), 2.12 (s, 3H), 1.69 (s, 3H), and 1.28 (t, 3H, $^3J = 7.5$ Hz). ^{13}C NMR (CDCl_3): δ

209.5, 193.0, 168.3, 138.3, 135.2, 129.2, 125.0, 50.8, 50.4, 49.6, 40.9, 23.8, and 21.5. IR (KBr): ν_{max} 3410, 2975, 2960, 1730, 1605, and 1160 cm^{-1}

Bicyclo[3.2.0]hept-2-ene **94 and **95**.**



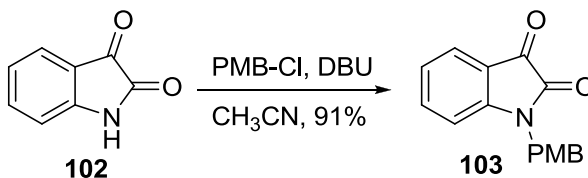
Sodium (200 mg) was added to dry ethanol (2 mL). The reaction was stirred for 3 hours to give a sodium ethoxide solution. The solution was added to a solution of cyclobutane (**93**, 8.1 mg, 0.026 mmol) in dry ethanol (2 mL). The reaction was stirred overnight at room temperature; and then it was quenched by HCl (1M, 10 mL). The water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na_2SO_4 . The organic solution was concentrated in vacuo. The residues were separated by a TLC plate with ethyl acetate and hexanes (1:10) to give the acid **95** (6.4 mg, 0.019 mmol, 75%) and the ester **94** (2.1 mg, 0.007 mmol, 25%).

94 ^1H NMR (CDCl_3): δ 7.12 (d, 1H, $^3J = 6.0$ Hz), 6.88 (s, 1H), 6.71 (s, 2H), 6.25 (d, 1H, $^3J = 6.0$ Hz), 4.18 (t, 2H, $^3J = 7.5$ Hz), 3.77 (d, 1H, $^3J = 9.0$ Hz), 3.10 (dd, 1H, $^3J = 9.0$ Hz, $^3J = 9.0$ Hz), 2.92 (d, 1H, $^3J = 5.5$ Hz), 2.29 (s, 6H), 1.56 (s, 3H), and 1.26 (t, 3H, $^3J = 7.5$ Hz). IR (KBr): ν_{max} 2950, 2905, 1730, 1709, 1460, and 1378 cm^{-1}

95 ^1H NMR (CDCl_3): δ 7.16 (d, 1H, $^3J = 5.0$ Hz), 6.89 (s, 1H), 6.71 (s, 2H), 6.27 (d, 1H, $^3J = 5.0$ Hz), 3.81 (s, 1H), 3.15 (s, 1H), 3.00 (s, 1H), 2.29 (s, 6H), and 1.50 (s, 3H). ^{13}C NMR (CDCl_3): δ 173.3, 168.0, 138.3, 135.2, 130.8, 129.0, 125.1, 124.9, 61.5, 50.5,

41.3, 29.9, 21.5, and 14.4. IR (KBr): ν_{max} 3455, 2960, 2935, 2806, 1730, 1709, 1460, and 1378 cm^{-1}

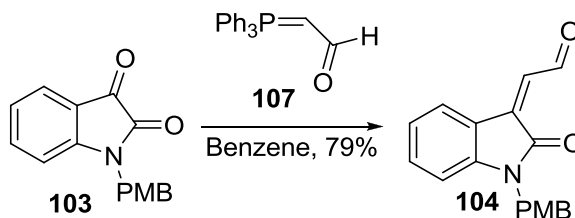
PMB protected isatin 103.



Isatin (**102**, 540 mg, 3.67 mmol) and p-methoxybenzyl chloride (0.54 mL, 4.2 mmol) in dry acetonitrile (20 mL) was added by 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU, 1.0 mL) dropwise at room temperature.⁶⁷ The solution was stirred for half an hour at 60 °C. Then the reaction was cooled down to 0 °C. HCl (1M, 10 mL) was added to quench the reaction. The organic phase was separated, and the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over Na₂SO₄. After concentration of organic solution, the residues were separated by a silica gel column with ethyl acetate and hexanes (1:10) to give protected isatin **103** (960 mg, 3.34 mmol, 91%).

¹HNMR (CDCl₃): δ 7.60 (d, 1H, ³J = 7.5 Hz), 7.48 (t, 2H, ³J = 7.5 Hz), 7.26 (d, 2H, ³J = 6.6 Hz), 7.09 (t, 1H, ³J = 7.5 Hz), 6.85 (t, 1H, ³J = 6.9 Hz), 4.82 (s, 2H), and 3.83 (s, 3H). ¹³CNMR (CDCl₃): δ 183.2, 159.3, 158.1, 150.6, 138.2, 128.8, 126.3, 125.2, 123.6, 117.5, 114.2, 110.8, 55.1, and 43.3. IR (KBr): ν_{max} 2955, 1737, 1691, 1612, 1513, 1467, 1246, and 1175 cm^{-1}

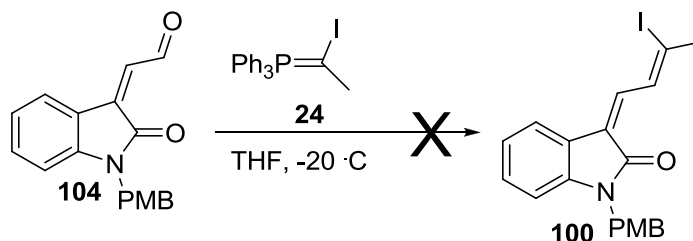
Aldehyde **104**.⁶⁸



A solution of PMB protected isatin (**103**, 130 mg, 0.5 mmol) in dry benzene (10 mL) was added by a solution of (formylmethylene)triphenylphosphorane (200 mg, 0.75 mmol) in dry benzene (5 mL). The solution was refluxed for 3 hours at 75 °C. The mixture was concentrated in vacuo. The residues were separated by a silica gel column with ethyl acetate and hexanes (1:4) to give the red aldehyde **104** (120 mg, 39 mmol, 79%).

^1H NMR (CDCl_3): δ 11.12 (d, 1H, $^3J = 7.8$ Hz), 7.46 (d, 1H, $^3J = 7.5$ Hz), 7.30 (m, 4H), 7.02 (t, 1H, $^3J = 7.5$ Hz), 6.85 (d, 2H, $^3J = 6.6$ Hz), 6.74 (d, 1H, $^3J = 8.1$ Hz), 6.71 (d, 1H, $^3J = 7.5$ Hz), 4.86 (s, 2H), and 3.78 (s, 3H). ^{13}C NMR (CDCl_3): δ 192.1, 165.9, 159.2, 144.5, 139.5, 132.9, 129.2, 128.7, 127.1, 122.8, 122.4, 121.2, 114.2, 109.7, 55.2, and 43.1. IR (KBr): ν_{max} 2958, 2867, 1791, 1718, 1664, 1615, 1470, 1341, 1215, and 1182 cm^{-1}

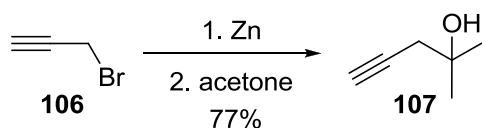
Attempt at iodo-diene **100**.



A suspension of (iodoethyl)triphenylphosphonium iodide (25 mg, 0.046 mmol) in THF (5 mL) was cooled down to -20 °C. Sodium hexamethyldisilazane (1 M, 0.05 mL) was added dropwise to the suspension by a syringe to give a red solution. After the solution was stirred for 10 minutes, a solution of the aldehyde (**104**, 13 mg, 0.044 mmol)

in THF (5 mL) was added dropwise to the mixture at -20 °C to give a green solution. After the reaction was stirred for 2 hours at -20 °C, it was quenched by sat. ammonium chloride solution (0.2 mL). The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over Na₂SO₄. After concentration of organic solution, the residues were separated by a silica gel column with ethyl acetate and hexanes (1:3). No desired product was given and the starting material was obtained.

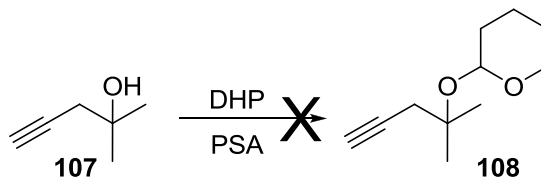
2-Methylpent-4-yn-2-ol **107**.⁶⁹



Acetone was distilled off from calcium chloride. Zinc was washed with HCl (0.2 g/L) and then ether. Propargyl bromide (**106**, 5 mL, 89 mmol) was added dropwise to zinc in THF at 0 °C. The reaction was stirred for 4 hours. Then acetone in THF was added the reaction at 0 °C. The reaction was stirred overnight and then the solid was filtered off. HCl (3M, 10 mL) was added to the filtrate. The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na₂SO₄. After concentration of organic solution, oil **107** (4.9 g, 69 mmol, 77%) was distilled out.

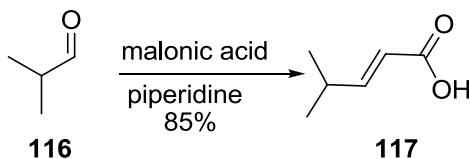
¹HNMR (CDCl₃): δ 2.37 (d, 2H, ³J = 2.7 Hz), 2.08 (t, 1H, ³J = 2.4 Hz), and 1.32 (s, 6H). IR (KBr): ν_{max} 3399, 2976, 2915, 2884, 1680, 1617, 1466, 1367, and 1299 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.⁶⁹

Attempt at THP protected alcohol 108.



p-Toluenesulfonic acid (61 mg, 0.35 mmol) was added to a mixture of the alcohol (**107**, 430 mg, 4.38 mmol) and dihydropyran (1.2 mL) in methylene chloride (25 mL). The mixture was stirred overnight to give a dark red solution. Then the reaction was quenched by sat. sodium bicarbonate solution (10 mL). The organic phase was separated; the water phase was extracted three times by ether. The organic phases were combined and dried over Na₂SO₄. After concentration of organic solution, the residues were separated by a silica gel column with ethyl acetate and hexanes (1:15) to give the starting material back.

(E)-4-Methylpent-2-enoic acid 117.⁷⁵

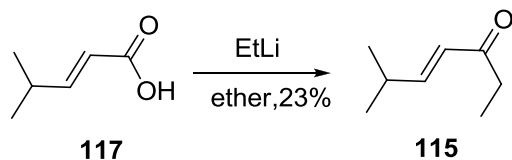


iso-Butanal (**116**, 5 mL, 64 mmol) was mixed with malonic acid (5.8 g, 58 mmol) and piperidine (20 mL). The mixture was stirred at 60 °C overnight and quenched by HCl (3 M, 100 mL). The water phase was extracted three times with methylene chloride. The organic phases were combined and dried over Na₂SO₄. After concentration of organic solution, the residues were separated by a silica gel column with acetone and hexanes (1:2) to give the liquid **117** (5.41 g, 54.4 mmol, 85%).

¹HNMR (CDCl₃): δ 7.02 (dd, 1H, ³J = 15.9 Hz, ³J = 6.6 Hz), 5.76 (d, 1H, ³J = 15.9 Hz), 2.43 (m, 1H), and 1.04 (d, 6H, ³J = 6.6 Hz). ¹³CNMR (CDCl₃): δ 171.9, 157.6, 118.5, 31.1, and 21.2. IR (KBr): ν_{max} 3414, 3092, 2966, 2874, 1701, 1656, 1467, 1387,

and 1214 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.⁷⁵

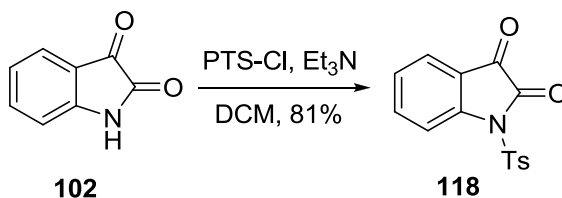
Ketone 115.⁷⁵



Under nitrogen, ethyl lithium (0.5 M, 20 mL, 10 mmol) was added dropwise to a mixture of the acid (**117**, 507 mg, 4.44 mmol) and dry ether (30 mL) at $-78\text{ }^{\circ}\text{C}$. The reaction was stirred for an hour at $-78\text{ }^{\circ}\text{C}$ and then for 3 hours at room temperature. The reaction was quenched by sat. NH_4Cl solution (5 mL). The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over Na_2SO_4 . After concentration of organic solution, the residue was separated by a silica gel column with ethyl acetate and hexanes (1:20) to give oil **115** (131 mg, 1.0 mmol, 23%).

^1H NMR (CDCl_3): δ 6.78 (dd, 1H, $^3J = 15.9\text{ Hz}$, $^3J = 6.6\text{ Hz}$), 6.03 (d, 1H, $^3J = 15.9\text{ Hz}$), 2.53 (dd, 2H, $^3J = 7.5\text{ Hz}$), 2.43 (m, 1H), and 1.05 (m, 9H). ^{13}C NMR (CDCl_3): δ 201.662, 153.244, 127.360, 33.365, 31.210, 21.459, and 8.286. IR (KBr): ν_{max} 2995, 2905, 2815, 1736, 1460, 1373, and 1188 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.⁷⁵

p-Toluenesulfonyl protected isatin 118.

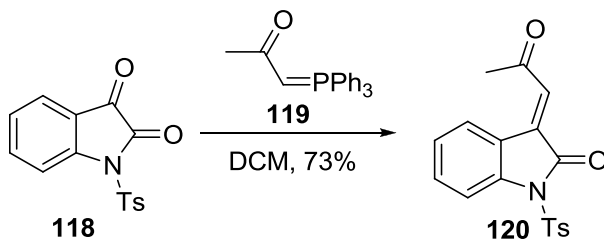


p-Toluenesulfonyl chloride was added to a mixture of isatin (**102**, 2 g, 13.6 mmol), triethylamine (10 mL) and methylene chloride (15 mL).⁷⁶ The mixture was stirred

overnight at room temperature to form a lot of yellow solid. Then the solid was filtered off. The yellow solid was washed with toluene once and ether twice. The solid was recrystallized in methylene chloride to give a yellow crystal **118** (3.9g, 11 mmol, 81%).

^1H NMR (CDCl_3): δ 8.08 (d, 1H, $^3J = 8.0$ Hz), 8.01 (d, 2H, $^3J = 8.5$ Hz), 7.72 (m, 2H), 7.36 (d, 2H, $^3J = 7.5$ Hz), 7.28 (t, 1H, $^3J = 7.5$ Hz), and 2.44 (s, 3H). ^{13}C NMR (CDCl_3): δ 179.1, 155.8, 147.7, 146.7, 139.5, 134.6, 130.3, 128.9, 128.1, 126.1, 119.0, 115.3, and 21.9. IR (KBr): ν_{max} 2978, 2738, 2604, 2530, 1774, 1605, 1475, 1397, 1232, and 1202 cm^{-1} .

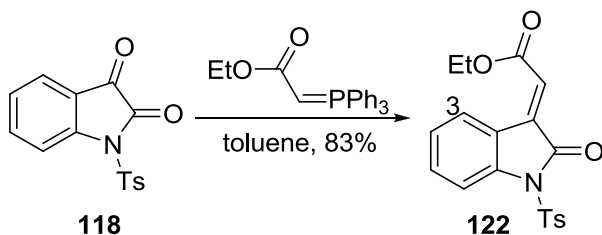
Ketone **120**.⁷⁷



A mixture of 1-(triphenylphosphoranylidene)-2-propanone (2.00 g, 6.3 mmol) and protected isatin (**118**, 3.06 g, 10 mmol) in methylene chloride (15 mL) was stirred overnight at room temperature. The solution was concentrated in vacuo. The residues were separated by a silica gel column with ethyl acetate and hexanes (3:1) to give an orange solid **120** (1.46 g, 4.6 mmol, 73%)

^1H NMR (CDCl_3): δ 8.56 (d, 1H, $^3J = 3.0$ Hz), 7.98 (dd, 3H, $^3J = 3.0$ Hz, $^3J = 8.5$ Hz), 7.48 (dt, 1H, $^3J = 8.5$ Hz, $^3J = 7.5$ Hz), 7.32 (d, 2H, $^3J = 7.5$ Hz), 7.20 (dt, 1H, $^3J = 8.5$ Hz, $^3J = 12.5$ Hz), 7.09 (s, 1H), 2.44 (s, 3H), and 2.42 (s, 3H). ^{13}C NMR (CDCl_3): δ 166.7, 146.1, 141.4, 135.3, 133.8, 133.1, 130.0, 129.2, 128.3, 128.1, 125.2, 120.7, 113.7, 32.4, and 21.9. IR (KBr): ν_{max} 3022, 2995, 1748, 1613, 1596, 1454, 1380, and 1086 cm^{-1}

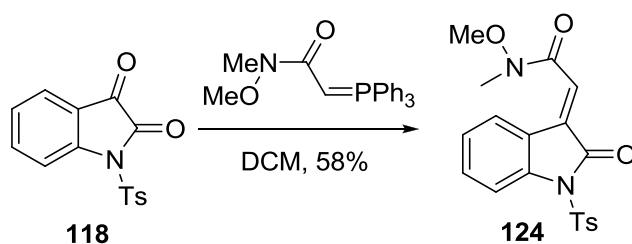
Ester **122**.



A mixture of (carbethoxymethylene)triphenylphosphorane (260 mg, 0.75 mmol) and protected isatin (**118**, 200 mg, 0.66 mmol) in toluene (20 mL) was stirred overnight at 70 °C. The mixture was concentrated in vacuo. The residues were separated by a silica gel column with ethyl acetate and hexanes (3:1) to give a lot of orange solid **122** (208 mg, 0.81 mmol, 83%)

^1H NMR (CDCl_3): δ 8.60 (d, 1H, $^3J = 7.8$ Hz), 7.32 (d, 1H, $^3J = 3.0$ Hz), 7.28 (d, 2H, $^3J = 8.1$ Hz), 7.04 (dd, 1H, $^3J = 7.5$ Hz), 6.90 (d, 2H, $^3J = 8.1$ Hz), 6.76 (d, 1H, $^3J = 8.1$ Hz), 4.39 (q, 2H, $^3J = 4.2$ Hz), 3.79 (s, 3H), and 1.43 (t, 3H, $^3J = 4.2$ Hz). ^{13}C NMR (CDCl_3): δ 167.8, 165.8, 145.3, 137.9, 132.5, 129.0, 127.7, 122.9, 120.2, 114.4, 109.3, 55.4, 43.5, and 14.4. IR (KBr): ν_{max} 2990, 2851, 1741, 1643, 1513, 1460, 1351, and 1246 cm^{-1} .

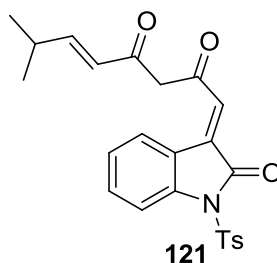
Weinreb amide **124**.⁷⁴



A mixture of N-methoxy-N-methyl(triphenylphosphoranylidene)acetamide (1.006 g, 2.77 mmol) and protected isatin (**118**, 1.54 g, 5 mmol) in dry methylene chloride (15 mL) was stirred overnight at room temperature. The mixture was concentrated and the residues were separated by a silica gel column with ethyl acetate and hexanes (1:2) to give a lot of white solid **124** (580 mg, 1.6 mmol, 58%).

^1H NMR (CDCl_3): δ 8.41 (d, 1H, $^3J = 7.6$ Hz), 7.99 (m, 3H), 7.43 (t, 1H, $^3J = 8.0$ Hz), 7.32 (m, 3H), 7.17 (t, 2H, $^3J = 7.6$ Hz), 3.72 (s, 3H), 3.33 (s, 3H), 2.42 (s, 3H), and 1.56 (s, 3H). ^{13}C NMR (CDCl_3): δ 166.1, 165.0, 145.9, 140.5, 135.1, 133.2, 132.4, 129.9, 127.9, 124.9, 124.3, 120.3, 113.4, 62.3, 32.3, and 21.7. IR (KBr): ν_{max} 2933, 2919, 1711, 1607, 1513, 1466, 1378, 1247, 1178, and 1031 cm^{-1}

Diketone **121**.⁷¹



1st try: The acid (**117**, 268 mg, 2.35 mmol) and carbonyl diimidazole (380 mg, 2.5 mmol) were mixed in dry THF (5 mL). The mixture was stirred for one hour at room temperature. A little gas was released. The ketone (**120**, 653 mg, 2.4 mmol) was added dropwise to a suspension of sodium hydride (120 mg) in dry THF (5 mL) in 15 minutes at 0 °C. The suspension was stirred for 15 minutes at room temperature and then was cooled down to 0 °C. The suspension was added dropwise by the solution of imidazole made in the first step. The reaction was stirred for 2 hours and quenched by water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over Na_2SO_4 . After concentration of organic solution, the crude product's NMR showed that only the starting materials were given.

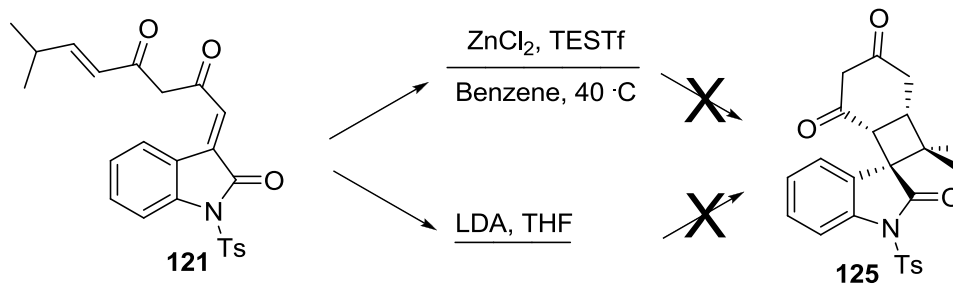
2nd try: A mixture of the ester (**122**, 0.81g, 2.2 mmol) and 1 M KOH in ethanol was stirred for an hour and quenched by 1M HCl. The water phase was extracted three times with methylene chloride. The organic phases were combined and dried over Na_2SO_4 . After concentration of organic solution, the residue was separated by a silica gel column with ethyl acetate and hexanes (1:1) to give an acid (157 mg, 21%). The acid (21 mg, 0.062 mmol) and carbonyl diimidazole (26 mg, 0.1 mmol) were dissolved in dry THF (5 mL). The mixture was stirred for one hour at room temperature. A little gas was released. 5-Methyl-3-hexen-2-one (**123**, 37 mg, 0.37 mmol) was added dropwise to a

mixture of lithium diisopropylamide (2 M, 0.2 mL) and hexamethylphosphoramide (0.2 mL) in THF (5 mL) at -78 °C. The reaction was stirred for half an hour at -78 °C to give a light red solution. The carbonyl imidazole solution made in the first step was added dropwise to the reaction in 15 minutes to give a dark yellow mixture. The mixture was stirred for 2 hours at -78 °C and quenched by sat. NH₄Cl solution (5 mL). The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na₂SO₄. After concentration of organic solution, the crude product's NMR showed that only the starting materials were given.

3rd try: Lithium diisopropylamide (2 M, 0.7 mL, 1.4 mmol) was added to a mixture of hexamethylphosphoramide (0.2 mL) and THF (5 mL) at -78 °C. 5-Methyl-3-hexen-2-one (75%, 88 mg, 0.88 mmol) in THF (2 mL) was added dropwise to the mixture in 15 minutes at -78 °C. The reaction was stirred for half an hour at -78 °C to give a light red solution. Winreb amide (**124**, 123 mg, 0.32 mmol) in THF (5 mL) was added dropwise to the reaction during 15 minutes to give a dark yellow mixture. The mixture was stirred for 6 hours at -78 °C and quenched by sat. NH₄Cl solution (5 mL). The organic phase was separated; the water phase was extracted three times by methylene chloride. The organic phases were combined and dried over Na₂SO₄. After concentration of organic solution, the residue was separated by a silica gel column with ethyl acetate and hexanes (1:6) to give yellow oil **121** (57 mg, 0.13 mmol, 41%).

¹HNMR (CDCl₃): δ 7.99 (m, 2H), 7.92 (d, 1H, ³J = 8.0 Hz), 7.32 (m, 3H), 7.13 (t, 1H, ³J = 7.5 Hz), 6.68 (d, 2H, ³J = 16 Hz), 5.87 (d, 1H, ³J = 16.5 Hz), 3.80 (s, 2H), 3.14 (s, 2H), 2.79 (m, 1H), 2.55 (m, 1H), 2.40 (s, 1H), and 1.02 (d, 6H, ³J = 7.5 Hz). ¹³CNMR (CDCl₃): δ 197.7, 174.6, 154.1, 145.6, 139.7, 129.9, 129.7, 129.1, 128.0, 126.7, 125.4, 124.7, 124.5, 113.6, 113.1, 61.6, 45.2, 38.7, 37.1, 31.1, 21.7, and 31.2. IR (KBr): ν_{max} 2975, 1770, 1660, 1460, 1370, 1230, 1180, and 1085 cm⁻¹.

3,5-Dioxobicyclo[4.2.0]octane **125**.



1st try: Zinc chloride powder (2 mg, 0.015 mmol) and triethyl amine (1 mL) were stirred together for an hour at room temperature to give a suspension. A solution of the diketone (**121**, 21 mg, 0.048 mmol) in benzene (1 mL) was added to the suspension. Then triethylsilyl triflate was added to this suspension at room temperature. The reaction was allowed to warm up to 40 °C in half an hour. The reaction was stirred for two days. The mixture was concentrated in vacuo. The residues were separated by a TLC plate with ethyl acetate and hexanes to give 3 fractions, but all of them were not the desired product.

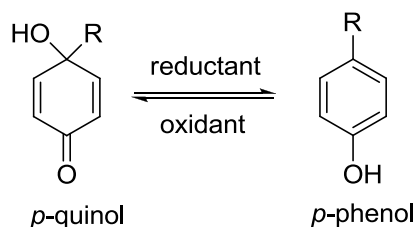
2nd try: A solution of the diketone (**121**, 36.8 mg, 0.084 mmol) in dry THF (5 mL) was cooled down to -78 °C. LDA (2 M, 0.7 mL, 1.4 mmol) was added dropwise to the mixture. Then the reaction was allowed to warm up to room temperature and stirred in the darkness for a day. The reaction was quenched by sat. NH_4Cl solution (5 mL). The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution, the residue was separated by a TLC plate with ethyl acetate and hexanes (1:2) to give 3 fractions, but all of them were not the product wanted and the starting material was recovered.

Chapter 2. 1,2-Addition of Organometallic Reagents to Unprotected Juglones and Progress Toward the Synthesis of Ravidomycins

2.1 Introduction

2.1.1 Addition to Juglone Derivatives

p-Quinols can undergo mild reductions to form *p*-phenols. Oxidations of *p*-phenols can reverse this reduction back to give the corresponding *p*-quinol (Scheme 39). Some biochemical compounds in nature such as coenzyme Q contain *p*-quinol substructures. *p*-Quinols have a variety of uses, principally associated with their actions as reduction or oxidation agents which are soluble in water. Thus *p*-quinols are very important synthetic precursors, pharmaceuticals, metabolic and biosynthetic intermediates, and chemical photoresists.⁷⁹ Quinols have been the subject of investigations, both synthetic and mechanistic, for many years.



Scheme 39. Redox reaction of the system of *p*-quinol and *p*-phenol.

The oxidation of *p*-phenols has been used in the synthesis of *p*-quinols. This transformation can be accomplished by a variety of reagents including peracetic acid, thallium(III) salts, and hypervalent iodine.⁸⁰ Most of these procedures suffer from limitations, such as poor yields and various side reactions. The other direct access to quinols is 1,2-addition of an organometallic reagent to a quinone. Alkylolithium and Grignard reagents have been used most frequently for this purpose and organocadmium reagents have been used, recently.⁷⁹

Although 1,2 addition to quinones by organometallic reagents has potential for the synthesis of quinols, its utility is complicated by the formation of double 1,2-addition products, Michael addition products, and/or electron-transfer products (hydroquinones).⁸¹ Electronic or steric differences between the two carbonyl groups in an unsymmetrical quinone may result in the formation of a single *p*-quinol; however, 1,2-addition to these systems usually produces mixtures of regioisomers.⁸² Approaches which employ protected quinones or masked quinones such as quinone monoketals and silylcyanohydrins have alleviated the above problems of direct quinone alkylation, but multistep procedures generally give low overall yields from commercially available materials.^{79,83}

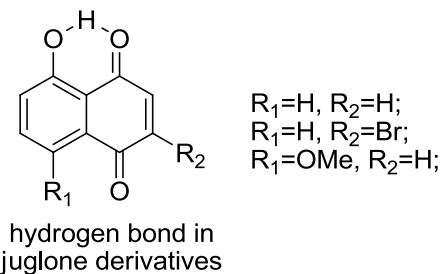
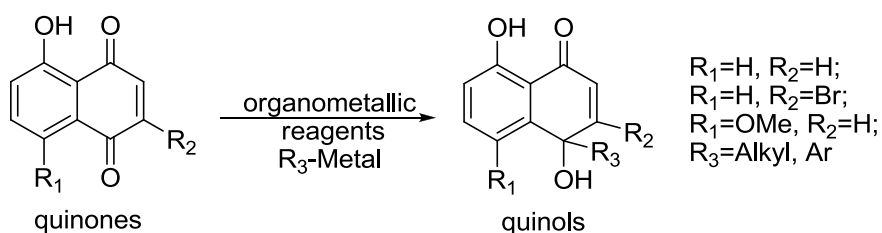


Figure 12. Juglone and derivatives.

Juglone and bromojuglone contain strong intramolecular hydrogen bonds (Figure 11). The intramolecular hydrogen bonds will be broken in Grignard or alkylolithium

reaction systems, but this process may be slower than addition of the organometallics to the non-hydrogen bonding carbonyl groups. Therefore, treatment of juglone or bromojuglone with one equivalent of Grignard reagent or alkyllithium could lead to a regioselective mono 1,2-adduct at the C-1 position (Scheme 40). 1,2-Addition to a 5-alkoxy naphthoquinone could also give a regioselective monoadduct due to a strong intramolecular hydrogen bond. The regioselective monoadducts of juglone derivatives are seen as key intermediates for the preparation of complex aryl C-glycosides such as gilvocarcins, pluramycins, and griseusins.⁸⁴



Scheme 40. Additions of organometallics to juglone derivatives.

2.1.1 Ravidomycin

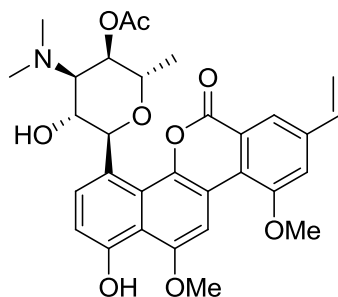


Figure 13. Ravidomycin.

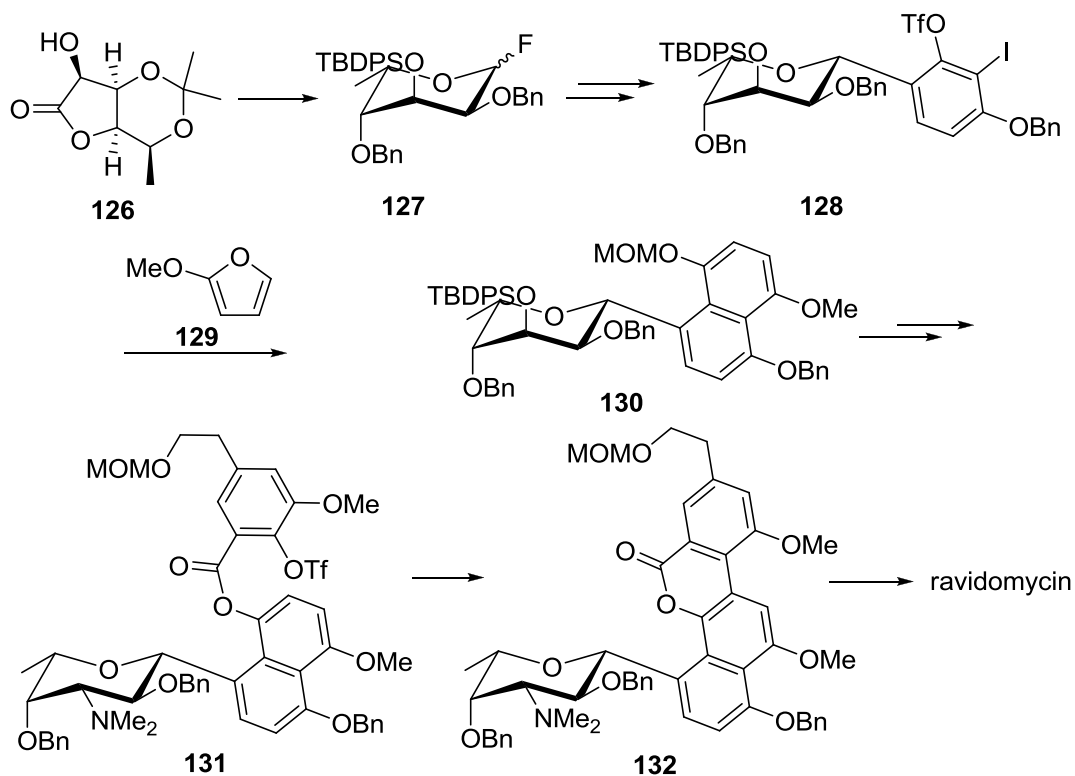
Ravidomycin was isolated from the mycelium of a streptomycete which was obtained from a soil sample from Guatemala (Figure 12).⁸⁵ Ravidomycin is an amino sugar congener of the gilvocarcin antibiotics.⁸⁶ It contains an amino sugar moiety, 3,6-dideoxy-3-N,N-dimethylamino pseudo altropyranose. The remaining aglycone moiety of

ravidomycin is identical with those of toromycinz and gilvocarcin. Since the complex polycyclic structure is a challenge to synthesize, ravidomycin attracted attention.

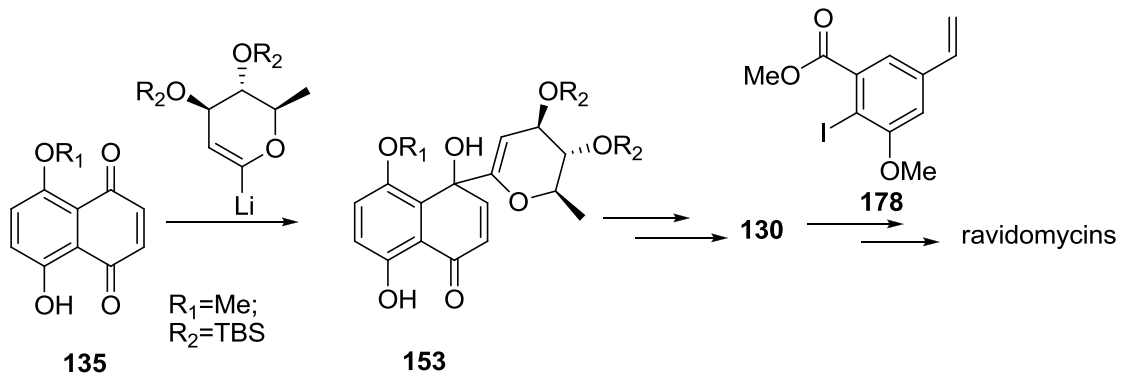
Ravidomycin exhibits strong activity against Gram-positive bacteria, weak activity against Gram-negative bacteria, and no activity against fungi.⁸⁶ It also shows strong activity against P388 lymphocytic leukemia, Colon 38 tumor, and CD8F1 mammary tumor in mice. The mechanism of action of ravidomycin is that it selectively inhibits DNA and RNA synthesis at low concentration.⁸⁷

Suzuki and coworkers finished a total synthesis of ravidomycin (Scheme 41).⁸⁸ The total synthesis of twenty-six steps started from making compound **126**. The overall yield from **126** is 2%. Actually, if the steps for making **126** are counted in, the synthesis took more than thirty two steps. Since compound **127** containing a fluorine atom was difficult to make, it took 17 steps to get the key intermediate **129**. Therefore, the synthetic route was not efficient. Thus development of a new strategy for preparation of ravidomycin is necessary (Scheme 42). Lithiated rhamnol could be added to quinone **135** to give the regioselective monoadduct **153**. The further transformations give **130** from **153**. Phenyl iodide could couple with **130** to offer the target compound.

The key steps of Suzuki's synthesis of ravidomycin



New strategy for preparation of ravidomycin



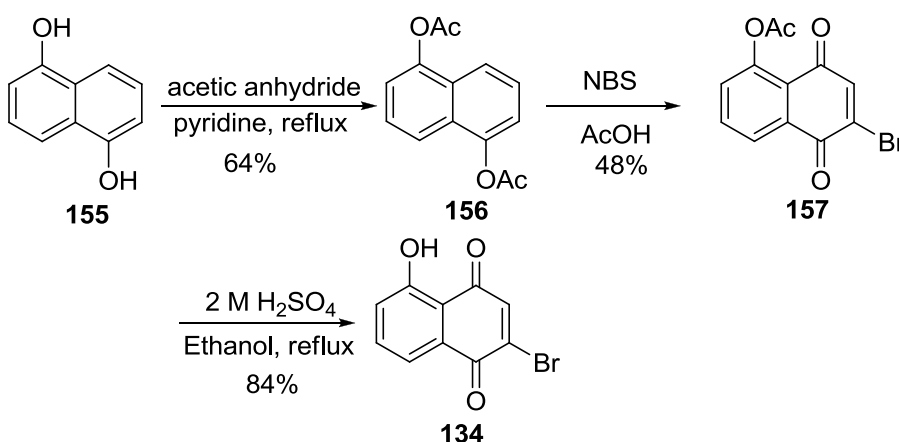
Scheme 41. The key steps of Suzuki's synthesis of ravidomycin and proposed new strategy for preparation of ravidomycin.

2.2 Results and Discussion

2.2.1 The Syntheses of Substrates for 1,2-Addition

2.2.1.1 The Preparation of Bromojuglone 134

Juglone is commercially available, but 2-bromojuglone (**134**) is not. In order to prepare compound **134**, first the hydroxyl groups of 1,5-dihydroxynaphthalene **155** were acylated by acetic anhydride to 1,5-diacetoxynaphthalene **156** in 64% yield (Scheme 42).⁸⁹ Then treatment of 1,5-diacetoxynaphthalene **156** with NBS in acetic acid afforded acetyl-protected bromojuglone **157** in 48% yield.⁸⁹ According to Grunwell's procedures, the acetyl protecting group was removed by strong acid to give bromojuglone **134** in 84% yield.⁸⁹

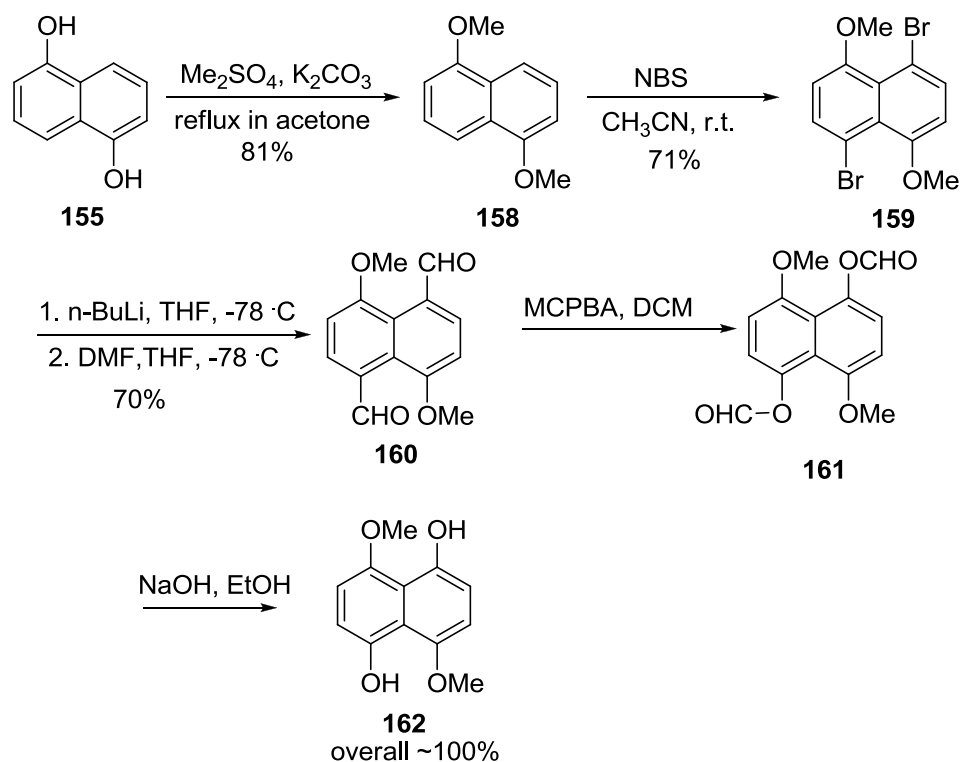


Scheme 42. Preparation of bromojuglone **134**.

2.2.1.2 The Preparation of 8-Hydroxy-5-methoxy-1,4-naphthoquinone 135

In order to synthesize 8-hydroxy-5-methoxy-1,4-naphthoquinone **135**, its precursor, 4,8-dimethoxynaphthalene-1,5-diol **162**, was prepared first (Scheme 43). The hydroxyl groups of 1,5-dihydroxynaphthalene **155** were methylated by dimethyl sulfate

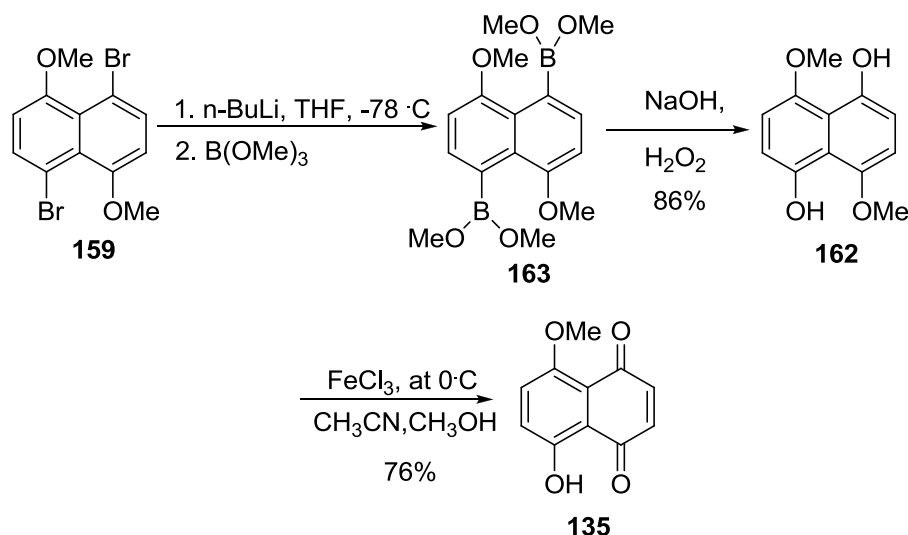
to 1,5-dimethoxynaphthalene **158** in 81% yield.⁹⁰ Treatment of 1,5-dimethoxynaphthalene **158** with NBS in acetonitrile afforded 1,5-dibromo-4,8-dimethoxynaphthalene **159** in 71% yield. Dibromide **159** was transformed to a dicarbanion by two equivalents of n-butyl lithium. Addition of the dicarbanion to DMF produced 4,8-dimethoxy-diformylnaphthalene **160** in 70% yield.⁹¹ According to Wipf's procedures, Baeyer-Villiger oxidation of **160** by 3-chloroperoxybenzoic acid yielded a diformate **161** which was hydrolyzed by sodium hydroxide to form 4,8-dimethoxynaphthalene-1,5-diol **162** in quantitative overall yield.⁹²



Scheme 43. Preparation of compound **162**.

A more convenient synthetic route for dimethoxynaphthalenediol **162** was developed (Scheme 44). 1,5-Dibromo-4,8-dimethoxynaphthalene **159** was transformed to a dicarbanion by two equivalents of n-butyl lithium. The reaction was quenched by trimethylborate to afford a dimethoxynaphthalenylborane; then the borane was oxidized

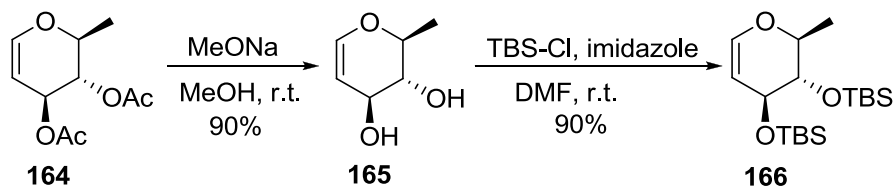
by hydrogen peroxide to yield a naphthalenyldimethoxyborate.⁹³ The resulting borate was hydrolyzed by sodium hydroxide to produce dimethoxynaphthalenediol **162** in overall 86% yield.⁹⁴ According to Wipf's procedures, oxidation of dimethoxynaphthalenediol **162** by iron(III) chloride afforded 8-hydroxy-5-methoxy-1,4-naphthoquinone **135** in 76% yield.⁹⁵



Scheme 44. Preparation of juglone **135** and the other method for preparation of compound **162**.

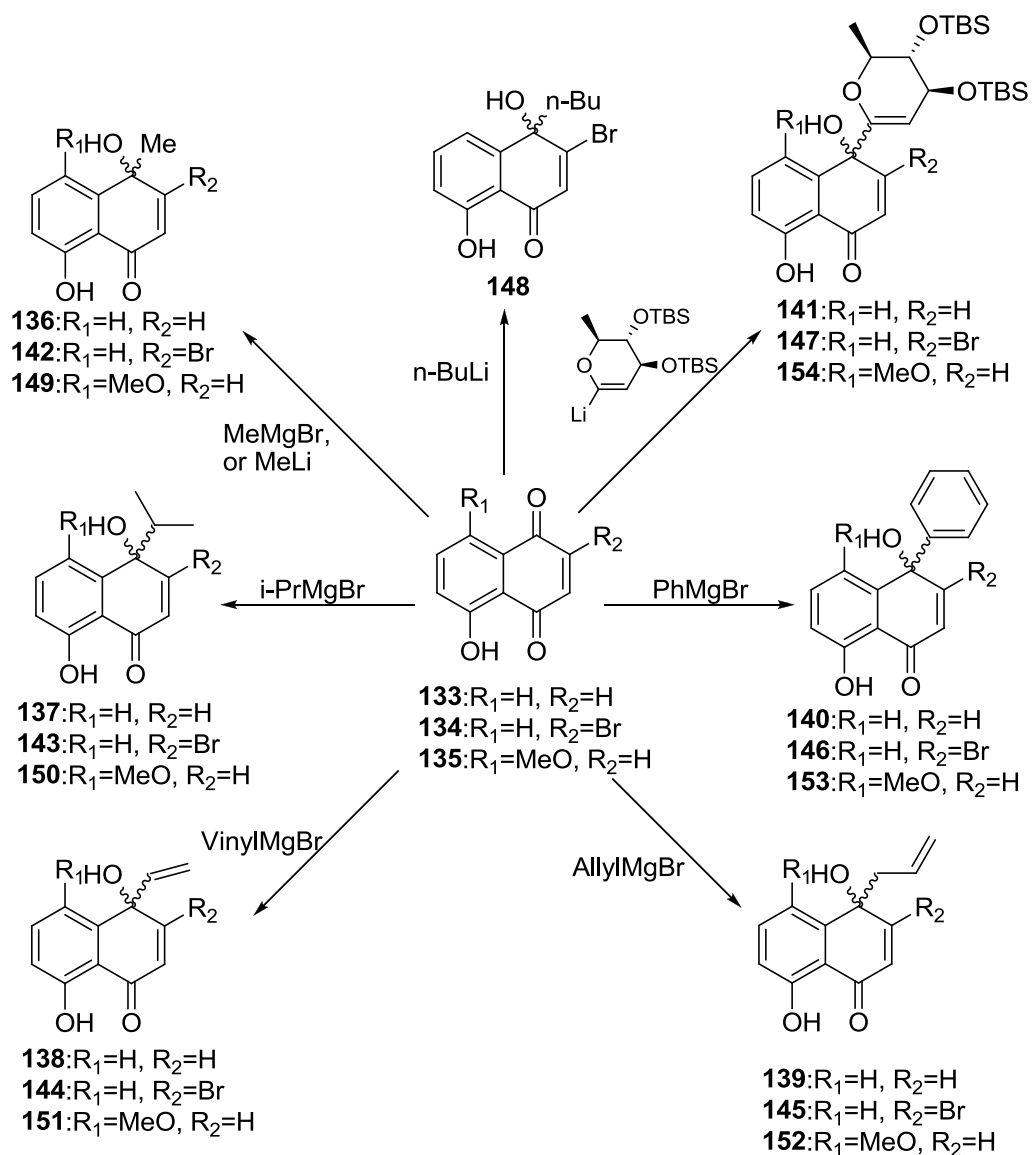
2.2.1.3 The Preparation of TBS-L-Rhamnal **166**

3,4-Di-O-acetyl-6-deoxy-L-glucal **164** was hydrolyzed by sodium methoxide to give L-rhamnal **165** in 90% yield (Scheme 45).⁹⁶ Treatment of L-rhamnal **165** with imidazole and t-butyldimethylsilyl chloride afforded TBS protected L-rhamnal **166** in 90% yield.⁹⁶



Scheme 45. Preparation of TBS protected L-rhamnal **166**.

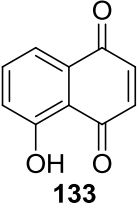
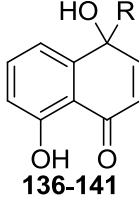
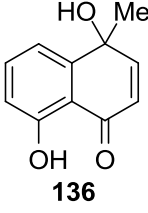
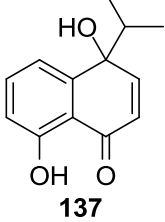
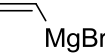
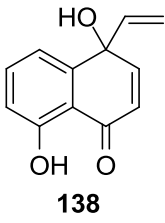
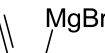
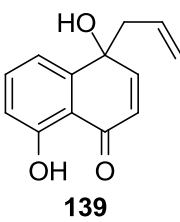
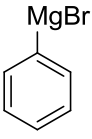
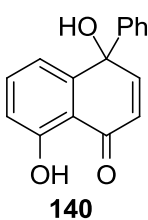
2.2.2-Addition to Juglone Derivatives.

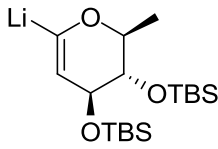
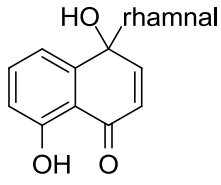


Scheme 46. 1,2-Additions to juglone derivatives

Juglone **133** was added 1.2 equivalents of Grignard reagents or alkyllithium under various reaction conditions to give the monoaddition products in modest yields (Scheme 46 and Table 6.) The organometallic reagents included methylmagnesium bromide, methyl lithium, isopropylmagnesium bromide, phenylmagnesium bromide, vinylmagnesium bromide, allylmagnesium bromide, and lithiated glycol **166**.

Table 6. 1,2-Additions to juglone **133**

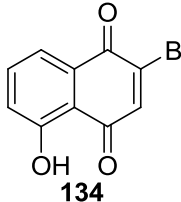
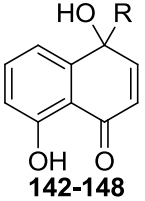
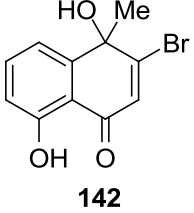
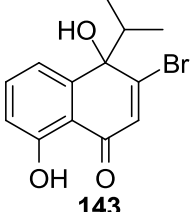
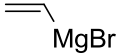
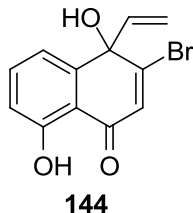
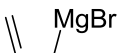
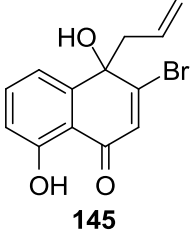
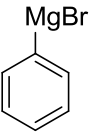
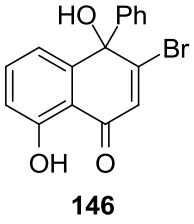
Organometallics	Yields and Conditions	Adducts
 <p>133</p>	Organometallics	 <p>136-141</p>
MeMgBr	46% 8 hours at -78°C	 <p>136</p>
iPrMgBr	41% 8 hours at -78°C	 <p>137</p>
	8% 0.5 hours at -95°C	 <p>138</p>
	19% 8 hours at -78°C	 <p>139</p>
	25% 8 hours at -78°C	 <p>140</p>

Organometallics	Yields and Conditions	Adducts
 <p>166</p>	<p>0% 48 hours at -78°C</p>	 <p>rhamnal 141</p>

The results showed that methyl and isopropyl magnesium bromide offered a moderate yield of around 45%. However phenyl, vinyl, and allyl magnesium bromide gave low yields of adducts; and lithiated glycol **166** could not give the desired product.

Similar procedures were applied to substrate **134** (Table 7) and **135** (Table 8). Organometallic reagents usually add to 2-haloquinones at the C-4 carbonyl group which is nonadjacent to the halogenated carbon.¹⁰⁸ But when 2-bromojuglone **134** was the substrate, organometallic reagents added to the carbonyl group adjacent to the halo substituent. When bromojuglone **134** was the substrate, lithiated glycol **166** was added to the non-hydrogen bonding carbonyl group to produce compound **154** in 51% yield.

Table 7. 1,2-Additions to 2-bromojuglone **134**

Organometallics	Yields and Conditions	Adducts
 <p>134</p>	Organometallics	 <p>142-148</p>
MeMgBr	41% 40 minutes at -93°C	 <p>142</p>
iPrMgBr	37% 8 hours at -78°C	 <p>143</p>
	8% 0.5 hours at -95°C	 <p>144</p>
	0% 8 hours at -78°C	 <p>145</p>
	16% 8 hours at -78°C 47% 48 hours at -78°C	 <p>146</p>

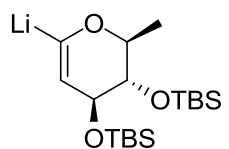
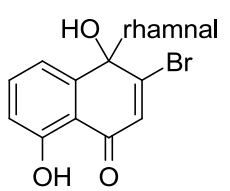
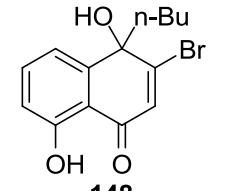
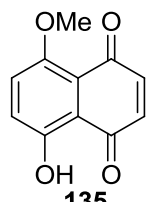
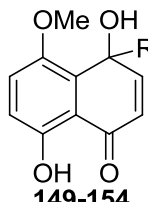
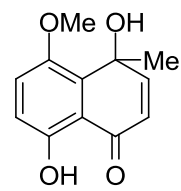
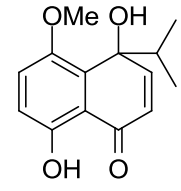
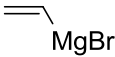
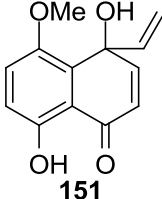
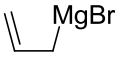
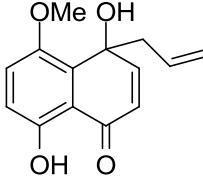
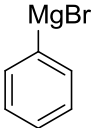
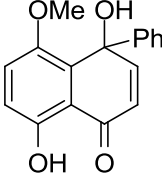
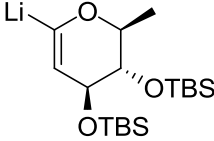
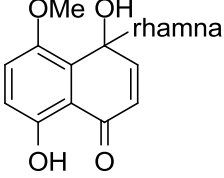
Organometallics	Yields and Conditions	Adducts
 166	0% 48 hours at -78°C	 147
n-BuLi	46% 0.5 hour at -93°C	 148

Table 8. 1,2-Additions to 8-hydroxy-5-methoxy-1,4-naphthoquinone **135**

Organometallics	Yields and Conditions	Adducts
 135	Organometallics	 149-154
MeMgBr	57% 12 hours at -72°C	 149
iPrMgBr	19% 12 hours at -78°C	 150

Organometallics	Yields and Conditions	Adducts
	31% 48 hours at -78°C	 151
	20% 12 hours at -78°C	 152
	46% 12 hours at -78°C	 153
	51% 48 hours at -78°C	 154

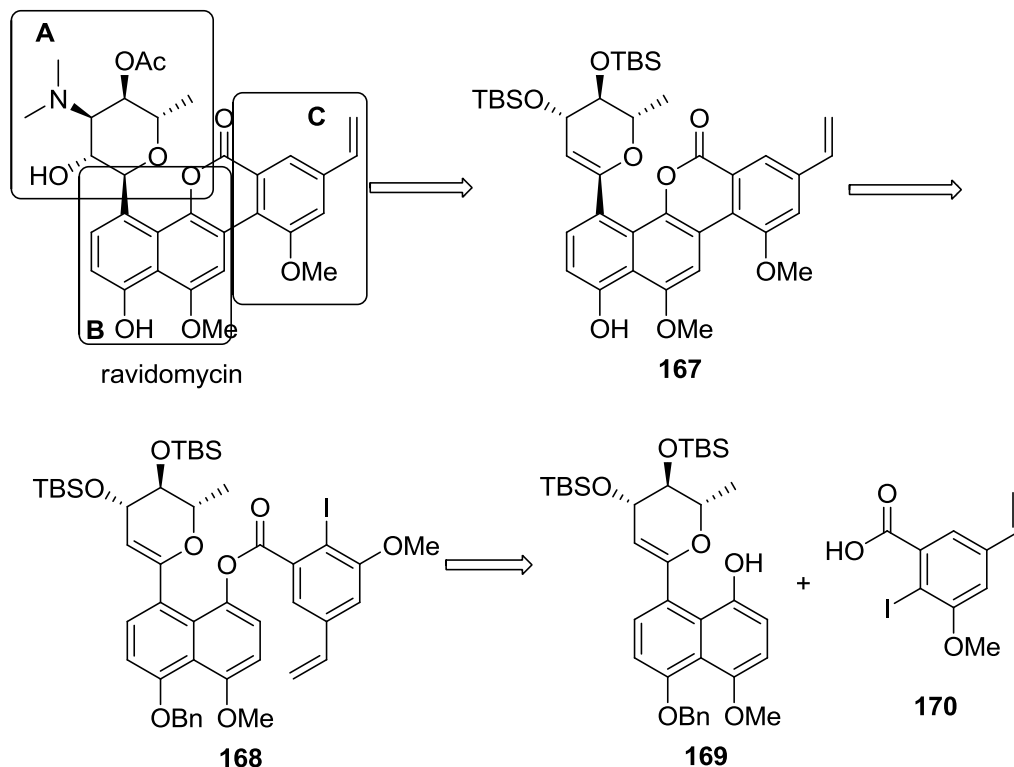
In summary, methylmagnesium bromide afforded the addition products in about 50% yields. With increasing steric effects for the organometallic reagents, the yields decreased. As a result, isopropylmagnesium bromide gave the adducts in about 40% yield. The yields of 1,2-additions of vinylmagnesium bromide and allylmagnesium bromide were low. Phenylmagnesium bromide gave about 20% yields. But when reaction times were lengthened, 1,2-addition of phenylmagnesium bromide gave much better yields. Lithiated rhamnol **165** only added to naphthoquinone **135**. In most cases, 1,2-additions to 8-hydroxy-5-methoxy-1,4-naphthoquinone **135** afforded better yields.

The results suggest that the addition reaction is more rapid than deprotonation of the hydroxyl group donating the intramolecular hydrogen bond. This hydrogen bonding hydroxyl group was involved in distinguishing the two carbonyl groups. Although the methoxy group on C-5 of 8-hydroxy-5-methoxy-1,4-naphthoquinone **135** has a steric effect, organometallic reagents regioselectively added to the carbonyl close to the methoxy group, and the addition to **135** produced higher yields than those to **133** and **134**. The result suggest that the hydrogen bond of 8-hydroxy-5-methoxy-1,4-naphthoquinone **135** may be stronger than those of **133** and **134**.

The strong intramolecular hydrogen bonding of juglone derivatives has not gone unrecognized and it has been the subject of a number of spectroscopic investigations.⁹⁷ Its effect as a director of regiochemistry in Diels-Alder reactions has been reported.⁹⁸ However, the hydrogen bond has not previously been applied for the control of regiochemistry in the addition of organometallic reagents. The regioselective addition of organometallic reagents to juglone derivatives could be the method of choice for the synthesis of densely functionalized naphthoquinols.

2.2.3 Progress toward the Synthesis of Ravidomycins.

As an application of the regioselective 1,2-additions of organometallic reagents to juglone derivatives, a new synthetic route for ravidomycin was proposed (Scheme 37). Ravidomycin intermediate **167** could be easily transformed to ravidomycin by treatment with trichloroacetonitrile and a subsequent borohydration.⁹⁹ Pd(II)-catalyzed intramolecular biaryl coupling of the naphthalenol ester **168** might lead to the polycyclic intermediate **167**.¹⁰⁰ Naphthalenol **169** might couple with the acid **170** to produce the ester **168**.

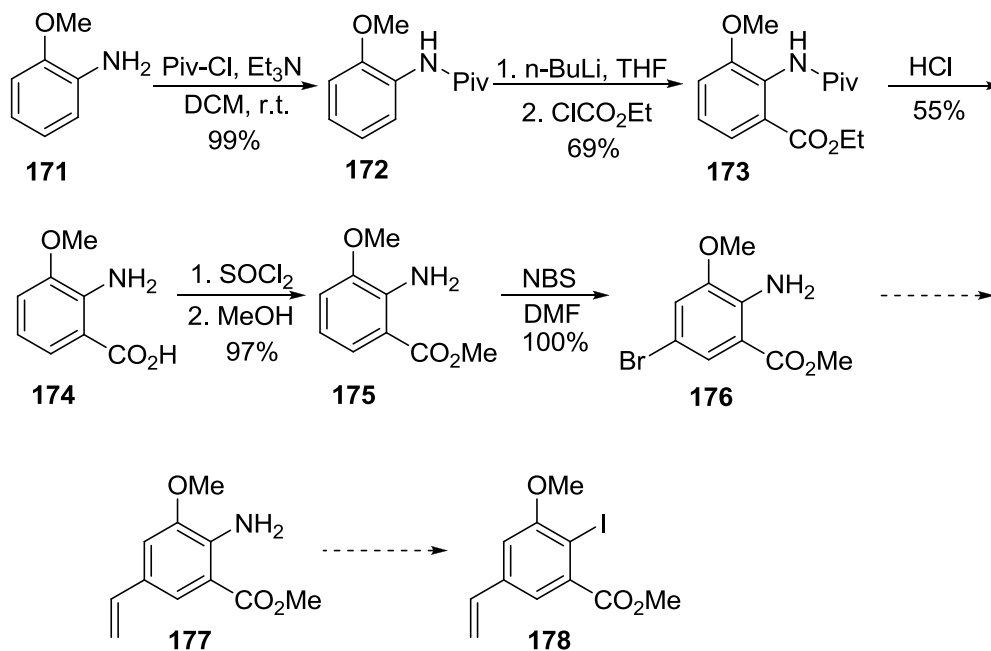


Scheme 47. Proposed synthetic route for ravidomycin

2.2.3.1 Toward Preparation of Iodide 178, Region C.

We planned to prepare phenyl iodide **178** to build the region C by the method shown in Scheme 48. The amino group of *o*-anisidine **171** was protected by pivaloyl chloride to produce compound **172** in 99% yield.¹⁰¹ *n*-Butyllithium deprotonated the Piv-protected *o*-anisidine **172** at C-3 position; and then the resulting carbanion was quenched by ethyl chloroacetate to form 2-amino-3-methoxybenzoic ester **173** in 69% yield.¹⁰² The Piv protecting group was removed by HCl to provide the amino acid **174** in 55% yield. Treatment with thionyl chloride converted the amino acid **174** to an acid chloride which was methoxylated to give the amino ester **175** in 97% yield. The amino ester **175** was oxidized by NBS at C-5 position to bromide **176** quantitatively.¹⁰¹ The bromide **176** was supposed to couple with tetravinyltin to produce the ester **177**. The amino group of **177**

could be oxidized to a diazo group which could be substituted by iodine to form *m*-iodobenzoic ester **178**.¹⁰⁰

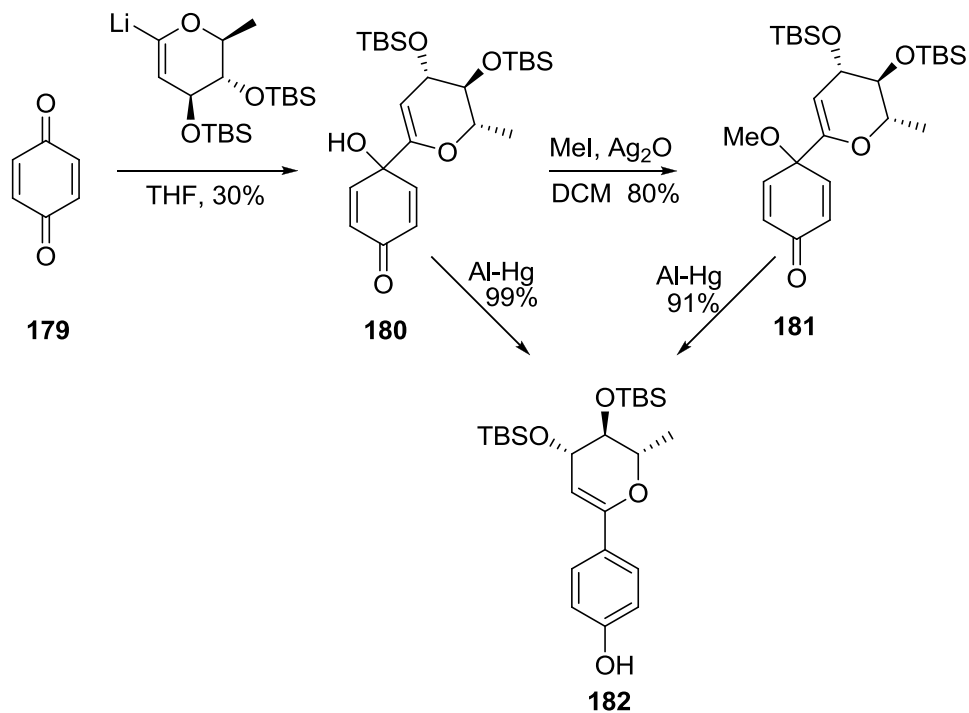


Scheme 48. Toward preparation of iodide **178**.

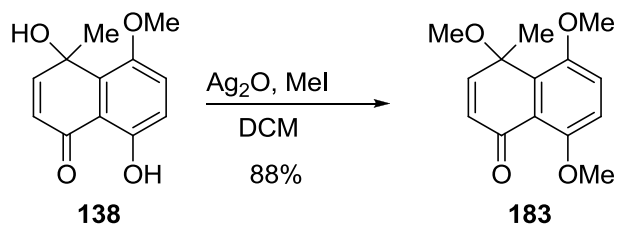
2.2.3.2 Models for Construction of Region A and B of Ravidomycin

2.2.3.2.1 Reduction of Quinols (Region B).

In order to build the region B of ravidomycin, a model system for preparing a carbohydrate substituted phenol was examined (Scheme 49). Lithiated rhamnol was added to quinone **179** to give the monoadduct **180** in 30% yield.⁹⁹ According to past work from Parker's group, mercury-aluminum amalgam reduced the *p*-quinol **180** to the phenol **182** quantitatively.¹⁰² The methylated quinol **181** was also reduced by mercury-aluminum amalgam to the phenol **182** in 91% yield.¹⁰³

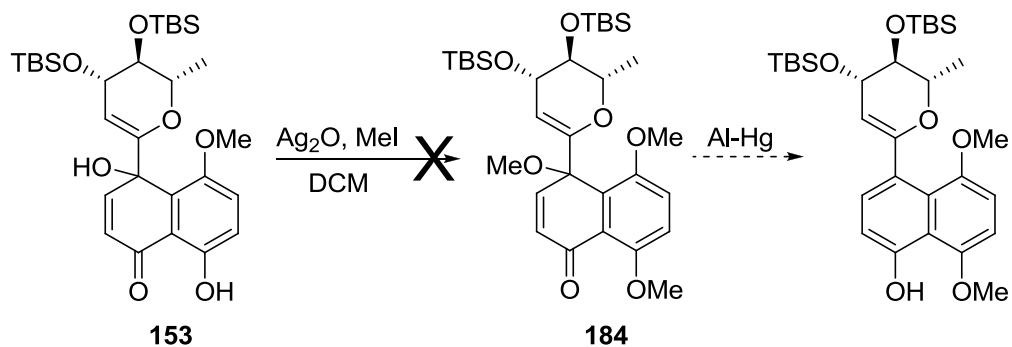


Scheme 49. A model for reduction of quinols.

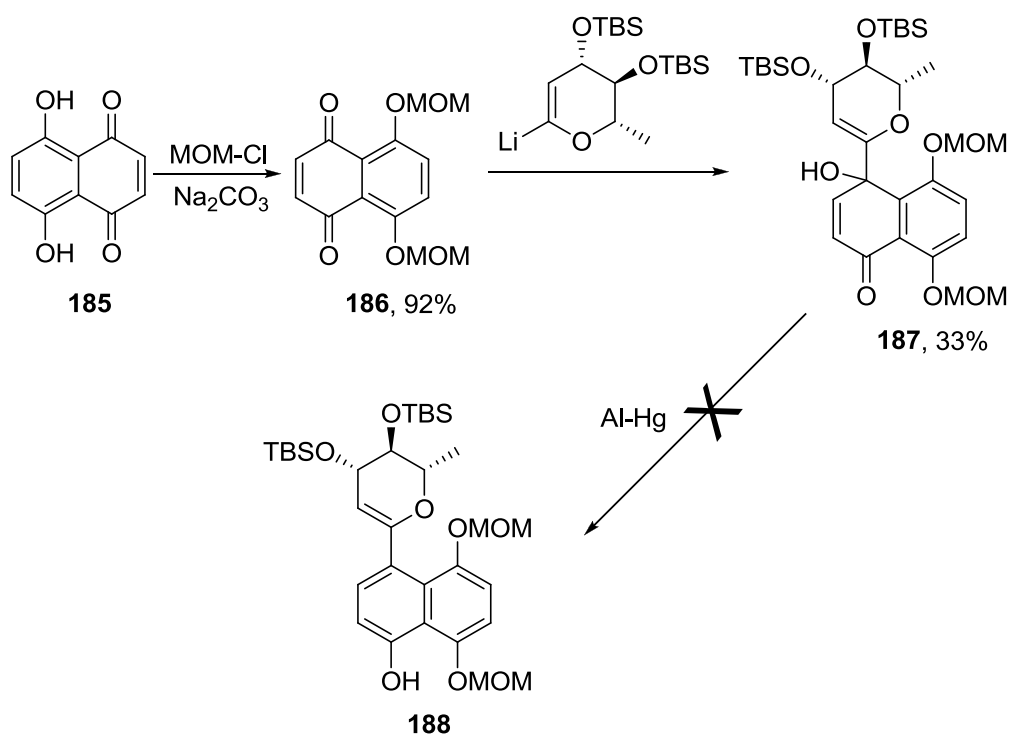


Scheme 50. Methylation of quinol **138**.

In order to examine the reduction method in the 5,8-dimethoxy-4-naphthoquinol system, methylation of carbohydrate substituted quinol **153** was studied. Treatment of L-rhamnal **165** with silver(I) oxide and iodomethane afforded the methylated quinol **183** in 88% yield (Scheme 50). Methylation of quinol **153** was also attempted using silver(I) oxide and iodomethane (Scheme 51). However, the target methylated quinol **184** was not obtained.



Scheme 51. Attempt at methylation of naphthylquinol **153**.

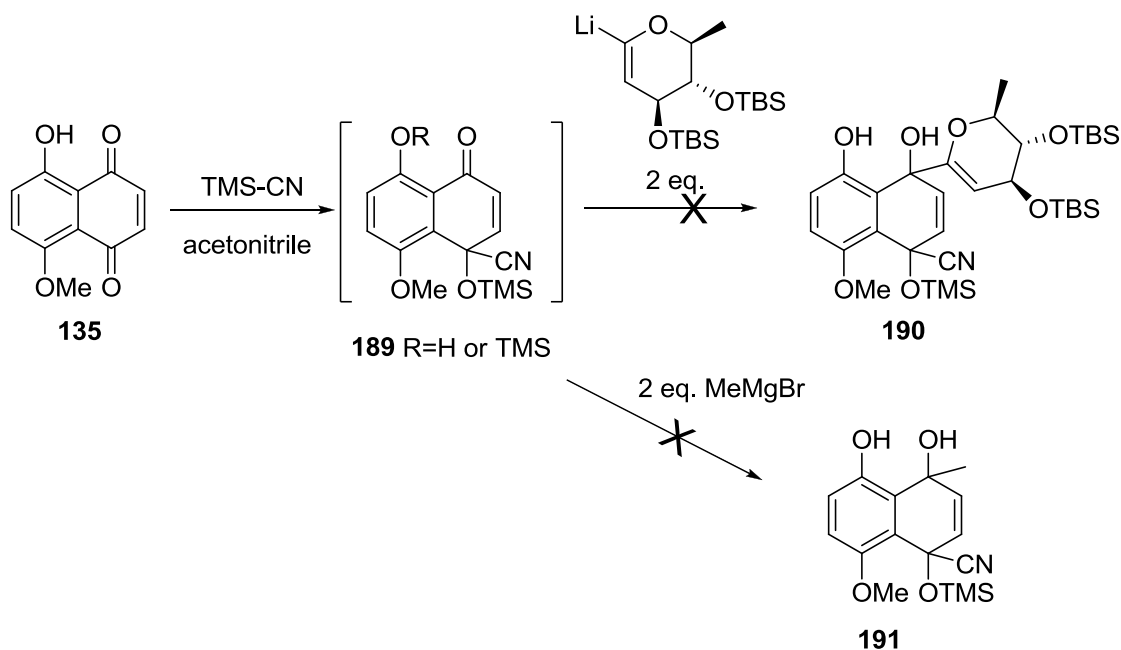


Scheme 52. Attempt at reduction of naphthylquinol **187**.

Since methylation of naphthylquinol **153** did not work, a direct reduction of **187** without methylation was tried (Scheme 52). Compound **186** was obtained in 92% yield by treatment of **185** with chloromethylmethylether (MOM-Cl) in the presence of sodium carbonate.¹⁰⁴ Then addition of lithiated rhamnal to **186** afforded compound **187** in 33%

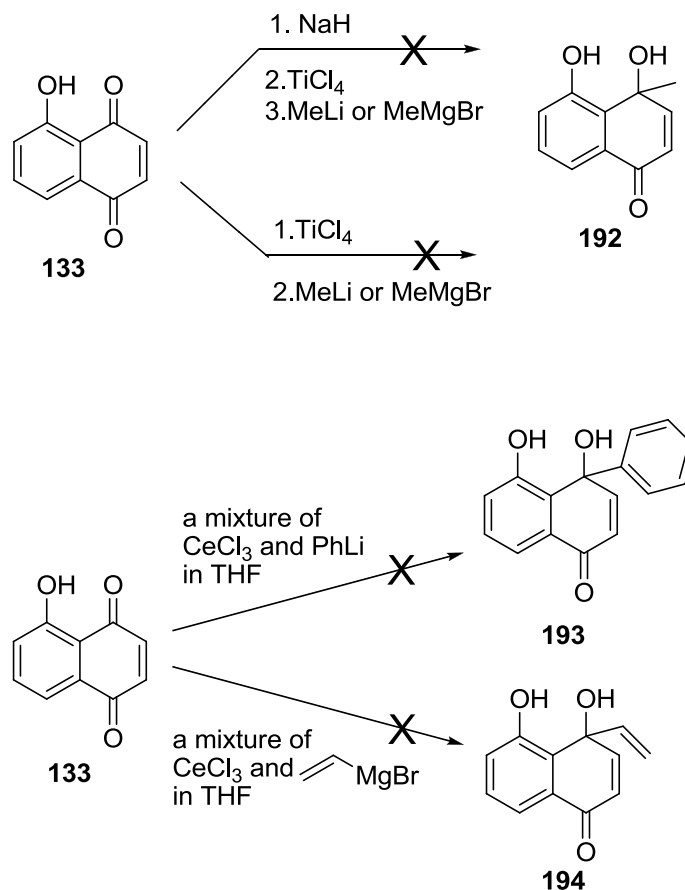
yield. However mercury-aluminum amalgam did not give any desired compound but the starting material.

2.2.3.2.2 Attempts at 1,2-Additions to the Hydrogen Bonded Carbonyl Groups of Juglone Derivatives (Region B).



Scheme 53. An umpolung strategy.

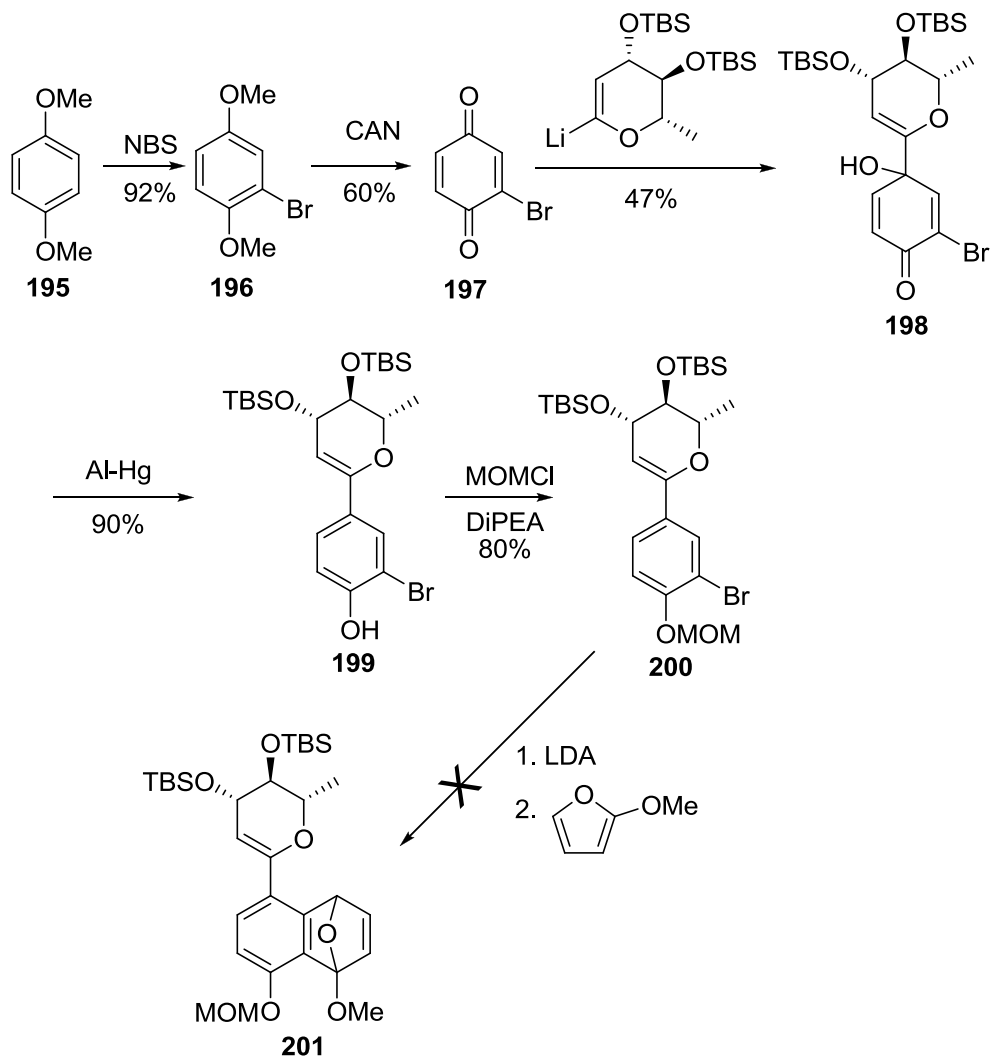
Selective 1,2-addition to hydrogen bonding C1 carbonyls of juglone derivatives might shorten the proposed synthetic route for ravidomycin. An umpolung strategy which is different from the previously described 1,2-addition was attempted (Scheme 53). The C5 carbonyl group of compound 135 was protected by trimethylsilyl cyanide to form an intermediate 189.¹⁰⁵ Then 189 was treated with lithiated rhamnol or methylmagnesium bromide. However, the desired product 190 or 191 was not obtained.



Scheme 54. Lewis acids mediated 1,2-addition attempts.

In order to get addition products at the C-4 position rather than the C-1 position, a series of models for Lewis acid-mediated 1,2-addition to the hydrogen bonded carbonyl of juglone **133** were tried (Scheme 54). In the presence of titanium(IV) chloride,¹⁰⁶ additions of methyllithium or methylmagnesium bromide to juglone **133** and deprotonated juglone did not afford the desired product **192**. When cerium(III) chloride was used as the Lewis acid,¹⁰⁷ addition of phenyl lithium or vinylmagnesium bromide to juglone did not give the corresponding adduct, either.

2.2.3.2.3 Diels-Alder Strategy.

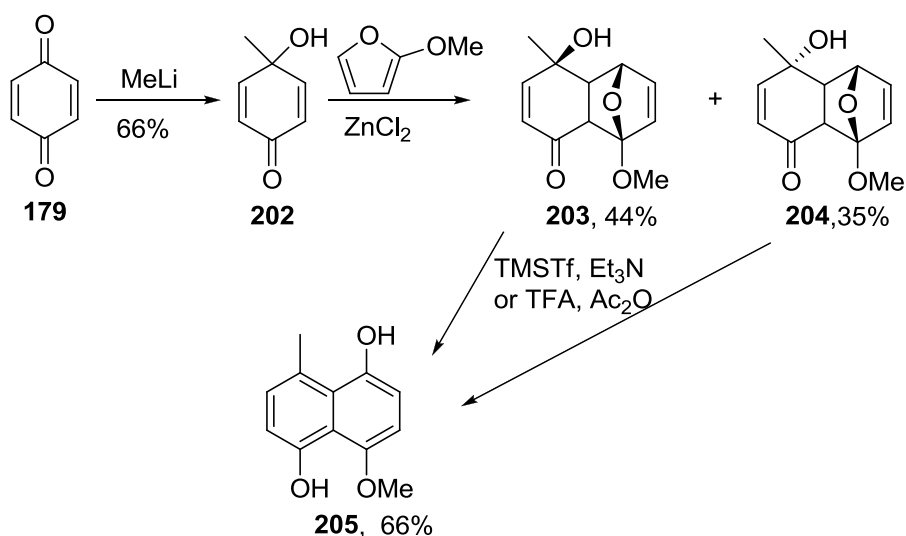


Scheme 55. Diels-Alder reaction of benzyne and 2-methoxyfuran.

Since the reduction of naphthylquinol to naphthalenol has not been solved, a Diels-Alder strategy was employed to construct the regions A and B of ravidomycin (Scheme 47 and 55). 1,4-Dimethoxybenzene **195** was oxidized by NBS to 2-bromo-1,4-dimethoxybenzene **196** in 92% yield. The further oxidation of **196** by ceric ammonium nitrate (CAN) gave the 2-bromoquinone **197** in 60% yield.¹⁰⁸ Then compound **198** was obtained in 47% yield by selective 1,2-addition of lithiated rhamnol to **197**. The presence

of the bromo group caused the selectivity. Mercury-aluminum amalgam reduced quinol **198** to phenol **199** in 90% yield. Chloromethylmethylether (MOMCl) was used to protect the hydroxyl group of **199**. However, the following Diels-Alder reaction did not work. Because the proton at the C3 position of compound **200** was distant to the MOMO group, it was not acidic enough to be deprotonated to form a benzyne by strong bases such as LDA.

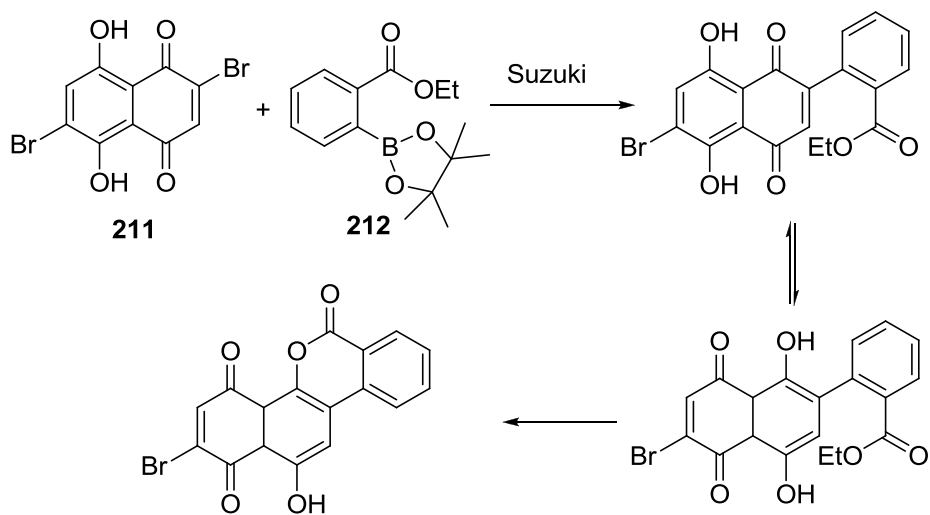
After two oxidations and two reductions, the oxidation state from compound **195** to **199** did not change (Scheme 55). The presence of the bromo group requests two more steps, one oxidation and one reduction. Therefore the bromo group was not necessary for the Diels-Alder strategy. A new model of Diels-Alder strategy was built (Scheme 56). Compound **202** was obtained by addition of methyllithium to benzoquinone **179**.¹⁰⁹ **202** was in the same oxidation state as compound **195**. In the presence of zinc chloride (ZnCl_2), a Diels-Alder cyclization between compound **202** and 2-methoxyfuran produced two isomers **203** and **204** in 44% and 35% yields, respectively. Treatment with trimethylsilyl triflate (TMSTf) or trifluoroacetic acid (TFA) opened the furan rings of both isomers to give the product **205**.



Scheme 56. Diels-Alder reaction of *p*-methylquinol and 2-methoxyfuran

2.2.3.3 A Model for Regions B and C of Ravidomycin, Suzuki Cross Coupling between Dibromo Juglone Derivatives and 2-Ethoxycarbonylphenylborate.

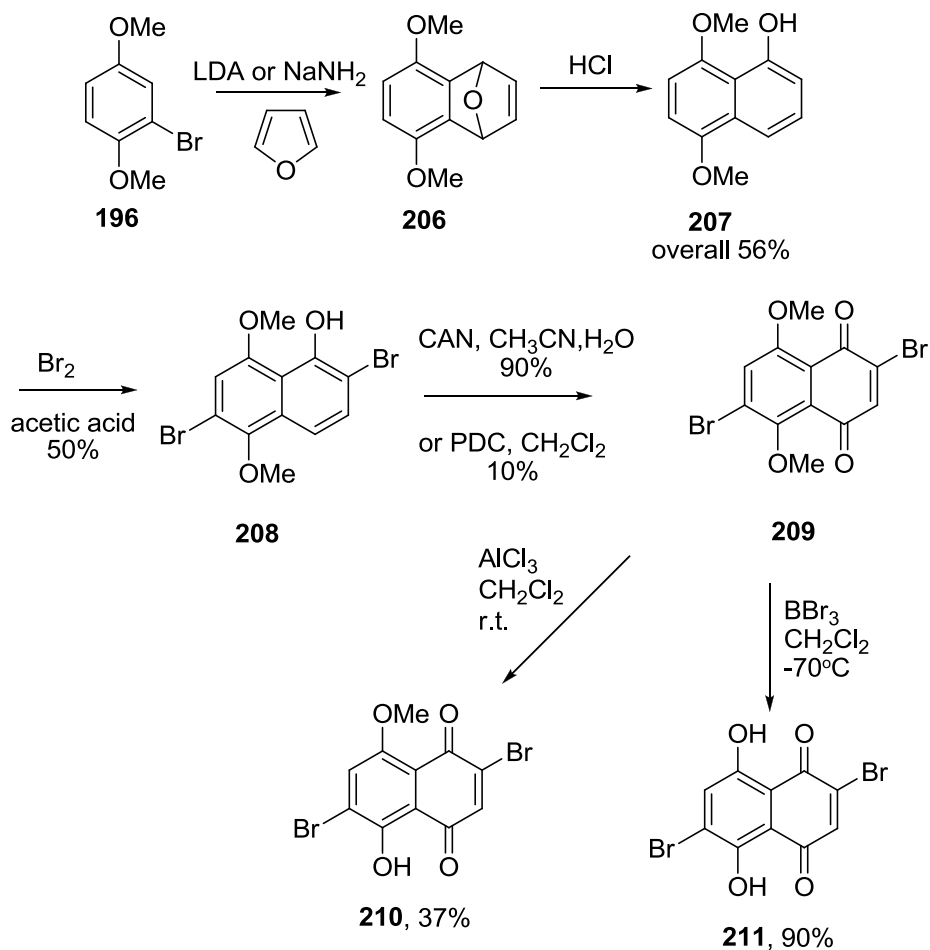
Suzuki cross coupling¹¹⁰ between a bromojuglone derivative and borate might shorten the proposed synthetic route for regions B and C of ravidomycin (Scheme 57). Considering that the presence of a bromo group might result in selective 1,2-addition of organometallics to the C-4 carbonyl of the naphthylquinone, a model of the Suzuki coupling between dibromo dibromonaphthylquinones and 2-ethoxycarbonylphenylborate was built.



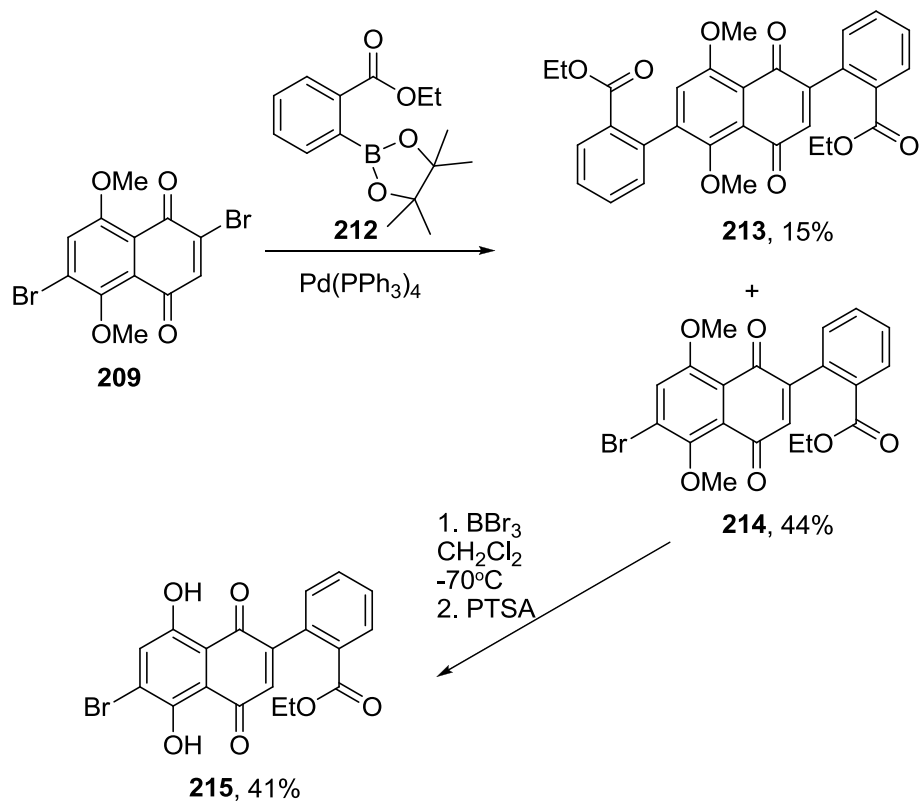
Scheme 57. Proposed Suzuki strategy.

A strong base removed the proton at the C3 position of **196** to form a benzyne which underwent a Diels-Alder cyclization with furan to yield compound **206** (Scheme 58).¹¹¹ Treatment with acid opened the furan ring of **206** to afford the compound **207** in overall 56% yield.¹¹² The oxidation of **207** by bromine in acetic acid gave the dibromo compound **208** in 50% yield.¹¹³ The further oxidation by pyridinium dichromate (PDC) transformed **208** to naphthylquinone **209** in very low yield. Ceric ammonium nitrate (CAN) as the oxidant increased the yield of the transformation to 90%.¹¹⁴ One of the two

methyl groups of naphthylquinone **209** was removed by aluminum chloride to produce monoprotected dibromonaphthylquinone **210** in 37% yield. Treatment with boron bromide deprotected both methoxy groups to give the dibromonaphthylquinone **211** in 90% yield.¹¹⁵

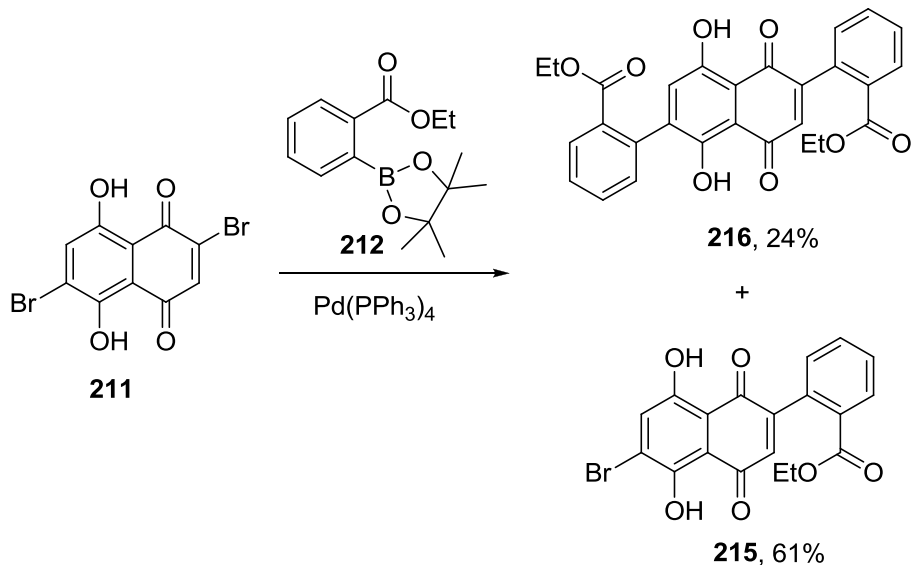


Scheme 58. Preparation of dibromo juglone derivatives **209**, **210**, and **211**.



Scheme 59. Suzuki coupling of **209** and 2-ethoxycarbonylphenylborate **212**.

Under the catalysis of tetrakis(triphenylphosphine)palladium(0) ($\text{Pd}(\text{PPh}_3)_4$), dibromonaphthylquinone **209** and 2-ethoxycarbonylphenylborate **212** coupled together to afford mono-coupling product **214** and di-coupling product **213** in 44% and 15% yields, respectively (Scheme 59). After demethylation of compound **200** by boron bromide, treatment with p-toluenesulfonic acid (PTSA) did not yield a lactone product but the deprotected compound **215** in 41% yield.



Scheme 60. Suzuki coupling of **211** and 2-ethoxycarbonylphenylborate **212**.

Under the catalysis of tetrakis(triphenylphosphine)palladium(0) (Pd(PPh₃)₄), dibromonaphthylquinone **211** and 2-ethoxycarbonylphenylborate **212** coupled together to afford mono-coupling product **215** and di-coupling product **216** in 61% and 24% yields, respectively (Scheme 60). The yields were much better than the coupling between **209** and **212**.

2.3 Experimental Section

General

For all compounds, ¹H, ¹³C and NOE NMR spectra were recorded on Varian Gemini 2300 (300 MHz), Inova 400 (400 MHz), Inova 500 (500 MHz), Inova 600 (600 MHz) spectrometer. Chemical shifts were measured relative to the residual solvent resonance for ¹H and internal reference. IR spectra were recorded on a Mattson Galaxy Series FTIR 3000 or 2020 spectrophotometer.

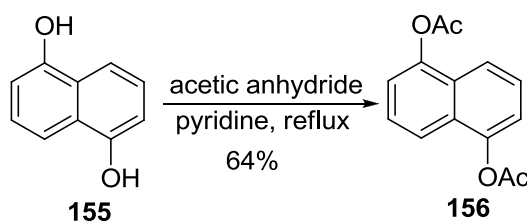
Reagents and solvents were purchased from Aldrich Chemical Company, Fisher Scientific or Acros Organics. Liquid reagents were purified by distillation prior to use.

Unless otherwise noted, solid reagents were used without further purification. Methylene chloride, acetonitrile, cyclohexane, and toluene were distilled from CaH₂ under Ar atmosphere. Diethyl ether and THF were distilled from Na and benzophenone under Ar atmosphere.

All reactions unless otherwise noted, were performed in oven dried glassware under positive pressure of Ar. All yields are isolated yields.

Chromatographic separations were performed using Fisher grade 1740 type 60 Å silica gel (170-400 mesh). Analytical thin layer chromatography was performed using Analtech Uniplate pre-coated glass 250 micron silica gel GHLF plates with 254 nm fluorescent indicator. Preparative thin layer chromatography was performed using Analtech Uniplate pre-coated glass 1000 or 2000 micron silica gel GHLF plates with 254 nm fluorescent indicator.

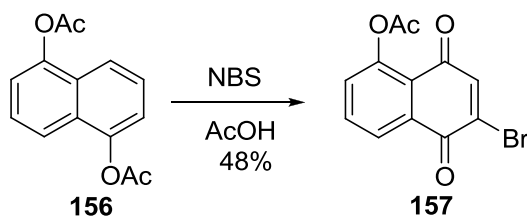
1,5-Dihydroxynaphthalene ester **156**.⁸⁹



1,5-Dihydroxynaphthalene **155** (21 g, 131 mmol), acetic anhydride (30 mL, 27mmol), and piperidine (25 mL) were mixed together in methylene chloride. After the mixture was refluxed overnight, it was poured into ice cold water (800 mL). The water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution, the residue was recrystallized in acetone to give a brown solid **156** (20.4 g, 83 mmol, 64%).

$^1\text{H-NMR}$ (CDCl_3): δ 7.78 (dd, 2H, $^3J = 7.7$ Hz, $^3J = 0.9$ Hz), 7.50 (dd, 2H, $^3J = 7.7$ Hz, $^3J = 0.9$ Hz), 7.29 (dd, 2H, $^3J = 7.7$ Hz, $^3J = 0.9$ Hz), and 2.47 (s, 6H). The spectroscopic and other physical data reported are consistent with values reported in the literature.⁸⁹

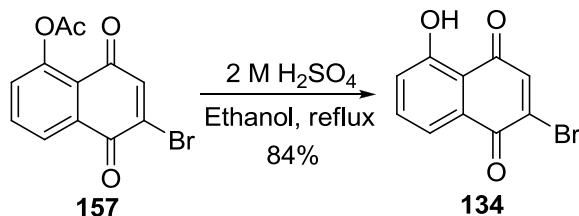
Acetyl bromojuglone **157**.⁸²



A solution of NBS (12.0 g, 67 mmol) in a mixture of acetic acid (50 mL) and water (120 mL) was warmed up to 55 °C. A solution of 1,5-dihydroxynaphthalene ester **156** (4.2 g, 17 mmol) in warm acetic acid (70 mL) was added dropwise to the NBS solution in 15 minutes at 55 °C. The reaction was stirred for 80 minutes and then it was poured into cold water (800 mL). The water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was recrystallized in ethanol to give a yellow solid **157** (2.41 g, 8.3 mmol, 48%).

$^1\text{H-NMR}$ (CDCl_3): δ 8.14 (dd, 1H, $^3J = 8.0$ Hz, $^3J = 1.0$ Hz), 7.77 (dd, 1H, $^3J = 7.5$ Hz, $^3J = 8.0$ Hz), 7.41 (d, 1H, $^3J = 7.0$ Hz), 7.39 (s, 1H), and 2.44 (s, 3H). The spectroscopic and other physical data reported are consistent with values reported in the literature.⁸²

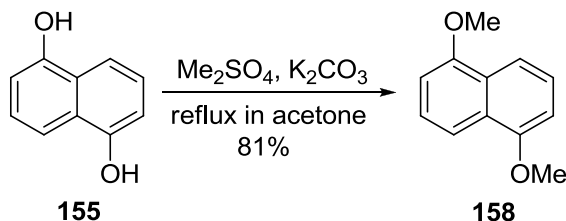
Bromojuglone **134**.⁸²



A solution of acetyl bromojuglone **157** (2.41 g, 8.3 mmol) in ethanol (95%, 120 mL) was added by sulfuric acid (30%, 30 mL). The reaction was refluxed carefully for 3 hours and then it was poured into ice cold water (400 mL). The water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution, the residues were separated by a silica gel column with ethyl acetate and hexanes (1:10) to give the yellow aldehyde **134** (1.74 g, 6.9 mmol, 84%).

¹H-NMR (CDCl₃): δ 11.77 (s, 1H), 7.73(d, 1H, ³J = 7.0 Hz), 7.63 (d, 1H, ³J = 8.0 Hz), 7.50 (s, 1H), and 7.31 (d, 1H, ³J = 8.0 Hz). ¹³CNMR (CDCl₃): δ 189.6, 187.7, 177.4, 161.9, 141.1, 140.5, 136.6, 130.9, 125.3, 121.1, and 114.8. IR (KBr): ν_{max} 1674, 1584, 1452, 1244, 1198, and 1088 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.⁸²

1,5-Dimethoxynaphthalene **158**.⁹⁰

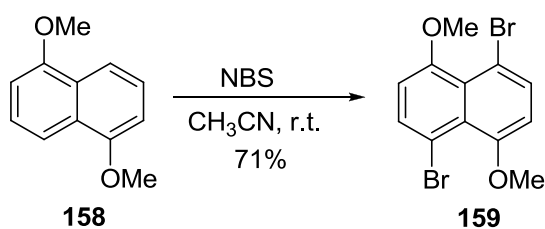


1,5-Dihydroxynaphthalene **155** (2.0 g, 13 mmol) and K₂CO₃ (4.0 g, 29 mmol) were mixed in 50 mL of dry acetone. Dimethyl sulfate (3.4 mL, 36 mmol) was added to the suspension at room temperature. After 12 hours reflux, the solid was filtered off. The

filtrate was concentrated in vacuo. The recrystallization of the residues gave 1.9 g of a yellow solid **158** (10.5 mmol, 81%).

$^1\text{H-NMR}$ (CDCl_3): δ 7.87 (d, 2H, $^3J = 8.5$ Hz), 7.41 (dd, 2H, $^3J = 8.5$ Hz, $^3J = 7.5$ Hz), 6.87 (d, 2H, $^3J = 7.5$ Hz), and 4.01 (s, 6H). The spectroscopic and other physical data reported are consistent with values reported in the literature.⁹⁰

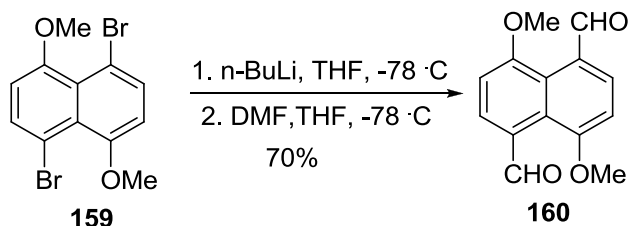
1,5-Dibromo-4,8-dimethoxynaphthalene **159**.⁹¹



1,5-Dimethoxynaphthalene **158** (1.89 g, 10.0 mmol) was dissolved in 30 mL of CH_3CN . N-Bromosuccinimide (4.05 g, 22.8 mmol) in 20 mL of CH_3CN was added to the solution at 0 °C. After addition, the reaction was allowed to warm to room temperature and it was stirred for 4 hours. Then 20 mL of CH_3CN was removed from the dark red solution by rotovap and the residue was crystallized at 0 °C. The crystal was filtered off and washed with 10 mL of CH_3CN and 5 mL of methanol to give a brown solid **159** (2.48 g, 7.1 mmol, 71%).

$^1\text{H-NMR}$ (CDCl_3): δ 7.70 (d, 2H, $^3J = 8.0$ Hz), 6.74 (d, 2H, $^3J = 8.0$ Hz), and 3.92 (s, 6H). $^{13}\text{CNMR}$ (CDCl_3): δ 155.3, 133.9, 126.4, 108.6, 107.3, and 55.9. IR (KBr): ν_{max} 2963, 2934, 1588, 1509, 1365, 1250, 1179, and 1060 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.⁹¹

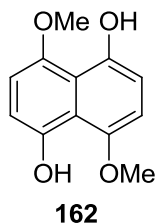
4,8-Dimethoxynaphthalene-1,5-dicarbaldehyde **160**.⁹¹



A solution of 1,5-dibromo-4,8-dimethoxynaphthalene **159** (580 mg, 1.69 mmol) in 20 mL of THF was cooled down to -78 °C. n-BuLi (2.5 M in hexanes, 1.8 mL, 4.5 mmol) was added to the solution at -78 °C and the reaction was stirred for 2 hours at -78 °C. Then the dark yellow suspension was added dimethylformamide (1.2 mL) dropwise. After the mixture was stirred for 3 hours, the reaction was allowed to warm to 0 °C and quenched by 1 M HCl. The yellow solid was filtered off and washed with ether to give the product (286 mg, 1.18 mmol, 70%).

¹H-NMR (CDCl₃): δ 10.90 (s, 2H), 8.01 (d, 2H, ³J = 8.0 Hz), 7.06 (d, 2H, ³J = 8.0 Hz), and 4.08 (s, 6H). IR (KBr): ν_{max} 2969, 2902, 2854, 1671, 1543, 1411, 1375, and 1271 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.⁹¹

4,8-Dimethoxy-1,5-naphthalenediol **162**.^{92, 94}



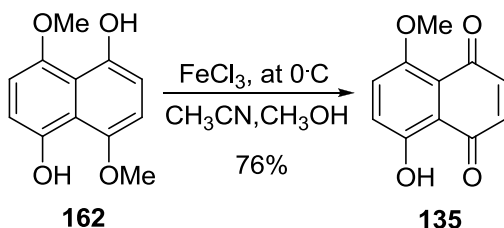
Method one:⁹² A solution of 4,8-dimethoxynaphthalene-1,5-dicarbaldehyde **160** (213 mg, 0.87 mmol) in 15 mL of methylene chloride was cooled down to 0 °C. A solution of m-chloroperbenzoic acid (520 mg, 3 mmol) in 6 mL of methylene chloride was added to the solution at 0 °C. After the reaction was stirred for 18 hours at room

temperature, it was quenched by sat. NaHCO_3 . The organic phase was separated; the water phase was extracted three times by ether. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residues were added a solution of NaOH (1 g) in EtOH (15 mL) and water (2 mL). The mixture was stirred overnight and poured into 1M HCl . The yellow solid was filtered off and washed with ether to give the crude product **162** (319 mg, 0.87 mmol, ~100%).

Method two:^{93, 94} A solution of 1,5-dibromo-4,8-dimethoxynaphthalene **159** (510 mg, 1.47 mmol) in 20 mL of THF was cooled down to -78°C . $n\text{-BuLi}$ (2.5 M in hexanes, 1.4 mL, 3.5 mmol) was added to the mixture at -78°C . The reaction was stirred for 2 hours at -78°C . The dark yellow suspension was added trimethyl borate (1.7 mL, 15.2 mmol) dropwise. The dark mixture resulted was stirred overnight. Then the reaction was allowed to warm to 0°C . A solution of NaOH (290 mg 7.25 mmol) in 2.9 mL of H_2O and 0.8 mL of H_2O_2 was added to the reaction. The mixture was stirred for 4 hours to give a lot of yellow solid. The suspension was poured into 50 mL of 1 M HCl . The yellow solid was filtered off and washed with ether to give the compound **162** (287 mg, 1.26 mmol, 86%).

$^1\text{H-NMR}$ (CDCl_3): δ 9.47 (s, 2H), 6.82 (d, 2H, $^3J = 3.0$ Hz), and 4.06 (s, 6H). IR (KBr): ν_{max} 2964, 2921, 2850, 1662, 1523, 1324, and 1224 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.^{92, 94}

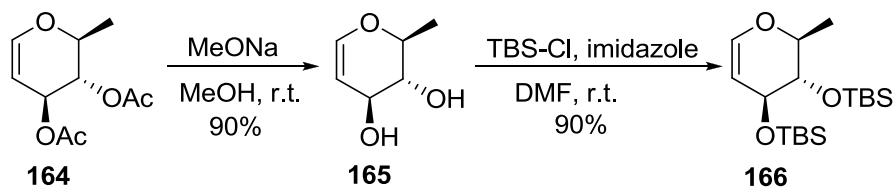
5-Hydroxy-8-methoxy-1,4-naphthalenedione **135**.⁹⁵



A solution of 4,8-dimethoxy-1,5-naphthalenediol **162** (256 mg, 1.16 mmol) in 22 mL of 1:1 MeCN and MeOH was cooled down to 0 °C in an ice bath. Iron (III) chloride hexahydrate (1.26 g, 4.66 mmol) in 2 mL of 4 N HCl was added to the mixture. The reaction was stirred for 15 minutes at 0 °C; and then it was quenched with 8 mL of 1 N NaOH. The red solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with methylene chloride and ethyl acetate (10:1) to give a red solid **135** (192 mg, 0.88 mmol, 76%).

¹H-NMR (CDCl₃): δ 12.44 (s, 1H), 7.37 (d, 1H, ³J = 8.5 Hz), 7.31 (d, 1H, ³J = 8.5 Hz), 6.88 (s, 2H), and 3.98 (s, 3H). ¹³CNMR (CDCl₃): δ 190.6, 183.4, 156.6, 154.1, 141.8, 136.5, 126.9, 123.6, 117.4, 114.9, and 57.0. IR (KBr): ν_{max} 2745, 1636, 1620, 1475, 1435, 1250, and 845 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.⁹⁵

Rhamnal **166**.⁹⁶



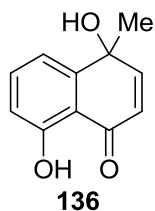
A solution of 3,4-di-O-acetyl-6-deoxy-L-glucal **164** (0.50 mL, 2.6 mmol) in 20 mL of methanol was added 3 drops of sodium methoxide. The reaction was stirred overnight at room temperature. The solvent was removed in vacuo, a white crystal **165** (298 mg, 2.3 mmol, 90%) was given.

A mixture of L-rhamnal **165** (290 mg, 2.23 mmol) and imidazole (570 mg, 8.38 mmol) in 20 mL of DMF was added t-butyl(chloro)dimethylsilane (790 mg, 5.27 mmol) at 0 °C. The reaction was stirred for 18 hours, and then quenched by CuSO₄ solution in

water. The water phase was extracted three times with ether; the organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ether (100:1) to give a lot transparent crystal **166** (715 mg, 2.0 mmol, 90%).

$^1\text{H-NMR}$ (CDCl_3): δ 6.27 (d, 1H, $^3J = 6.0$ Hz), 4.65 (dd, 1H, $^3J = 6.0$ Hz, $^3J = 3.0$ Hz), 4.08 (dd, 1H, $^3J = 5.0$ Hz, $^3J = 4.0$ Hz), 3.92 (dt, 1H, $^3J = 6.5$ Hz, $^3J = 7.0$ Hz), 3.55 (dd, 1H, $^3J = 5.5$ Hz, $^3J = 6.5$ Hz), 1.31 (d, 3H, $^3J = 6.5$ Hz), 0.90 (s, 18H), and 0.09 (m, 12H). IR (KBr): ν_{max} 2955, 2887, 2858, 1251, 1115, and 1073 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.⁹⁶

Quinol **136**.

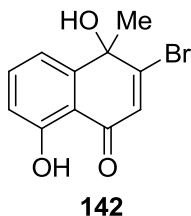


Juglone **133** (103 mg, 0.75 mmol) was dissolved in 20 mL of dry THF. Methylmagnesium bromide (3.0 M in diethyl ether, 0.30 mL, 0.9 mmol) was added to the red solution at -78 °C. After the solution was stirred for eight hours, the reaction was quenched with 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **136** (51 mg, 3.4 mmol, 46%).

$^1\text{H-NMR}$ (CDCl_3): δ 12.26 (s, 1H), 7.44 (bt, 1H, $^3J = 8.0$ Hz), 7.15 (dd, 1H, $^3J = 7.7$ Hz, $^4J = 0.9$ Hz), 6.96 (d, 1H, $^3J = 10.2$ Hz), 6.80 (dd, 1H, $^3J = 8.3$ Hz, $^4J = 0.9$ Hz),

6.16 (d, 1H, $^3J = 10.2$ Hz), 2.88 (s, 1H), and 1.50 (s, 3H). ^{13}C NMR (CDCl_3): δ 190.3, 162.0, 154.8, 149.2, 136.3, 126.6, 117.5, 116.8, 114.1, 68.8, and 31.3. IR (KBr): ν_{max} 3405, 2980, 2920, 1655, 1605, 1455, 1345, 1240, 1050, 850, and 750 cm^{-1} .

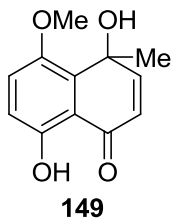
Bromoquinol **142**.



A solution of bromojuglone **134** (155 mg, 0.62 mmol) in 20 mL of dry THF was cooled down to $-93\text{ }^\circ\text{C}$. Methylmagnesium bromide (3.0 M in diethyl ether, 0.25 mL, 0.75 mmol) was added to the red solution at $-93\text{ }^\circ\text{C}$. After the solution was stirred for 40 minutes, the reaction was quenched with 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **142** (68 mg, 26 mmol, yield 41%).

^1H -NMR (CDCl_3): δ 12.05 (s, 1H), 7.52 (dt, 1H, $^3J = 7.5$ Hz, $^3J = 8.5$ Hz), 7.29 (t, 1H, $^3J = 8.0$ Hz, $^3J = 6.0$ Hz), 6.92 (d, 1H, $^3J = 9.0$ Hz), 6.73 (s, 1H), 2.88 (s, 1H), and 1.69 (s, 3H). ^{13}C NMR (CDCl_3): δ 187.0, 161.8, 157.2, 147.7, 136.2, 130.9, 118.2, 117.2, 113.108, 72.4, and 33.0. IR (KBr): ν_{max} 3428, 3418, 1641, 1590, 1485, 1325, and 1240 cm^{-1}

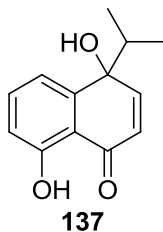
Quinol 149.



A solution of 5-hydroxy-8-methoxy-1,4-naphthalenedione **135** (77 mg, 0.37 mmol) in 25 mL of dry THF was cooled down to $-72\text{ }^{\circ}\text{C}$. Methylmagnesium bromide (3.0 M in diethyl ether, 0.15 mL, 0.45 mmol) was added to the red solution at $-72\text{ }^{\circ}\text{C}$. The solution was stirred for 18 hours. The reaction was quenched with 1 mL of saturated aq. NH_4Cl . The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (20:1) to give a red solid **149** (47 mg, 21 mmol, 57%).

^1H NMR (CDCl_3): δ 12.13 (s, 1H), 7.16 (d, 1H, $^3J = 9.2$ Hz), 7.99 (d, 1H, $^3J = 10.4$ Hz), 6.91 (d, 1H, $^3J = 9.2$ Hz), 6.24 (d, 2H, $^3J = 10.4$ Hz), 4.63 (s, 1H), 3.93 (s, 3H), and 1.68 (s, 1H). ^{13}C NMR (CDCl_3): δ 190.3, 156.7, 154.3, 151.9, 148.9, 134.0, 130.1, 125.6, 119.9, 117.1, 113.8, 69.8, 56.5, and 29.5. ^1H NMR (CDCl_3 , D_2O): δ 12.13 (s, 1H), 7.16 (d, 1H), 7.99 (d, 1H), 6.91 (d, 1H), 6.24 (d, 2H), 3.93 (s, 3H), 1.68 (s, 1H). IR (KBr): ν_{max} 3545, 2965, 2805, 1665, 1570, 1480, 1275, and 1070 cm^{-1} . ESI-MS ($\text{M}-\text{H}^+$): 219.11, and 220.13, g/mol.

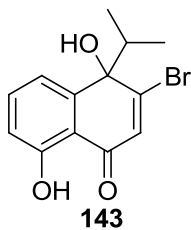
Quinol 137.



A solution of juglone **133** (186.4 mg, 1.1 mmol) in 20 mL of dry THF was cooled down to -78 °C. iso-Propylmagnesium bromide (2.0 M in diethyl ether, 0.7 mL, 1.4 mmol) was added to the red solution at -78 °C. The solution was stirred for eight hours. The reaction was quenched with 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **137** (96 mg, 0.45 mmol, 41%).

¹HNMR (CDCl₃): δ 12.30 (s, 1H), 7.45 (dt, 1H, ³J = 9.2 Hz, ³J = 7.2 Hz), 7.15 (d, 1H, ³J = 10.4 Hz), 6.97 (d, 1H, ³J = 9.2 Hz), 6.92 (d, 1H, ³J = 7.2 Hz), 6.32 (d, 1H, ³J = 10.4 Hz), 2.64 (s, 1H), 2.17 (m, 1H), 1.09 (d, 3H, ³J = 6.8 Hz), and 0.52 (d, 3H, ³J = 6.8 Hz). ¹³CNMR (CDCl₃): δ 189.9, 161.7, 151.4, 148.4, 136.0, 128.9, 117.0, 116.8, 115.2, 74.2, 41.3, 17.4, and 17.0. IR (KBr): ν_{max} 3410, 2925, 1650, 1605, 1460, 1365, 1275, and 1220 cm⁻¹.

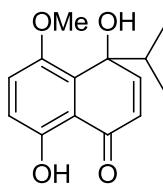
Bromoquinol 143.



A solution of bromojuglone **134** (165 mg, 0.65 mmol) in 20 mL of dry THF was cooled down to -78 °C. iso-Propylmagnesium bromide (2.0 M in diethyl ether, 0.40 mL, 0.8 mmol) was added to the red solution at -78 °C. The solution was stirred for eight hours. The reaction was quenched with 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give the starting material back (37 mg, 0.15 mmol, 23%) and yellow oil **143** (73 mg, 0.24 mmol, 37%).

¹HNMR (CDCl₃): δ 12.01 (s, 1H), 7.47 (d, 1H, ³J = 8.0 Hz), 7.21 (d, 1H, ³J = 8.0 Hz), 6.94 (d, 1H, ³J = 8.0 Hz), 6.86 (s, 1H), 2.69 (s, 1H), 2.33 (m, 1H), 1.00 (d, 3H, ³J = 7.0 Hz), and 0.61 (d, 3H, ³J = 6.5 Hz). ¹³CNMR (CDCl₃): δ 187.5, 161.9, 156.4, 143.6, 134.8, 132.7, 129.9, 119.3, 117.5, 115.2, 78.4, 39.0, 17.6, and 15.7. IR (KBr): ν_{max} 3429, 1639, 1604, 1452, and 1248 cm⁻¹

Quinol **150**.



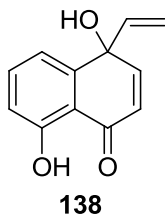
150

A solution of 5-hydroxy-8-methoxy-1,4-naphthalenedione **135** (160 mg, 0.78 mmol) in 25 mL of dry THF was cooled down to -78 °C. iso-Propylmagnesium bromide (2.0 M in diethyl ether, 0.36 mL, 0.72 mmol) was added to the red solution at -78 °C. The solution was stirred overnight. Then the reaction was allowed to warm up to -65 °C and stirred for 2 days. The reaction was quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The

organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **150** (36 mg, 0.15 mmol, 19%).

¹HNMR (CDCl₃): δ 12.15 (s, 1H), 7.15 (d, 1H, ³J = 9.2 Hz), 6.95 (d, 1H, ³J = 16.4 Hz), 6.91 (d, 1H, ³J = 14 Hz), 6.41 (d, 1H, ³J = 10.8 Hz), 4.63 (s, 1H), 3.91 (s, 3H), 2.55 (m, 1H), 1.15 (d, 3H, ³J = 6.8 Hz), and 0.52 (d, 3H, ³J = 7.2 Hz). ¹³CNMR (CDCl₃): δ 190.4, 156.6, 150.5, 133.8, 128.3, 119.9, 117.1, 115.5, 76.1, 56.5, 38.5, 18.1, and 17.4. IR (KBr): ν_{max} 3520, 2965, 1655, 1470, 1220, and 1040 cm⁻¹. ESI-MS (M⁻): 247.15 g/mol.

Quinol 138.

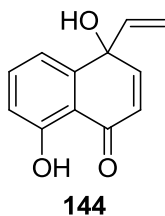


A solution of juglone **133** (85 mg, 0.49 mmol) in 10 mL of dry THF was cooled down to -95 °C. Vinylmagnesium bromide (1.0 M in diethyl ether, 0.60 mL, 0.6 mmol) was added to the red solution at -95 °C. The solution was stirred for half an hour. Then the reaction was allowed to warm up to -78 °C and stirred for 2 hours. The reaction was quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **138** (8 mg, 0.04 mmol, 8%).

¹HNMR (CDCl₃): δ 12.30 (s, 1H), 7.49 (dt, 1H, ³J = 7.0 Hz, ³J = 8.5 Hz), 7.08 (d, 1H, ³J = 7.0 Hz), 6.98 (m, 2H), 6.31 (d, 1H, ³J = 10.5 Hz), 5.80 (dd, 1H, ³J = 17.5 Hz, ³J

= 10.5 Hz), 5.46 (d, 1H, $^3J = 17$ Hz), 5.23 (d, 1H, $^3J = 10.5$ Hz), and 2.51 (s, 1H). ^{13}C NMR (CDCl_3): δ 190.0, 177.5, 162.1, 157.6, 151.5, 139.7, 136.2, 127.0, 118.6, 117.3, 115.2, 114.0, and 71.3. IR (KBr): ν_{max} 3409, 3054, 1653, 1606, 1454, 1345, and 1234cm^{-1}

Bromoquinol **144**.



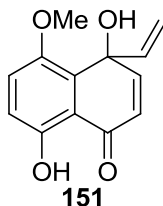
A solution of bromojuglone **134** (102 mg, 0.41 mmol) in 10 mL of dry THF was cooled down to -78 °C. Vinylmagnesium bromide (1.0 M in diethyl ether, 0.50 mL, 0.50 mmol) was added to the red solution at -78 °C. The solution was stirred for eight hours. The reaction was quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give the starting material back (69 mg, 0.27 mmol, 66%) and yellow oil **144** (28.3 mg, 0.1 mmol, 25%).

A solution of bromojuglone **134** (103 mg, 0.41 mmol) in 10 mL of dry THF was cooled down to -78 °C. Vinylmagnesium bromide (1.0 M in diethyl ether, 0.55 mL, 0.55 mmol) was added to the red solution at -78 °C. The solution was stirred for a day. Then the reaction was allowed to warm up to -68 °C for a day. The reaction was quenched with 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a

silica gel column with hexanes and ethyl acetate (15:1) to give the starting material back (61 mg, 0.24 mmol, 59%) and yellow oil **144** (24 mg, 0.085 mmol, 21%).

^1H NMR (CDCl_3): δ 12.04 (s, 1H), 7.50 (dt, 1H, $^3J = 8.0$ Hz, $^3J = 8.0$ Hz), 7.16 (d, 1H, $^3J = 7.5$ Hz), 6.96 (d, 1H, $^3J = 8.5$ Hz), 6.86 (s, 1H), 5.78 (dd, 1H, $^3J = 16.5$ Hz, $^3J = 11.5$ Hz), 5.53 (d, 1H, $^3J = 17$ Hz), 5.33 (d, 1H, $^3J = 10.5$ Hz), and 2.77 (s, 1H). ^{13}C NMR (CDCl_3): δ 187.1, 162.1, 153.7, 144.9, 140.0, 137.2, 136.2, 131.7, 125.7, 119.5, 117.6, 115.9, and 75.0. IR (KBr): ν_{max} 2910, 2845, 1642, 1607, 1453, 1359, and 1250 cm^{-1}

Quinol **151**.

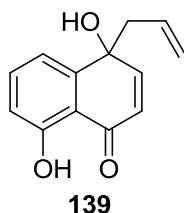


A solution of 5-hydroxy-8-methoxy-1,4-naphthalenedione **135** (80 mg, 39 μmol) in 10 mL of dry THF was cooled down to -78 $^{\circ}\text{C}$. Vinylmagnesium bromide (1.0 M in diethyl ether, 0.5 mL, 0.5 mmol) was added to the red solution at -78 $^{\circ}\text{C}$. The solution was stirred for 2 days and then quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **151** (32 mg, 0.14 mmol, 35%).

^1H NMR (CDCl_3): δ 12.11 (s, 1H), 7.16 (d, 1H, $^3J = 11.5$ Hz), 6.95 (d, 1H, $^3J = 11.5$ Hz), 6.81 (d, 1H, $^3J = 12.5$ Hz), 6.27 (d, 1H, $^3J = 12.5$ Hz), 5.89 (dd, 1H, $^3J = 16.5$ Hz, $^3J = 13$ Hz), 5.33 (d, 1H, $^3J = 16.5$ Hz), 5.20 (d, 1H, $^3J = 13$ Hz), 4.78 (s, 1H), and 3.85 (s, 3H). ^{13}C NMR (CDCl_3): δ 190.4, 156.8, 151.3, 149.2, 140.2, 132.2, 125.9, 120.8,

117.7, 114.9, 113.9, 72.4, and 56.9. IR (KBr): ν_{max} 3435, 1657, 1480, 1395, 1265, 1220, 1160, and 1040 cm^{-1}

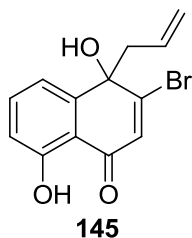
Quinol 139.



A solution of juglone **133** (100 mg, 57 μmol) in 25 mL of dry THF was cooled down to $-78\text{ }^{\circ}\text{C}$. Allylmagnesium bromide (1.0 M in diethyl ether, 0.7 mL, 0.7 μmol) was added to the red solution at $-78\text{ }^{\circ}\text{C}$. The black solution was stirred for eight hours and then quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **139** (24 mg, 11 μmol , 19%).

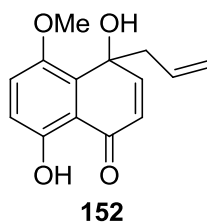
^1H NMR (CDCl_3): δ 12.32 (s, 1H), 7.51 (dt, 1H, $^3J = 16\text{ Hz}$, $^3J = 8.0\text{ Hz}$), 7.19 (d, 1H, $^3J = 8.0\text{ Hz}$), 6.96 (d, 1H, $^3J = 12\text{ Hz}$), 6.90 (d, 1H, $^3J = 8.4\text{ Hz}$), 6.30 (d, 1H, $^3J = 10.4\text{ Hz}$), 5.42 (m, 1H), 5.01 (m, 2H), and 2.61 (m, 2H). ^{13}C NMR (CDCl_3): δ 190.0, 162.1, 153.0, 147.3, 136.3, 136.1, 131.1, 127.8, 120.5, 117.2, 117.2, 114.7, 71.3, and 48.8. IR (KBr): ν_{max} 3420, 3040, 2915, 1740, 1675, 1630, 1420, 1240, and 1150 cm^{-1} .

Bromoquinol 145.



A solution of bromojuglone **134** (158 mg, 0.63 mmol) in 20 mL of dry THF was cooled down to $-78\text{ }^{\circ}\text{C}$. Allylmagnesium bromide (1.0 M in diethyl ether, 0.87 mL, 0.87 mmol) was added to the red solution at $-78\text{ }^{\circ}\text{C}$. The solution was stirred for eight hours and then quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give the starting material back (43 mg, 17 mmol, 0.26%).

Quinol 152.

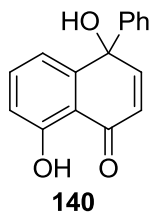


A solution of 5-hydroxy-8-methoxy-1,4-naphthalenedione **135** (110 mg, 0.54 mmol) in 25 mL of dry THF was cooled down to $-78\text{ }^{\circ}\text{C}$. Allylmagnesium bromide (1.0 M in diethyl ether, 0.69 mL, 0.69 mmol) was added to the red solution at $-78\text{ }^{\circ}\text{C}$. The solution was stirred overnight. Then the reaction was allowed to warm up to $-65\text{ }^{\circ}\text{C}$ and stirred for 12 hours. The reaction was quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic

solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **152** (27 mg, 11 mmol, 20%).

^1H NMR (CDCl_3): δ 12.12 (s, 1H), 7.17 (d, 1H, $^3J = 8.8$ Hz), 6.90 (dd, 2H, $^3J = 10$ Hz, $^3J = 8.4$ Hz), 6.32 (d, 1H, $^3J = 10$ Hz), 6.39 (m, 1H), 4.92 (d, 1H, $^3J = 10.4$ Hz), 4.85 (d, 1H, $^3J = 17.2$ Hz), 4.75 (s, 1H), 3.95 (s, 3H), 2.86 (dd, 1H, $^3J = 13.2$ Hz, $^3J = 7.6$ Hz), and 2.72 (dd, 1H, $^3J = 13.2$ Hz, $^3J = 7.2$ Hz). ^{13}C NMR (CDCl_3): δ 190.5, 156.7, 152.6, 148.8, 132.5, 131.8, 127.0, 119.8, 119.5, 117.4, 114.8, 72.8, 56.5, and 46.4. IR (KBr): ν_{max} 3320, 3070, 2980, 2820, 1705, 1662, 1440, and 1340 cm^{-1} . ESI-MS (M-H): 245.14 g/mol.

Quinol 140.

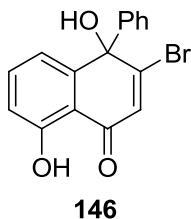


A solution of juglone **133** (167.3 mg, 0.95 mmol) in 25 mL of dry THF was cooled down to -78 $^{\circ}\text{C}$. Phenylmagnesium bromide (3.0 M in diethyl ether, 0.39 mL, 1.2 mmol) was added to the red solution at -78 $^{\circ}\text{C}$. The solution was stirred for eight hours and then quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **140** (61 mg, 0.24 mmol, 25%).

^1H NMR (CDCl_3): δ 12.36 (s, 1H), 7.32 (m, 6H), 6.95 (d, 1H, $^3J = 6.0$ Hz), 6.90 (m, 2H), 6.27 (d, 1H, $^3J = 10$ Hz), and 3.07 (s, 1H). ^{13}C NMR (CDCl_3): δ 190.5, 161.8,

152.9, 147.8, 142.2, 136.5, 128.8, 128.7, 128.0, 125.6, 125.6, 119.7, 117.2, 114.3, and 72.2. IR (KBr): ν_{max} 3435, 3040, 2945, 1650, 1490, 1400, 1360, 1230, and 1145 cm^{-1} .

Bromoquinol **146**.

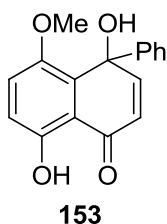


A solution of bromojuglone **134** (136 mg, 0.54 mmol) in 20 mL of dry THF was cooled down to $-78\text{ }^{\circ}\text{C}$. Phenylmagnesium bromide (3.0 M in diethyl ether, 0.2 mL, 0.6 mmol) was added to the red solution at $-78\text{ }^{\circ}\text{C}$. The black solution resulted was stirred for eight hours and then quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **146** (29 mg, 0.09 mmol, 16%) and the starting material (69 mg, 0.27 mmol, 50%).

A solution of bromojuglone **134** (104 mg, 0.41 mmol) in 20 mL of dry THF was cooled down to $-78\text{ }^{\circ}\text{C}$. Phenylmagnesium bromide (3.0 M in diethyl ether, 0.2 mL, 0.6 mmol) was added to the red solution at $-78\text{ }^{\circ}\text{C}$. The solution was stirred for 2 days and then quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give yellow oil **146** (64 mg, 19 mmol, 47%) and the starting material (22 mg, 0.086 mmol, 21%).

^1H NMR (CDCl_3): δ 12.10 (s, 1H), 7.30 (m, 6H), 6.94 (m, 3H), and 3.03 (s, 1H).
 ^{13}C NMR (CDCl_3): δ 187.6, 161.7, 155.3, 147.2, 141.7, 136.5, 131.3, 129.8, 129.8, 128.8, 128.3, 125.2, 120.8, 120.5, 117.3, and 75.8. IR (KBr): ν_{max} 3440, 1640, 1607, 1452, 1340, 1253, 1230, and 1195cm^{-1} .

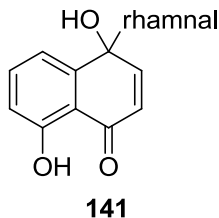
Quinol **153**.



A solution of 5-hydroxy-8-methoxy-1,4-naphthalenedione **135** (100 mg, 0.50 mmol) in 25 mL of dry THF was cooled down to $-78\text{ }^\circ\text{C}$. Phenylmagnesium bromide (3.0 M in diethyl ether, 0.2 mL, 0.6 mmol) was added to the red solution at $-78\text{ }^\circ\text{C}$. The solution was stirred overnight. Then the reaction was allowed to warm up to $-65\text{ }^\circ\text{C}$ and stirred for 36 hours. The reaction was quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes chloride and ethyl acetate (15:1) to give yellow oil **153** (63 mg, 23 mmol, 46%).

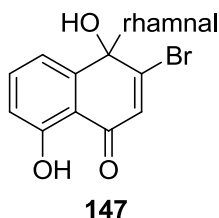
^1H NMR (CDCl_3): δ 12.18 (s, 1H), 7.28 (m, 5H), 7.13 (d, 1H, $^3J = 9.2\text{ Hz}$), 7.00 (d, 1H, $^3J = 9.2\text{ Hz}$), 6.87 (d, 1H, $^3J = 10\text{ Hz}$), 6.20 (d, 1H, $^3J = 10\text{ Hz}$), 4.92 (s, 1H), and 3.46 (s, 3H). ^{13}C NMR (CDCl_3): δ 190.7, 156.7, 152.6, 149.1, 143.7, 134.1, 128.6, 127.9, 125.326, 124.4, 122.0, 118.0, 114.6, 73.2, and 57.1. IR (KBr): ν_{max} 3520, 1665, 1590, 1480, 1435, 1285, 1220, and 1640 cm^{-1} . ESI-MS (M-H $^-$): 281.16, and 283.31 g/mol.

Quinol 141.



A solution of rhamnal **166** (309 mg, 0.86 mmol) in 10 mL of dry THF was added t-BuLi (1.7 M, 0.7 mL, 1.2 mmol) dropwise at -78 °C. After addition, the yellow solution was allowed to warm up to 0 °C and stirred for 100 minutes. The pale yellow solution was cooled down to -78 °C again. Then it was transferred to a solution of juglone **133** (166 mg, 0.95 mmol) in 5 mL of dry THF by a cannula at -78 °C. The reaction was stirred for 24 hours at -78 °C and 24 hours at -70 °C to form a dark green mixture. The reaction was quenched by 1 mL of water. The organic phase was separated; the water phase was extracted three times with diethyl ether. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give the starting juglone (96 mg, 55 mmol, 58%) back.

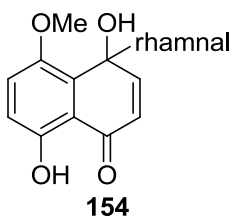
Bromoquinol 147.



A solution of rhamnal **166** (355 mg, 1 mmol) in 10 ml of dry THF was added t-BuLi (1.7 M, 0.8 mL, 1.4 mmol) dropwise at -78 °C. After addition, the yellow solution was allowed to warm up to 0 °C and stirred for 2 hours. The pale yellow solution was cooled down to -78 °C again. Then it was transferred to a solution of bromojuglone **134** (131 mg, 0.52 mmol) in 5 mL of dry THF by a cannula at -78 °C. The reaction was

stirred for 24 hours at -78 °C and 24 hours at -70 °C to form a dark green mixture. The reaction was quenched by 1 mL of water. The organic phase was separated; the water phase was extracted three times with ether. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (15:1) to give the starting juglone (82 mg) back.

Quinol **154**.

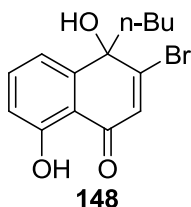


A solution of rhamnal **166** (370 mg, 1.03 mmol) in 15 mL of dry THF was added t-BuLi (1.7 M, 1.4 mL, 2.38 mmol) dropwise at -78 °C. After addition, the yellow solution was allowed to warm up to 0 °C and stirred for 2 hours. The pale yellow solution resulted was cooled down to -78 °C again. Then it was transferred to a solution of 5-hydroxy-8-methoxy-1,4-naphthalenedione **135** (135 mg, 0.661 mmol) in 10 mL of dry THF by a cannula at -78 °C. The reaction was stirred for 12 hours at -78 °C and 48 hours at -60 °C to form a dark green mixture. The reaction was quenched by 1 mL of water. The organic phase was separated; the water phase was extracted three times with ether. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (10:1) to give a yellow solid **154** (190 mg, 0.33 mmol, 51%).

¹HNMR (CDCl₃): δ 12.07 (s, 1H), 7.11 (dd, 1H, ³J = 6.8 Hz, ³J = 3.0 Hz), 6.93 (dd, 1H, ³J = 9.0 Hz, ³J = 2.5 Hz), 6.75 (d, 1H, ³J = 9.5 Hz), 6.29 (dd, 1H, ³J = 10 Hz, ³J = 4.0 Hz), 5.18 (d, 1H, ³J = 4.0 Hz), 4.90 (s, 1H), 3.93 (d, 1H, ³J = 3.5 Hz), 3.82 (s, 3H),

3.51 (s, 1H), 1.08 (s, 3H), 0.85 (m, 18H), and 0.10 (m, 12H). ^{13}C NMR (CDCl_3): δ 190.9, 156.8, 153.0, 152.5, 150.4, 149.2, 132.0, 127.3, 121.7, 120.2, 117.8, 115.3, 98.0, 76.0, 73.8, 72.1, 68.9, 67.9, 58.1, 57.0, 26.0, 18.2, 16.1, 15.5, and -4.2. IR (KBr): ν_{max} 3505, 2945, 2810, 1665, 1610, 1470, 1250, 1105, and 1060 cm^{-1} . ESI-MS (M-H): 561.26, 562.33, and 563.31 g/mol.

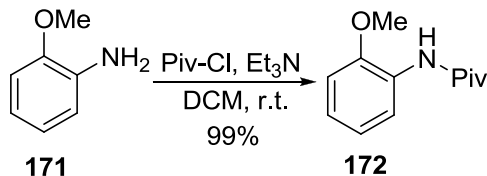
Bromophenol **148**.



A solution of bromojuglone **134** (104 mg, 0.41 mmol) in 20 mL of dry THF was cooled down to $-93\text{ }^\circ\text{C}$. n-Butyl lithium (1.6 M in diethyl ether, 0.30 mL, 0.48 mmol) was added to the red solution at $-93\text{ }^\circ\text{C}$. The black solution resulted was stirred for half an hour and then quenched by 0.4 mL of water. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with methylene chloride and ethyl acetate (15:1) to give yellow oil **148** (52 mg, 19 mmol, yield 46%).

^1H NMR (CDCl_3): δ 12.10 (s, 1H), 7.53 (dt, 1H, $^3J = 8.0\text{ Hz}$, $^3J = 8.0\text{ Hz}$), 7.27 (s, 1H), 6.95 (d, 1H, $^3J = 8.5\text{ Hz}$), 6.87 (s, 1H), 2.56 (s, 1H), 2.22 (m, 1H), 1.91 (m, 1H), 1.17 (m, 2H), 0.93 (m, 2H), and 0.77 (t, 3H, $^3J = 7.0\text{ Hz}$). ^{13}C NMR (CDCl_3): δ 188.9, 163.8, 157.6, 147.9, 137.5, 134.2, 119.7, 118.6, 116.1, 77.1, 45.8, 27.3, and 23.9. IR (KBr): ν_{max} 3393, 2998, 1642, 1608, 1453, 1362, 1251, and 1040 cm^{-1} .

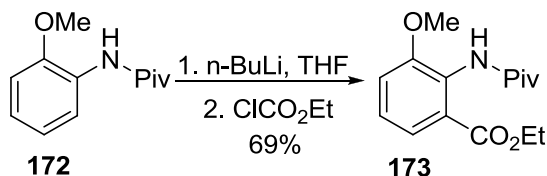
Protected anisidine **172**.¹⁰¹



A solution of o-anisidine **171** (21 g, 170 mmol) and trimethylamine (12 g, 1.2 mmol) in methylene chloride was cooled down to 0 °C. Pivaloyl chloride (15 g, 1.25 mmol) was added to the mixture at 0 °C. The reaction was stirred overnight at room temperature and then quenched by 1M HCl. The organic phase was separated; the water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (5:1) to give a lot crystal **172** (35 g, 99%).

¹HNMR (CDCl₃): δ 7.53 (d, 1H, ³J = 7.2 Hz), 6.89 (dd, 1H, ³J = 8.4 Hz), 6.77 (m, 2H), 3.73 (s, 3H), and 1.24 (s, 9H). IR (KBr): ν_{max} 3442, 2961, 2870, 1680, 1601, 1522, 1460, 1334, 1249, and 1160 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹⁰¹

Amino ester **173**.¹⁰²

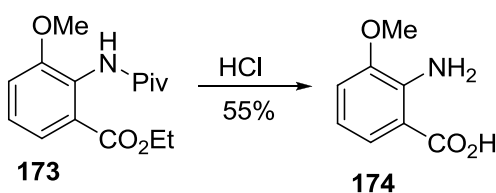


n-BuLi (1.6 M, 11.7 mL, 18.7 mmol) was added to a solution of the protected anisidine **172** (1.53 g, 7.3 mmol) in THF (20 mL) at room temperature. The mixture was stirred for 2 hours at room temperature to give a lot yellow solid. The reaction was allowed to cool down to 0 °C and then quenched by ethyl chloroformate (1.7 mL). After the solution was stirred for 2 hours at room temperature, it was poured into sat. NH₄Cl solution. The organic phase was separated; the water phase was extracted three times by

methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (6:1) to give a crystal **173** (1.43g, 12.9 mmol, 69%).

¹HNMR (CDCl₃): δ 7.55 (dd, 1H, ³J = 8.0 Hz, ⁴J = 1.6 Hz), 7.30 (dd, 2H, ³J = 8.0 Hz, ⁴J = 1.6 Hz), 7.06 (dd, 2H, ³J = 8.4 Hz, ⁴J = 1.6 Hz), 4.10 (q, 2H, ³J = 7.2 Hz), 3.77 (s, 3H), and 1.11 (t, 3H, ³J = 7.2 Hz). ¹³CNMR (CDCl₃): δ 183.4, 165.4, 129.2, 128.4, 122.7, 120.7, 115.4, 111.8, 62.5, 61.1, 56.0, 27.7, and 14.1. IR (KBr): ν_{max} 2967, 2909, 2841, 1794, 1584, 1474, 1393, 1263, and 1059 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹⁰²

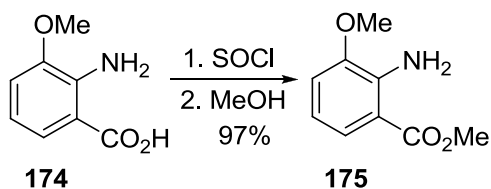
Amino acid **174**.



Amino ester **173** (700 mg, 2.5 mmol) was added a mixture of concentrated hydrochloric acid (12 mL) and water (3 mL). The mixture was refluxed overnight and then quenched by sat. NaHCO₃ solution. The water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (3:1) to give oil **174** (230 mg, 1.38 mmol, 55%).¹⁰²

¹HNMR (CDCl₃): δ 7.55 (dd, 1H, ³J = 8.4 Hz, ⁴J = 1.2 Hz), 6.88 (d, 1H, ³J = 7.2 Hz), 6.60 (dt, 1H, ³J = 8.0 Hz, ³J = 1.2 Hz), and 3.88 (s, 3H). ¹³CNMR (CDCl₃): δ 173.7, 147.2, 142.7, 123.4, 114.9, 113.7, 109.0, and 55.9. IR (KBr): ν_{max} 3455, 2960, 1680, 1595, 1465, 1410, and 1270 cm⁻¹.

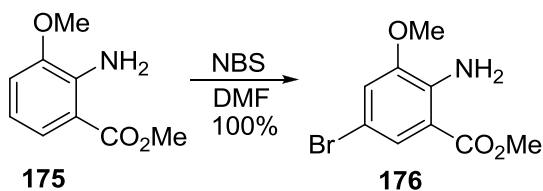
Amino ester **175**.¹⁰²



Amino acid **174** (119 mg, 0.71 mmol) was added by thionyl chloride (5 mL). The mixture was refluxed for 3 hours. Then extra thionyl chloride was removed in vacuo. The residues were added dry methanol (25 mL). The reaction was stirred overnight and quenched by water. The water phase was extracted three times by methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (5:1) to give a crystal **175** (125 mg, 0.69 mmol, 97%).

¹HNMR (CDCl₃): δ 7.47 (d, 1H, ³J = 8.0 Hz), 6.85 (d, 1H, ³J = 7.6 Hz), 6.58 (dt, 1H, ³J = 8.0 Hz, ³J = 8.0 Hz), and 3.87 (s, 6H). ¹³CNMR (CDCl₃): δ 185.4, 173.9, 147.2, 142.7, 123.4, 114.9, 113.7, 109.2, 55.9, 38.8, 27.2, and 22.8. IR (KBr): ν_{max} 3455, 2960, 1680, 1595, 1465, 1410, and 1270 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹⁰²

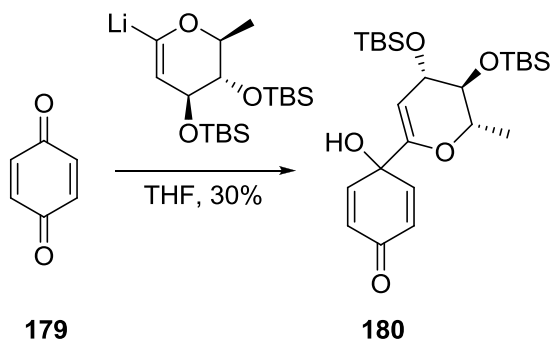
Bromobenzoic ester **176**.



Amino ester **175** (100 mg, 0.55 mmol) and NBS (111 mg, 0.66 mmol) were dissolved in DMF (5 mg) at 0 °C. Then the mixture was stirred for 4 hours at room temperature. The mixture was washed with a copper sulfate solution. The water phase was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (5:1) to give a crystal **176** (144 mg, 0.55 mmol, 100%).⁹⁸

^1H NMR (CDCl_3): δ 7.59 (d, 1H, $^3J = 1.5$ Hz), 6.89 (d, 1H, $^3J = 2.0$ Hz), 6.01 (br, 2H), 3.85 (s, 3H), and 3.84 (s, 3H). ^{13}C NMR (CDCl_3): δ 167.8, 147.9, 141.1, 124.6, 116.1, 105.9, 56.1, and 51.8. IR (KBr): ν_{max} 3497, 1695, 1547, 1477, 1439, 1301, 1264, and 1230 cm^{-1} .

Quinol **180**.⁹⁶

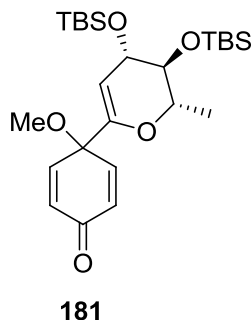


A solution of rhamnol **166** (954 mg, 2.7 mmol) in 30 mL of dry THF was added *t*-BuLi (1.7 M, 3.1 mL, 5.2 mmol) dropwise at -78 °C. The yellow solution was allowed to warm up to 0 °C. After the pale yellow solution was stirred for 2 hours at 0 °C, it was cooled down to -78 °C; and then it was transferred to a solution of quinone **179** (660 mg, 0.661 mmol) in 30 mL of dry THF at -78 °C by a cannula. The reaction was stirred overnight at -78 °C and then quenched by 1 mL of water. The organic phase was separated; the water phase was extracted three times with diethyl ether. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (7:3) to give a solid **180** (366 mg, 0.81 mmol, 30%).

^1H NMR (CDCl_3): δ 6.82 (m, 2H), 6.20 (m, 2H), 4.94 (d, 1H, $^3J = 4.0$ Hz), 4.12 (m, 1H), 3.98 (dt, 1H, $^3J = 1.0$ Hz, $^3J = 1.0$ Hz), 3.58 (dd, 1H, $^3J = 5.0$ Hz, $^3J = 3.5$ Hz), 2.98 (s, 1H), 1.31 (d, 1H, $^3J = 7.0$ Hz), 0.86 (s, 18H), and 0.80 (m, 12H). ^{13}C NMR (CDCl_3): δ 185.5, 150.0, 147.8, 131.5, 100.1, 74.0, 52.5, 26.1, 18.1, 16.4, -3.8 , -4.00 , and -4.3 . IR (KBr): ν_{max} 3315, 2954, 2886, 1602, 1514, 1255, 1169, and 1098 cm^{-1} . The

spectroscopic and other physical data reported are consistent with values reported in the literature.⁹⁶

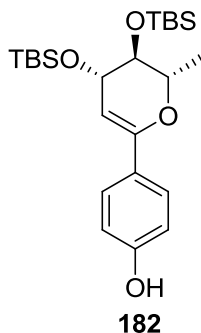
Methylated quinol **181**.



A solution of quinol **180** (230 mg, 0.5 mmol) in methylene chloride was added silver oxide (360 mg) and iodomethane (400 mg). The reaction was stirred for 5 hours at 40 °C. The solid was filtered off; the filtrate was concentrated in vacuo to give oil **181** (194 mg, 0.4 mmol, 80%).¹⁰³

¹HNMR (CDCl₃): δ 6.76 (m, 2H), 6.36 (m, 2H), 5.05 (d, 1H, ³J = 3.9 Hz), 4.02 (m, 1H), 3.54 (dd, 1H, ³J = 4.8 Hz, ³J = 4.2 Hz), 3.28 (s, 3H), 1.25 (d, 1H, ³J = 6.6 Hz), 0.84 (m, 18H), and 0.61 (m, 12H). ¹³CNMR (CDCl₃): δ 185.6, 150.0, 148.3, 147.8, 143.2, 131.5, 128.8, 100.1, 76.2, 74.9, 74.0, 69.0, 52.6, 26.0, 18.1, 16.6, and -3.9. IR (KBr): ν_{max} 3432, 2953, 2889, 1669, 1472, 1386, 1254, 1105, and 1069 cm⁻¹.

Quinol **182**.⁹⁹



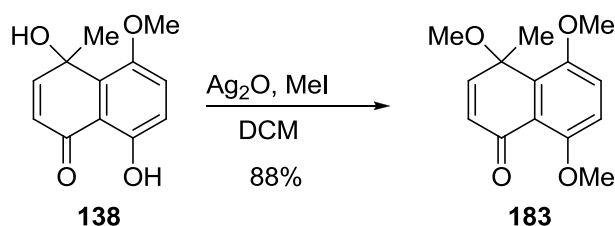
(a) Aluminum was merged in a 2% solution of mercury(II) chloride in water for 10 sec. The aluminum was cut into small pieces. Methylated quinol **181** (110 mg, 0.24) in

a mixture of 2 mL of THF and 0.3 mL of water was added by mercury-aluminum amalgam (60 mg). The reaction was refluxed for 2 hours. The solid was filtered off; the filtrate was concentrated in vacuo to give oil (93mg, 0.22 mmol, 91%)

(b) Quinol **180** (104 mg, 0.23 mmol) in a mixture of 2 mL of THF and 0.3 mL of water was added by the mercury-aluminum amalgam (60 mg). The reaction was refluxed for 2 hours. The solid was filtered off; the filtrate was concentrated in vacuo to give oil **182** (97mg, 0.23 mmol, 99%)

^1H NMR (CDCl_3): δ 7.44 (dd, 2H, $^3J = 6.8$ Hz, $^4J = 2.0$ Hz), 6.79 (dd, 2H, $^3J = 6.8$ Hz, $^4J = 2.0$ Hz), 5.10 (d, 1H, $^3J = 3.6$ Hz), 4.30 (dd, 1H, $^3J = 5.2$ Hz, $^4J = 3.6$ Hz), 4.08 (dd, 1H, $^3J = 6.8$ Hz, $^3J = 6.8$ Hz), 3.64 (dd, 1H, $^3J = 6.8$ Hz, $^3J = 5.2$ Hz), 1.42 (d, 3H, $^3J = 6.4$ Hz), 0.86 (m, 18H), and 0.18 (m, 12H). ^{13}C NMR (CDCl_3): δ 185.9, 150.3, 148.6, 148.3, 128.7, 128.3, 116.3, 98.9, 76.6, 69.3, 68.7, 25.9, 18.1, 16.5, and -4.0. IR (KBr): ν_{max} 3396, 2930, 2888, 1669, 1512, 1472, 1388, 1255, 1104, and 1071 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.⁹⁹

Quinol **183**.

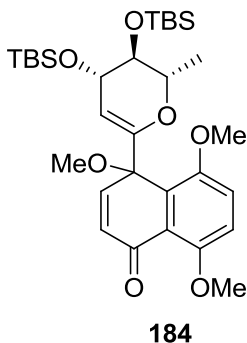


A solution of quinol **138** (30 mg, 0.14 mmol) in methylene chloride was added silver oxide (320 mg) and iodomethane (0.40 mL). The reaction was stirred for 24 hours at 40 °C. The solid was filtered off; the filtrate was concentrated in vacuo to give the methylation product **183** (26mg, 0.12 mmol, 88%)¹⁰³

^1H NMR (CDCl_3): δ 7.09 (d, 1H, $^3J = 9.2$ Hz), 6.93 (d, 1H, $^3J = 9.2$ Hz), 6.56 (d, 1H, $^3J = 10.4$ Hz), 6.28 (d, 1H, $^3J = 10.4$ Hz), 3.84 (s, 3H), 3.82 (s, 3H), 2.94 (s, 3H), and

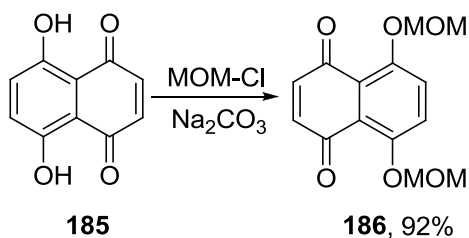
1.67 (s, 3H). ^{13}C NMR (CDCl_3): δ 150.8, 130.5, 117.8, 113.2, 74.3, 56.8, 56.7, 52.7, and 26.0. IR (KBr): ν_{max} 2930, 2865, 1680, 1490, 1277, and 1092 cm^{-1} .

Attempt at quinol **184**.



A solution of quinol **153** (60 mg, 0.11 mmol) in methylene chloride was added silver oxide (200 mg) and iodomethane (0.30 mL). The reaction was stirred for 5 hours at 40 °C. The solid was filtered off; the filtrate was concentrated in vacuo. The NMR of crude product showed that the starting material decomposed.

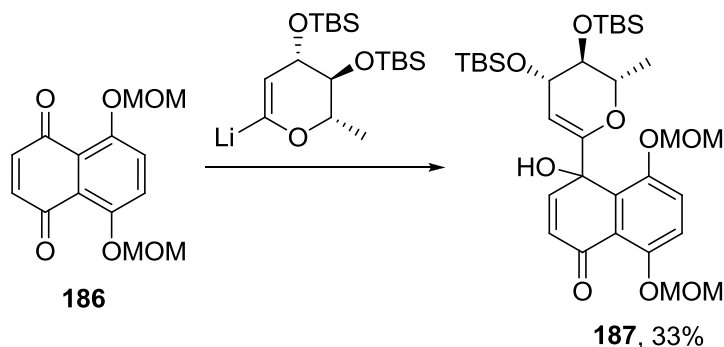
Naphthylquinone **186**.



A solution of naphthylquinone **185** (200 mg, 1.05 mmol) in 10 mL of methylene chloride was added 300 mg of sodium carbonate and 200 mg of chloromethylmethyl ether at 0 °C. The reaction was stirred overnight at 0 °C. The solid was filtered off; the filtrate was concentrated in vacuo to give the MOM protected product (260 mg, 0.93 mmol, 92%).

^1H NMR (CDCl_3): δ 7.43 (s, 2H), 6.73 (s, 2H), 5.21 (s, 4H), and 3.49 (s, 6H). ^{13}C NMR (CDCl_3): δ 184.5, 152.2, 138.4, 125.2, 96.0, and 56.6. IR (KBr): ν_{max} 3384, 2935, 2836, 1736, 1612, 1586, 1463, 1363, 1247, and 1175 cm^{-1} .

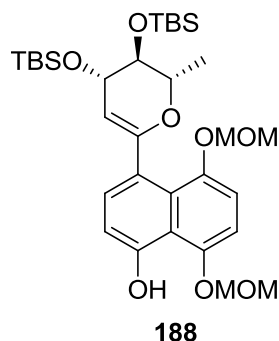
Naphthylquinol **187**.



A solution of rhamnal **166** (370 mg, 1.03 mmol) in 15 mL of dry THF was added *t*-BuLi (1.7 M, 1.4 mL, 2.38 mmol) dropwise at -78 °C. After addition, the yellow solution was allowed to warm up to 0 °C and stirred for 2 hours. The pale yellow solution was cooled down to -78 °C again. Then it was transferred to a solution of naphthalenedione **187** (260 mg, 0.94 mmol) in 10 mL of dry THF by a cannula at -78 °C. The reaction was stirred for 12 hours at -78 °C and 48 hours at -60 °C to form a dark green mixture. The reaction was quenched by 1 mL of water. The organic phase was separated; the water phase was extracted three times with ether. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (10:1) to give a yellow solid (196 mg, 0.31 mmol, 33%).

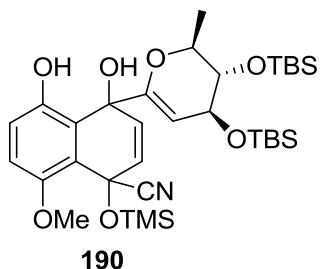
¹HNMR (CDCl₃, diastereomers): δ 7.30 (m, 1H), 7.14 (m, 1H), 6.58 (m, 1H), 6.23 (m, 1H), 5.23 (s, 1H), 5.17 (s, 1H), 5.06 (m, 1H), 4.99 (s, 1H), 4.95 (s, 1H), 3.91 (m, 2H), 3.54 (m, 1H), 3.51 (d, 6H), 1.05 (q, 3H), 0.83 (q, 18H), and 0.00 (q, 12H). ¹³CNMR (CDCl₃): δ 184.6, 153.1, 150.0, 145.9, 135.0, 129.0, 122.6, 120.5, 119.1, 98.4, 96.4, 76.3, 75.6, 74.3, 74.0, 71.8, 70.0, 67.9, 56.5, 26.1, 25.9, 18.1, -4.0, and -4.5. IR (KBr): ν_{max} 3399, 2954, 2856, 1715, 1597, 1463, 1254, and 1102 cm⁻¹.

Attempt at naphthol 188.



Aluminum was merged in a 2% solution of mercury(II) chloride in water for 10 sec. The mercury-aluminum amalgam was cut into small pieces. A solution of naphthylquinol **187** (50 mg, 0.078 mmol) in a mixture of 2 mL of THF and 0.3 mL of water was added the mercury-aluminum amalgam (30 mg). The reaction was refluxed for 2 hours. The solid was filtered off; the filtrate was concentrated in vacuo. The NMR of crude product showed that the desired product was not obtained.

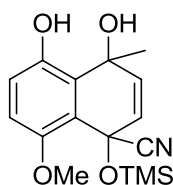
Attempt at naphthylquinol 190.



t-Butyl lithium (1mL, 1.7 M, 1.7 mmol) was added to a solution of 420 mg (1.17 mmol) of TBS protected rhamnol **166** in THF at -78 °C. The mixture was allowed to warm up to room temperature and it was stirred for 2 hours to give a yellow solution. 0.350 mL of trimethylsilyl cyanide was added to a solution of 103 mg (0.51 mmol) of quinone **135** in 5 mL of acetonitrile under argon. The reaction was stirred for an hour. The solution was concentrated in vacuo. The residues were dissolved in 50 mL of THF. The solution was added the lithiated rhamnol solution prepared in the first step at -78 °C. The reaction was stirred for 24 hours at -78 °C and then it was quenched by water. The water solution was extracted three times with diethyl ether. The organic phases were

combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 10:1 hexanes and ethyl acetate. The desired product was not obtained.

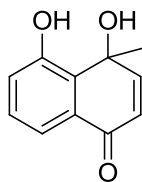
Attempt at naphthylquinol **191**.



191

0.300 mL of trimethylsilyl cyanide was added to a solution of 88 mg (0.44 mmol) of quinone **135** in 25 mL of acetonitrile under argon at 0 °C. The reaction was stirred for an hour. The solution was concentrated in vacuo. The residues were dissolved in 20 mL of THF. The solution was added by 0.400 mL of 3 M methylmagnesium bromide (1.2 mmol) at -78 °C. The reaction was stirred for 24 hours at -78 °C and then it is quenched with water. The water solution was extracted three times with diethyl ether. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 10:1 hexanes and ethyl acetate. The desired product was not obtained.

Attempt at quinol **192**.



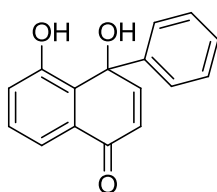
192

(a) 23 mg (1 mmol) of sodium hydride was added to a solution of 174 mg (1 mmol) of juglone in 10 mL of THF. 0.070 mL of titanium chloride was added to the black solution. Some solid appeared. The reaction was stirred for half an hour at -30 °C. Then the reaction was cooled down to -78 °C. 1.2 mL of 1.6 M (1.9 mmol) methyl lithium was added to the reaction. The reaction was tracked by TLC. After 4 hours, there

was not reaction occurred. 3.6 mL of 1.6 M (5.8 mmol) methyl lithium was added. The reaction was stirred overnight. The TLC shown there was still much starting material in the reaction. The reaction was quenched with water. The water solution was extracted three times with ethyl acetate. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 15: 1 hexanes and ethyl acetate to give starting material back.

(b) 0.06 mL of titanium chloride was added to a solution of 174 mg (1.0 mmol) of juglone in THF. The reaction turned black. The reaction was stirred for 1 hour at room temperature. The solution was cooled down to $-78\text{ }^\circ\text{C}$. 10 equivalents of methylmagnesium bromide were added to the reaction. The reaction became dark yellow. The reaction was allowed to warm up to room temperature and stirred for 24 hours. The reaction was quenched by water. The water solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 15:1 hexanes and ethyl acetate to give starting material back.

Attempt at quinol 193

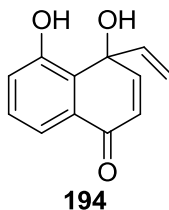


193

1.5 mL (1.5 mmol) of 1M phenyl lithium was added to a solution of 246 mg (1 mmol) of CeCl_3 in 5 mL of THF at $-78\text{ }^\circ\text{C}$. Then, the solution was allowed to warm up to room temperature and stirred for 2 hours to give a black solution. The solution was cooled down to $-78\text{ }^\circ\text{C}$. A solution of 87 mg (0.5 mmol) of juglone in 5 mL of THF was added to the reaction. The reaction was stirred overnight and then it was warmed up to $0\text{ }^\circ\text{C}$. After the reaction was stirred for 2 more hours at $0\text{ }^\circ\text{C}$, it was quenched with sat.

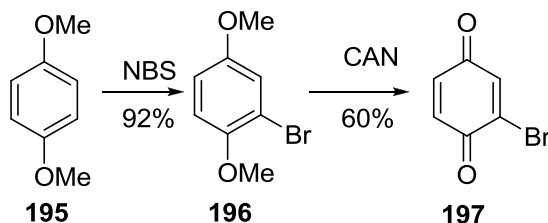
NH₄Cl solution. The water solution was extracted three times with diethyl ether. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 10:1 hexanes and ethyl acetate to give 74 mg of starting material.

Attempt at quinol 194.



1.5 mL (1.5 mmol) of 1M vinylmagnesium bromide was added to a solution of 298 mg (1.2 mmol) of dry CeCl₃ in 8 mL of THF at -78 °C. Then, the solution was warmed up to -20 °C and stirred for 2 hours to give a brown solution. The solution was cooled down to -78 °C. A solution of 87 mg (0.5 mmol) of juglone in 1 mL of THF was added to the reaction. The reaction was stirred overnight and then it was warmed up to 0 °C. After the reaction was stirred for 2 more hours at 0 °C, it was quenched by sat. NH₄Cl solution. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the organic solutions were dried, combined and concentrated. The residues were separated by a silica gel column with 10:1 hexanes and ethyl acetate to give 77 mg of starting material back.

Bromoquinone 197.¹⁰⁸



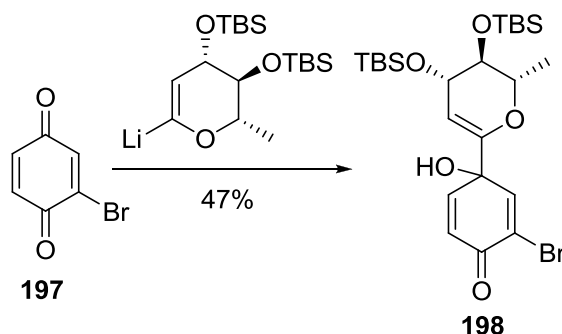
84 g (0.61 mol) of 1,4-dimethoxybenzene and 114 g (0.64 mmol) of N-bromosuccinimide were dissolved in 700 mL methylene chloride at room temperature. The mixture was stirred for 48 hours at room temperature. The solution was washed with a solution of sodium thiosulfate in water. The organic solution was separated and dried

over MgSO_4 . After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 30:1 hexanes and ethyl acetate to give 122 g of compound **196** (0.56 mmol, 92%).

In a 100 mL round bottom flask, 30 g of silica gel was added a solution of 11.25 g (20.5 mmol) of ammonium cerium (IV) nitrate in 2.5 mL of water. The powder was stirred to a free flowing yellow solid. A solution of 1.86 g (8.6 mmol) of 2-bromo-1,4-dimethoxybenzene **196** in 10 mL of methylene chloride was added to the solid. The reaction was stirred for 15 minutes. The solid was filtered off; the filtrate was concentrated in vacuo. The residues were separated by a silica gel column with 10:1 hexanes and ethyl acetate to give 1.10 g of compound **197** (7.8 mmol, 90%).

$^1\text{H NMR}$ (CDCl_3): δ 7.30 (d, 1H, $^4J = 2.4$ Hz), 6.96 (d, 1H, $^3J = 20.0$ Hz), and 6.82 (dd, 1H, $^3J = 20.0$ Hz, $^4J = 2.4$ Hz). $^{13}\text{C NMR}$ (CDCl_3): δ 184.7, 179.3, 138.3, 137.7, 136.8, and 136.0. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹⁰⁸

Quinol **198**.

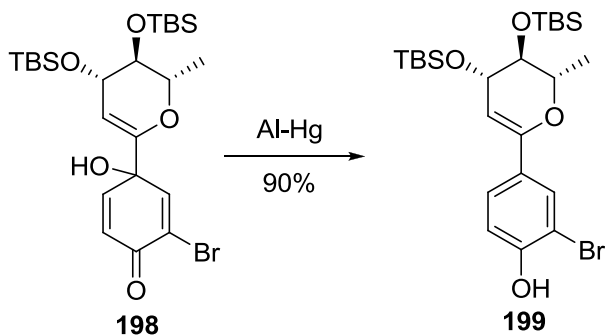


A solution of rhamnal **166** (622 mg, 2.06 mmol) in 1 mL of dry THF was added *t*-BuLi (1.7 M, 2.8 mL, 4.8 mmol) dropwise at -78 °C. After addition, the yellow solution was allowed to warm up to 0 °C and it was stirred for 2 hours. The pale yellow solution resulted was cooled down to -78 °C again. Then it was transferred to a solution of bromoquinone **197** (711 mg, 3.82 mmol) in 10 mL of dry THF by a cannula at -78 °C. The reaction was stirred for 4 hours at -78 °C to form a dark green mixture. The reaction

was quenched by 1 mL of water. The organic phase was separated; the water phase was extracted three times with ether. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (10:1) to give a yellow solid **198** (723 mg, 1.3 mmol, 63%).

Mixture of diastereomers $^1\text{H NMR}$ (CDCl_3): δ 7.25 (m, 1H), 6.85 (m, 1H), 6.30 (m, 1H), 5.10 (m, 1H), 4.09 (m, 1H), 3.98 (m, 1H), 3.57 (m, 1H), 1.26 (m, 3H), 0.863 (m, 18H), and 0.06 (m, 12H).

Phenol **199**.

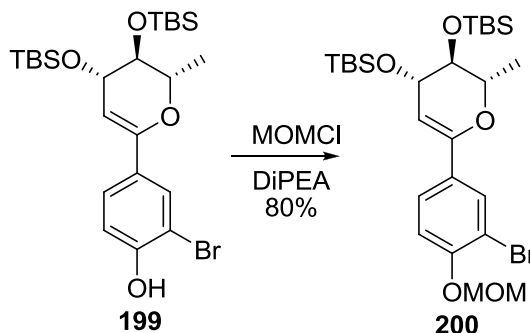


Aluminum was merged in a 2% solution of mercury(II) chloride in water for 10 sec. The mercury-aluminum amalgam was cut into small pieces. Quinol **187** (364 mg, 0.34 mmol) in a mixture of 2 mL of THF and 0.3 mL of water was added mercury-aluminum amalgam (130 mg). The reaction was stirred for 2 hours at room temperature. The solid was filtered off; the filtrate was concentrated in vacuo. The residue was separated by a silica gel column with hexanes and ethyl acetate (5:1) to give a yellow solid **199** (314 mg, 0.31 mmol, 90%).

$^1\text{H NMR}$ (CDCl_3): δ 8.14 (d, 1H, $^4J = 2.0$ Hz), 7.87 (dd, 1H, $^3J = 8.8$ Hz, $^4J = 2.0$ Hz), 7.05 (d, 1H, $^3J = 8.8$ Hz), 4.60 (dd, 1H, $^3J = 8.0$ Hz, $^3J = 4.0$ Hz), 3.98 (m, 1H), 3.48 (dd, 1H, $^3J = 8.8$ Hz, $^3J = 4.0$ Hz), 3.35 (dd, 1H, $^3J = 16.0$ Hz, $^3J = 8.8$ Hz), 1.25 (d, 3H, $^3J = 6.0$ Hz), 0.86 (d, 18H), and 0.10 (d, 12H). $^{13}\text{C NMR}$ (CDCl_3): δ 196.2, 157.1, 133.3, 131.8, 129.9, 116.1, 110.9, 81.2, 75.5, 73.5, 70.0, 25.9, 25.8, 21.0, -4.0, -4.4, -4.5, and -

4.7. IR (KBr): ν_{max} 3391, 2955, 2895, 1733, 1683, 1596, 1496, 1361, 1290, 1170, and 1055 cm^{-1} .

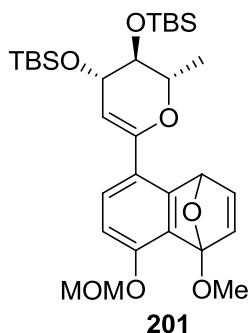
MOM-phenol **200**.



A solution of phenol **199** (59 mg, 0.11 mmol) in 2 mL of methylene chloride was added 30.8 mg of diisopropylethylamine and 11.8 mg (0.15 mmol) of chloromethylmethyl ether at 0 °C. The reaction was stirred overnight at 0 °C. The solution was concentrated in vacuo. The residue was separated by a silica gel column with hexanes and ethyl acetate (5:1) to give the MOM protected compound **200** (52mg, 0.088 mmol, 80%).

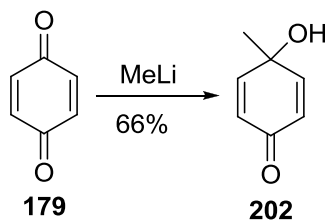
^1H NMR (CDCl_3): δ 7.76 (d, 1H, $^4J = 2.4$ Hz), 7.44 (dd, 1H, $^3J = 8.8$ Hz, $^4J = 2.4$ Hz), 7.09 (d, 1H, $^3J = 8.8$ Hz), 5.25 (s, 2H), 5.12 (d, 1H, $^3J = 3.2$ Hz), 4.25 (dd, 1H, $^3J = 4.0$ Hz, $^3J = 4.0$ Hz), 4.10 (dq, 1H, $^3J = 6.4$ Hz, $^3J = 6.4$ Hz), 3.63 (dq, 1H, $^3J = 6.4$ Hz, $^3J = 4.0$ Hz), 3.51 (s, 1H), 1.41 (d, 3H, $^3J = 6.4$ Hz), 0.93 (s, 9H), 0.91 (s, 9H), 0.14 (s, 6H), and 0.11 (s, 6H). ^{13}C NMR (CDCl_3): δ 153.9, 149.6, 130.4, 115.7, 112.8, 99.1, 95.2, 75.9, 75.0, 70.7, 56.5, 26.2, 26.1, 18.4, 18.2, 17.3, -3.4, and -3.7. IR (KBr): ν_{max} 2927, 2854, 1713, 1650, 1463, 1257, 1140, and 1088 cm^{-1} .

Attempt at MOM-phenol **201**.



1.5 mL of 2M LDA (3 mmol) was added to a solution of 23 mg (0.04 mmol) of MOM-phenol **200** in 2 mL of THF at $-78\text{ }^{\circ}\text{C}$. After the reaction was stirred for 1 hour at $-78\text{ }^{\circ}\text{C}$, it was allowed to warm up to room temperature. The reaction was stirred for 24 hours and then quenched with water. The organic phase was separated; the water phase was extracted three times with ether. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (5:1) to give the starting material back.

p-Methylquinol **202**.¹⁰⁹

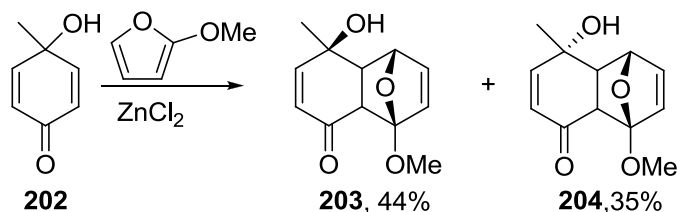


4.7 mL (8 mmol) of 1.7 M solution of methyl lithium in diethyl ether was added to a solution of 660 mg (6.1 mmol) of benzoquinone **179** in THF at $-95\text{ }^{\circ}\text{C}$. After the reaction was stirred for an hour, it was quenched by water. The organic phase was separated; the water phase was extracted three times with diethyl ether. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a silica gel column with hexanes and ethyl acetate (5:1) to give 500 mg of compound **202** (4.0 mmol, 66 %).

^1H NMR (CDCl_3): δ 6.81 (d, 2H, $^3J = 10.5\text{ Hz}$), 7.87 (d, 2H, $^3J = 10.5\text{ Hz}$), 4.07 (s, 1H), and 1.37 (s, 3H). ^{13}C NMR (CDCl_3): δ 186.0, 153.3, 126.6, 67.0, and 26.8. IR (KBr):

ν_{max} 3387, 2978, 1666, 1620, 1396, 1306, 1246, 1177, and 1092 cm^{-1} . The spectroscopic and other physical data reported are consistent with values reported in the literature.¹⁰⁹

Compound **203** and **204**.

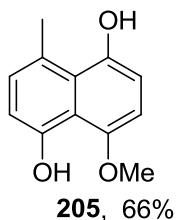


A solution of 150 mg (1.2 mmol) of p-methylquinol **202** in 2 mg of benzene was added 143 mg (1.46 mmol) of furan and 15 mg (0.11 mmol) of zinc chloride at room temperature. The reaction was stirred overnight at 45-50 °C in a well-sealed vial. The solution was concentrated in vacuo. The residue was separated by a silica gel column with hexanes and ethyl acetate (10:1) to give 119 mg of compound **203** (0.54 mmol, 44%) and 95 mg of compound **204** (0.43 mmol, 35%).

¹HNMR (CDCl_3): δ 6.41 (d, 1H, $^3J = 10.4$ Hz), 5.80 (d, 1H, $^3J = 11.2$ Hz), 5.78 (d, 1H, $^3J = 10.4$ Hz), 5.57 (dd, 1H, $^3J = 11.2$ Hz, $^3J = 10.4$ Hz), 3.68 (s, 3H), 3.38 (s, 1H), 3.33 (dd, 1H, $^3J = 10.4$ Hz, $^3J = 8.0$ Hz), 2.09 (d, 2H, $^3J = 8.0$ Hz), and 1.50 (s, 3H). ¹³CNMR (CDCl_3): δ 194.8, 166.8, 149.2, 147.8, 126.3, 120.0, 66.6, 51.4, 36.4, 33.9, 29.2, and 27.0. IR (KBr): ν_{max} 3453, 2952, 1778, 1715, 1666, 1444, 1370, 1203, and 1006 cm^{-1} .

¹HNMR (CDCl_3): δ 6.44 (d, 1H, $^3J = 10.4$ Hz), 5.84 (m, 2H), 5.66 (d, 1H, $^3J = 10.4$ Hz), 3.76 (s, 1H), 3.69 (s, 3H), 3.11 (dd, 1H, $^3J = 10.4$ Hz, $^3J = 4.4$ Hz), 2.16 (m, 1H), 1.97 (m, 1H), and 1.47 (s, 3H). ¹³CNMR (CDCl_3): δ 194.0, 167.6, 151.8, 148.0, 124.0, 120.9, 67.5, 51.7, 34.5, 33.8, 29.6, and 27.5. IR (KBr): ν_{max} 3453, 2952, 1778, 1715, 1666, 1444, 1370, 1203, 1006, and 922 cm^{-1} .

Naphthol **205**.

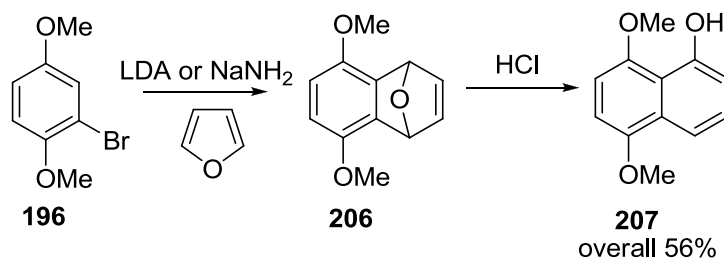


(a) A solution of 40 mg (0.18 mmol) of **203** or **204** in methylene chloride was added 80 mg of triethylamine and 200 μ L of trimethylsilyl triflate at 0 $^{\circ}$ C. The reaction was stirred overnight at room temperature. Then the reaction was quenched with 1 M HCl solution. The organic phase was separated; the water phase was extracted three times with diethyl ether. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a TLC plate with hexanes and ethyl acetate (10:1) to give 23 mg of compound **205** (0.12 mmol, 66 %).

(b) A solution of 40 mg (0.18 mmol) of **203** or **204** in methylene chloride was added 100 μ L of trifluoroacetic acid and 200 μ L of acetic anhydride at 0 $^{\circ}$ C. The reaction was stirred overnight at room temperature. Then the reaction was quenched by water. The organic phase was separated; the water phase was extracted three times with ether. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was separated by a TLC plate with hexanes and ethyl acetate (10:1) to give 21 mg of product (10.8 μ mol, 60 %).

^1H NMR (CDCl_3): δ 7.31 (d, 1H, $^3J = 10.4$ Hz), 7.31 (s, 1H), 7.05 (d, 1H, $^3J = 10.4$ Hz), 6.55 (s, 1H), 3.82 (s, 2H), 3.75 (s, 3H), and 2.42 (s, 3H). ^{13}C NMR (CDCl_3): δ 169.5, 153.5, 148.2, 132.3, 129.9, 128.8, 125.3, 120.8, 110.7, 105.0, 52.6, and 34.7. IR (KBr): ν_{max} 3453, 2954, 2854, 1742, 1677, 1546, 1491, 1262, 1121, and 1014 cm^{-1} . GC-MS (M^+): 205.12, 204.04, 146.10, 144.96, 116.06, 115.01, 91.03 and 72.14 g/mol.

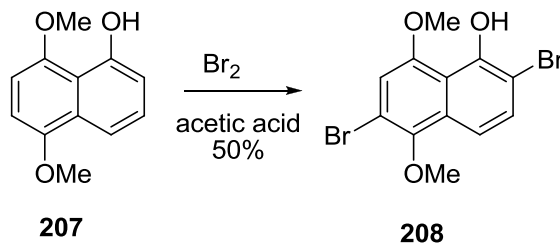
Compound **207**.¹¹²



50 g of sodium amide was added to a mixture of 200 mL of furan and 300 mL of THF at room temperature. The reaction was allowed to warm up to 50 °C. 122 g (0.56 mol) of bromoquinone **197** in 200 mL of THF was added to the reaction. After the reaction was stirred for 24 hours, it was cooled down to room temperature. The solid was filtered off and the filtrate was concentrated in vacuo. 350 mL of methanol was added to the crude product at 0 °C. Then 15 mL of concentrated HCl was added. The solution was stirred for 12 hours at room temperature and refluxed for 3 hours. The reaction was poured into ice cold water. The water solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 10:1 hexanes and ethyl acetate to give 64 g of compound **207** (0.31 mmol, 56%).

¹HNMR (CDCl₃): δ 9.46 (s, 1H), 7.71 (d, 1H, ³J = 8.4 Hz), 7.37 (dd, 1H, ³J = 8.4 Hz, ³J = 8.0 Hz), 6.93 (d, 1H, ³J = 8.4 Hz), 6.66 (m, 2H), 4.01 (s, 3H), and 3.94 (s, 3H). The spectroscopic and other physical data reported are consistent with values reported in the literature.¹¹²

Dibromonaphthol **208**.

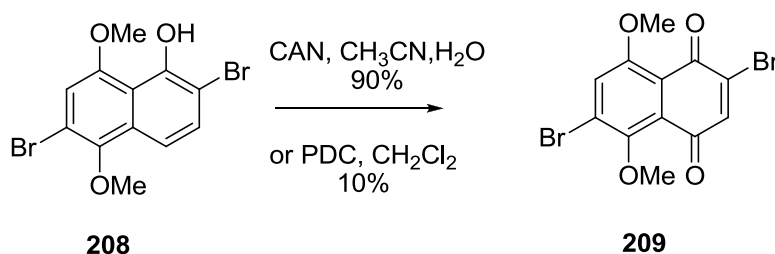


A solution of 21 g (0.13 mol) of bromine in 50 mL of acetic acid was added to a solution of 14 g (68 mmol) of compound **207** in 150 mL of acetic acid at 60 °C. The

reaction was stirred for 3 hours and then quenched by water. The water solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residues were crystallized in methanol to give 12.54 g of compound **208** (34 mmol, 50%).

¹HNMR (CDCl₃): δ 9.89 (s, 1H), 7.61 (d, 1H, ³J = 9.2 Hz), 7.46 (d, 1H, ³J = 9.2 Hz), 6.92 (s, 1H), 4.07 (s, 3H), and 3.91 (s, 3H). ¹³CNMR (CDCl₃): δ 151.9, 151.2, 148.1, 132.6, 130.4, 115.6, 114.4, 112.2, 109.3, 105.8, 61.5, and 57.1. IR (KBr): ν_{max} 3431, 2963, 1634, 1588, 1498, 1330, 1265, 1164, and 1064 cm⁻¹

Compound 209



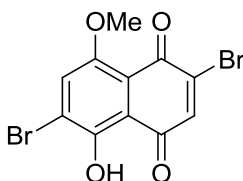
(a) A solution of 818 mg (1.5 mmol) of ammonium cerium (IV) nitrate in 2 mL of water was added to a solution of 130 mg (0.36 mmol) of **208** in 20 mL of acetonitrile at room temperature. The reaction was stirred for 15 minutes and then quenched by ice cold water. The water solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 8:1 hexanes and ethyl acetate to give 121 mg of compound **209** (0.32 mmol, 90%).

(b) 1g (4.6 mmol) of pyridinium dichromate was added to a solution of 312 mg (0.86 mmol) of **208** in 20 mL of methylene chloride at 0 °C. The reaction was stirred for 2 hours at room temperature. The solid was filtered off. The organic solution was washed with sodium bisulfite. Then it was dried over MgSO₄ and concentrated in vacuo. The

residues were separated by a silica gel column with 8:1 hexanes and ethyl acetate to give 34 mg of compound **209** (0.086 mmol, 10%).

$^1\text{H NMR}$ (CDCl_3): δ 7.55 (s, 1H), 7.35 (s, 1H), 3.98 (s, 1H), and 3.89 (s, 3H).
 $^{13}\text{C NMR}$ (CDCl_3): δ 181.2, 176.2, 156.8, 150.8, 140.3, 139.4, 129.2, 128.4, 126.3, 123.3, 62.0, and 57.2. IR (KBr): ν_{max} 2942, 1758, 1665, 1565, 1466, 1379, 1216, 1191, and 1089 cm^{-1} .

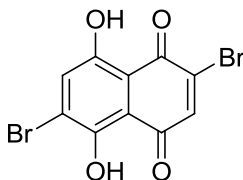
Compound **210**.



A solution of 100 mg (0.27 mmol) of compound **209** in 5 mL of methylene chloride was cooled down to 0 °C. 54 mg (0.4 mmol) of aluminum chloride was added to the solution at 0 °C. The reaction was stirred overnight at room temperature and then quenched by water. The water solution was extracted three times with methylene chloride. The organic solution was washed with sodium bisulfite. Then it was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 3:1 hexanes and ethyl acetate to give 36 mg of compound **210** (0.1 mmol, 37%).

$^1\text{H NMR}$ (CDCl_3): δ 12.94 (s, 1H), 7.64 (s, 1H), 7.47 (s, 1H), and 3.99 (s, 3H).
 $^{13}\text{C NMR}$ (CDCl_3): δ 187.7, 175.1, 154.7, 153.8, 143.9, 138.1, 126.4, 122.2, 115.7, 114.8, and 57.4. IR (KBr): ν_{max} 3399, 3059, 1662, 1587, 1450, 1355, 1266, 1192, and 1094 cm^{-1} .

Dibromonaphthylquinone **211**.¹¹⁵

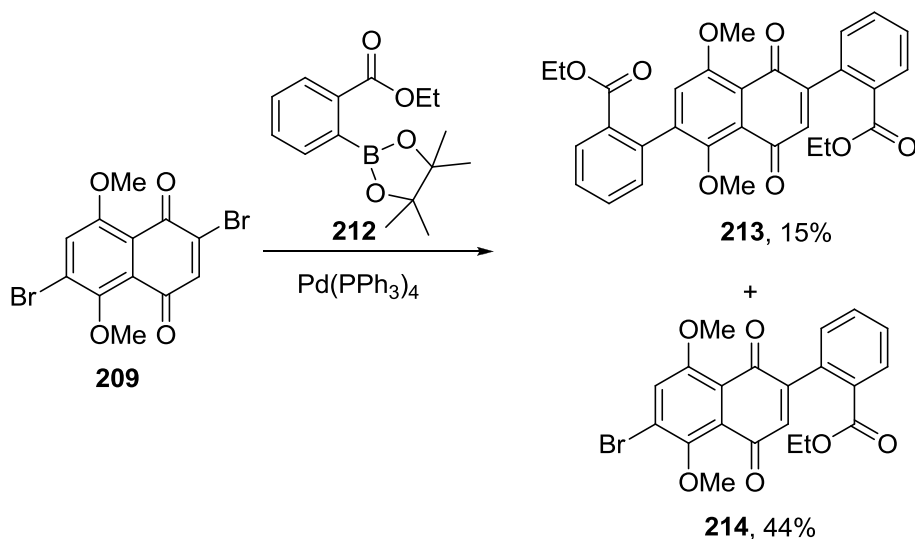


211, 90%

A solution of 29 mg (0.078 mmol) of compound **209** in 5 mL of methylene chloride was cooled down to -70 °C. 0.1 mL of boron bromide was added to the solution at -70 °C. The reaction was stirred for 2 hours at -70 °C and then it was allowed to warm up to room temperature. The reaction was quenched by water. The water solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 10:1 hexanes and ethyl acetate to give 24 mg of product (0.07 mmol, 90%).

¹HNMR (CDCl₃): δ 12.57 (s, 2H), and 7.61 (s, 2H). The spectroscopic and other physical data reported are consistent with values reported in the literature.¹¹⁵

Naphthylquinone **213** and **214**.



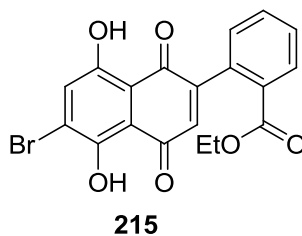
1 g (10 mmol) of sodium carbonate was added to a solution of 147 mg (0.4 mmol) of compound **209** in a mixture of 8 mL of benzene, 3 mL of 1,4-dioxane, and 4 mL of water. Then 89 mg (0.53 mmol) of 2-ethoxycarbonylphenylboronic acid pinacol ester was added to the reaction. Finally, 80 mg of tetrakis(triphenylphosphine)palladium was added. After the reaction was stirred for 48 hours at 55 °C, it was poured into water. The water solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 5:1 hexanes and ethyl acetate to

give 76 mg of compound **214** (0.18 mmol, 44%) and 32 mg of compound **213** (0.06 mmol, 15%).

213 $^1\text{H NMR}$ (CDCl_3): δ 8.05 (d, 1H, $^3J = 7.6$ Hz), 7.58 (dd, 1H, $^3J = 7.6$ Hz, $^3J = 6.4$ Hz), 7.52 (s, 1H), 7.49 (dd, 1H, $^3J = 7.2$ Hz, $^3J = 6.4$ Hz), 7.32 (d, 1H, $^3J = 7.2$ Hz), 6.73 (s, 1H), 4.19 (q, 2H, $^3J = 7.2$ Hz), 3.90 (s, 6H), and 1.23 (t, 3H, $^3J = 7.2$ Hz). $^{13}\text{C NMR}$ (CDCl_3): δ 184.0, 183.1, 166.5, 156.0, 152.4, 150.2, 135.5, 132.6, 131.8, 130.4, 130.3, 129.6, 127.4, 126.9, 123.2, 121.5, 61.8, 61.4, 57.0, and 14.1. IR (KBr): ν_{max} 3515, 2978, 1721, 1659, 1586, 1467, 1390, 1284, 1135, and 1085 cm^{-1} .

214 $^1\text{H NMR}$ (CDCl_3): δ 8.06 (d, 1H, $^3J = 8.0$ Hz), 8.01 (d, 1H, $^3J = 7.6$ Hz), 7.59 (dd, 2H, $^3J = 8.0$ Hz, $^3J = 7.6$ Hz), 7.50 (dd, 2H, $^3J = 8.0$ Hz, $^3J = 7.6$ Hz), 7.37 (d, 1H, $^3J = 8.0$ Hz), 7.34 (d, 1H, $^3J = 7.6$ Hz), 7.18 (s, 1H), 6.75 (s, 1H), 4.21 (q, 2H, $^3J = 7.2$ Hz), 4.14 (q, 2H, $^3J = 7.2$ Hz), 3.90 (s, 3H), 3.43 (s, 3H), 1.25 (t, 3H, $^3J = 7.2$ Hz), and 1.09 (t, 3H, $^3J = 7.2$ Hz). $^{13}\text{C NMR}$ (CDCl_3): δ 184.9, 183.5, 167.2, 166.6, 155.9, 152.2, 150.7, 145.1, 137.3, 135.8, 132.7, 132.5, 131.7, 131.3, 130.9, 130.6, 130.3, 130.2, 129.4, 128.5, 126.0, 121.1, 120.3, 61.4, 61.2, 61.2, 56.9, 14.1, and 13.9. IR (KBr): ν_{max} 3389, 2930, 1711, 1659, 1467, 1382, 1264, 1194, 1138, and 1097 cm^{-1} .

Naphthylquinone **215**.

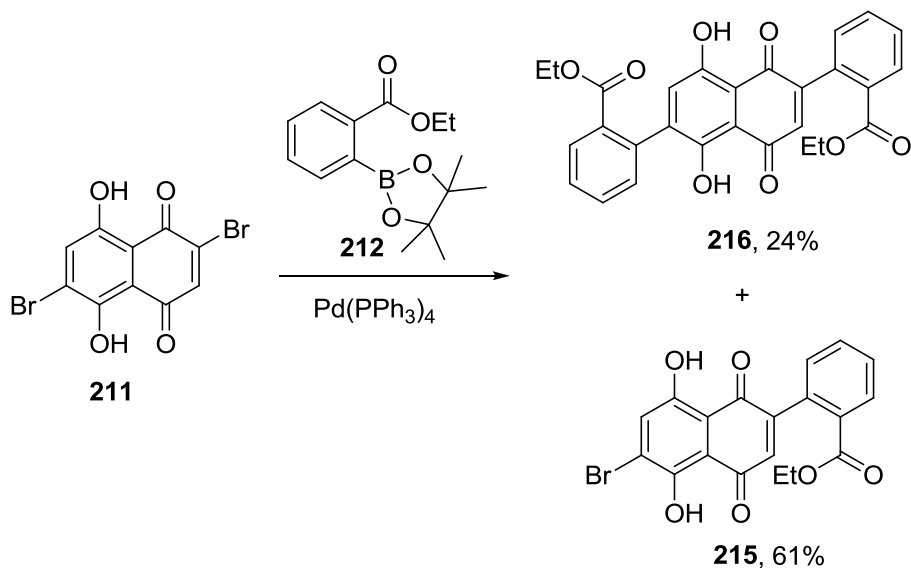


A solution of 12 mg (0.027 mmol) of **214** in 3 mL of methylene chloride was cooled down to -70 $^{\circ}\text{C}$. 0.1 mL of boron bromide was added to the solution at -70 $^{\circ}\text{C}$. The reaction was stirred for 2 hours at -70 $^{\circ}\text{C}$ and then quenched by water. The water solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO_4 . After concentration of organic solution in vacuo, the residue was dissolved in methylene chloride. The solution was added 4 mg of p-

toluenesulfonic acid. The reaction was stirred overnight at room temperature and quenched by water. The water solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residues were separated by a silica gel column with 5:1 hexanes and ethyl acetate to 5 mg of compound **215** (0.012 mmol, 45%).

¹HNMR (CDCl₃): δ 12.98 (d, 1H), 12.28 (d, 1H), 8.16 (d, 1H, ³J = 7.6 Hz), 7.64 (m, 3H), 7.31 (d, 1H, ³J = 7.6 Hz), 7.03 (d, 1H), 4.23 (m, 2H), and 1.26 (m, 3H). ¹³CNMR (CDCl₃): δ 181.7, 179.5, 166.6, 160.8, 159.5, 157.7, 153.3, 135.7, 134.2, 133.5, 132.9, 132.0, 130.8, 130.3, 129.9, 129.4, 128.7, 61.7, and 14.2. IR (KBr): ν_{max} 3451, 2923, 2852, 1720, 1658, 1481, 1366, 1262, 1188, and 1096 cm⁻¹.

Naphthylquinone **215** and **216**.



1 g (10 mmol) of sodium carbonate was added to a solution of 98 mg (0.28 mmol) of **211** in a mixture of 6 mL of benzene, 2 mL of 1,4-dioxane, and 3 mL of water. Then 64 mg (0.23 mmol) of 2-ethoxycarbonylphenylboronic acid pinacol ester was added to the reaction. Finally, 70 mg of tetrakis(triphenylphosphine)palladium was added. After the reaction was stirred for 48 hours at 55 °C, it was washed by water. The water solution was extracted three times with methylene chloride. The organic phases were combined and dried over MgSO₄. After concentration of organic solution in vacuo, the residues

were separated by a silica gel column with 5:1 hexanes and ethyl acetate to give 72 mg of **215** (0.17 mmol, 61%) and 33 mg of **216** (0.068 mmol, 24%).

216 ^1H NMR (CDCl_3): δ 12.81 (s, 2H), 8.09 (d, 2H, $^3J = 7.6$ Hz), 7.64 (dd, 2H, $^3J = 7.6$ Hz, $^3J = 7.6$ Hz), 7.55 (dd, 2H, $^3J = 7.6$ Hz, $^3J = 7.6$ Hz), 7.34 (d, 2H, $^3J = 7.6$ Hz), 7.10 (s, 2H), 4.24 (q, 4H, $^3J = 7.2$ Hz), and 1.25 (t, 6H, $^3J = 7.2$ Hz). ^{13}C NMR (CDCl_3): δ 173.2, 171.4, 166.8, 148.8, 135.1, 132.5, 131.3, 130.5, 130.3, 129.5, 128.4, 61.4, and 14.1. IR (KBr): ν_{max} 3372, 3053, 2923, 1721, 1590, 1481, 1392, 1258, 1183, and 1074 cm^{-1} .

Chapter 3. Synthesis of Cis-Alkenes via a Novel Copper-Mediated Cross-Coupling of 1,1- Dibromoalkenes with Halides.

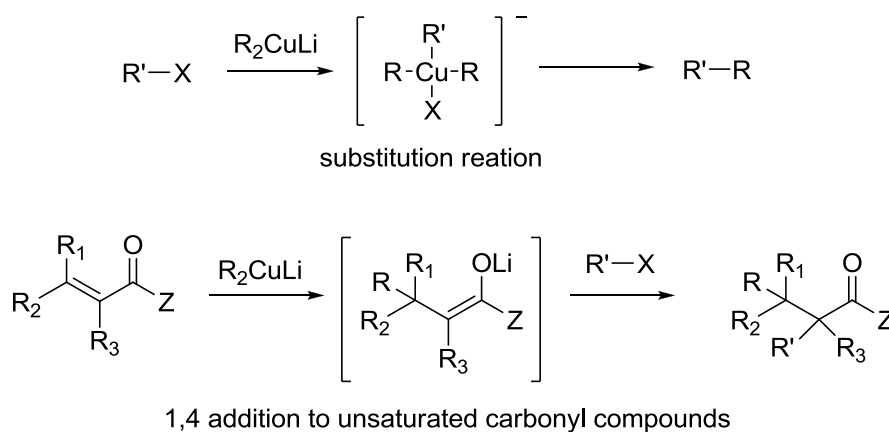
3.1 Introduction

3.1.1 Background: Organocopper Reagents

Since the first organo-copper compound, copper(I) acetylide, was synthesized by Bottger in 1859,¹¹⁶ various organo-copper compounds and their reactions have been studied by chemists. Methylcopper was prepared by Henry Gilman in 1936.¹¹⁷ In 1941 Kharash discovered 1,4-addition instead of 1,2-addition of a Grignard reagent to cyclohexenone in presence of Cu(I).¹¹⁸ In 1952 Gilman synthesized dialkylcuprates for the first time.¹¹⁹ Now organo-copper reagents are very frequently used in organic syntheses because of selective reactivity and lower nucleophilicity for carbon.¹²⁰ However, organo-copper compounds are thermally unstable.¹²¹ Thus they can not undergo consecutive cross-coupling processes like some other transition metal complexes.

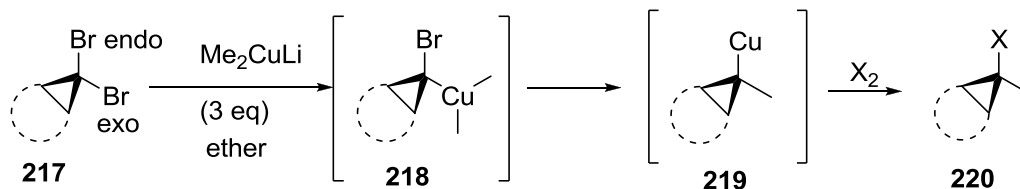
Because of the instability of organocopper reagents, they are usually prepared in situ by adding organolithium reagents or Grignard reagents to copper(I) halides. Divalent cuprates R_2CuLi are also known as Gilman reagents. In diethyl ether, lithium dimethyl cuprate, a Gilman reagent, is a dimer forming an eight-membered ring with two lithium atoms coordinating between two methyl groups.¹²² In different solvent, the Gilman reagent exhibits different structures.¹²³

Organocopper reagents are mainly involved in two kinds of reactions. One is substitution with halides; the other is 1,4-addition to α,β -unsaturated carbonyl compounds (Scheme 61). In various organocopper reactions, the oxidation state of copper can be changed from +1 to +2. In an organocopper intermediate, the oxidation state of copper can be +3.



Scheme 61. Coupling and 1,4 addition reactions of organocopper reagents.

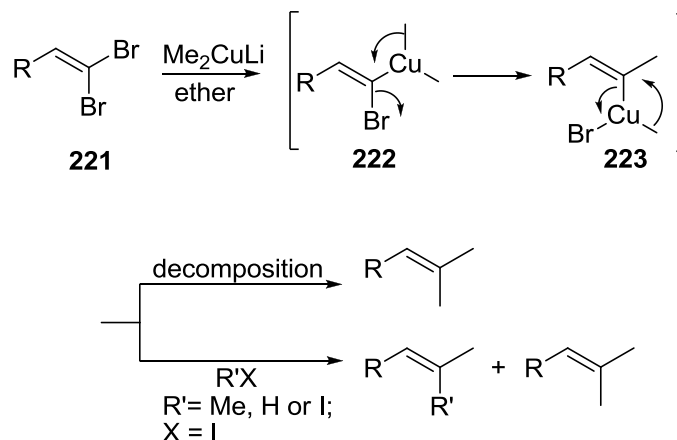
3.1.2 A New Method to Generate (Z)-Vinylcopper Species from 1,1-Dibromoalkenes



Scheme 62. The reaction of Gilman reagent and 1,1-dibromocyclopropane.

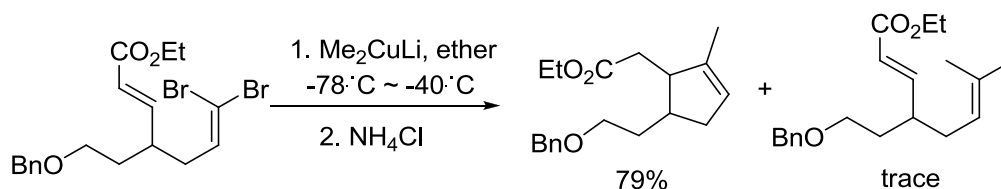
Lithium dimethyl cuprate may replace the less hindered exo-bromo group of a 1,1-dibromocyclopropane **217** to give the intermediate **218** (Scheme 62).¹²⁴ Then a 1,2-shift of the alkyl group with inversion of configuration affords an endo cyclopropyl

copper intermediate **219** which can be quenched by halogen to form an endo cyclopropyl halide **220**.



Scheme 63. The reaction of Gilman reagent and 1,1-dibromoalkene.

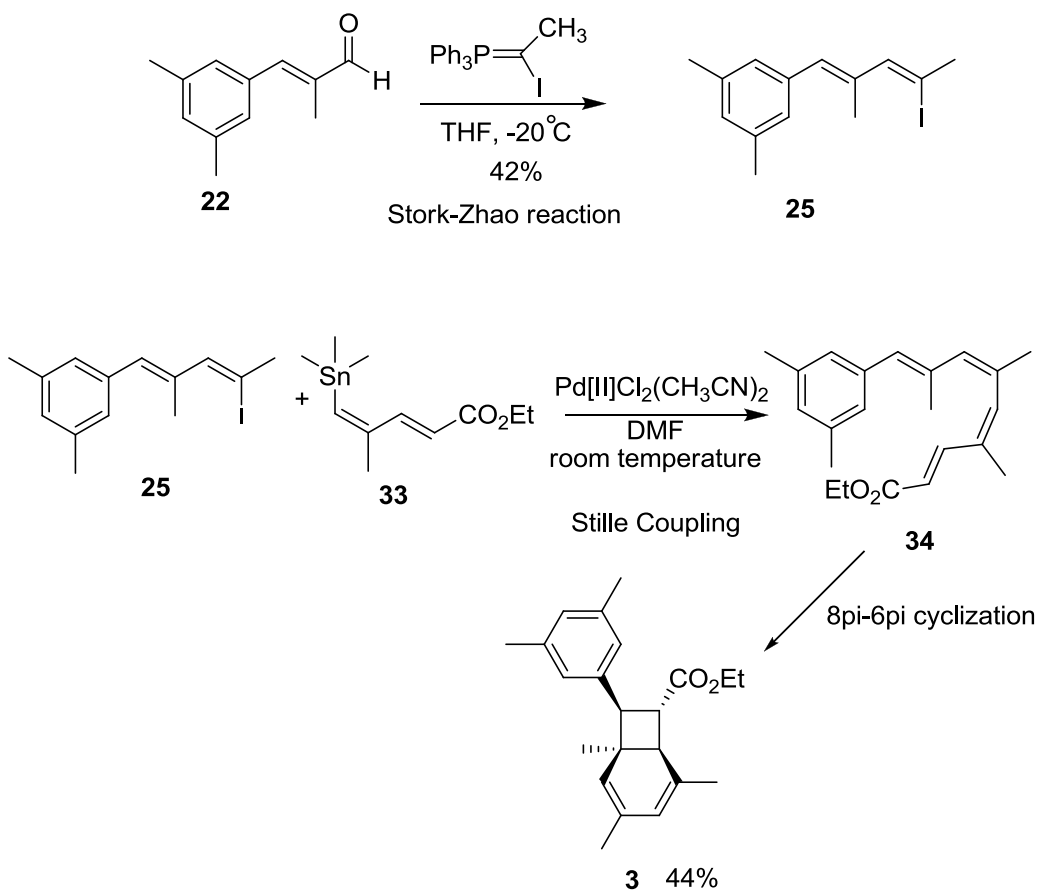
According to the literature,¹²⁵ lithium dimethyl cuprate may act as a nucleophile to substitute the less hindered (E)-bromo group of a 1,1-dibromoalkene (Scheme 63). Then a 1,2-shift of the methyl group gives a (Z)-vinyl copper intermediate **223**. The intermediate may decompose to the dimethylalkene product. Quenching the reaction with MeI may improve the yield of dimethyl product.¹²⁶ Recently, Masaaki Miyashita reported that a (Z)-vinylcopper intermediate undergoes an intramolecular 1,4-addition to an α,β -unsaturated ester moiety (Scheme 64).¹²⁷ Additionally, the vinylcopper intermediate was quenched with iodine to prepare (Z)-1-iodo-1-methylalkenes (Scheme 63).



Scheme 64. (Z)-vinylcopper intermediate involved 1,4-addition.

In one of the key steps for the synthesis of SNF analogues, the low-yield Stork-Zhao reaction has been used to make a (Z)-iodoalkene (Scheme 65). The other key step is Stille cross-coupling for the preparation of tetraenes. The Stille reaction not only afforded

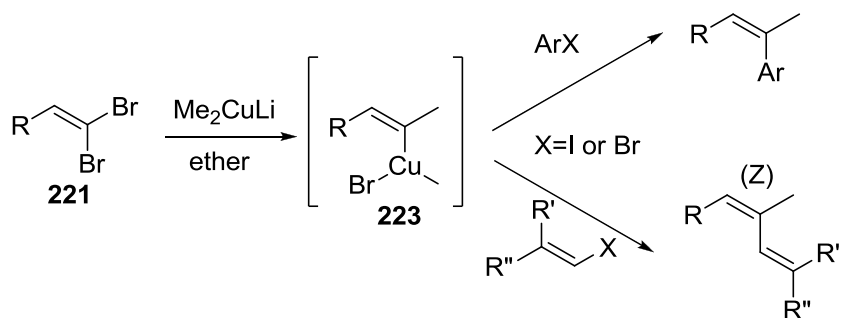
a product in low yield but also used a toxic tin starting material. We investigated some new olefination strategies to avoid the low-yield Stork-Zhao reaction and the toxic Stille coupling. We wondered whether the vinylcopper intermediates could be trapped by coupling with an aryl halide or a vinyl halide.



Scheme 65. Low-yield Stork-Zhao reaction and Stille cross-coupling in the synthesis of SNF analogs.

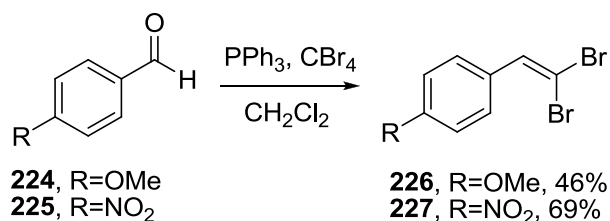
3.2 Results and Discussion

Thus, we proposed a new synthetic route for polyenes (Scheme 66). Lithium dimethyl cuprate replaces the less hindered exo-bromo group of a 1,1-dibromoalkene **221**. Then a 1,2-shift of the methyl group affords an (*Z*)-vinyl copper intermediate **223**. The intermediate may couple with aryl halide to provide a (*Z*)-1-aryl-1-methylalkene; the intermediate also may couple with vinyl halide to give a (*Z*)-polyene.



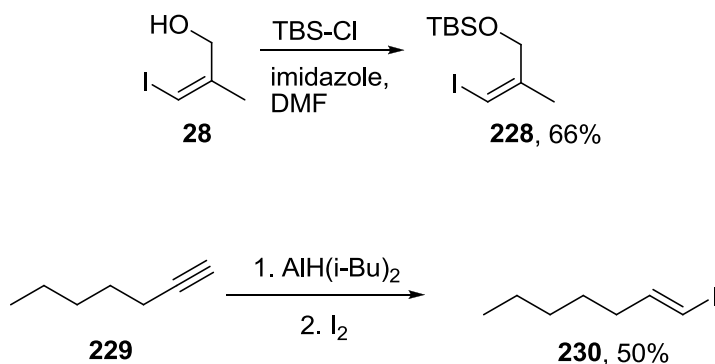
Scheme 66. A new proposed synthetic route for polyenes.

3.2.1 The Syntheses of Substrates



Scheme 67. Preparation of 1,1-dibromoalkene **226** and **227**.

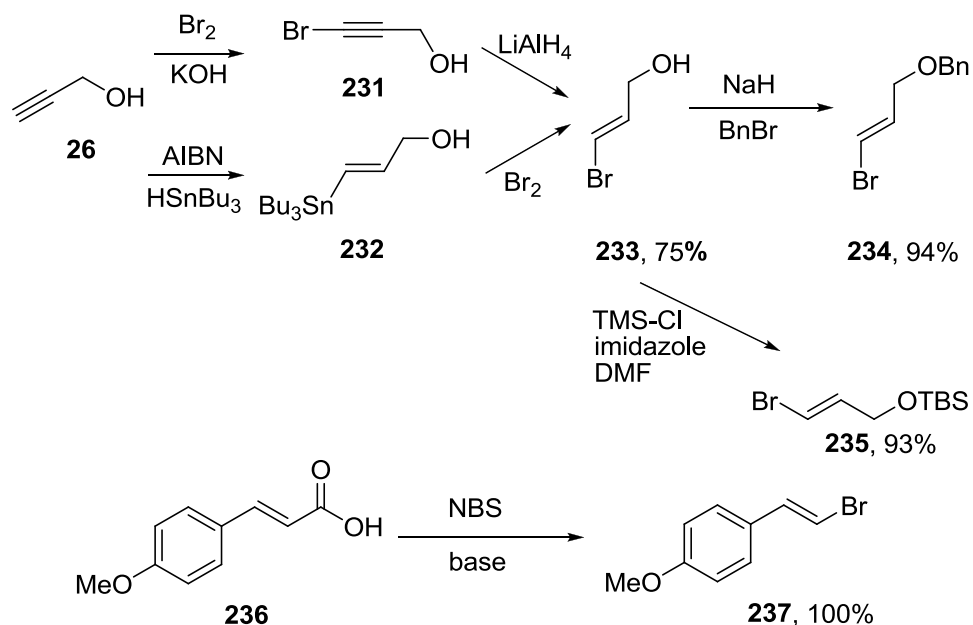
To examine the proposed methodology, 1,1-dibromoalkene **226** and **227** were designed as the substrates (Scheme 67). Treatment with tetrabromomethane and triphenylphosphine transformed *p*-methoxybenzaldehyde or *p*-nitrobenzaldehyde to the corresponding dibromoalkene.¹²⁸



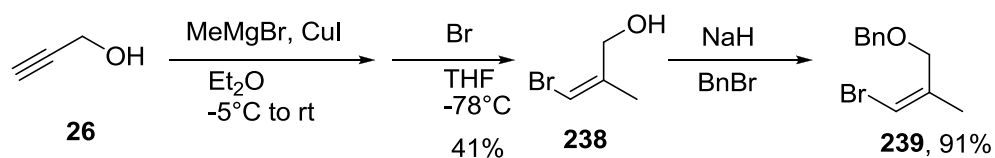
Scheme 68. Preparation of vinyl iodide **228** and **230**.

In order to examine the proposed coupling of the vinylcopper intermediate **223** and vinyl iodides, two vinyl iodides were prepared. TBS protection of (Z)-iodoallylic alcohol **28** provided the vinyl iodide **228** (Scheme 68).¹²⁹ 1-Heptyne **228** was reduced by diisobutylaluminum hydride to form a trans-vinylaluminum intermediate.¹³⁰ Treatment with iodine quenched the intermediate to give the (E)-vinyl iodide **230** in 50% yield.

In order to examine the proposed coupling of the vinylcopper intermediate **223** and vinyl bromides, several vinyl bromides were prepared. Bromoallylic alcohol **233** was synthesized in two ways (Scheme 69).¹³¹ Propargyl alcohol was oxidized by bromine in the presence of strong base to compound **231**. Reduction of **231** by lithium aluminum hydride produced compound **233** in 75% yield.¹³² In the other synthetic route, propargyl alcohol **26** was reduced by tributylstannane to compound **232**. The bromine oxidized **232** to bromoallylic alcohol **233** in 46% yield.¹³³ Benzyl protection of **233** provided the vinyl bromide **234** in 91% yield.¹³⁴ Treatment of bromoallylic alcohol **233** with tert-butyltrimethylsilyl chloride afforded the TBS-protected (E)-vinyl bromide **235** in 93% yield.¹³⁵ Base catalyzed decarboxylation of acid **236** and subsequent oxidation by NBS gave the (E)-vinyl bromide **237** in 94% yield (Scheme 69).¹³⁶



Scheme 69. Preparation of (E)-vinyl bromides **234**, **235**, and **237**.

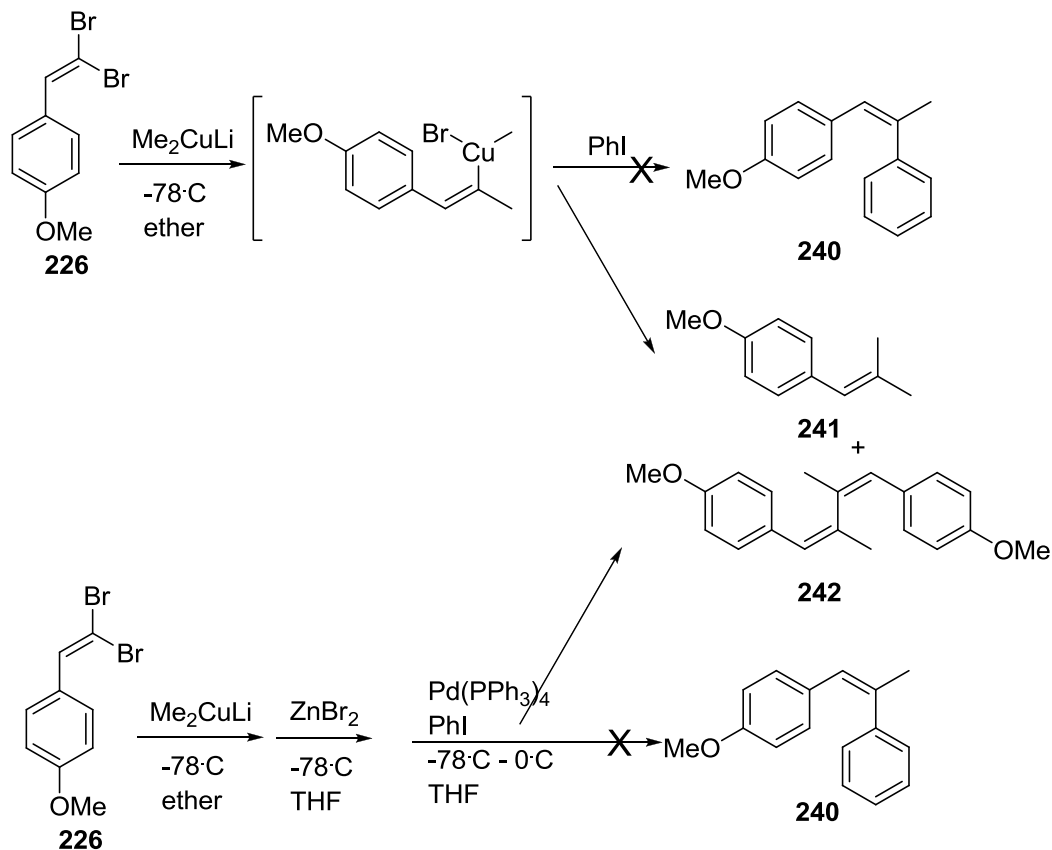


Scheme 70. Preparation of (Z)-vinyl bromide **239**.

The (Z)-bromoallylic alcohol **238** was obtained in 41% yield from propargyl alcohol **26** through a stereoselective carbocupration/bromination sequence (Scheme 70).¹³⁷ Benzyl protection of **238** provided the (Z)-vinyl bromide **239** in 91% yield.

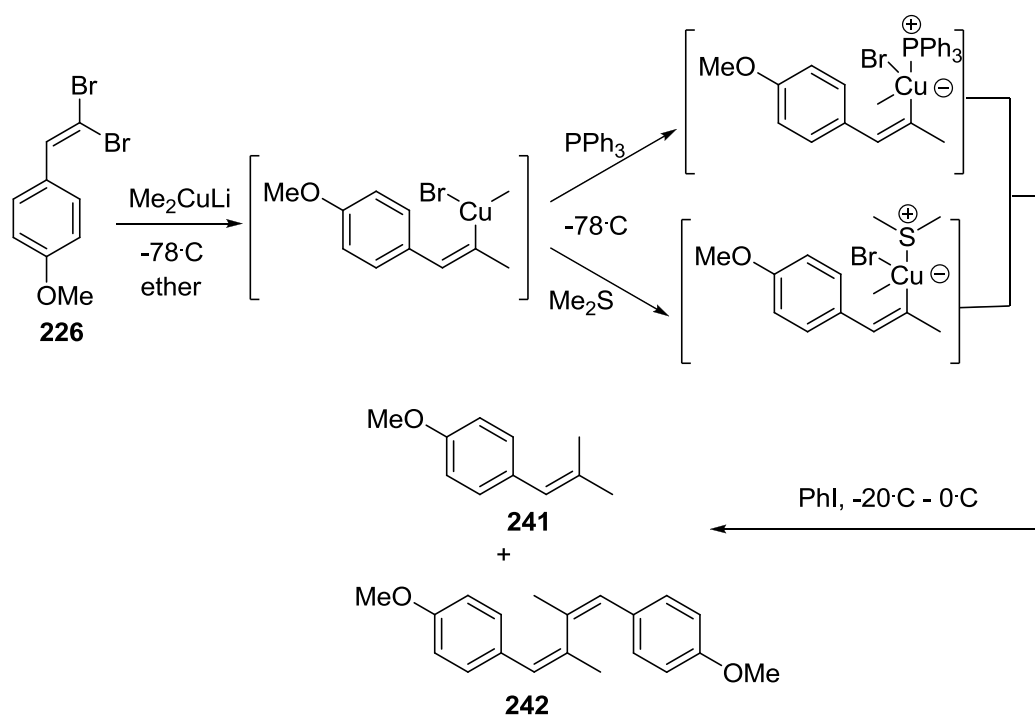
3.2.2 A New One-Pot Coupling Reaction via Organocopper

Intermediates



Scheme 71. Attempt at coupling of dibromoalkene **226** and iodobenzene.

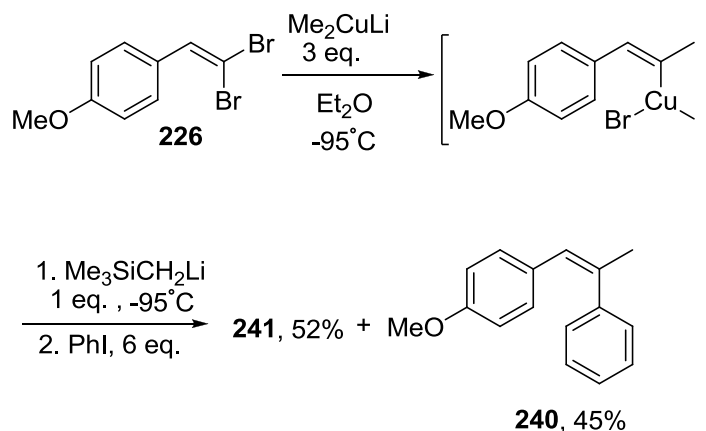
The substrate **226** was treated with lithium dimethyl cuprate to form a copper intermediate. Iodobenzene was added directly to the reaction mixture at low temperature (Scheme 71). Since nothing happened at $-95\text{ }^{\circ}\text{C}$, the reaction was allowed to warm up to $-20\text{ }^{\circ}\text{C}$. With increasing temperature, the decomposition of the intermediate accelerated to give lots of black solid. The reaction did not produce the desired 1-phenyl-1-methylalkene **240** but 1,1-dimethylalkene **241**¹³⁸ as the major product and a little dimerized product **242**. A Negishi strategy also was employed. Addition of zinc bromide to the copper intermediate was supposed to offer a (Z)-vinyl zinc reagent. However, Negishi condition only gave dimethylalkene **241** and dimerized product **242**. The results suggested that the Cu(I) intermediates were not stable enough to make a cross-coupling.



Scheme 72. Attempt at coupling of dibromoalkene **226** and iodobenzene in the presence of ligands.

Dimethyl sulfide, triphenylphosphine, and cyanide are known as good ligands for stabilizing organocopper reagents.¹³⁹ In order to stabilize the proposed (Z)-vinylcopper intermediates, dimethyl sulfide or triphenylphosphine was added as ligand to the reaction (Scheme 72). However, the reaction also only gave compounds **241** and **242** as products.

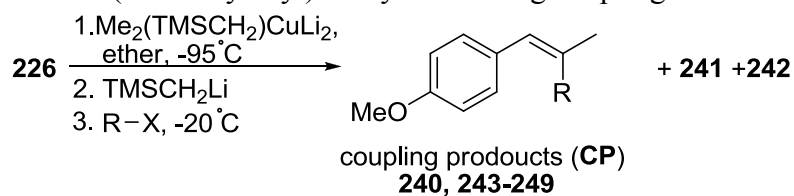
The results showed that dimethyl sulfide and triphenylphosphine as ligands could not make the copper intermediates thermally stable enough for a further coupling. In addition, those ligands could not activate the intermediates so well that the desired coupling of the copper intermediate and phenyl iodide could not compete with the decomposition of the copper intermediate.



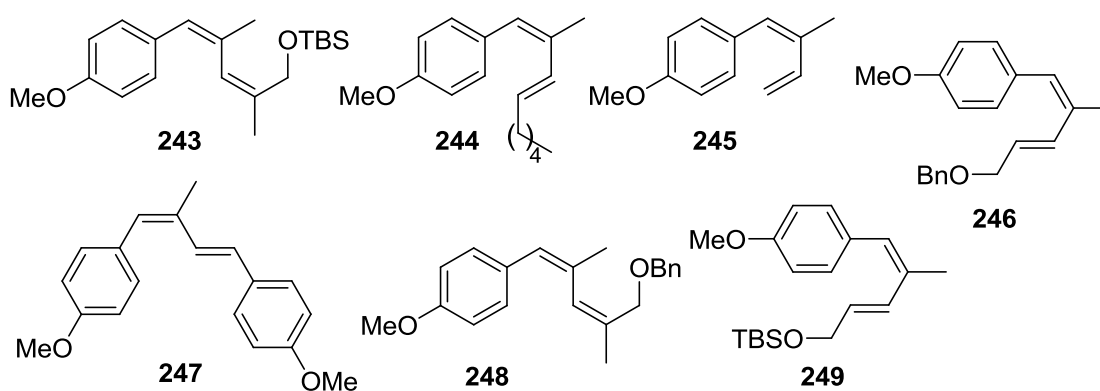
Scheme 73. Coupling of dibromoalkene **226** and iodobenzene in the presence of (trimethylsilyl)methyl lithium.

Since (trimethylsilyl)methyl group ($\text{TMS-CH}_2\text{-}$) can stabilize and activate cuprate reagents, (trimethylsilyl)methyl was chosen as the ligand.¹⁴⁰ The procedures described above were modified by adding 1 equivalent of (trimethylsilyl)methyl lithium to the reaction at -95°C after 1,1-dibromoalkene was added to dimethyl cuprate. The reaction was allowed to slowly warm up to -20°C and then 6 equivalents of iodobenzene was added (Scheme 73). After work-up, 1,1-dimethylalkene **241** and the coupling product, 1-methyl-1-phenylalkene **240**, were obtained in 52% and 45% yields, respectively. The chemical shifts of NMR spectra of compound **240** are different from those of trans-1-(4-methoxyphenyl)-2-phenylpropene.¹⁴¹ In addition, the NOE spectra indicated that compound **240** is cis.

Table 9. (Trimethylsilyl)methyl stabilizing coupling reactions.



entry	R-X	yields% 241,242,CP	coupling products (CP)
a	PhI	27, 0, 68	240
b	PhOSO ₂ CF ₃	80, 17, 0	240
c	PhBr	85, 14, 0	240
d	228	88, 12, 0	243
e	230	85, 0, 13	244
f	234	49, 0, 43	245
g	234	38, 0, 56	246
h	237	35, 0, 57	247
i	239	60, 10, 29	248
j	235	81, 0, 12	249



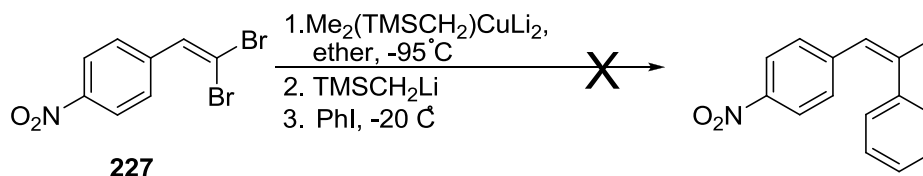
In order to keep copper intermediates stable from the beginning of the reaction, a mixed 2:1:1 cuprate, dimethyl (trimethylsilyl)methyl cuprate ($\text{Me}_2(\text{TMS-CH}_2)\text{CuLi}_2$), was used instead of dimethylcuprate.¹⁴² Compound **226** was added to 3 equivalents of $\text{Me}_2(\text{TMS-CH}_2)\text{CuLi}_2$ at -95°C . After the reaction was warmed up to -20°C , one more equivalent of $\text{TMS-CH}_2\text{Li}$ followed by 6 equivalents of various halides was added to the reaction mixture (Table 9).

The results showed that iodobenzene (entry a) gave the relatively best yield of the consecutive cross-coupling product **240** at 68% and 1,1-dimethyl product **241** in 27% yield. However, phenyl triflate (entry b), bromobenzene (entry c), and TBS protected cis-iodoallyl alcohol (entry d) did not produce cross-coupling products; they only yielded 1,1-dimethyl product **241** as major product and dimerized product **242** as the minor product. When trans-1-iodoalkene (entry e) was used as the vinyl halide to examine the reaction, the yield of compound **244** was low.

A series of bromoalkenes were employed to examine the copper mediated reaction. Vinyl bromide (entry f) and trans-bromoalkenes (entry g, h) offered coupling products **245**, **246**, and **247** in about 50% yields. The benzyl protected (*Z*)-3-bromo-2-methylallyl alcohol (entry i) gave the product **248** in low yield at 29% presumably because of the steric effect of (*Z,Z*)-transition state. Little of coupling product **249** was obtained when TBS-protected (*E*)-bromoallylic alcohol (entry j) was the substrate.

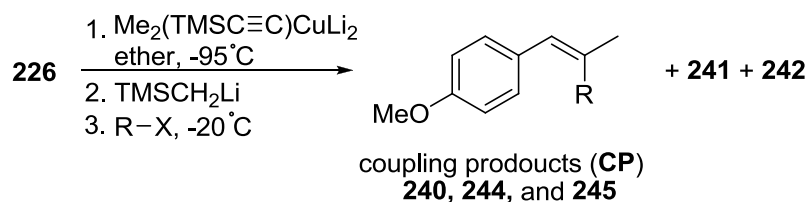
The results suggested that phenyl iodide has a better activity in the reaction than phenyl bromide does; however, vinyl bromides worked better in the reaction than vinyl iodide. Presumably because of steric effects, cis-vinyl bromides do not give good yields of cross coupling products. TBS protection of allylic alcohol may also lower the yield.

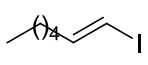

The dibromo compound **227** was also used as a substrate to examine the reaction (Scheme 74). Under the conditions that produced **240** from **226**, the reaction did not successfully afford the coupling product and the starting material decomposed. We concluded that the nitro group is not compatible with this reaction.



Scheme 74. Attempt at coupling using 1,1-dibromoalkene **227** as the substrate.

Table 10. (Trimethylsilyl)ethynyl stabilizing coupling reactions.

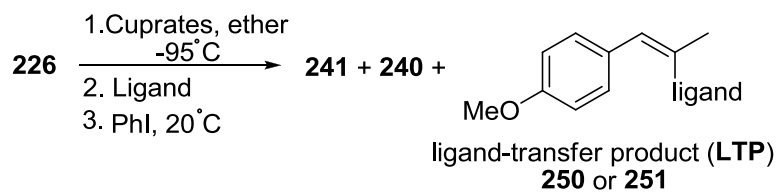


entry	R-X	yields% 241,242,CP	coupling products (CP)
a	PhI	46, 0, 43	240
b	PhOSO ₂ CF ₃	86, 10, 0	240
c	PhBr	84, 14, 0	240
d	 230	89, 0, 10	244
e	 Br	65, 0, 31	245

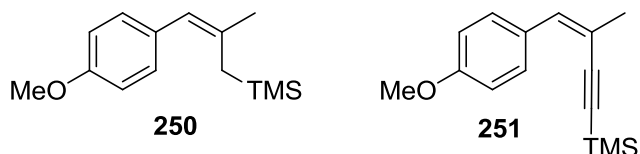
(Trimethylsilyl)ethynyl group (TMS-C≡C-) is also known as a ligand that stabilizes mixed cuprate reagents.¹⁴³ (Trimethylsilyl)ethynyllithium was applied to make dimethyl (trimethylsilyl)ethynyl cuprate (Me₂(TMS-C≡C)CuLi₂) which was examined in consecutive cross-coupling reactions (Table 10). (Trimethylsilyl)ethynyl ligand acted as trimethylsilyl ligand but was not very good. The (trimethylsilyl)ethynyl coordinated copper intermediate did not couple with phenyl triflate (entry b) and bromobenzene (entry c), either; but it coupled with iodobenzene (entry a), cis-iodoallyl alcohol (entry d), and vinyl bromide (entry f) to offer coupling products **240**, **244**, and **245** in lower yields than (trimethylsilyl)methyl copper intermediates did.

We wondered whether or not the reaction gave a better yield of the coupling product when we reduced the proportion of methyl group in the mixed cuprate reagents. Two mixed 1:2:1 cuprates, $\text{Me}(\text{TMS-CH}_2)_2\text{CuLi}_2$ and $\text{Me}(\text{TMSC}\equiv\text{C})_2\text{CuLi}_2$, were studied in the reaction (Table 11). Both of them produced ligand-transfer products **250** and **251**; and only $\text{Me}(\text{TMSCH}_2)_2\text{CuLi}_2$ gave the desired product **240** in 33% yield.

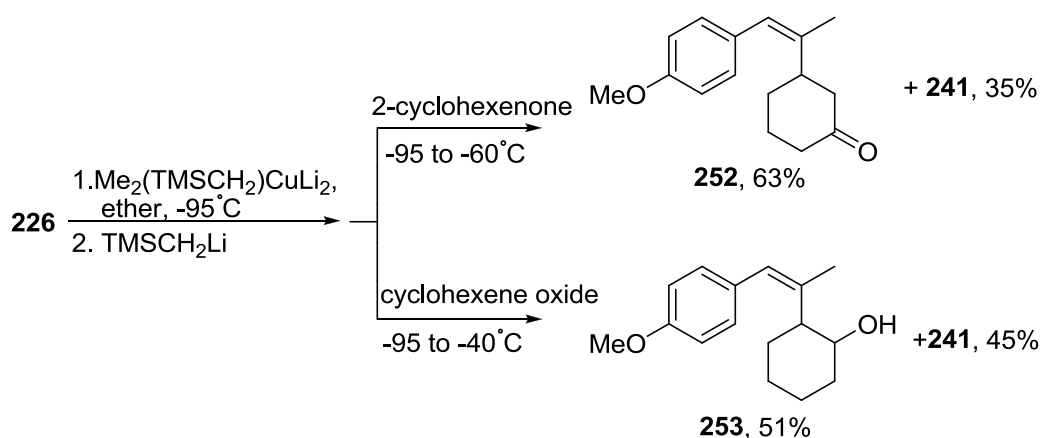
Table 11. Coupling using $\text{Me}(\text{TMSCH}_2)_2\text{CuLi}_2$ and $\text{Me}_2(\text{TMSC}\equiv\text{C})_2\text{CuLi}_2$.



Cuprates	Ligand	yields% 241, 240, LTP	LTP
$\text{Me}(\text{TMSCH}_2)_2\text{CuLi}_2$	TMSCH_2Li	33, 33, 33	250
$\text{Me}(\text{TMSC}\equiv\text{C})_2\text{CuLi}_2$	$\text{TMSC}\equiv\text{CLi}$	76, 0, 20	251



We expected that the (trimethylsilyl)methyl stabilized copper intermediate **223** was able to carry out an intermolecular 1,4-addition to α,β -unsaturated ketones. At -60°C , 6 equivalents of 2-cyclohexen-1-one was added to the (trimethylsilyl)methyl stabilizing intermediate (Scheme 75). After work-up, ketone **252** was obtained in 63% yield. The intermolecular epoxide-opening was also investigated. The addition of 6 equivalents of cyclohexene oxide to the reaction gave alcohol **253** in 51% yield.



Scheme 75. Intermolecular 1,4 addition to α,β -unsaturated ketone and intermolecular ring-opening of epoxide.

The new copper mediated consecutive cross-coupling may stereospecifically produce (*Z*)-2-methyl-alkenes in moderate yields in two steps from aldehydes.¹⁴⁴ It can potentially be applied in syntheses of polyene natural products, such as SNF 4435 C/D and superstolides.¹⁴⁵

3.3 Experimental Section

General

For all compounds, ^1H , ^{13}C and NOE NMR spectra were recorded on Varian Gemini 2300 (300 MHz), Inova 400 (400 MHz), Inova 500 (500 MHz), Inova 600 (600 MHz) spectrometer. Chemical shifts were measured relative to the residual solvent resonance for ^1H and internal reference. IR spectra were recorded on a Mattson Galaxy Series FTIR 3000 or 2020 spectrophotometer.

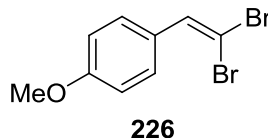
Reagents and solvents were purchased from Aldrich Chemical Company, Fisher Scientific or Acros Organics. Liquid reagents were purified by distillation prior to use. Unless otherwise noted, solid reagents were used without further purification. Methylene chloride, acetonitrile, cyclohexane, and toluene were distilled from CaH_2 under Ar

atmosphere. Diethyl ether and THF were distilled from Na and benzophenone under Ar atmosphere.

All reactions unless otherwise noted, were performed in oven dried glassware under positive pressure of Ar. All yields are isolated yields.

Chromatographic separations were performed using Fisher grade 1740 type 60 Å silica gel (170-400 mesh). Analytical thin layer chromatography was performed using Analtech Uniplate pre-coated glass 250 micron silica gel GHLF plates with 254 nm fluorescent indicator. Preparative thin layer chromatography was performed using Analtech Uniplate pre-coated glass 1000 or 2000 micron silica gel GHLF plates with 254 nm fluorescent indicator.

1,1-Dibromoalkene **226**.¹²⁸

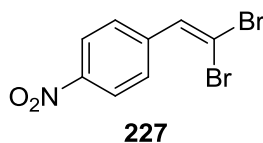


2.0 g (6.6 mmol) of tetrabromomethane was added to a solution of 4.0 g (15 mmol) of triphenylphosphine in 50 mL of methylene chloride at 0 °C. The mixture was stirred for 15 minutes at 0 °C. Then 5.0 g (37 mmol) of 4-methoxybenzaldehyde was added to the reaction. The reaction was stirred for an hour at 0 °C and half an hour at room temperature. The reaction was quenched by ice cold water. The water solution was extracted three times with methylene chloride. The organic solutions were combined, dried over Mg₂SO₄, and concentrated in vacuo. The residues were separated by a silica gel column with 40:1 hexanes and ethyl ether to give 1.89 g of compound **226** (98%).

¹HNMR (CDCl₃): δ 7.37 (d, 2H, ³J = 8.8 Hz), 7.41(s, 1H), 6.75 (d, 2H, ³J = 8.8 Hz), and 3.66 (s, 3H). ¹³CNMR (CDCl₃): δ 159.7, 136.4, 129.9, 127.8, 113.8, 87.3, and 55.3. IR (KBr): ν_{max} 3056, 3023, 1948, 1885, 1594, 1444, 1337, 1267, 1157, and 1029,

cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹²⁸

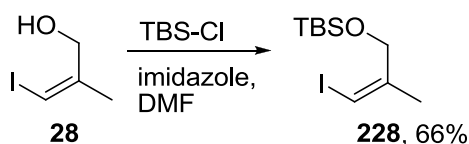
1,1-Dibromoalkene **227**.¹²⁸



2.0 g (6.6 mmol) of tetrabromomethane was added to a solution of 4 g (15 mmol) of triphenylphosphine in 50 mL of methylene chloride at 0 °C. The mixture was stirred for 15 minutes at 0 °C. Then 4.0 g (26 mmol) of 4-nitro-benzaldehyde was added to the reaction. The reaction was stirred for an hour at 0 °C and half an hour at room temperature. The reaction was quenched by ice cold water. The water solution was extracted three times with methylene chloride. The organic solutions were combined, dried over Mg₂SO₄, and concentrated in vacuo. The residues were separated by a silica gel column with hexanes and ethyl acetate (10:1) to give 1.6 g of compound **227** (79%).

¹HNMR (CDCl₃): δ 8.22 (d, 2H, ³J = 9.0 Hz), 7.69 (d, 2H, ³J = 9.0 Hz), and 7.55 (s, 1H). ¹³CNMR (CDCl₃): δ 147.4, 141.6, 135.1, 129.3, 123.9, and 94.2. IR (KBr): ν_{max} 3029, 2968, 2729, 1633, 1596, 1494, 1346, 1248, 1178, and 1054 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹²⁸

TBS protected iodoallylic alcohol **228**.¹²⁹

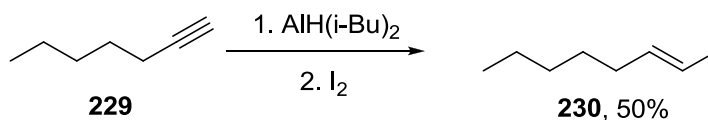


1.8 g (12 mmol) of t-butyldimethylchlorosilane was added to a mixture of 1.45 g (7.3 mmol) of iodoallylic alcohol **28** and 2.2 g (32 mmol) of imidazole in 15 mL of DMF at 0 °C. The reaction was stirred overnight and then it was quenched by water. The water solution was extracted three times with diethyl ether. The organic solutions were combined, dried over Mg₂SO₄, and concentrated in vacuo. The residues were separated

by a silica gel column with 100:1 hexanes and diethyl ether to give 1.89 g of compound **228** (4.9 mmol, 66%).

^1H NMR (CDCl_3): δ 5.86 (s, 1H), 4.25 (s, 2H), 1.91 (s, 3H), 0.91 (s, 9H), and 0.10 (s, 6H). ^{13}C NMR (CDCl_3): δ 146.9, 136.1, 68.8, 26.0, 21.6, 18.5, and -5.0. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹²⁹

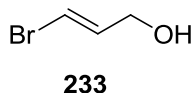
Vinyl iodide 230.¹³⁰



45 mL (45 mmol) of 1 M solution of diisobutylaluminum hydride in hexanes was added to 4.6 g (47 mmol) of heptyne at 0 °C. Then the reaction was allowed to warm up to 50 °C and stirred overnight. The reaction was cooled down to -50 °C and quenched by a solution of 11.4 g (0.045 mol) of iodine in 25 mL of THF. The solution was warmed up to 0 °C. 200 mL of 20% sulfuric acid was added to the solution. The water solution was extracted three times with diethyl ether. The organic solutions were washed with sodium thiosulfate solution. The organic solution was separated, dried over Mg_2SO_4 , and concentrated in vacuo. The residues were distilled at 120 °C in vacuo to give 5.34 g of compound **230** (23 mmol, 50%). Red copper was added to stabilize the product.

^1H NMR (CDCl_3): δ 6.51 (dt, 1H, $^3J = 15.0$ Hz, $^3J = 7.0$ Hz), 5.97 (d, 1H, $^3J = 15.0$ Hz), 2.05 (dt, 2H, $^3J = 8.0$ Hz, $^3J = 7.0$ Hz), 1.39 (tt, 2H, $^3J = 8.0$ Hz, $^3J = 7.0$ Hz), 1.29 (m, 4H), and 0.89 (t, 3H, $^3J = 6.8$ Hz). ^{13}C NMR (CDCl_3): δ 147.0, 74.4, 36.2, 31.3, 28.2, 22.6, and 14.1. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹³⁰

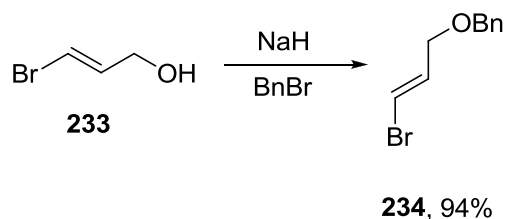
(E)-3-Bromo-2-propene-1-ol 233.¹³¹



Br₂ (12.00 g, 75 mmol) was added to a vigorously stirring solution of KOH (11.00 g, 0.2 mol) in water (40 mL) at -5 °C. The yellow solution was kept at 0 °C and added dropwise for 3 h to a solution of propargyl alcohol (4.5 g, 80 mmol, 1.06 equiv; freshly distilled) in water (10 mL) at -7 °C - 0 °C. The reaction mixture was warmed up to 10 °C and then extracted 4 times with diethyl ether. The organic layer was washed with Na₂S₂O₃, dried over K₂CO₃. The solvent was removed to provide the crude 1-bromopropyne-3-ol. A 500 mL flask charged with LiAlH₄ (5.1 g, 150 mmol, 2.0 equiv) was carefully added anhydrous ether (50 mL) at -5 °C with stirring. Then a solution of AlCl₃ (13.50 g, 75 mmol, 1.0 equiv) in anhydrous ether (50 mL) was carefully added. The crude 1-bromopropyne-3-ol, prepared in the first step, was added dropwise and the mixture was refluxed for 4 h. The reaction was quenched with wet ether (water saturated, 200 mL), followed by water (20 mL) and 5 % NaOH (20 mL) at -10 °C. Then water (60 mL) was added to break up the solid mass. The liquid was decanted and extracted with ether. The combined organic layer was dried over K₂CO₃ and concentrated. The residue was distilled at reduced pressure to yield the alcohol **233** (7.46 g, 60 mmol, 75 % from propargyl alcohol).

¹HNMR (CDCl₃): δ 6.35 (m, 2H), and 4.09 (m, 2H). ¹³CNMR (CDCl₃): δ 136.6, 107.979, and 63.0. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹³¹

Benzyl protected allylic alcohol **234**.¹³⁴



NaH (0.55 g, 60 %, 13.8 mmol) was added to a solution of (*E*)-3-bromo-2-propene-1-ol **233** (1.33 g, 10 mmol) in THF (20 mL) at 0 °C. After the mixture was stirred for 30 minutes, benzyl bromide (1.96 g, 11.5 mmol) and trace amounts of iodine was added. The reaction mixture was stirred for 4 hours at room temperature and it was

poured into water. The mixture was extracted with dichloromethane. The combined organic layer was dried over Na₂SO₄ and concentrated in vacuo. The residue was purified by flash chromatography on silica gel (hexanes : ethyl acetate = 20:1) to give the ether **234** (2.07 g, 9.4 mmol, 94 %).

¹HNMR (CDCl₃): δ 7.37 (m, 5H), 6.37 (m, 2H), 4.54 (s, 2H), and 4.00 (d, 2H, ³J = 5.0 Hz). IR (KBr): ν_{max} 3063, 2855, 1721, 1619, 1453, 1357, 1269, 1108, and 1071 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹³⁴

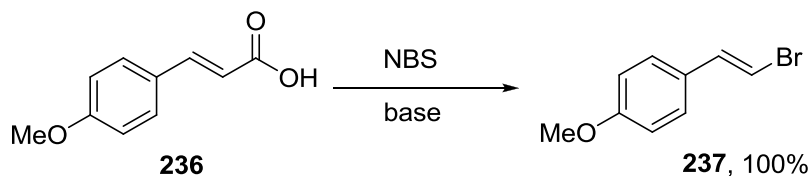
TBS protected allylic alcohol **235**.¹³⁵



5 g of t-butyldimethylchlorosilane (33.3 mmol, 1.5 eq.) was added to a mixture of 3.00 g (22 mmol) of bromoallylic alcohol and 2.50 g (37 mmol) of imidazole in 20 mL of DMF at 0 °C. The reaction was stirred overnight and then quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over Na₂SO₄ and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 5.17 g of compound **235** (20.5 mmol, 93%).

¹HNMR (CDCl₃): δ 6.28 (m, 2H), 4.12 (m, 2H), 0.89 (s, 9H), and 0.07 (s, 6H).
¹³CNMR (CDCl₃): δ 136.9, 106.2, 63.4, 26.0, and -5.1. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹³⁵

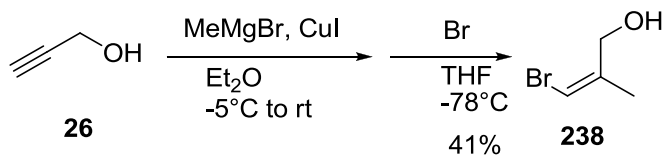
(E)-Vinyl bromide 237.¹³⁶



Triethylamine (3.0 mL, 27 mmol) was added to a solution of 4-methoxycinnamic acid (3.0 g, 17 mmol) in 20 mL of dichloromethane. After the mixture was stirred for 5 minutes at room temperature, *N*-bromosuccinimide (3.0 g, 17 mmol) was added. The solution was stirred for 5 minutes, and then the solvent was removed in vacuo. The mixture was subjected to column chromatography over silica gel (eluent 1% ethyl acetate in hexane) to afford the halostyrenes **237** (3.6 g, 17 mmol, 100%).

¹HNMR (CDCl₃): δ 7.22 (d, 2H, ³*J* = 7.6 Hz), 7.05 (d, 1H, d, 2H, ³*J* = 9.0 Hz), 6.86 (d, 2H, ³*J* = 7.6 Hz), 6.62 (d, 1H, d, 2H, ³*J* = 9.0 Hz), and 3.81 (s, 3H). ¹³CNMR (CDCl₃): δ 159.8, 136.7, 128.9, 127.5, 114.3, 104.1, and 55.5. IR (KBr): ν_{max} 3066, 2957, 2836, 1605, 1512, 1441, 1305, 1254, 1181, and 1029 cm⁻¹. The spectroscopic and other physical data reported are consistent with values reported in the literature.¹³⁶

(Z)-3-Bromo-2-methyl-2-Propen-1-ol 238.

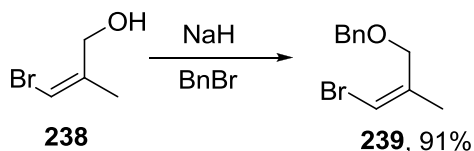


In 1 L round bottom flask, 1.5 mL (27 mmol) of propargyl alcohol **26** and 2 g (10.5 mmol) of dry CuI were mixed in 200 mL of Et₂O. 14 mL (35 mmol) of 2.5 M methyl magnesium bromide solution in diethyl ether (MeMgBr) was added dropwise to the reaction at 0 °C.¹³⁷ The first 6 mL was added very carefully. After the CH₄ bubbling stopped, the left 8 mL was added a little bit fast. After addition of MeMgBr, the reaction became very viscous. After 20 hours continuous stirring, the reaction mixture turned dark

green. 15 mL (16.5) of 1.1 M Br₂ solution in THF was added dropwise to the reaction mixture at -78 °C. The color of the reaction mixture first turned red from dark green, and then became yellow. The reaction mixture was stirred over night and then it was quenched by the saturated NH₄Cl solution. The organic layer was separated and the water layer was extract three times with CH₂Cl₂. The organic layers were combined and dried over Na₂SO₄. After solvents were removed by rotovap, the organic products were separated by a silica gel column with 5:1 hexanes and AcOEt to give 1.67 g (11 mmol) of yellow liquid **238** in 41% yield.

¹HNMR (CDCl₃): δ 5.947 (s, 1H), 4.21 (s, 2H), and 1.95 (s, 1H). ¹³CNMR (CDCl₃): δ 146.2, 108.8, 68.2, and 21.7. IR (KBr): ν_{max} 3337, 2912, 1629, 1432, 1375, 1223, 1134, and 1063 cm⁻¹.

Benzyl protected allylic alcohol **239**.

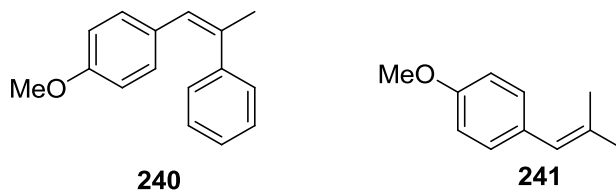


NaH (0.800 g, 60%, 20 mmol) was added to a solution of alcohol **238** (2.00 g, 13 mmol) in THF (50 mL) at 0 °C. After the mixture was stirred for 30 minutes, benzyl bromide (2.50 g, 14.6 mmol) and trace amounts of iodine were added. The mixture was stirred for 4 hours at room temperature and then it was poured into water. The mixture was extracted with diethyl ether. The combined organic layer was dried over Na₂SO₄ and concentrated in vacuo. The residue was purified by flash chromatography on silica gel (hexanes: ethyl acetate = 20:1) to give the benzyl ether **239** (2.92 g, 11.8 mmol, 91 %).

¹HNMR (CDCl₃): δ 7.35 (m, 5H), 6.09 (s, 1H), 4.51 (s, 2H), 4.19 (s, 2H), and 1.98 (s, 3H). ¹³CNMR (CDCl₃): δ 144.4, 138.2, 128.5, 127.9, 127.8, 76.2, 75.0, 72.3, and

22.0. IR (KBr): ν_{max} 3062, 2913, 2855, 1699, 1558, 1454, 1352, 1281, 1139, and 1093 cm^{-1} .

Coupling with iodobenzene to generate 240.



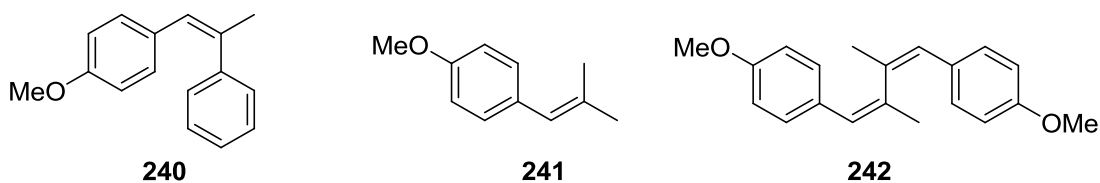
To a suspension of 275 mg (1.44 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 2 mL (3.2 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 1.5 mL (1.5 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 146 mg (0.50 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.5 mL (0.5 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 760 mg of iodobenzene was added to the reaction. The reaction solution turned white and then yellow. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 76 mg of compound **240** (0.34 mmol, 68%) and 22 mg of compound **241** (0.13 mmol, 27%).

240 $^1\text{H NMR}$ (CDCl_3): δ 7.53 (d, 2H, $^3J = 7.6$ Hz), 7.35 (m, 5H), 6.93 (d, 1H, $^3J = 7.6$ Hz), 6.80 (s, 1H), 3.85 (s, 3H), and 2.30 (s, 3H). $^{13}\text{C NMR}$ (CDCl_3): δ 158.4, 144.4,

126.1, 131.1, 130.5, 128.5, 127.4, 127.1, 126.1, 113.8, 55.4, and 17.6. IR (KBr): ν_{max} 2953, 2835, 1677, 1573, 1463, 1378, 1297, 1108, and 1034 cm^{-1} .

241 $^1\text{H NMR}$ (CDCl_3): δ 7.18 (d, 2H, $^3J = 8.8$ Hz), 6.89 (d, 2H, $^3J = 8.8$ Hz), 6.24 (s, 1H), 3.82 (s, 3H), 1.91 (s, 3H), and 1.88 (s, 3H).

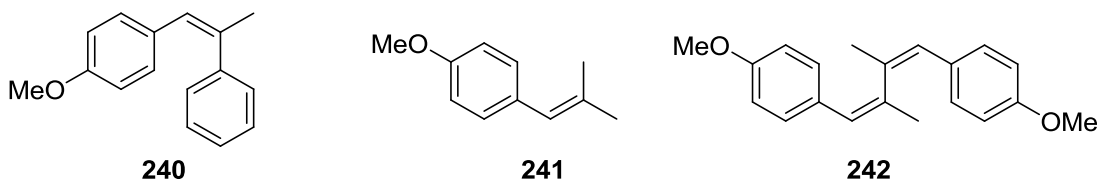
Attempt at coupling with phenyl triflate to produce 240.



To a suspension of 275 mg (1.44 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 2 mL (3.2 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 1.5 mL (1.5 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 146 mg (0.5 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.5 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 780 mg phenyl triflate was added to the reaction. The reaction solution became white and then yellow. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 65 mg of dimethyl substitution compound **241** (0.4 mmol, 80%) and 13 mg of dimerized compound **242** (0.08 mmol, 17%).

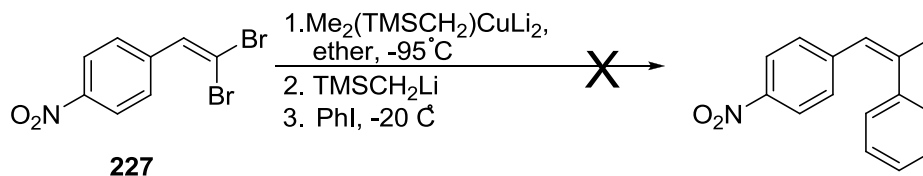
242 $^1\text{H NMR}$ (CDCl_3): δ 7.26 (d, 4H, $^3J = 8.8$ Hz), 6.91 (d, 4H, $^3J = 8.8$ Hz), 6.72 (s, 2H), 3.87 (s, 6H), and 2.13 (s, 3H). $^{13}\text{C NMR}$ (CDCl_3): δ 158.2, 138.0, 131.4, 130.7, 126.53, 113.7, 55.4, and 15.9. IR (KBr): ν_{max} 2952, 2835, 1604, 1509, 1462, 1371, 1248, 1178, and 1030 cm^{-1} .

Attempt at coupling with bromobenzene to generate 240.



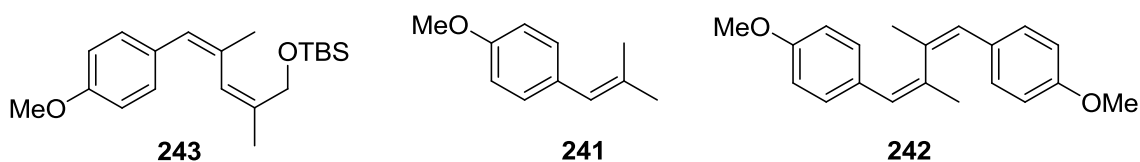
To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at $0\text{ }^\circ\text{C}$ till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at $0\text{ }^\circ\text{C}$. The solution was stirred for 15 min at $0\text{ }^\circ\text{C}$ and then cooled down to $-95\text{ }^\circ\text{C}$. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to $-79\text{ }^\circ\text{C}$ for 15 min and then 0.25 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to $-20\text{ }^\circ\text{C}$, 350 mg of bromobenzene was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 34 mg dimethyl substitution compound **241** (0.21 mmol, 85%) and 5 mg dimerized compound **242** (0.04 mmol, 14%).

Attempt at coupling using 1-(2,2-dibromovinyl)-4-nitrobenzene 227 as the substrate.



To a suspension of 275 mg (1.44 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 2 mL (3.2 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 1.5 mL (1.5 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 169 mg (0.5 mmol) of nitro-dibromide in 1 mL of diethyl ether was added to the reaction dropwise to give a green intermediate. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.5 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 760 mg of iodobenzene was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO₄ and concentrated in vacuo. The residues were separated by a silica gel column with 10:1 hexanes and ethyl acetate. The starting material decomposed.

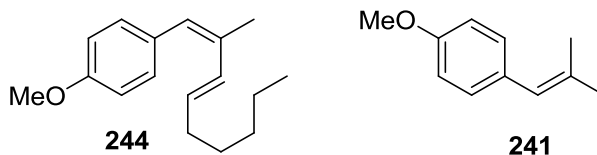
Attempt at coupling with TBS protected iodoallylic alcohol 228 to generate 243.



To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution

was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 480 mg of TBS iodoallylic alcohol **228** was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO₄ and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 36 mg of dimethyl substitution compound **241** (0.22 mmol, 88%) and 4.5 mg of dimerized compound **242** (0.3 mmol, 12%).

Diene **244**.

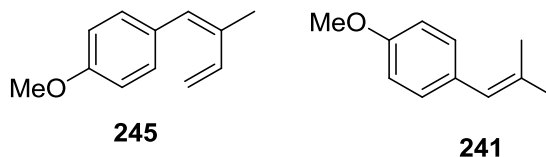


To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 400 mg of iodo-heptene was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO₄ and concentrated in

vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 44 mg of mixture of dimethyl substitution compound **241** and methyl vinyl product compound in ratio 6:1. After further purification, 6.5 mg of compound **244** (0.033 mmol, 13%) and 36 mg of compound (21 mmol, 85%) were obtained.

$^1\text{H NMR}$ (CDCl_3): δ 7.21 (d, 2H, $^3J = 9.0$ Hz), 6.87 (d, 2H, $^3J = 9.0$ Hz), 6.36 (s, 1H), 6.22 (d, 1H, $^3J = 16.0$ Hz), 5.75 (dt, 1H, $^3J = 8.0$ Hz, $^3J = 7.0$ Hz), 3.81 (s, 3H), 2.15 (dt, 2H, 1H, $^3J = 16.0$ Hz, $^3J = 8.0$ Hz), 1.97 (s, 3H), 1.31 (m, 6H), and 0.91 (t, 3H, $^3J = 7.0$ Hz). $^{13}\text{C NMR}$ (CDCl_3): δ 158.1, 135.3, 134.6, 130.9, 130.6, 129.9, 128.8, 113.7, 55.5, 33.2, 31.7, 30.6, 29.6, 22.8, and 14.3. IR (KBr): ν_{max} 2955, 2855, 1606, 1573, 1464, 1294, 1105, and 1037 cm^{-1} .

Diene **245**.

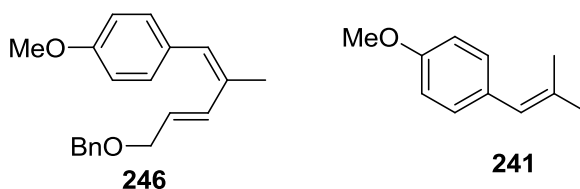


To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 300 mg of vinylbromide was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in

vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 43mg of mixture of dimethyl substitution compound **241** and vinyl substitution compound **245**. The further separation by TLC gave 19 mg of compound **245** (0.108 mmol, 43%) and 19 mg of compound **241** (0.123 mmol, 49%).

$^1\text{H NMR}$ (CDCl_3): δ 7.28 (d, 2H, $^3J = 8.8$ Hz), 6.91 (d, 2H, $^3J = 8.8$ Hz), 6.57 (dd, 1H, $^3J = 17.2$ Hz, $^3J = 10.8$ Hz), 6.50 (s, 1H), 5.28 (d, 1H, $^3J = 17.2$ Hz), 5.12 (d, 1H, $^3J = 10.8$ Hz), 3.84 (s, 3H), and 2.02 (d, 3H, $^4J = 1.2$ Hz). $^{13}\text{C NMR}$ (CDCl_3): δ 158.4, 142.1, 133.9, 133.7, 131.3, 130.5, 128.6, 113.7, 55.4, and 13.4. IR (KBr): ν_{max} 3001, 2953, 2835, 1605, 1572, 1441, 1376, 1299, 1110, and 1035 cm^{-1} .

Diene **246**.

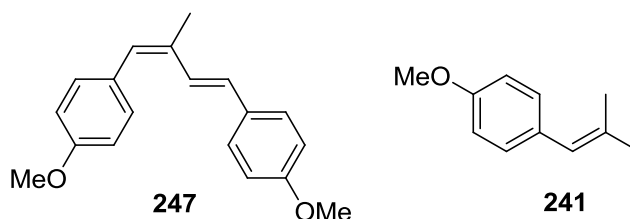


To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 350 mg of vinylbromide **234** was added to the reaction. The reaction solution turned white and then yellow. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined

organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 41 mg of vinyl methyl substitution compound **246** (0.14 mmol, 56%) and 15 mg of dimethyl substitution compound **241** (0.095 mmol, 38%).

$^1\text{H NMR}$ (CDCl_3): δ 7.37 (m, 5H), 7.25 (d, 2H, $^3J = 9.0$ Hz), 6.89 (d, 2H, $^3J = 9.0$ Hz), 6.47 (s, 1H), 6.44 (d, 1H, $^3J = 15.5$ Hz), 5.85 (dt, 1H, $^3J = 15.5$ Hz, $^3J = 6.0$ Hz), 4.56 (s, 2H), 4.15 (d, 1H, $^3J = 6.0$ Hz), 3.82 (s, 3H), and 2.01 (s, 3H). $^{13}\text{C NMR}$ (CDCl_3): δ 158.5, 138.5, 133.8, 131.5, 130.7, 130.6, 130.5, 128.6, 128.0, 127.8, 124.6, 113.8, 72.3, 71.2, 55.4, and 14.0. IR (KBr): ν_{max} 3425, 3029, 2931, 2854, 1606, 1509, 1453, 1249, 1176, and 1090 cm^{-1} .

Diene **247**.

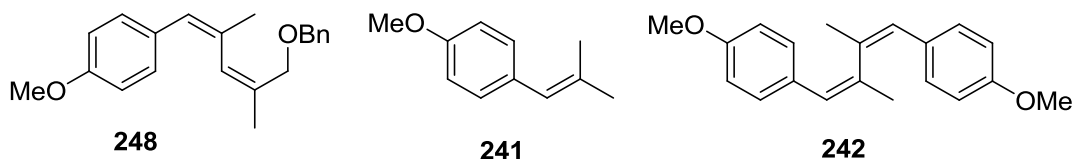


To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at $0\text{ }^\circ\text{C}$ till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at $0\text{ }^\circ\text{C}$. The solution was stirred for 15 min at $0\text{ }^\circ\text{C}$ and then cooled down to $-95\text{ }^\circ\text{C}$. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to $-79\text{ }^\circ\text{C}$ for 15 min and then 0.25 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to $-20\text{ }^\circ\text{C}$, 350 mg of vinylbromide **237** was added to the reaction. The reaction solution turned white and then yellow. The

reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO₄ and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 27 mg of vinyl methyl substitution compound **247** (0.14 mmol, 57%) and 14 mg of dimethyl substitution compound **241** (0.087 mmol, 35%).

¹HNMR (CDCl₃): δ 7.40 (d, 2H, ³J = 8.8 Hz), 7.28 (d, 2H, ³J = 8.8 Hz), 6.89 (m, 5H), 6.58 (d, 1H, ³J = 15.6 Hz), 6.47 (s, 1H), 3.83 (s, 3H), 3.82 (s, 3H), and 2.12 (d, 3H, ⁴J = 0.8 Hz). ¹³CNMR (CDCl₃): δ 159.1, 158.4, 134.7, 132.7, 131.1, 130.9, 130.8, 130.6, 127.6, 126.9, 114.3, 113.8, 55.53, 55.50, and 14.1. IR (KBr): ν_{max} 3375, 2954, 1604, 1510, 1462, 1303, 1255, 1178, and 1029 cm⁻¹.

Diene **248**.

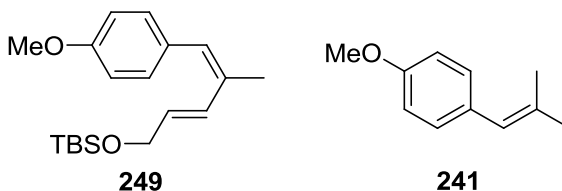


To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 400 mg of vinylbromide **239** was added to the reaction. The reaction solution turned white and then yellow. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by

water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 22 mg of vinyl methyl substitution compound **248** (0.073 mmol, 29%), 24 mg of dimethyl substitution compound **241** (0.15 mmol, 60%), and 4 mg of dimerized compound **242** (0.025 mmol, 10%).

$^1\text{H NMR}$ (CDCl_3): δ 7.32 (m, 5H), 7.18 (2H, $^3J = 8.8$ Hz), 6.88 (d, 2H, $^3J = 8.8$ Hz), 6.23 (s, 1H), 6.01 (s, 1H), 4.48 (s, 2H), 4.21 (s, 2H), 3.83 (s, 3H), 1.93 (s, 3H), and 1.92 (s, 3H). $^{13}\text{C NMR}$ (CDCl_3): δ 158.3, 138.7, 134.3, 134.0, 133.2, 130.7, 130.4, 128.9, 128.5, 128.0, 127.7, 113.7, 72.3, 69.7, 55.4, 22.3, and 19.0. IR (KBr): ν_{max} 2930, 2852, 1605, 1547, 1452, 1296, 1176, and 1035 cm^{-1} .

Diene **249**.

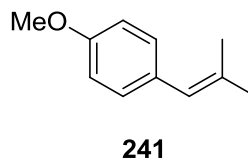
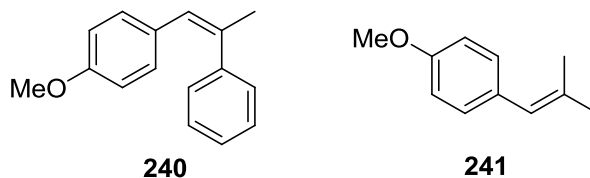


To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1 M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 420 mg of vinylbromide **235** was added to the reaction. The reaction solution turned white and then yellow. The

reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO₄ and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 47 mg mixture of vinyl methyl substitution product and dimethyl substitution product. The mixture was dissolved in 2 mL of THF and then the solution was added 1 mL of 1 M tetrabutylammonium fluoride solution in tetrahydrofuran at 0 °C. The reaction was stirred for 6 hours at 0 °C and then it was quenched by water. The water solution was extracted three times with diethyl ether. The organic solutions were combined and dried over MgSO₄. The solution was concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 6.1 mg of deprotected vinyl methyl substitution compound **249** (deprotected compound, 0.03 mmol, 12%) and 33 mg of dimethyl substitution compound **241** (0.2 mmol, 81%).

¹HNMR (CDCl₃): δ 7.22 (d, 2H, ³J = 8.8 Hz), 6.86 (d, 2H, ³J = 8.8 Hz), 6.45 (s, 1H), 6.40 (d, 1H, ³J = 16.0 Hz), 5.87 (dt, 1H, ³J = 16.0 Hz, ³J = 5.6 Hz), 4.25 (d, 2H, ³J = 5.6 Hz), 3.79 (s, 3H), and 2.00 (s, 3H). ¹³CNMR (CDCl₃): δ 197.0, 185.8, 137.3, 131.6, 130.6, 127.1, 113.8, 94.6, 64.1, 55.4, and 20.4. IR (KBr): ν_{max} 3388, 2956, 2919, 1708, 1665, 1510, 1462, 1364, 1299, 1175, and 1035 cm⁻¹.

(Trimethylsilyl)acetyl stabilizing coupling to generate **240**.

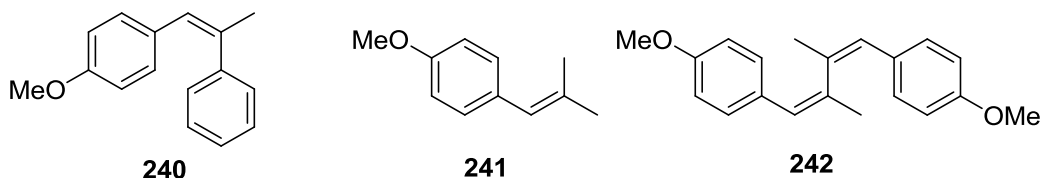


To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of lithium

(trimethylsilyl)acetylide in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added. After the reaction was slowly warmed up to -20 °C, 400 mg of iodobenzene was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO₄ and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 19 mg of dimethyl substitution compound **241** (0.12 mmol, 46%) and 24 mg of methyl phenyl compound **240** (0.11 mmol, 43%).

¹HNMR (CDCl₃): δ 7.53 (d, 2H, ³J = 7.6 Hz), 7.35 (m, 5H), 6.93 (d, 1H, ³J = 7.6 Hz), 6.80 (s, 1H), 3.85 (s, 3H), and 2.30 (s, 3H). ¹³CNMR (CDCl₃): δ 158.4, 144.4, 136.1, 131.1, 130.5, 128.5, 127.4, 127.1, 126.1, 113.8, 55.4, and 17.6. IR (KBr): ν_{max} 2953, 2835, 1677, 1509, 1463, 1378, 1249, 1176, and 1034 cm⁻¹.

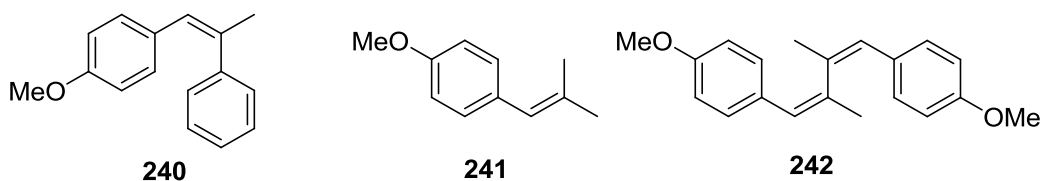
Attempt at (trimethylsilyl)acetyl stabilizing coupling with phenyl triflate to generate **240.**



To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added to the reaction at 0 °C. The solution was

stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added. After the reaction was slowly warmed up to -20 °C, 380 mg of phenyl triflate was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO₄ and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 31 mg of dimethyl substitution compound **241** (0.18 mmol, 72%) and 6 mg of dimerized compound **242** (0.04 mmol, 16%).

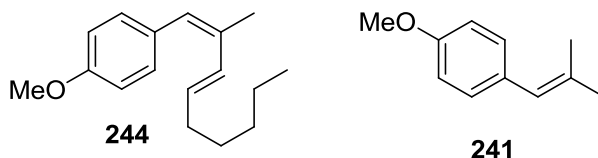
Attempt at (trimethylsilyl)acetyl stabilizing coupling with bromobenzene to generate 240.



To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added. After the reaction was slowly warmed up to -20 °C, 350 mg of bromobenzene was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The

reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 34 mg of dimethyl substitution compound **241** (0.21 mmol, 84%) and 5 mg of dimerized compound **242** (0.035 mmol, 14%).

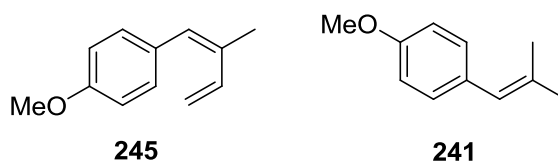
(Trimethylsilyl)acetyl stabilizing coupling to generate 244.



To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added. After the reaction was slowly warmed up to -20 °C, 400 mg of trans 1-iodo-1-heptene was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 43 mg of a mixture of dimethyl substitution compound **241** and methyl vinyl compound **244** in ratio 9:1. The further purification by TLC gave 4.6 mg of compound **244** (0.025 mmol, 10%) and 37 mg of compound **241** (0.22 mmol, 89%).

^1H NMR (CDCl_3): δ 7.21 (d, 2H, $^3J = 9.0$ Hz), 6.87 (d, 2H, $^3J = 9.0$ Hz), 6.36 (s, 1H), 6.22 (d, 1H, $^3J = 16.0$ Hz), 5.75 (dt, 1H, $^3J = 8.0$ Hz, $^3J = 7.0$ Hz), 3.81 (s, 3H), 2.15 (dt, 2H, 1H, $^3J = 16.0$ Hz, $^3J = 8.0$ Hz), 1.97 (s, 3H), 1.31 (m, 6H), and 0.91 (t, 3H, $^3J = 7.0$ Hz). ^{13}C NMR (CDCl_3): δ 158.1, 135.3, 134.6, 130.9, 130.6, 129.9, 128.8, 113.7, 55.5, 33.2, 31.7, 30.6, 29.6, 22.8, and 14.3. IR (KBr): ν_{max} 2955, 2855, 1606, 1573, 1441, 1294, 1105, and 1037 cm^{-1} .

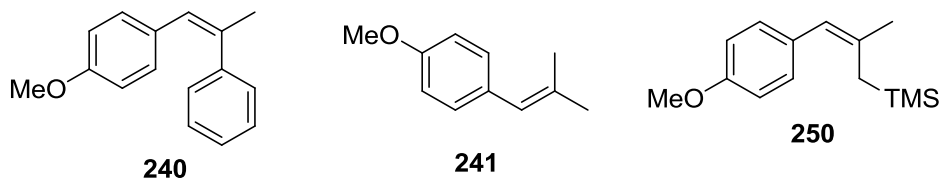
(Trimethylsilyl)acetyl stabilizing coupling to generate 245.



To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added. After the reaction was slowly warmed up to -20 °C, 300 mg of vinylbromide was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 41 mg of a mixture of dimethyl substitution compound **241** and vinyl substitution compound **245**. The further separation by TLC gave 14 mg of compound **245** (0.078 mmol, 31%) and 26 mg of compound **241** (0.12 mmol, 49%).

^1H NMR (CDCl_3): δ 7.28 (d, 2H, $^3J = 8.8$ Hz), 6.91 (d, 2H, $^3J = 8.8$ Hz), 6.57 (dd, 1H, $^3J = 17.2$ Hz, $^3J = 10.8$ Hz), 6.50 (s, 1H), 5.28 (d, 1H, $^3J = 17.2$ Hz), 5.12 (d, 1H, $^3J = 10.8$ Hz), 3.84 (s, 3H), and 2.02 (d, 3H, $^4J = 1.2$ Hz). ^{13}C NMR (CDCl_3): δ 158.4, 142.1, 133.9, 133.7, 131.3, 130.5, 128.6, 113.7, 55.4, and 13.4. IR (KBr): ν_{max} 3001, 2835, 1605, 1572, 1463, 1376, 1299, 1176, and 1035 cm^{-1} .

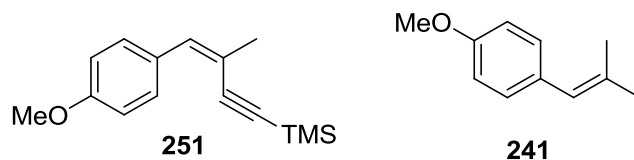
Ligand-transfer coupling product 250.



To a suspension of 275 mg (1.44 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (3.2 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 3.0 mL (3.0 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 146 mg (0.75 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.5 mL of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -20 °C, 760 mg of iodobenzene was added to the reaction. The reaction solution became white and then yellow. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give a mixture of dimethyl substitution compound **241**, methyl TMS-methylene substitution compound **250**, and methyl phenyl substitution compound **240** in total 99% yield. The ratio determined by NMR is 1:1:1. Further purification by TLC gave 30 mg of compound **250** (0.25 mmol, 33%), 29 mg of compound **240** (0.25 mmol, 33%), and 21 mg of compound **241** (0.25 mmol, 33%).

^1H NMR (CDCl_3): δ 7.14 (d, 2H, $^3J = 8.8$ Hz), 6.85 (d, 2H, $^3J = 8.8$ Hz), 6.036 (s, 1H), 3.81 (s, 3H), 1.84 (s, 3H), 1.26 (s, 2H), and 0.06 (s, 9H). ^{13}C NMR (CDCl_3): δ 158.8, 129.8, 127.5, 126.3, 122.1, 113.6, 55.4, 29.9, 19.7, and -1.2. IR (KBr): ν_{max} 2953, 1727, 1607, 1509, 1464, 1287, 1174, and 1037 cm^{-1} .

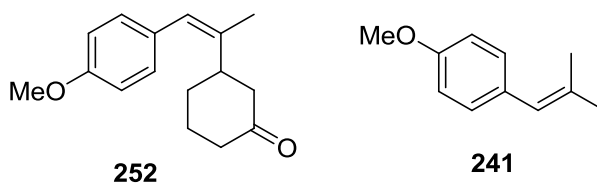
Ligand-transfer coupling product **251**.



To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 0.5 mL (0.8 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 1.5 mL (1.5 mmol) of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL of 1M solution of lithium (trimethylsilyl)acetylide in hexanes was added. After the reaction was slowly warmed up to -20 °C, 400 mg of iodobenzene was added to the reaction. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO_4 and concentrated in vacuo. The residues were separated by a silica gel column with 100:1 hexanes and diethyl ether to give 43 mg of a mixture of dimethyl substitution compound **241** and ligand-transfer compound **251** in ratio 4:1 (NMR). The further separation by TLC gave 11 mg of compound **251** (0.05 mmol, 20%) and 31 mg of compound **241** (0.19 mmol, 76%).

^1H NMR (CDCl_3): δ 7.23 (d, 2H, $^3J = 8.7$ Hz), 6.88 (d, 2H, $^3J = 8.7$ Hz), 6.85 (s, 1H), 3.81 (s, 3H), 2.06 (s, 3H), and 0.21 (s, 9H). ^{13}C NMR (CDCl_3): δ 158.8, 136.7, 130.4, 129.5, 117.9, 113.8, 109.6, 92.2, 55.5, 19.4, and 0.3. IR (KBr): ν_{max} 2957, 2835, 2139, 1604, 1509, 1464, 1250, 1177, and 1036 cm^{-1} .

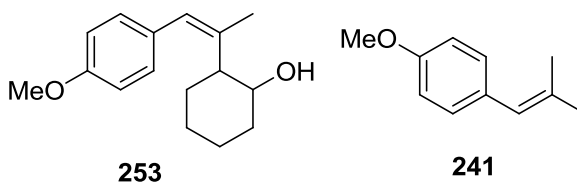
Ketone 252.



To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL (0.25 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -40 °C, 400 mg of 1-cyclohexenone was added to the reaction. The reaction solution turned white and then yellow. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO₄ and concentrated in vacuo. The residues were separated by a silica gel column with 40:1 hexanes and diethyl ether to give 38 mg of 1,4-adduct **252** (0.16 mmol, 63%) and 14 mg of dimethyl substitution compound **241** (0.087 mmol, 35%).

¹HNMR (CDCl₃): δ 7.16 (d, 1H, ³J = 8.8 Hz), 6.86 (d, 1H, ³J = 8.8 Hz), 6.25 (s, 1H), 3.80 (s, 3H), 2.47 (m, 3H), 2.35 (m, 1H), 2.20 (m, 1H), 1.98 (m, 1H), 1.85 (s, 3H), 1.84(m, 1H), and 1.71 (m, 2H). ¹³CNMR (CDCl₃): δ 211.7, 158.1, 139.0, 130.7, 130.2, 124.4, 113.7, 55.4, 48.5, 47.0, 41.5, 30.3, and 25.4. IR (KBr): ν_{max} 2927, 2855, 1737, 1263, 1201, 1121, and 1017 cm⁻¹

Alcohol 253.



To a suspension of 145 mg (0.75 mmol) of copper (I) iodide in 1.5 mL of diethyl ether, 1 mL (1.6 mmol) of 1.6 M solution of methyl lithium in diethyl ether was added at 0 °C till a transparent solution was given. 0.75 mL (0.75 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added to the reaction at 0 °C. The solution was stirred for 15 min at 0 °C and then cooled down to -95 °C. A solution of 73 mg (0.25 mmol) of compound **226** in 1 mL of diethyl ether was added to the reaction dropwise to give a pale yellow solution. The reaction was allowed to slowly warm up to -79 °C for 15 min and then 0.25 mL (0.25 mmol) of 1M solution of trimethylsilylmethyl lithium in hexanes was added. After the reaction was slowly warmed up to -60 °C, 400 mg of cyclohexene oxide was added to the reaction. The reaction solution turned white and then yellow. The reaction was stirred for 4 hours to give a black suspension. The reaction was quenched by water. The water solution was extracted three times with diethyl ether. The combined organic layer was dried over MgSO₄ and concentrated in vacuo. The residues were separated by a silica gel column with 10:1 hexanes and diethyl ether to give 31 mg of ring-opening compound **253** (0.13 mmol, 51%) and 14 mg of dimethyl substitution compound **241** (0.11 mmol, 45%).

¹HNMR (CDCl₃): δ 7.21 (d, 2H, ³J = 8.4 Hz), 6.87 (d, 2H, ³J = 8.4 Hz), 6.38 (s, 1H), 3.81s (s, 3H), 3.54 (m, 1H), 2.06 (m, 2H), 1.84 (s, 3H), 1.77 (m, 2H), 1.42 (m, 2H), and 1.35 (m, 3H). ¹³CNMR (CDCl₃): δ 158.2, 137.8, 130.5, 130.3, 127.5, 113.7, 70.8, 57.3, 55.4, 34.3, 30.3, 25.9, 25.1, and 14.5. IR (KBr): ν_{max} 3425, 2929, 2855, 1607, 1573, 1449, 1295, 1109, and 1062 cm⁻¹

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- 2 Parker, K. A.; Lim, Y.-H. "The Total Synthesis of (-)-SNF4435 C and (+)-SNF4435 D." *J. Am. Chem. Soc.*, **2004**, *126*, 15968–69
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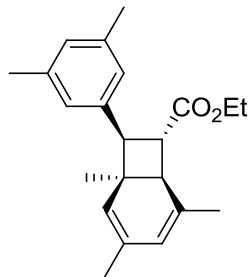
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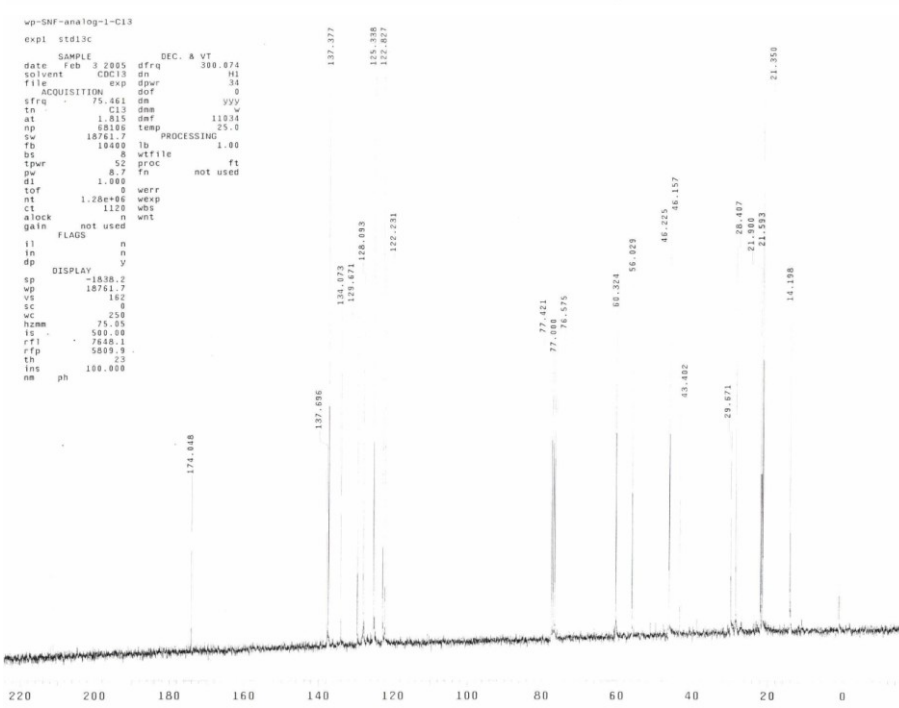
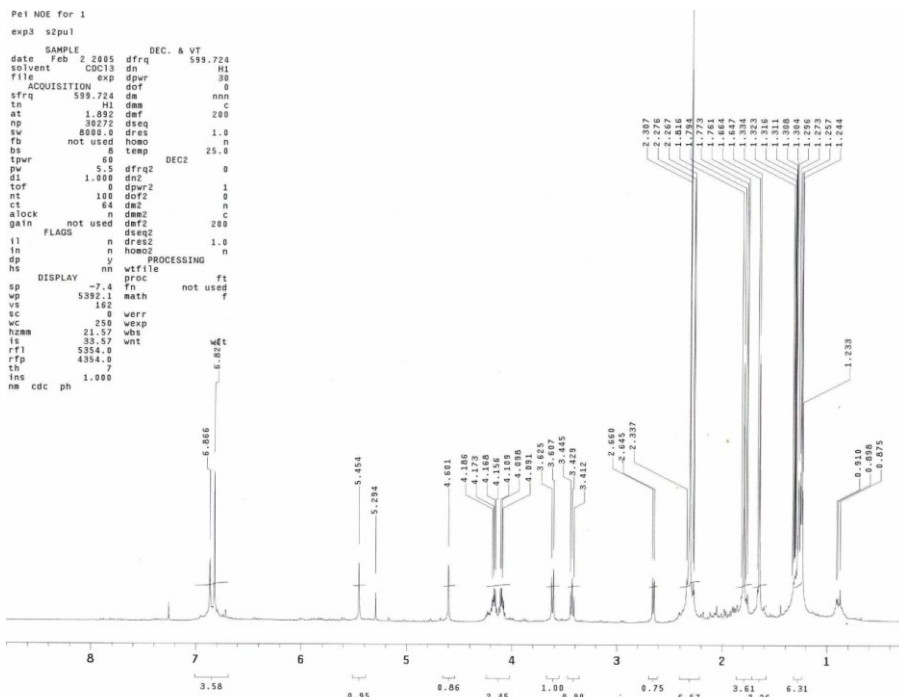
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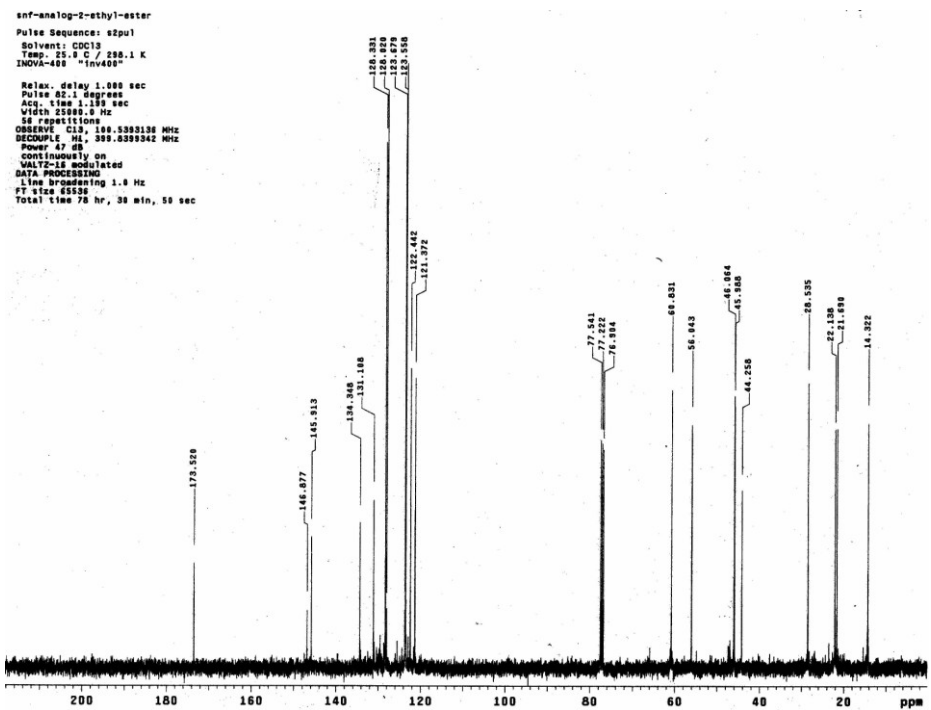
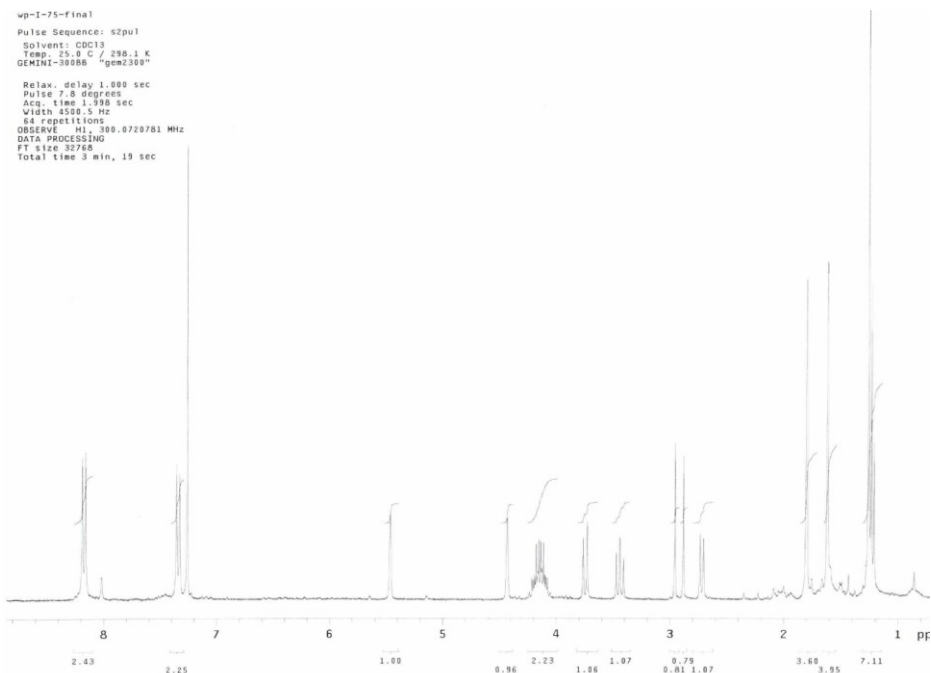
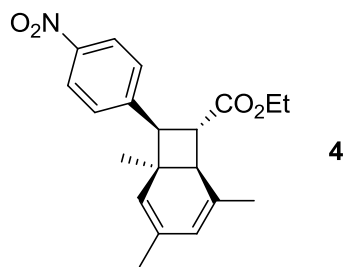
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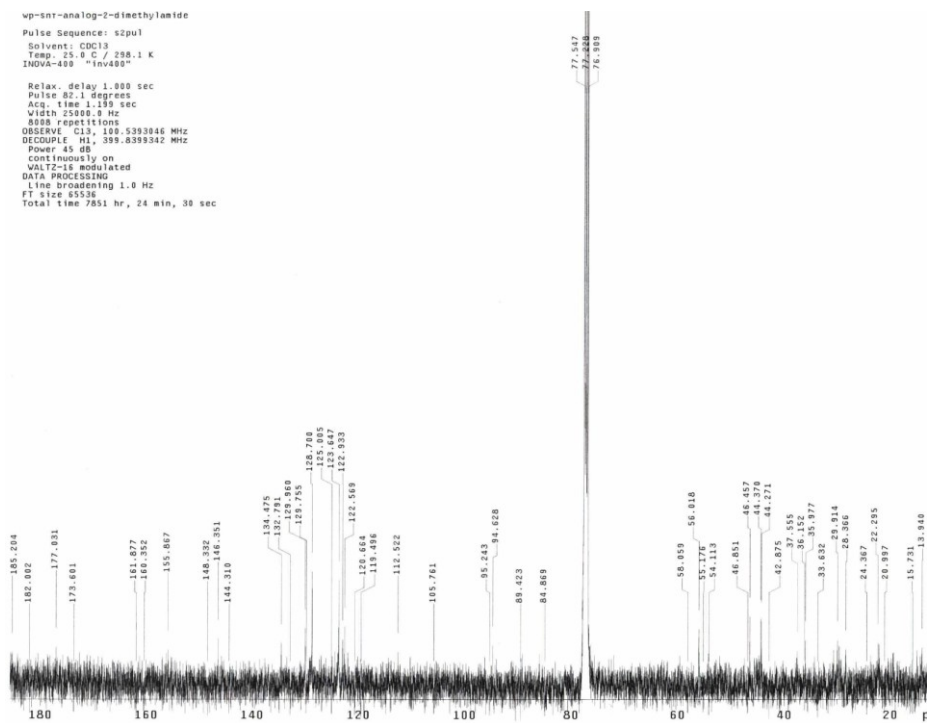
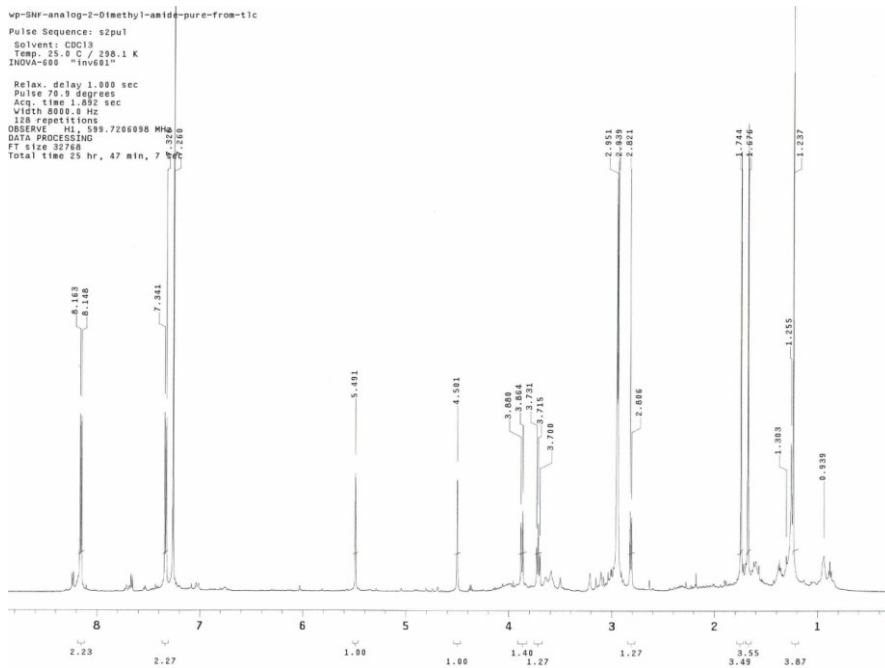
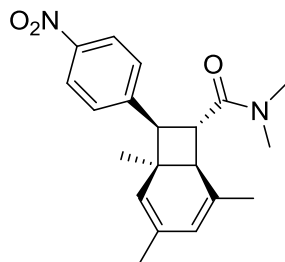
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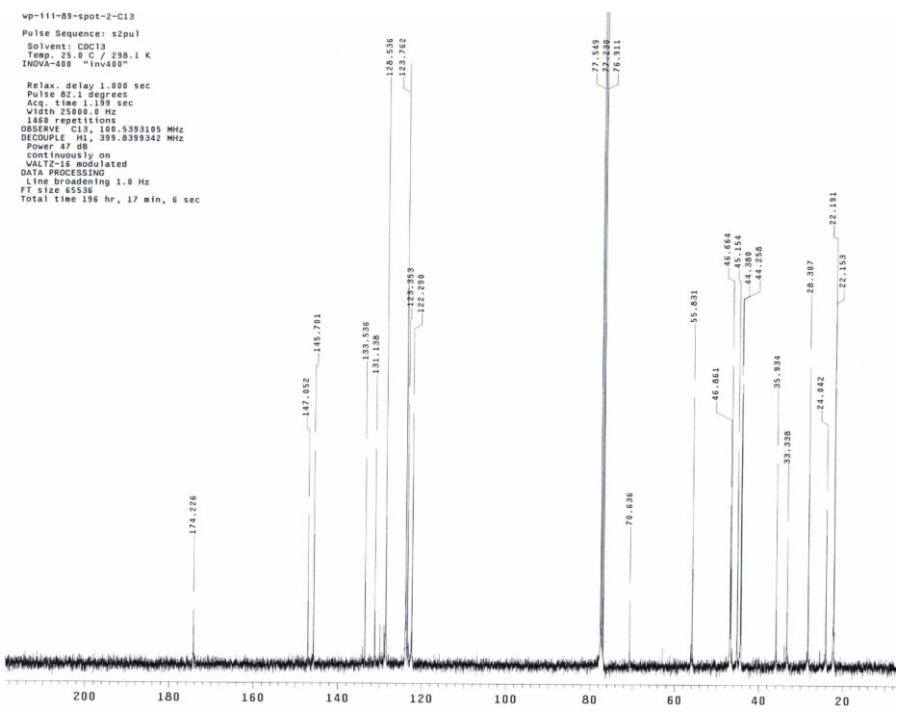
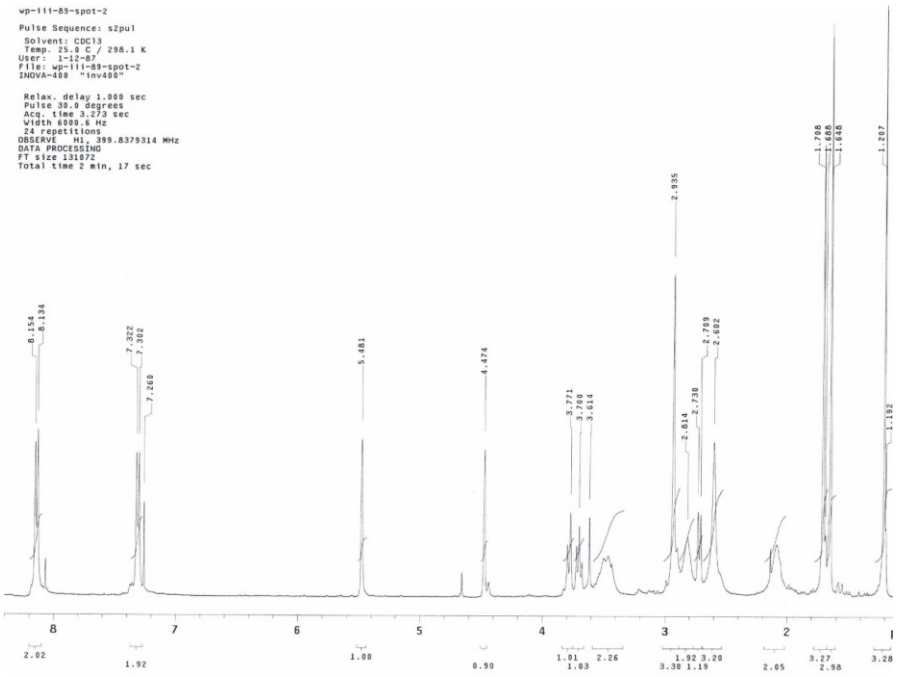
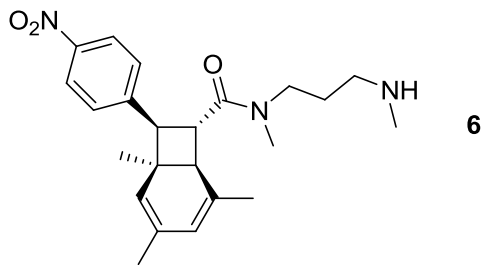


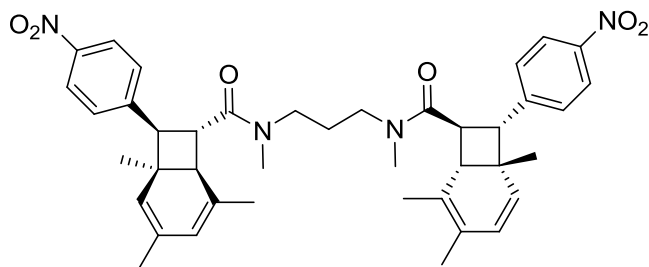
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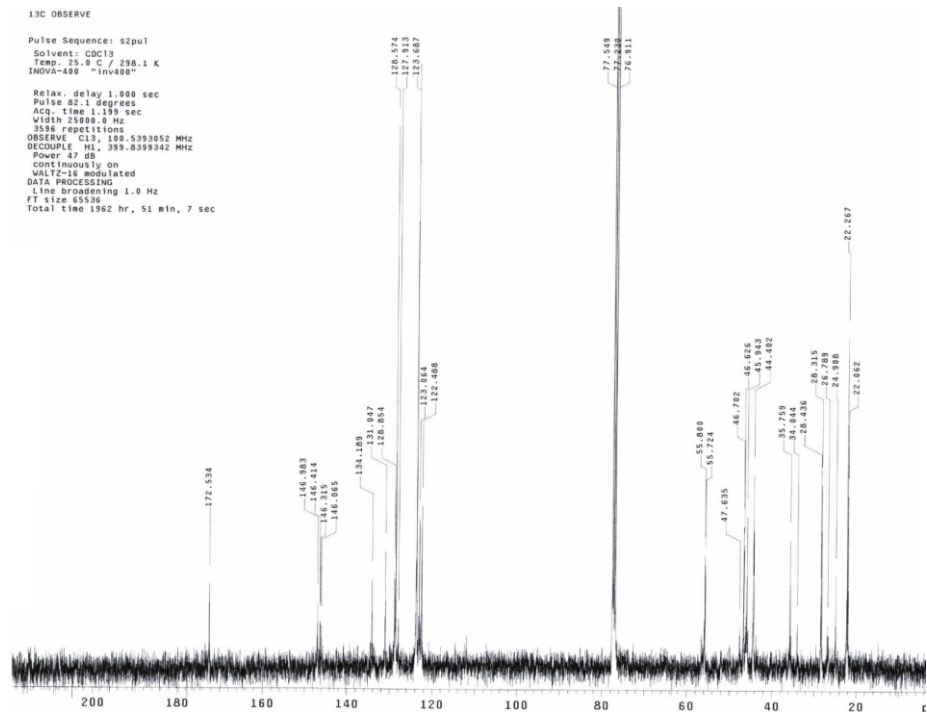
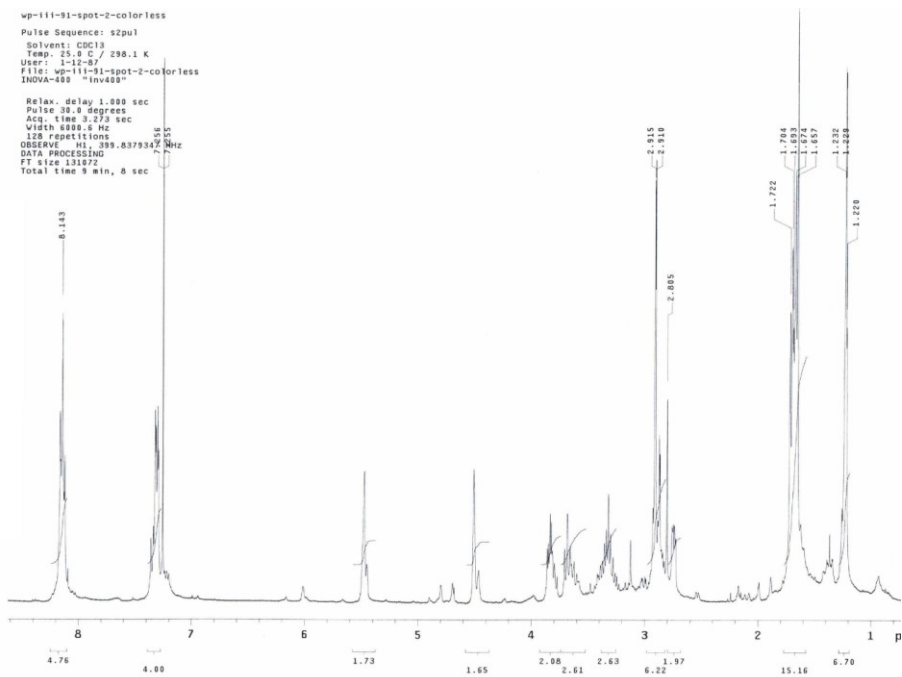


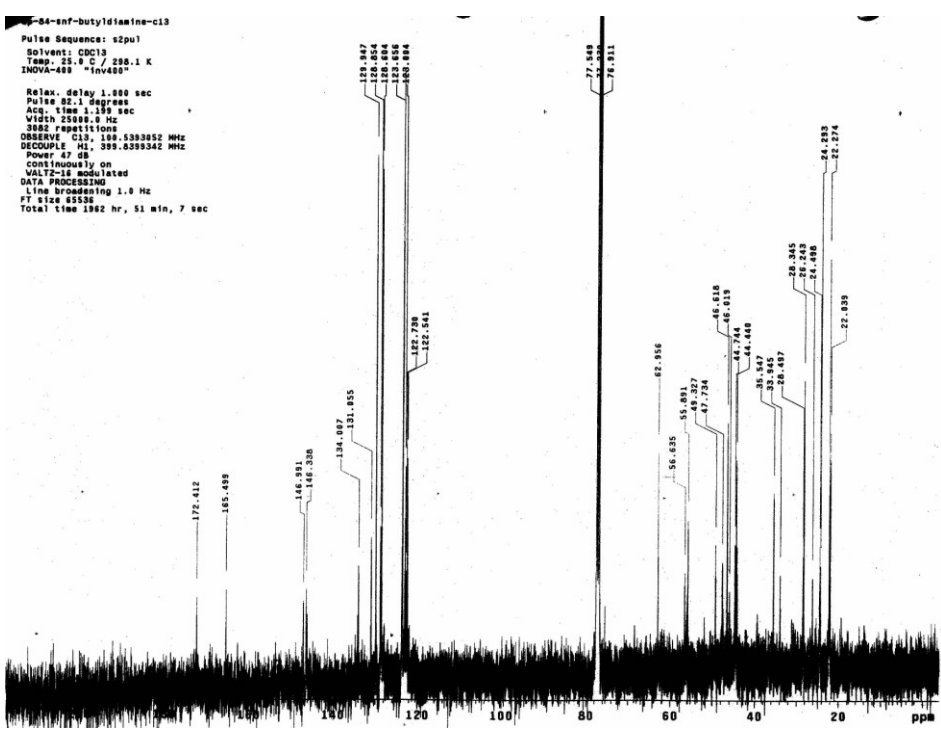
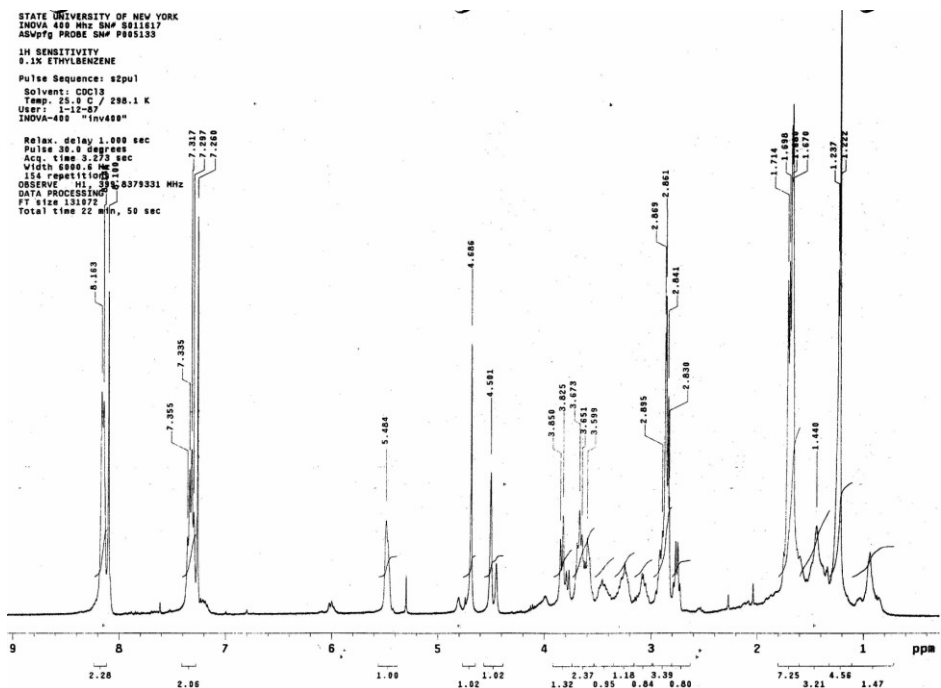
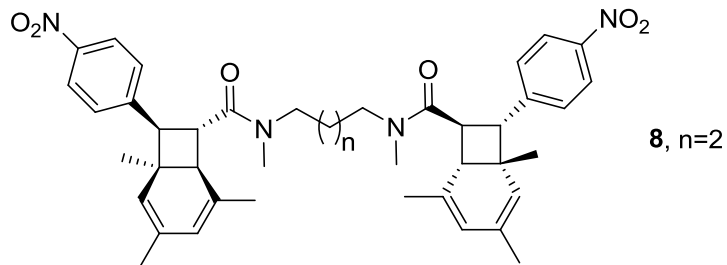


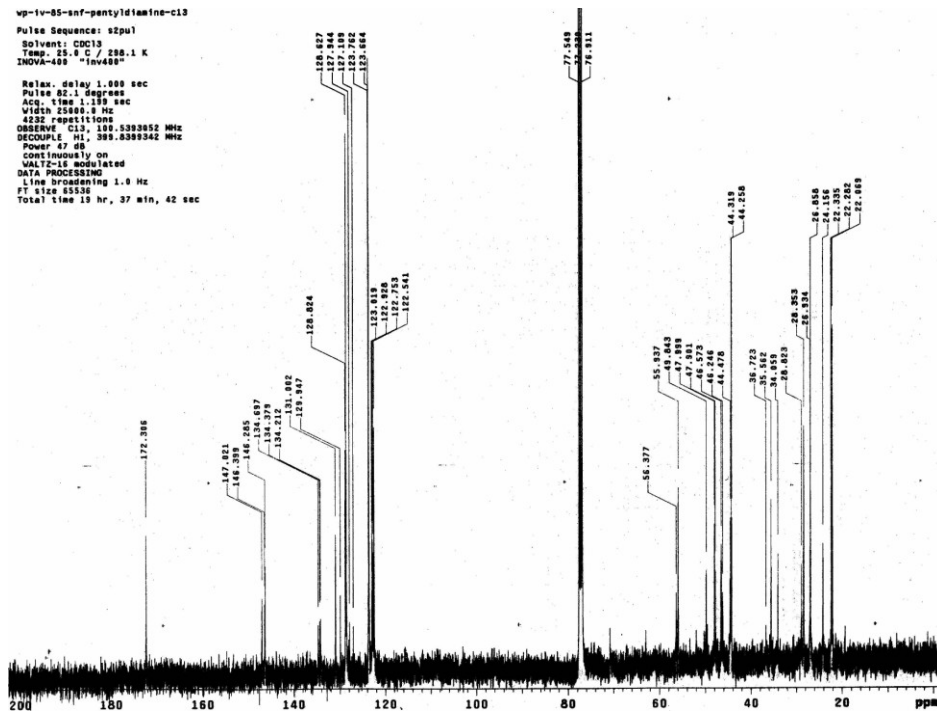
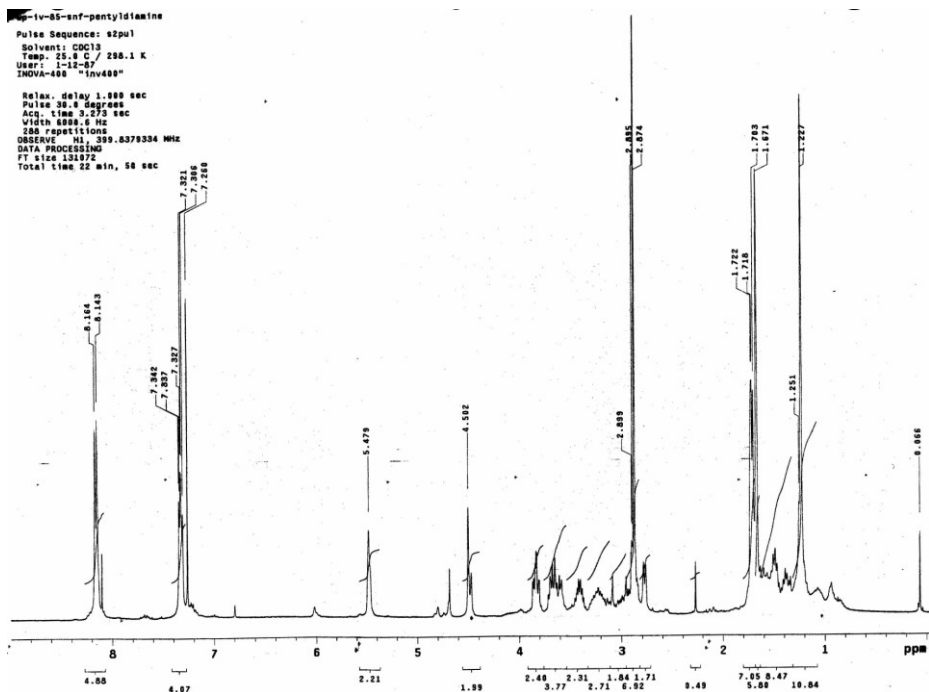
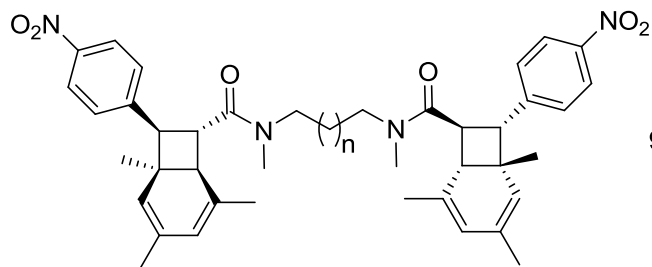


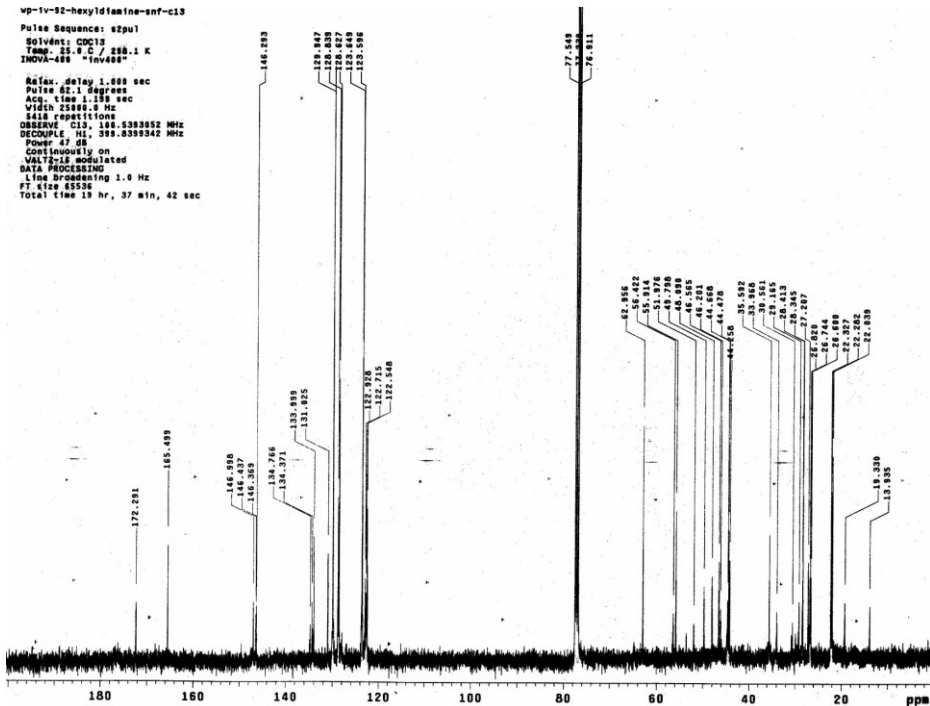
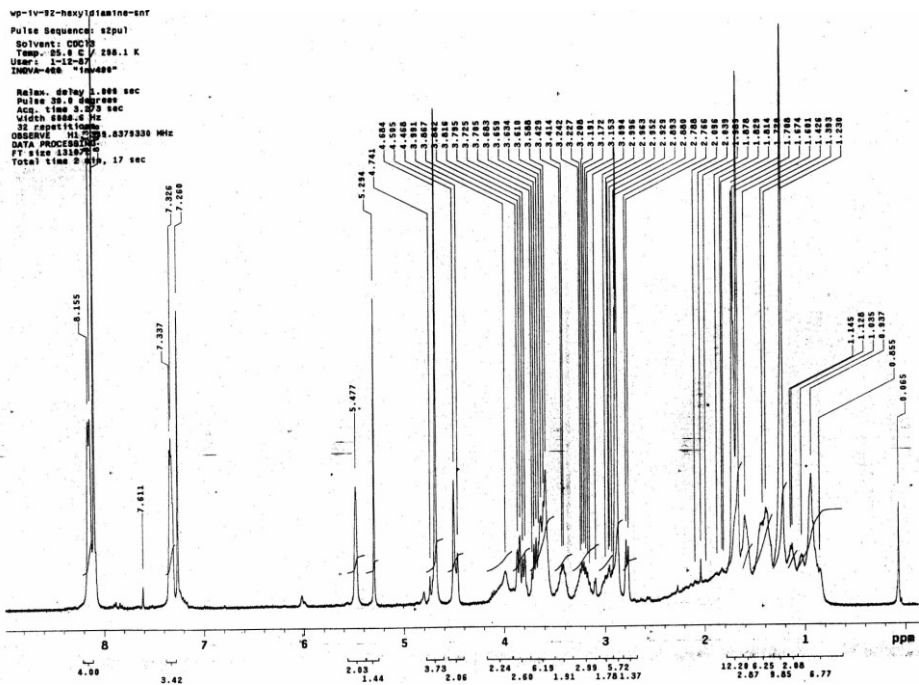
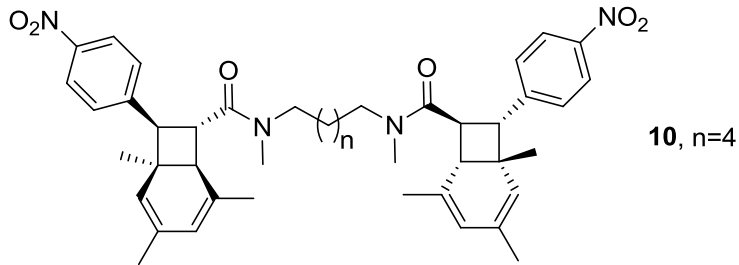


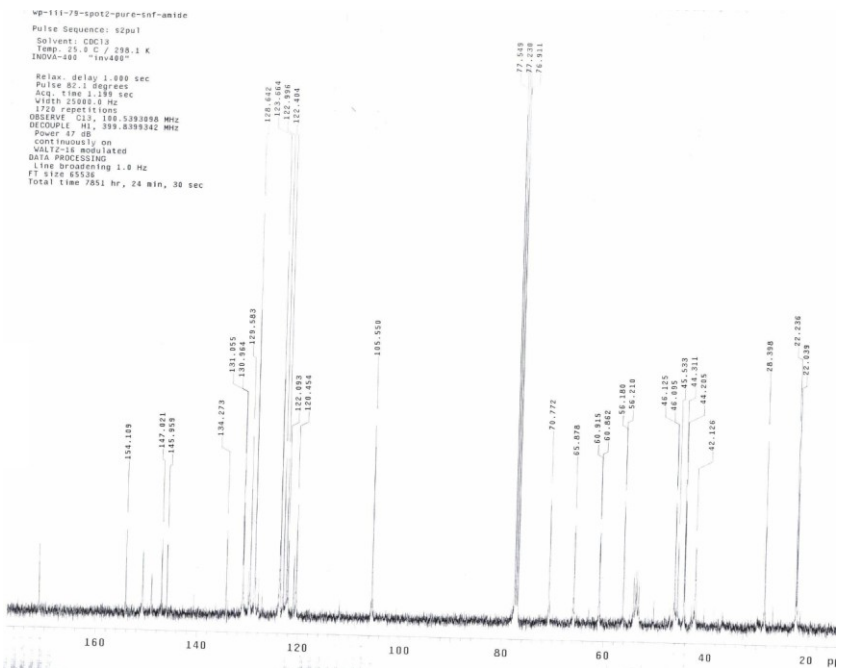
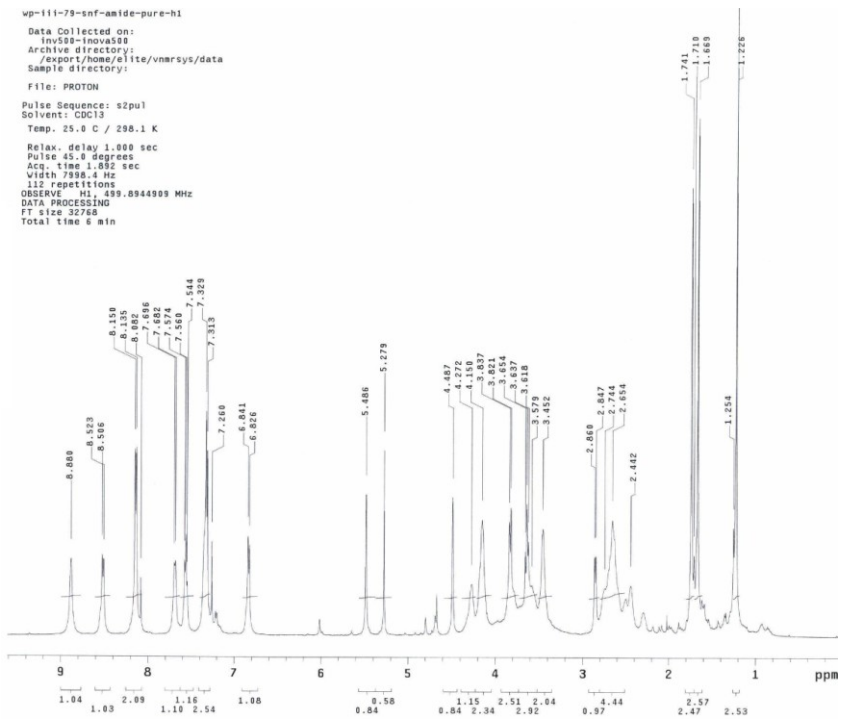
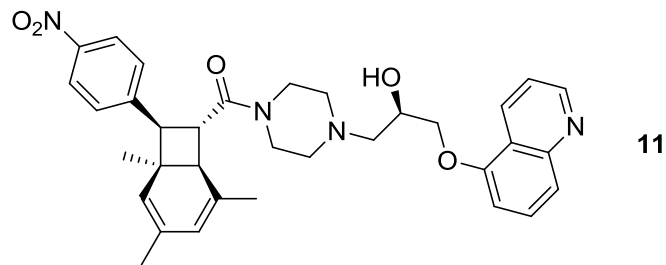
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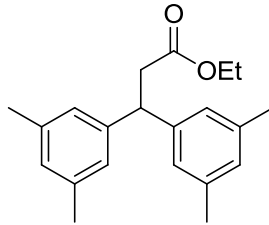




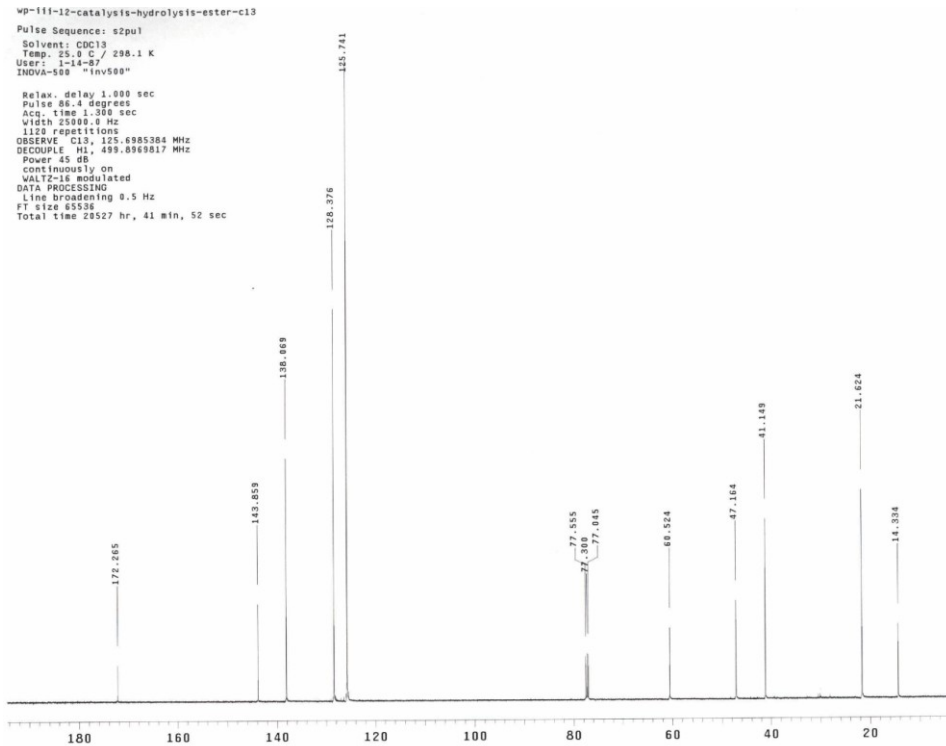
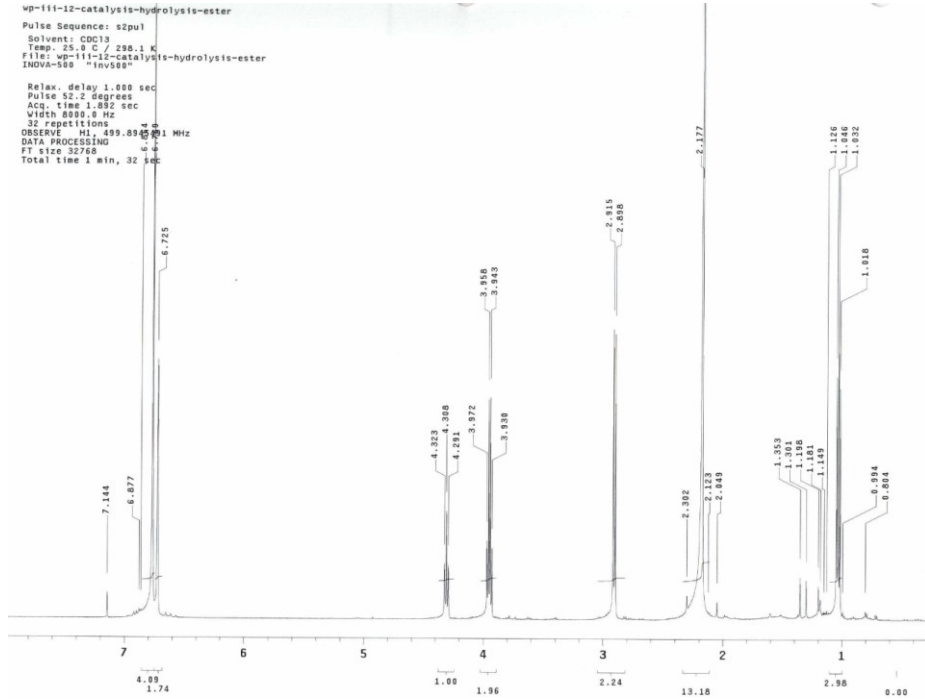


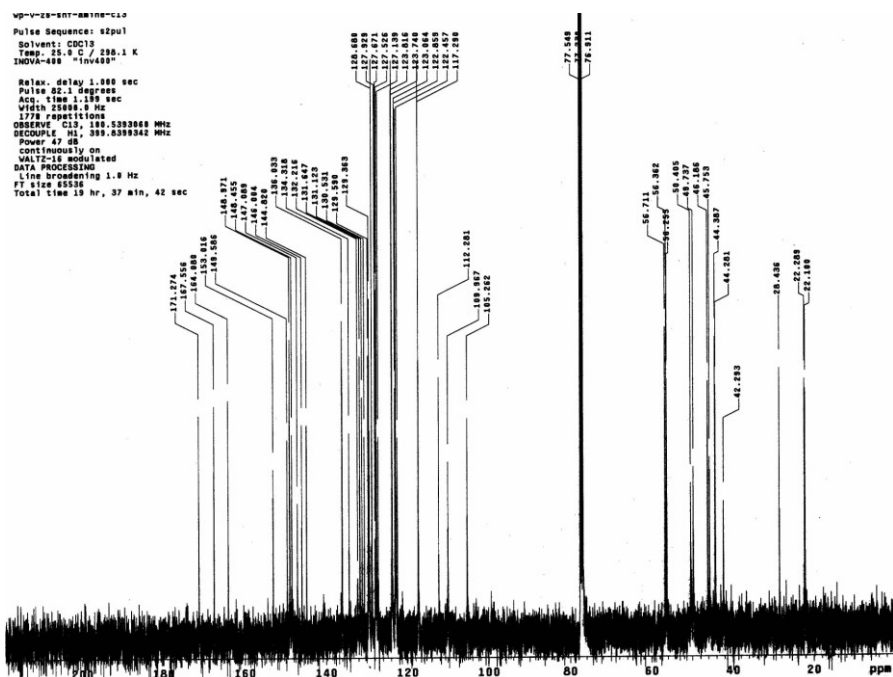
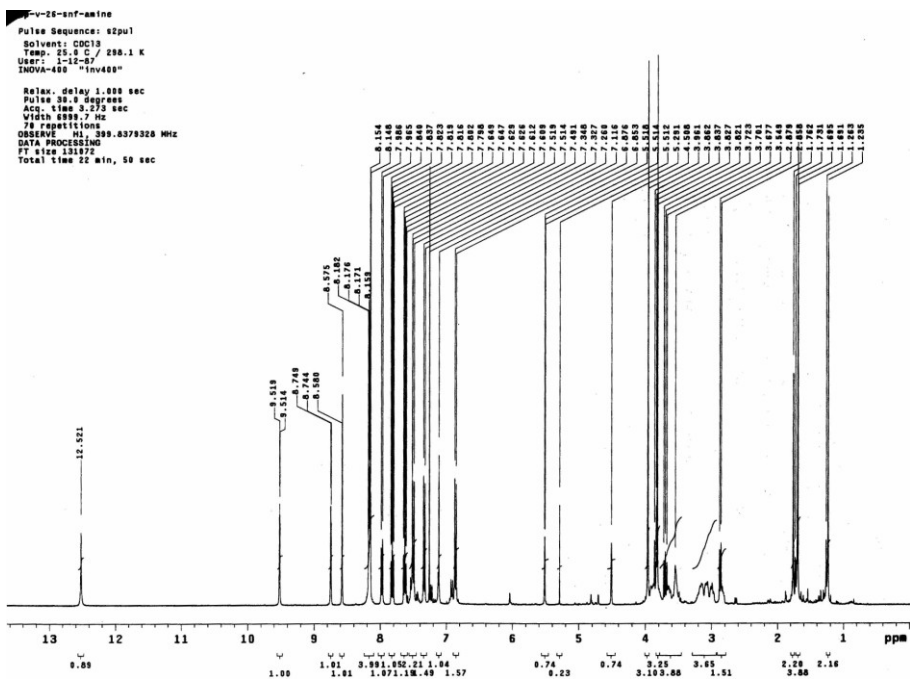
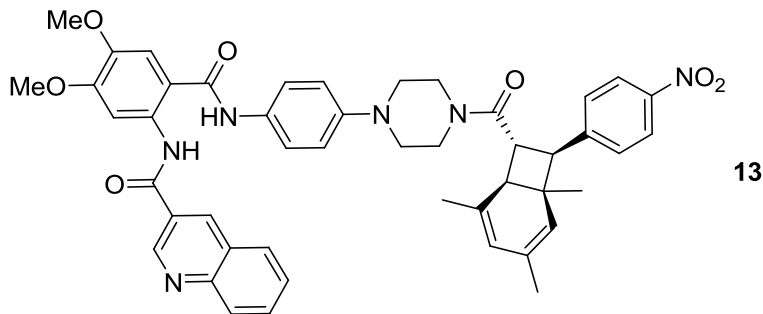


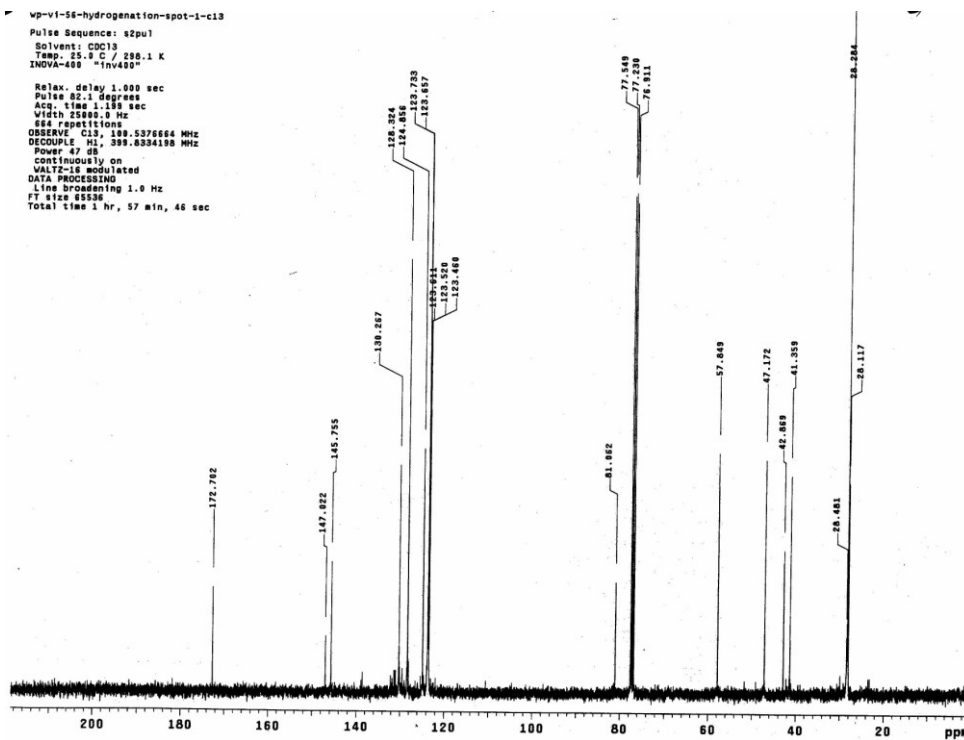
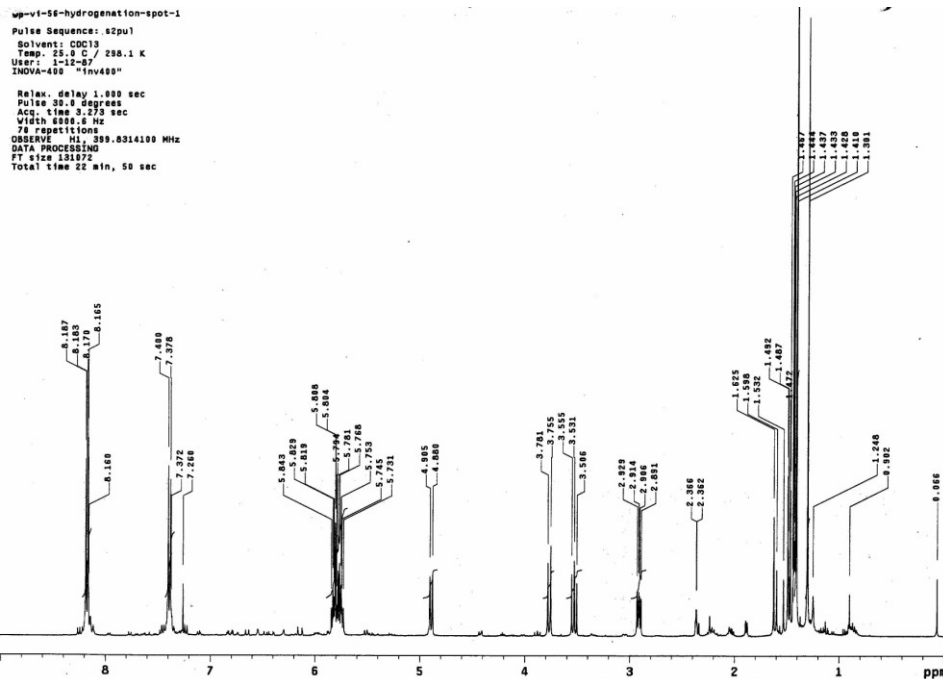
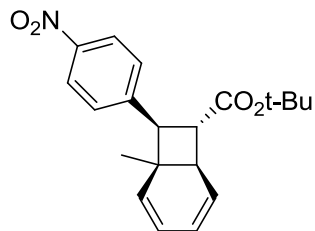


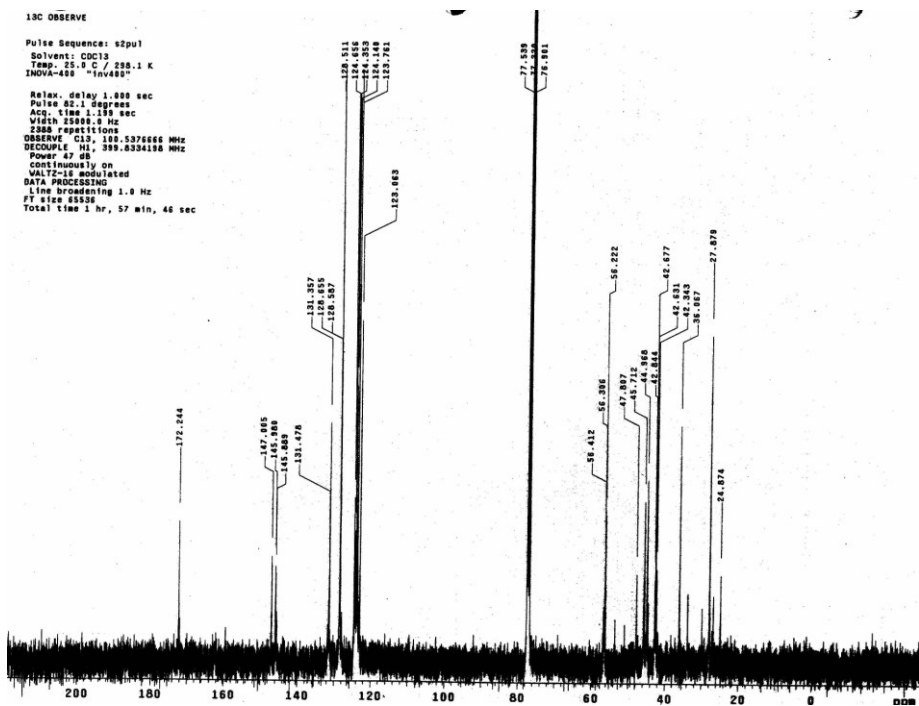
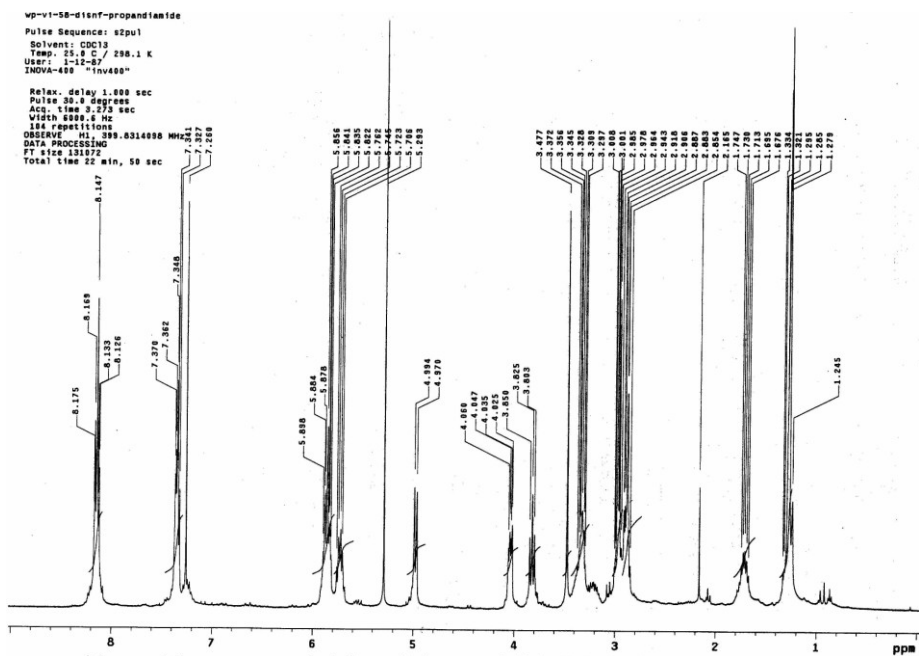
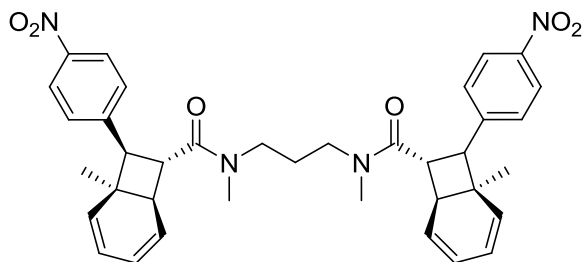


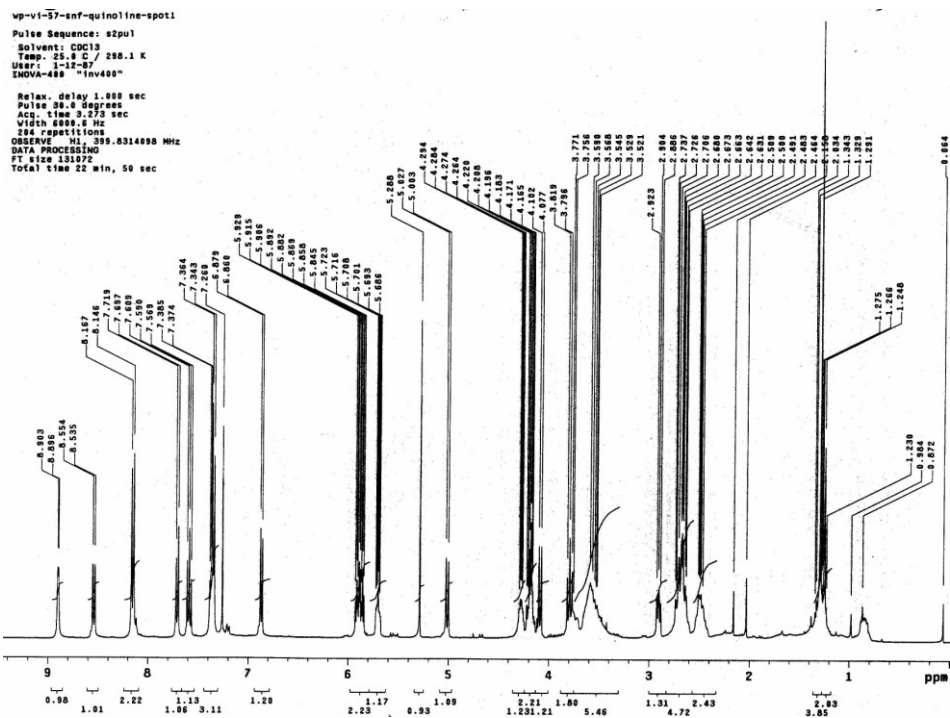
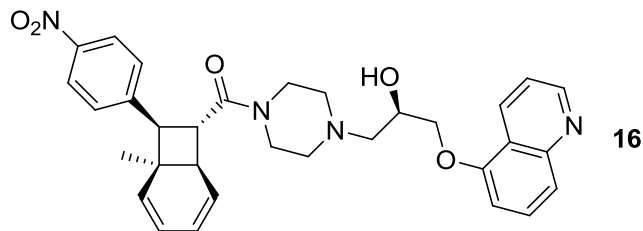
12

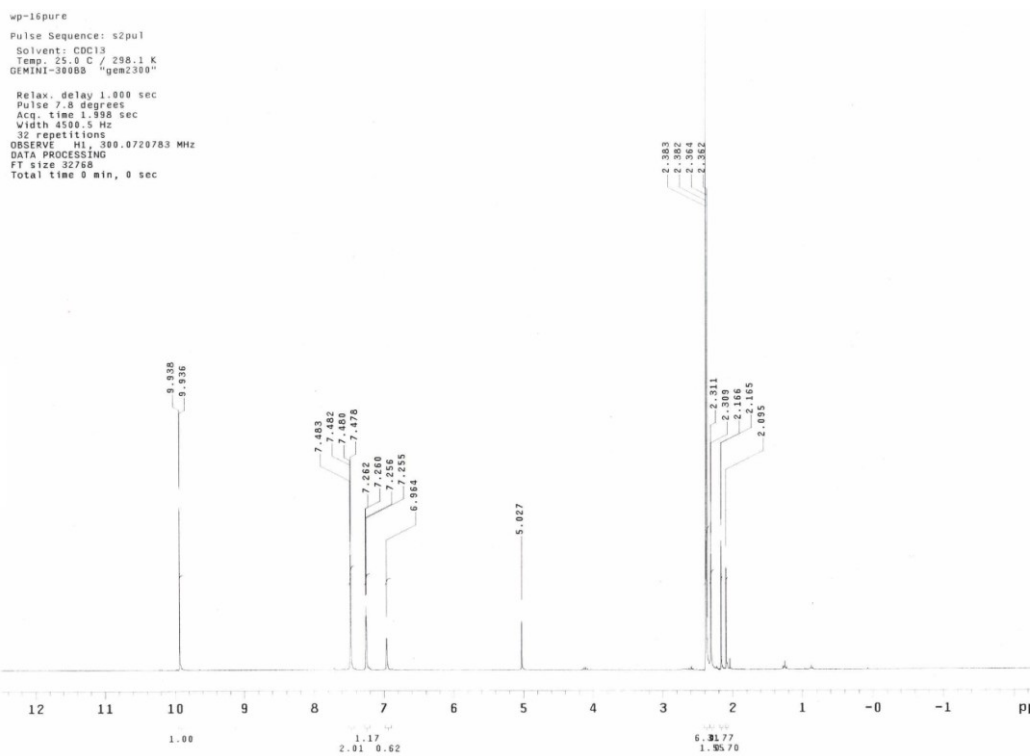
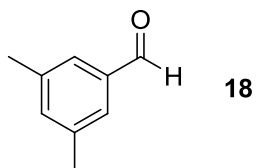


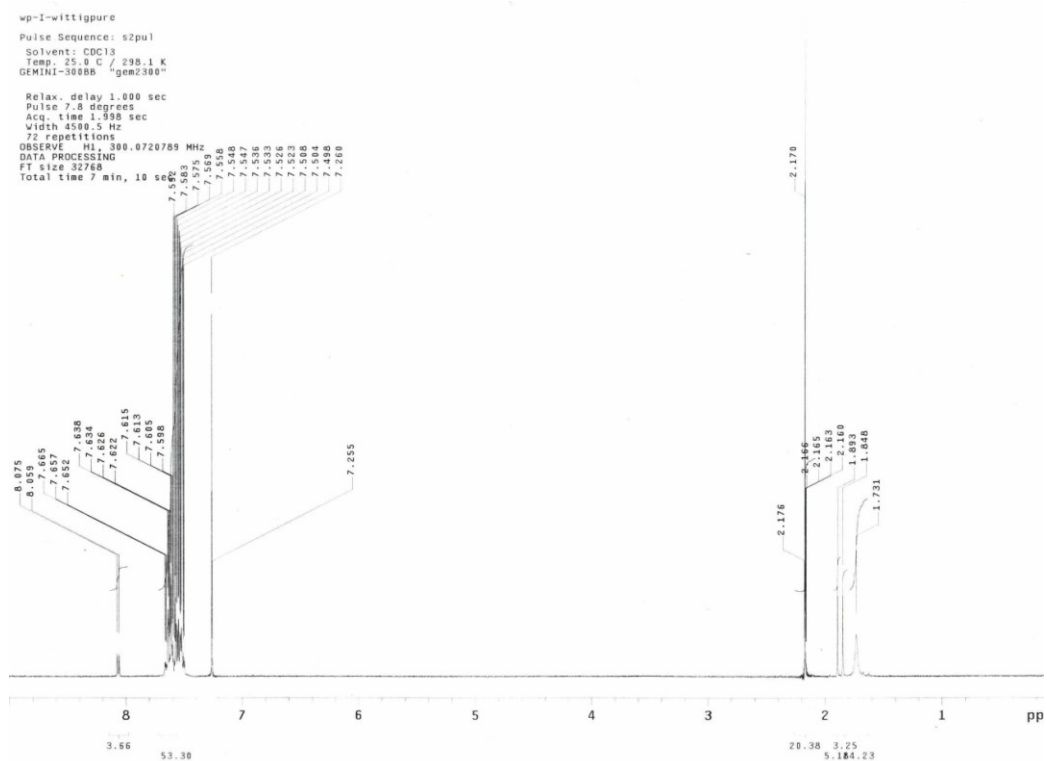
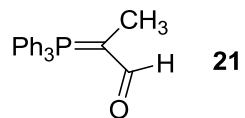


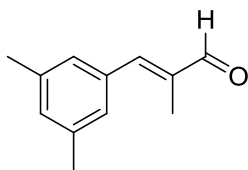




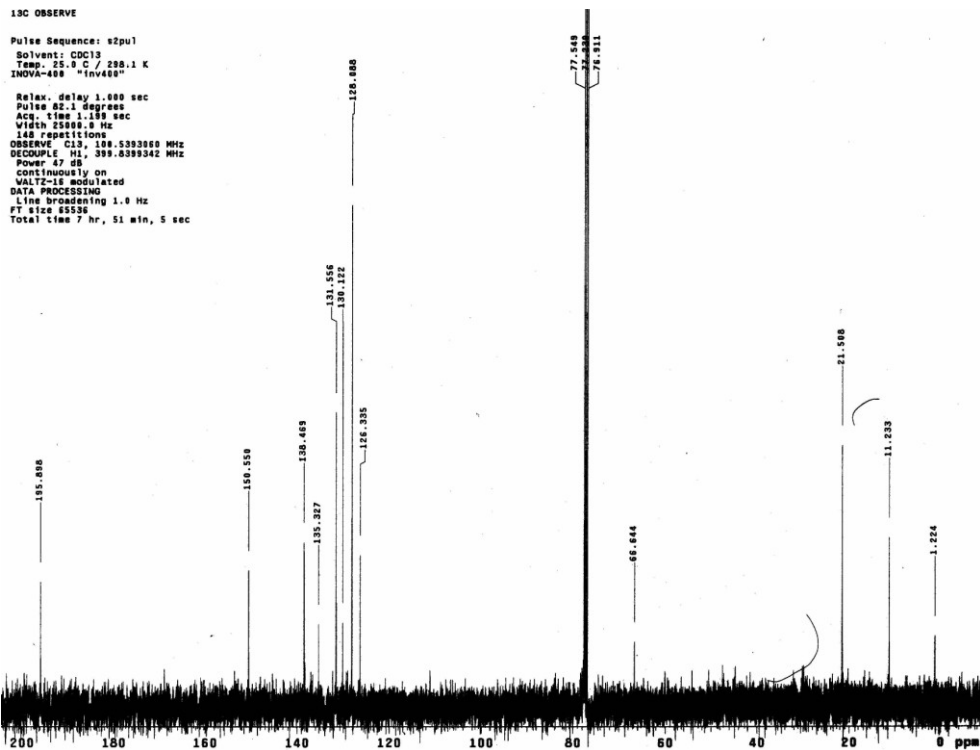
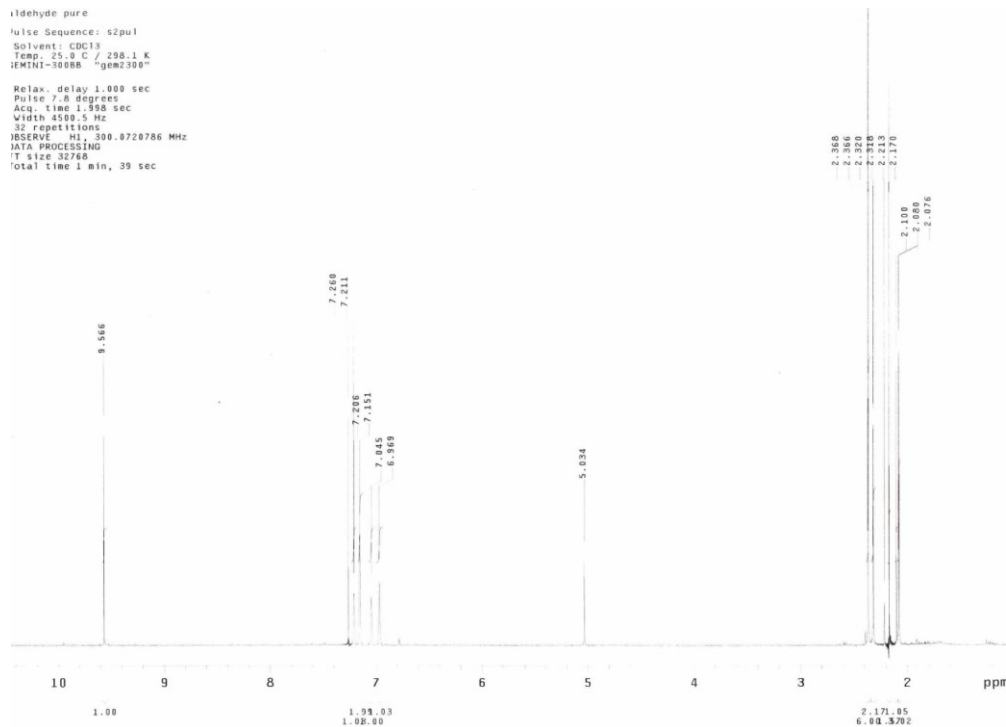


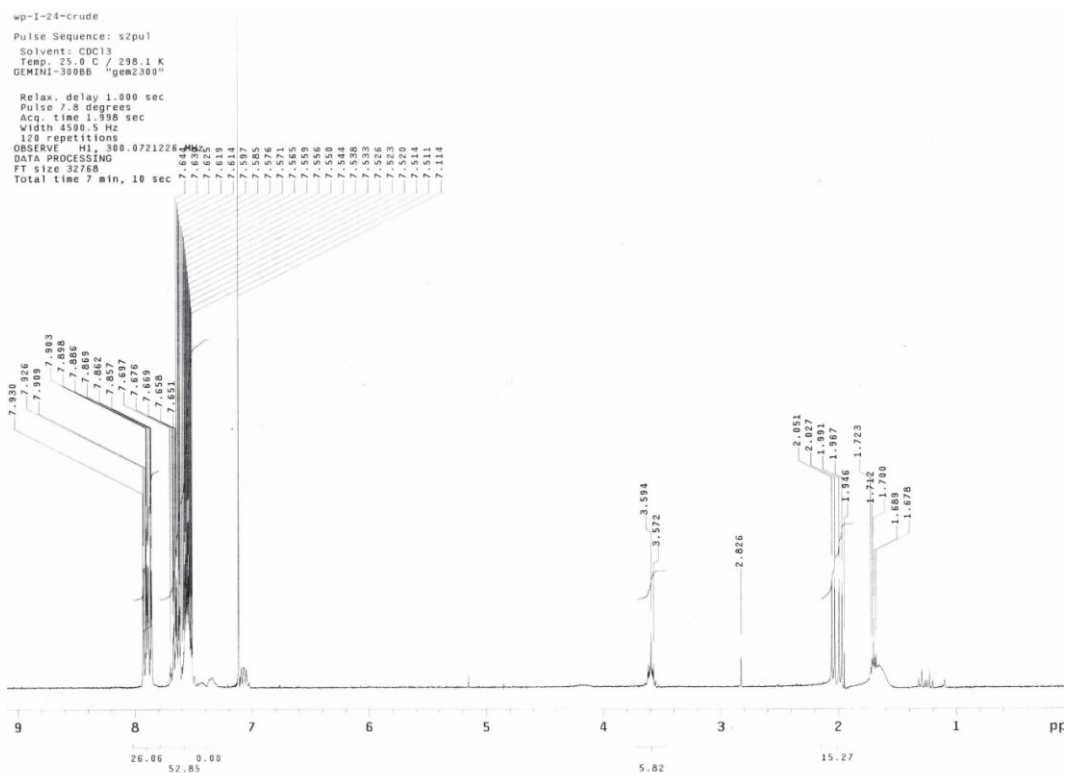
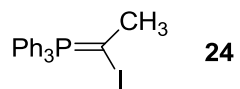


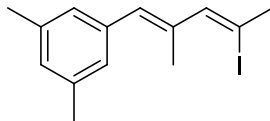




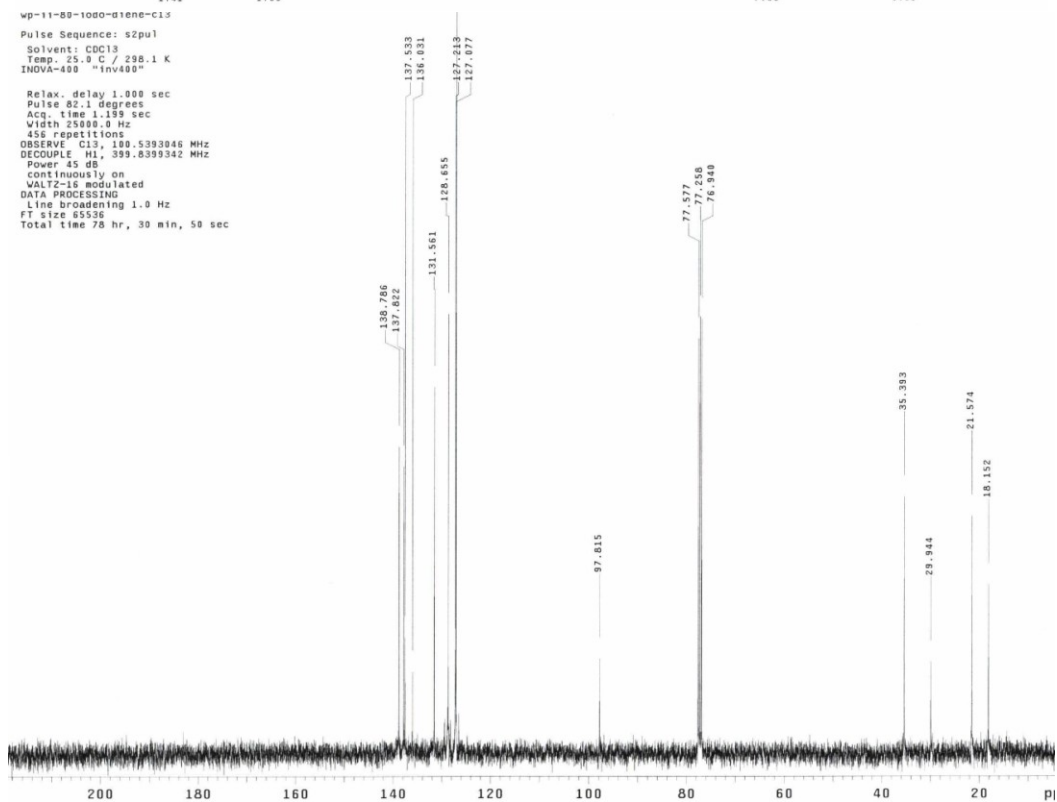
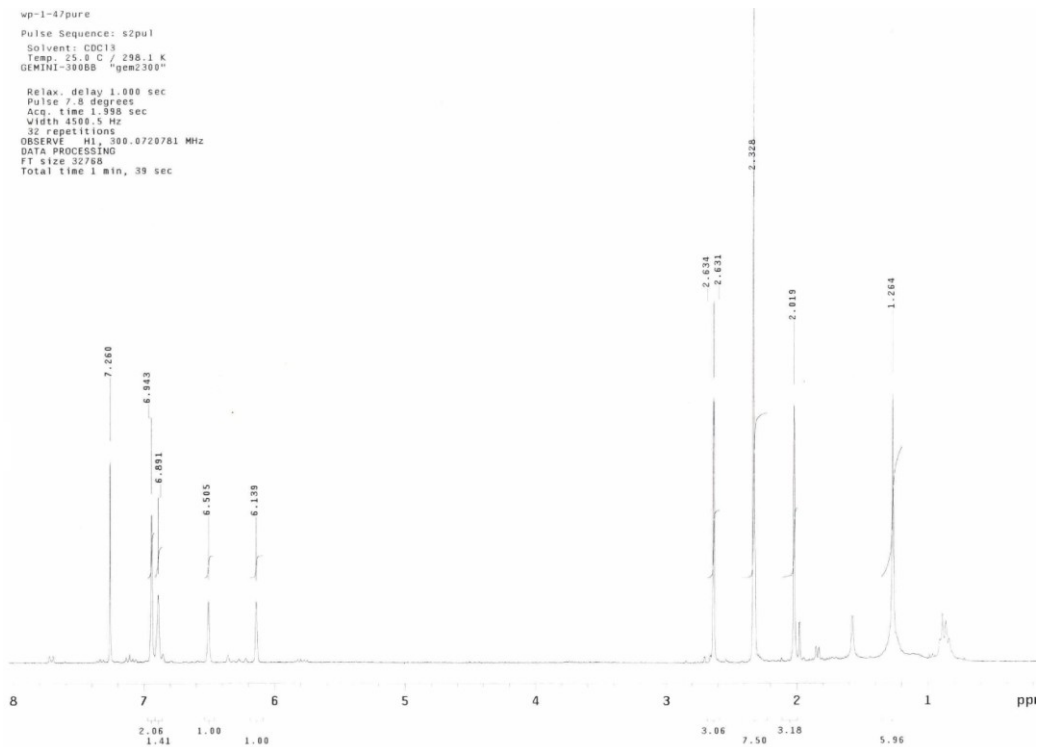
22

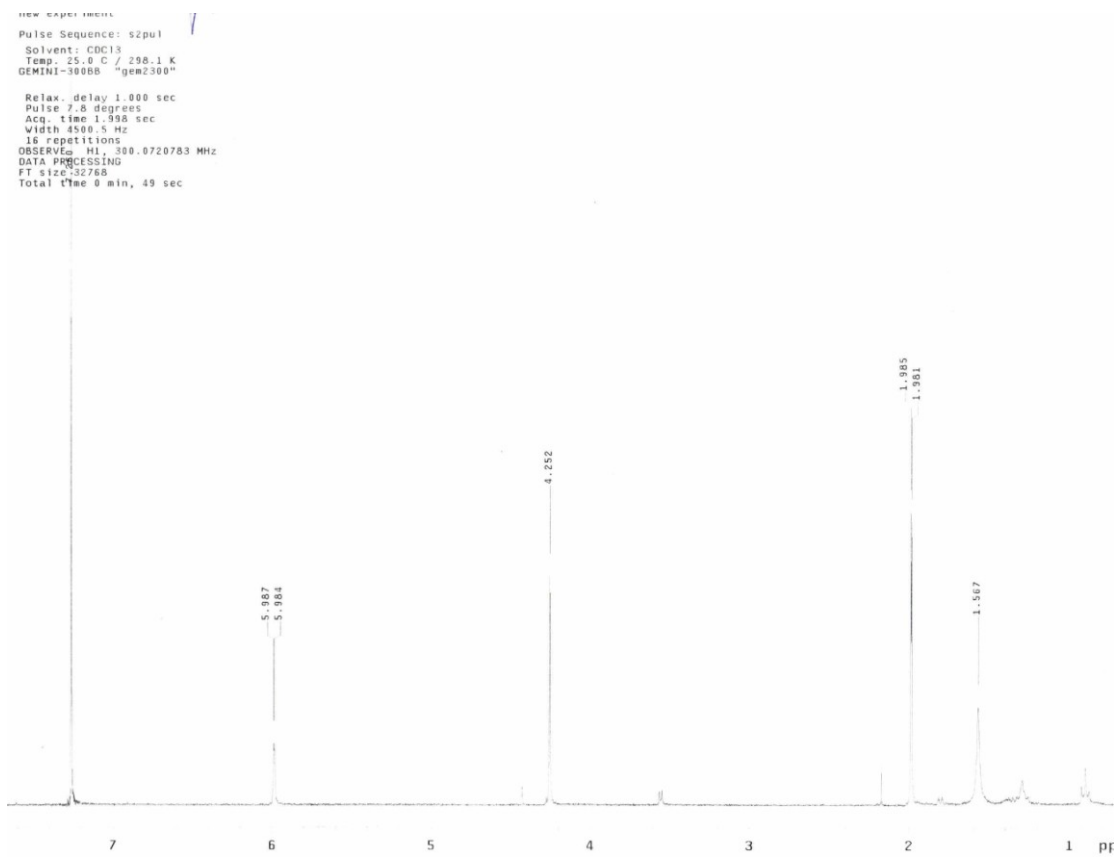
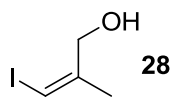


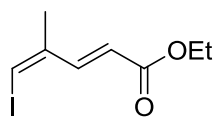




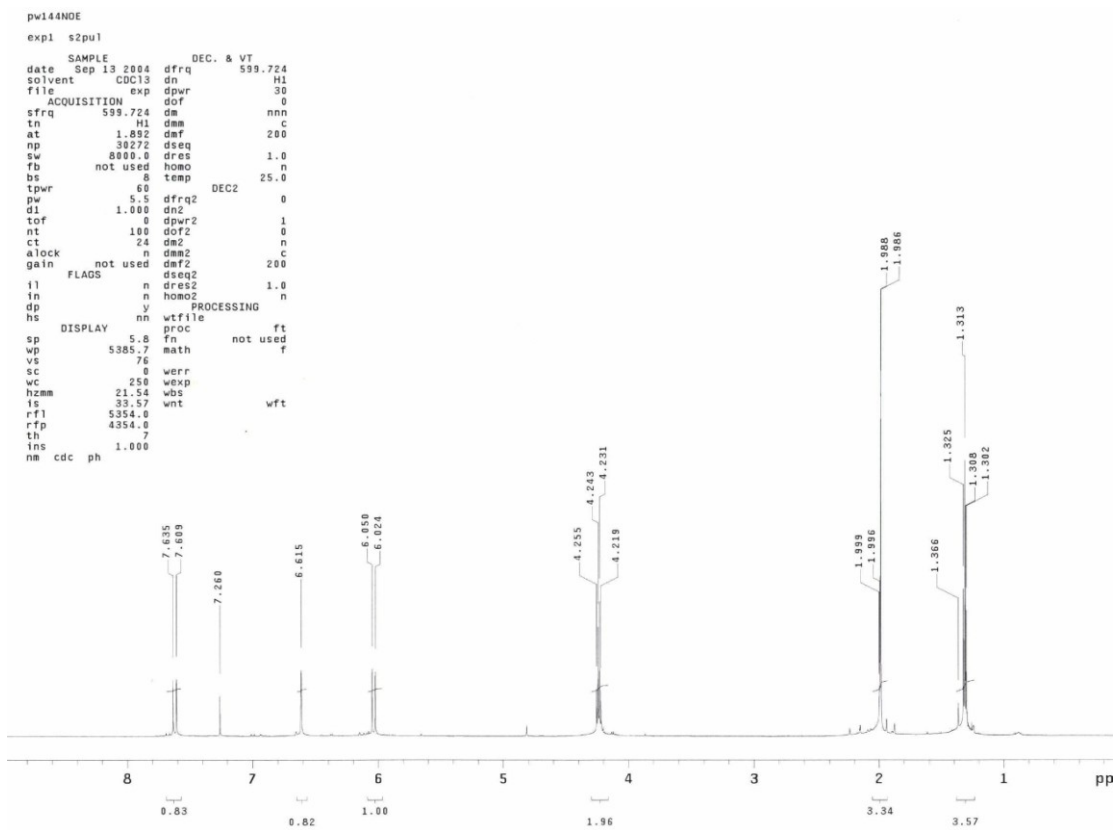
25

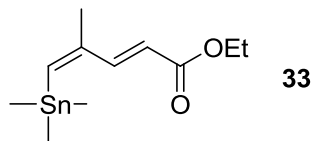






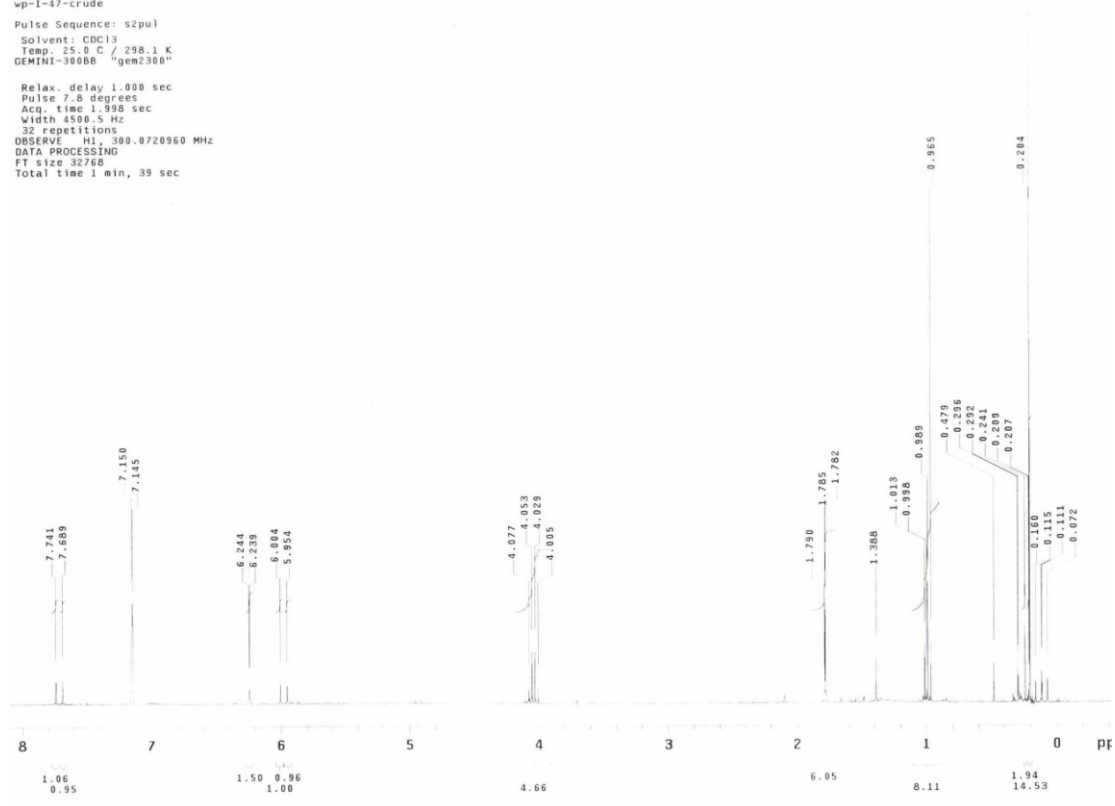
32

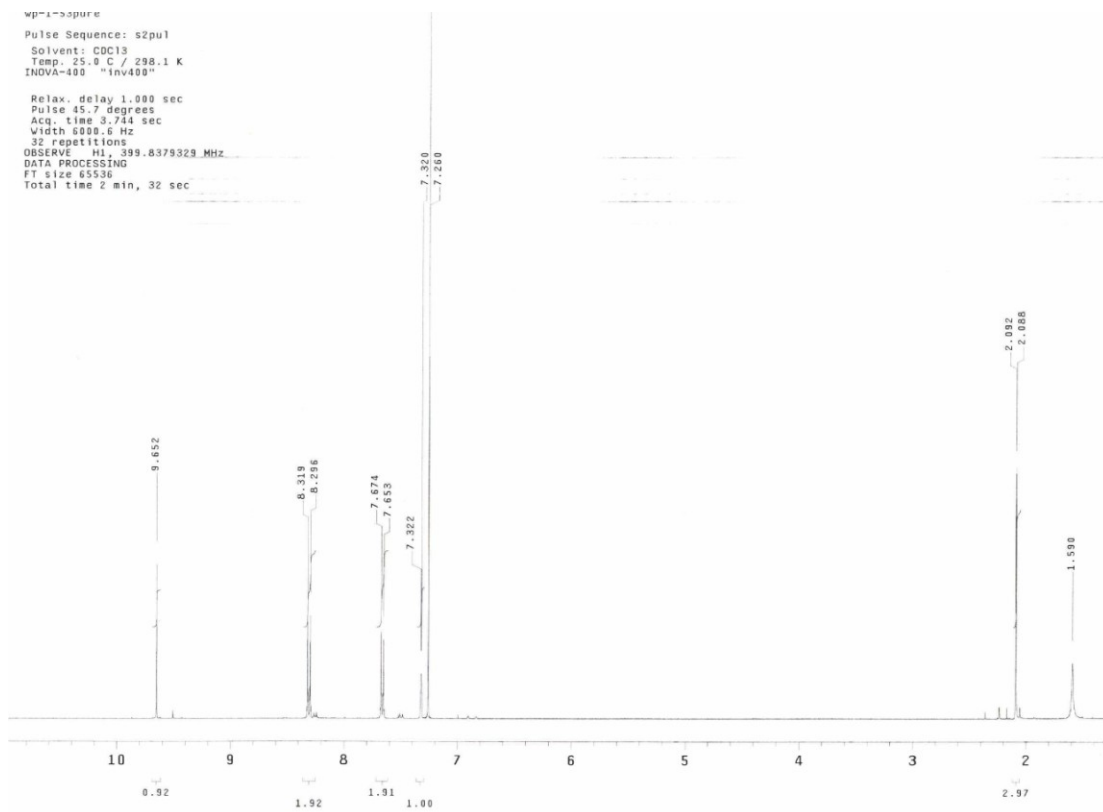
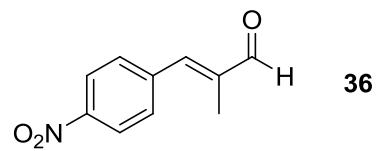


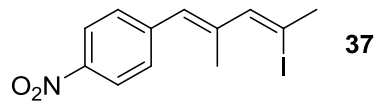


wp-1-47-crude
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 GEMINI-300BB "gen2300"

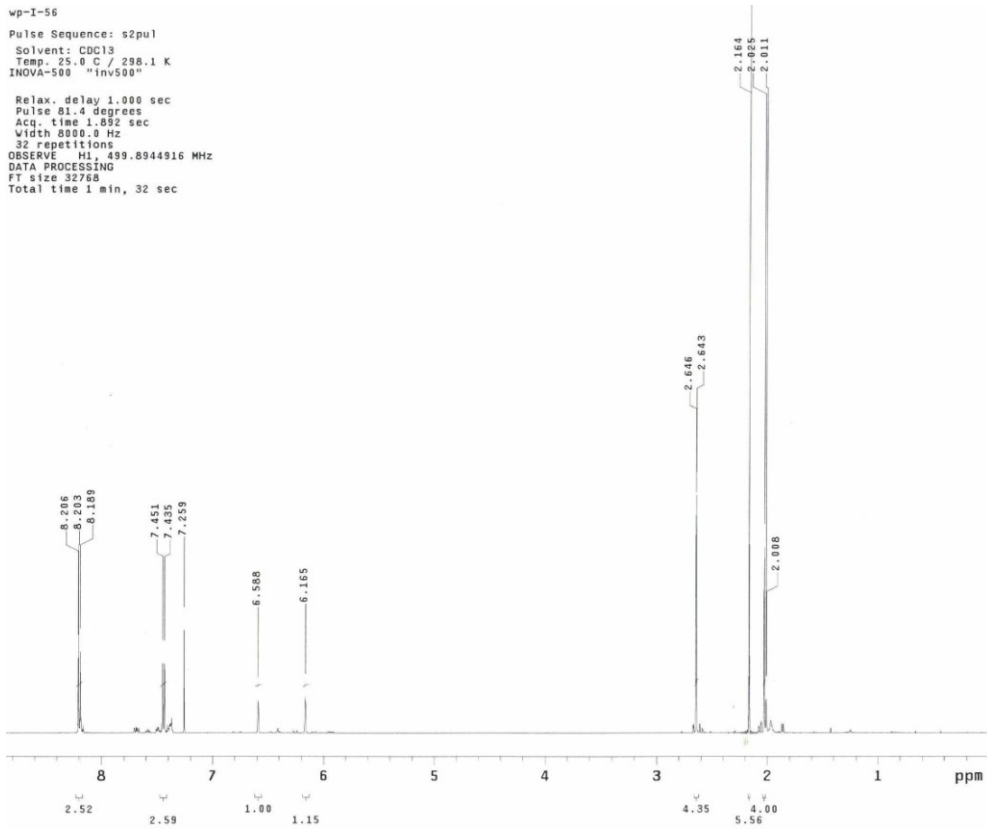
Relax. delay 1.009 sec
 Pulse 7.8 degrees
 Acq. time 1.998 sec
 Width 4500.5 Hz
 32 repetitions
 OBSERVE H1 300.8720960 MHz
 DATA PROCESSING
 FT size 32768
 Total time 1 min, 39 sec

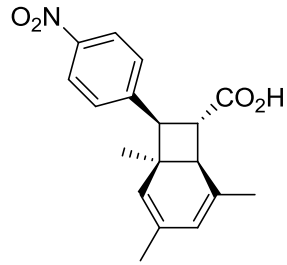






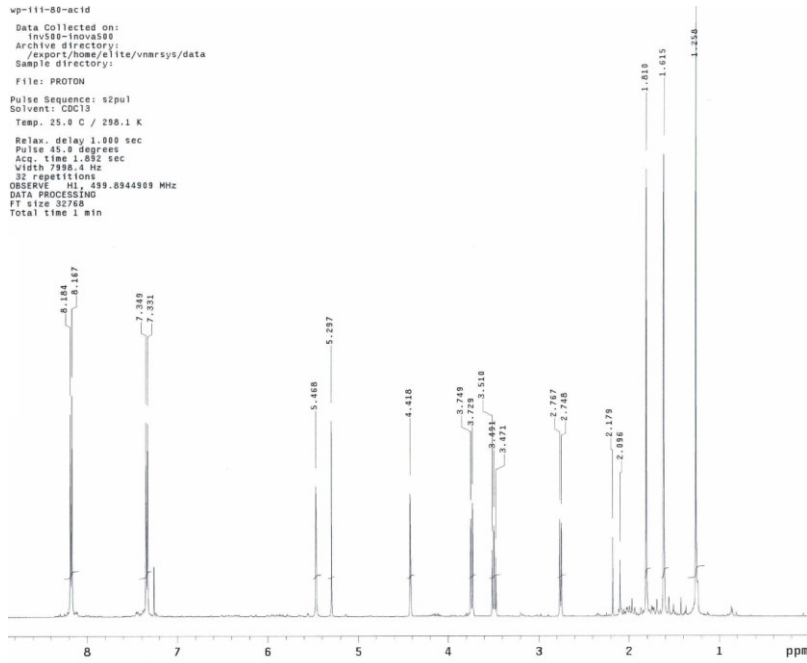
wp-I-56
Pulse Sequence: s2pul
Solvent: CDCl₃
Temp: 25.0 C / 298.1 K
INOVA-500 "inv500"
Relax. delay 1.000 sec
Pulse 81.4 degrees
Acq. time 1.892 sec
Width 3000.0 Hz
32 repetitions
OBSERVE H1, 499.8944916 MHz
DATA PROCESSING
FT size 32768
Total time 1 min, 32 sec



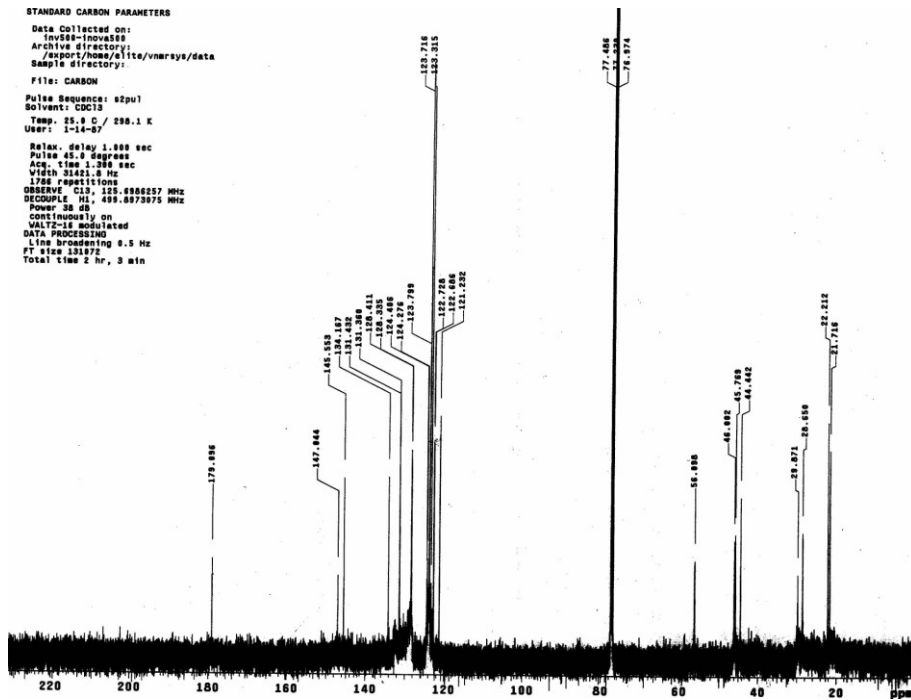


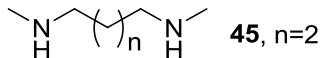
38

vg-111-80-acid
 Data Collected on:
 Inv599-inova500
 Archive directory:
 /export/home/elite/vmrsys/data
 Sample directory:
 File: PROTON
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.892 sec
 Width 7390.4 Hz
 32 repetitions
 OBSERVE H1, 499.8944889 MHz
 DATA PROCESSING
 FT size 32768
 Total time 1 min

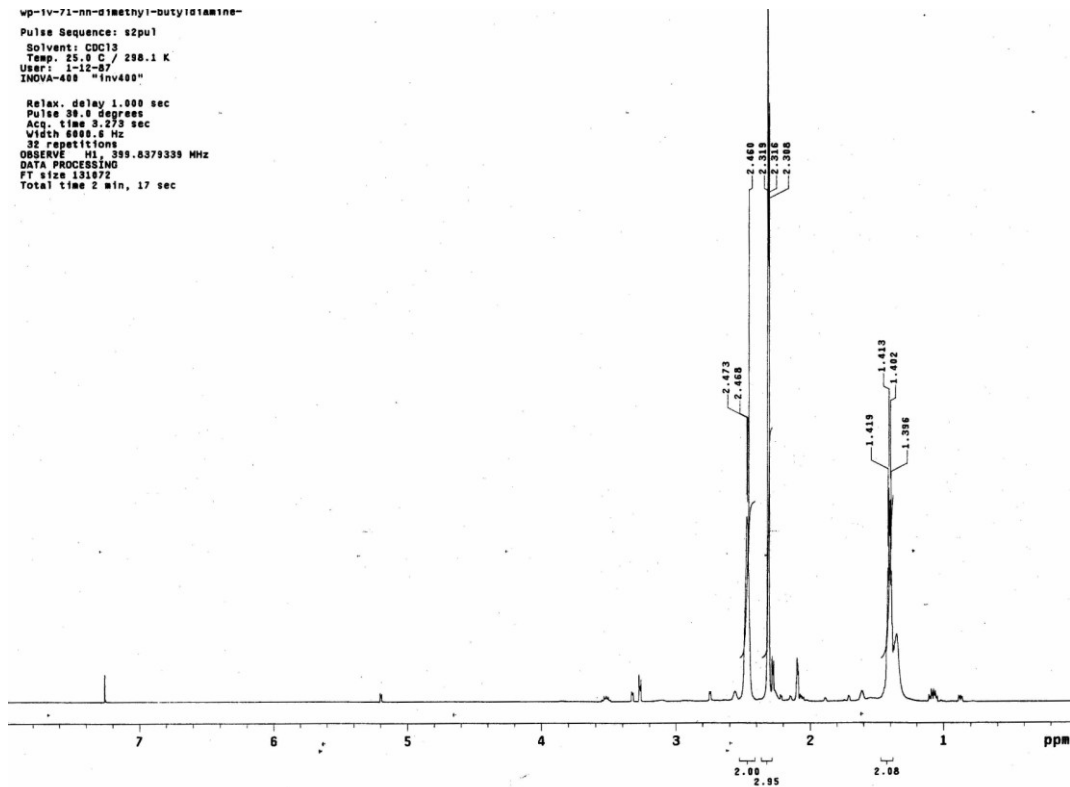


STANDARD CARBON PARAMETERS
 Data Collected on:
 Inv599-inova500
 Archive directory:
 /export/home/elite/vmrsys/data
 Sample directory:
 File: CARBON
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-14-87
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.286 sec
 Width 31421.8 Hz
 1768 repetitions
 OBSERVE C13, 125.898257 MHz
 DECOUPLE H1, 499.8973979 MHz
 Power 28 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 0.5 Hz
 FT size 131872
 Total time 2 hr, 3 min

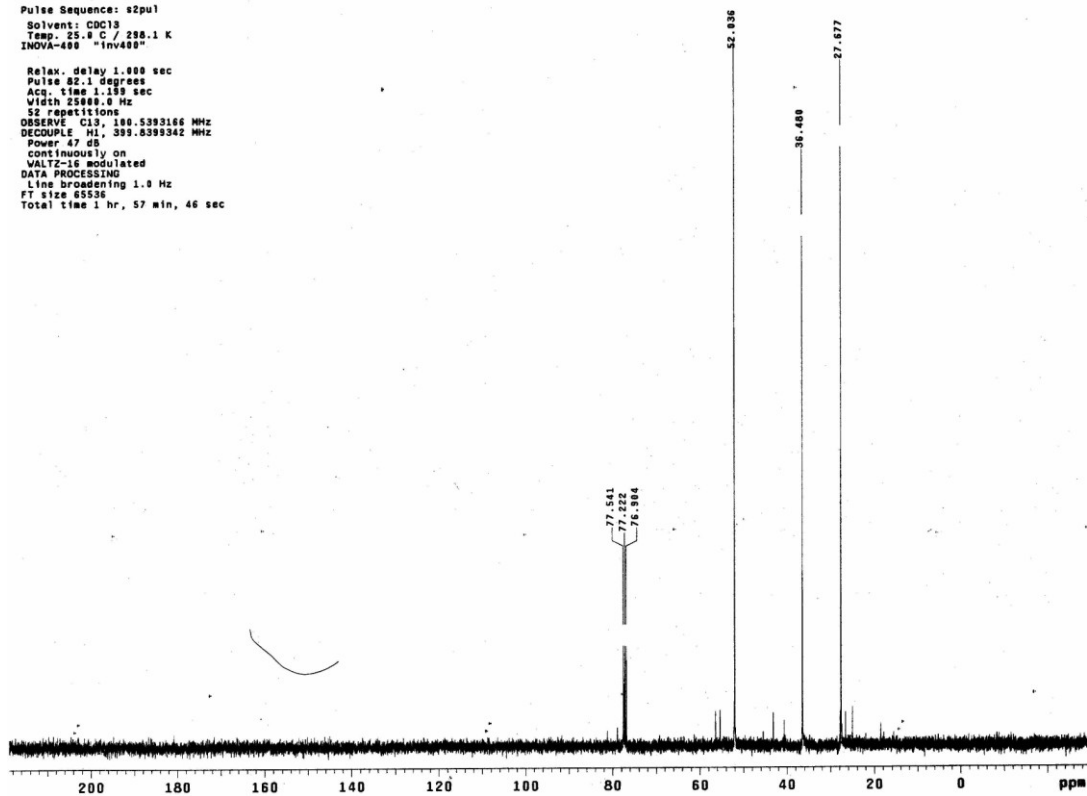


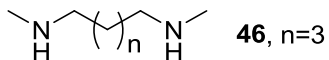


wp-1v-71-nn-dimethyl-butylidiamine-
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 User: 1-12-87
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 38.0 degrees
 Acq. time 3.273 sec
 Width 8888.0 Hz
 32 repetitions
 OBSERVE H1, 399.8379339 MHz
 DATA PROCESSING
 FT size 131872
 Total time 2 min, 17 sec

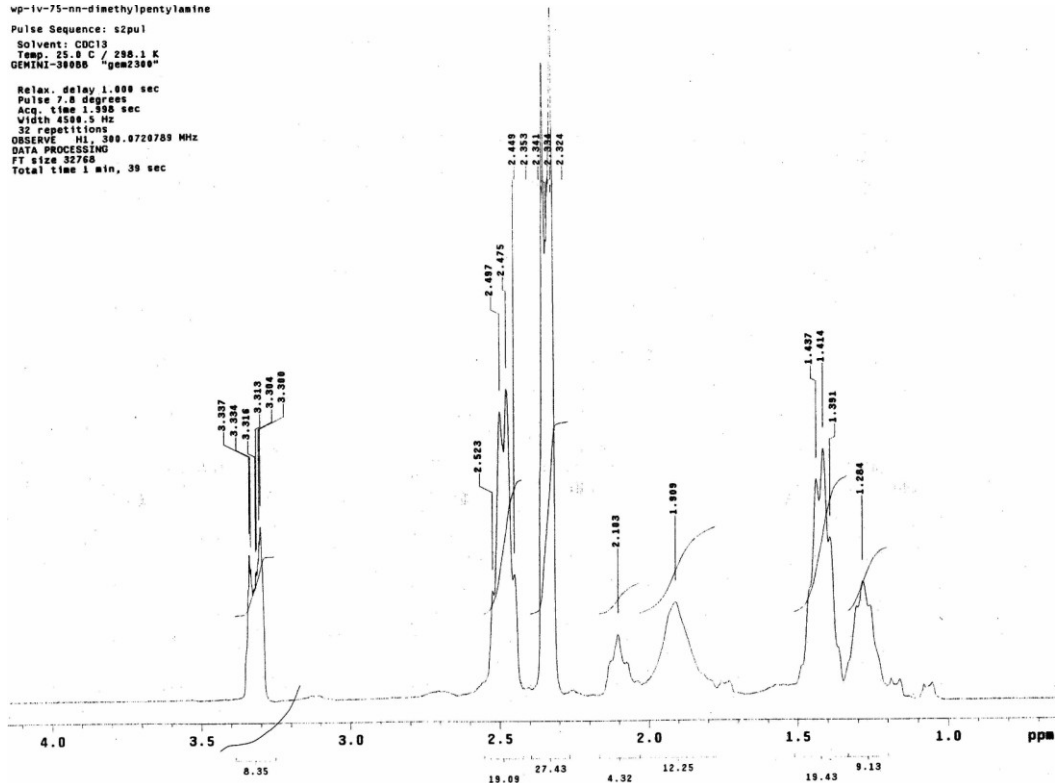


wp-1v-71-nn-dimethyl-butylidiamine-c13
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.159 sec
 Width 25888.0 Hz
 32 repetitions
 OBSERVE c13, 100.5393166 MHz
 DECOUPLE H1, 399.8399342 MHz
 Power 47 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 1 hr, 57 min, 46 sec



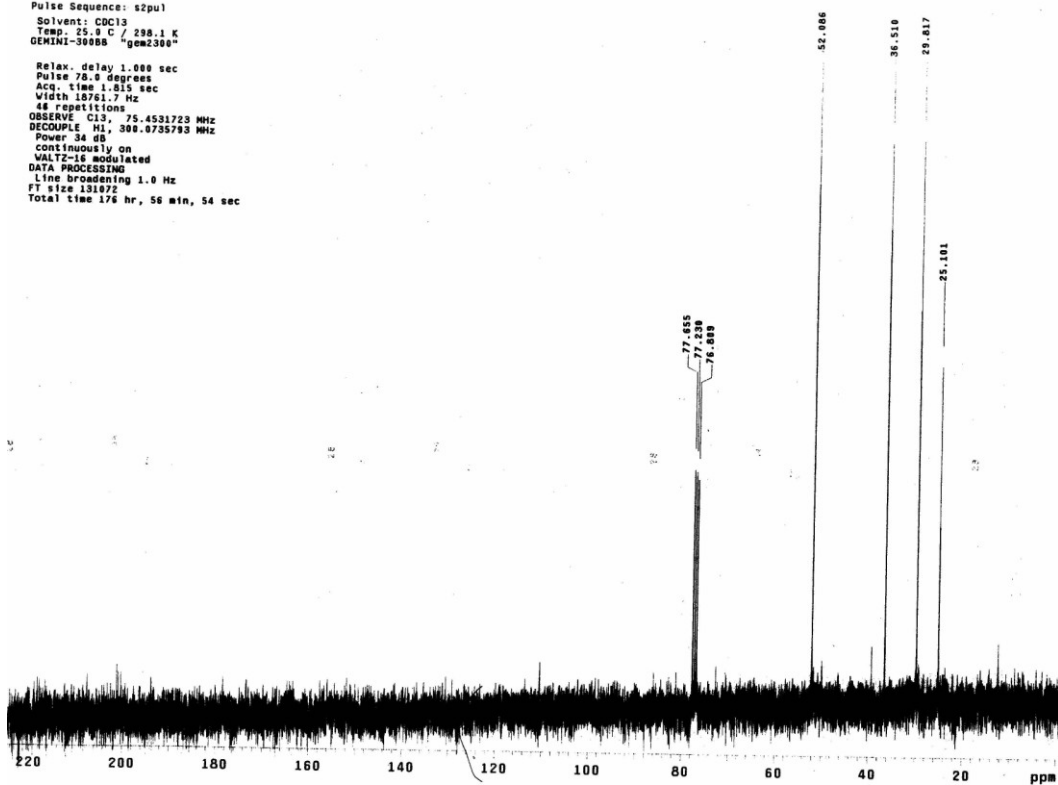


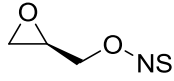
wp-1v-75-nn-dimethylpentylamine
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 GEMINI-30006 "gem2300"
 Relax. delay 1.000 sec
 Pulse 7.8 degrees
 Acq. time 1.390 sec
 Width 4500.5 Hz
 32 repetitions
 OBSERVE H1, 300.0720789 MHz
 DATA PROCESSING
 FT size 32768
 Total time 1 min, 39 sec



13C OBSERVE

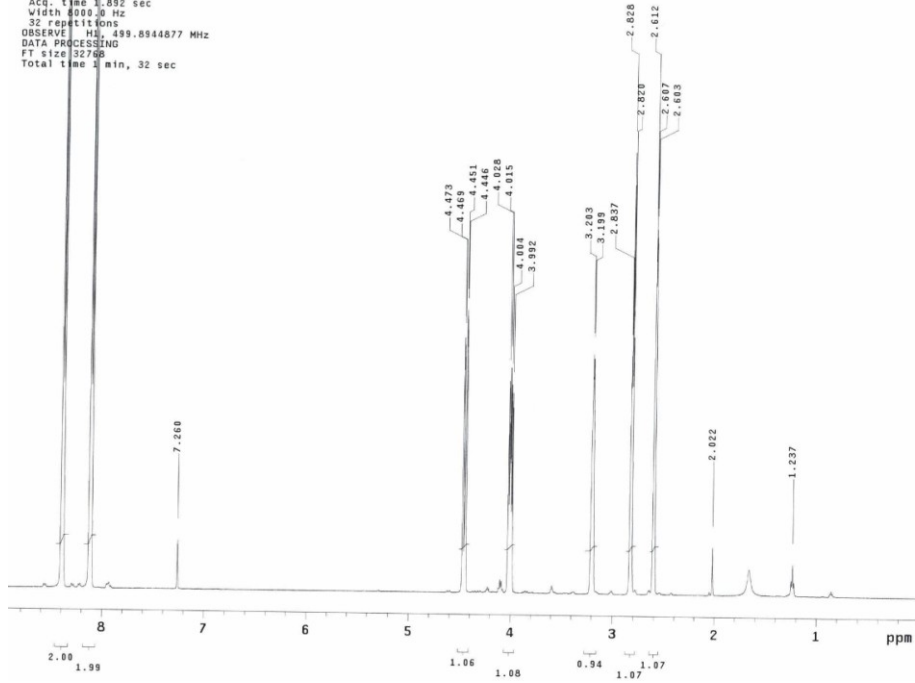
Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 GEMINI-30006 "gem2300"
 Relax. delay 1.000 sec
 Pulse 76.0 degrees
 Acq. time 1.015 sec
 Width 18761.7 Hz
 48 repetitions
 OBSERVE C13, 75.4531723 MHz
 DECOUPLE H1, 300.0735793 MHz
 Power 34 dB
 Continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 131072
 Total time 176 hr, 56 min, 54 sec



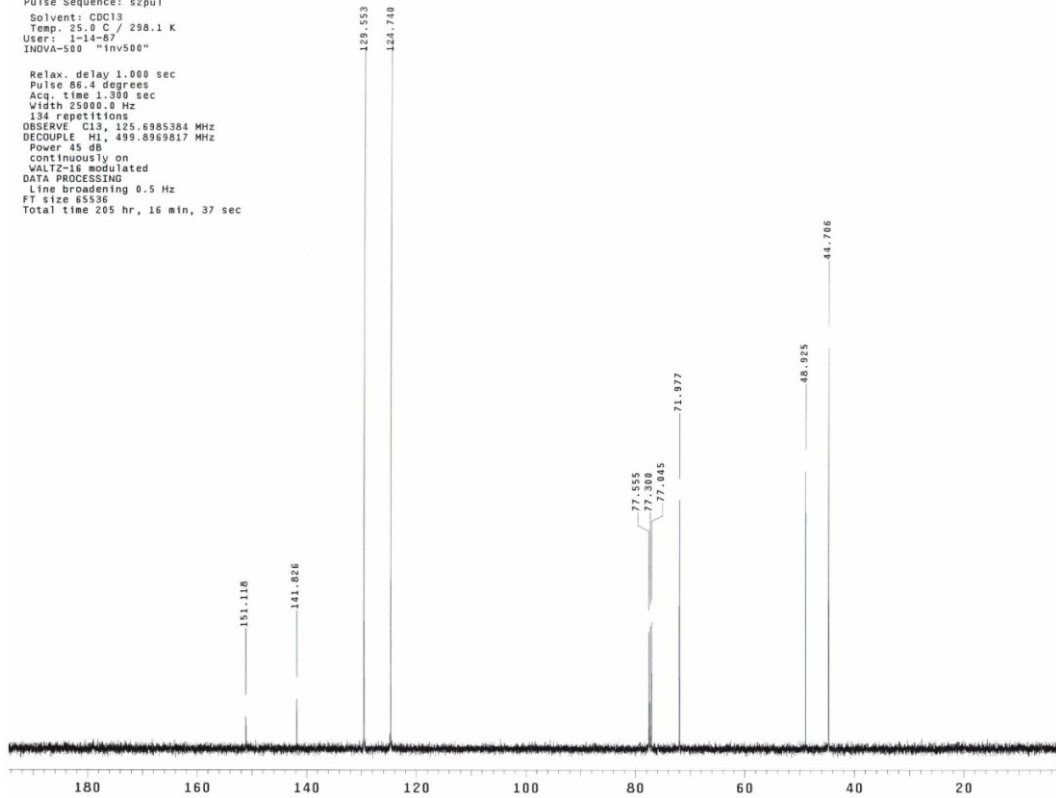


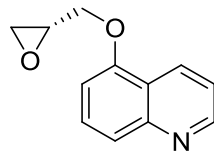
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wp-111-10-NS-glycidol-pure
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-500 "inv500"
 Relax. delay 1.000 sec
 Pulse 52.2 degrees
 Acq. time 1.892 sec
 Width 8000.0 Hz
 32 repetitions
 OBSERVE H1, 499.8944877 MHz
 DATA PROCESSING
 FT size 32768
 Total time 8 min, 32 sec

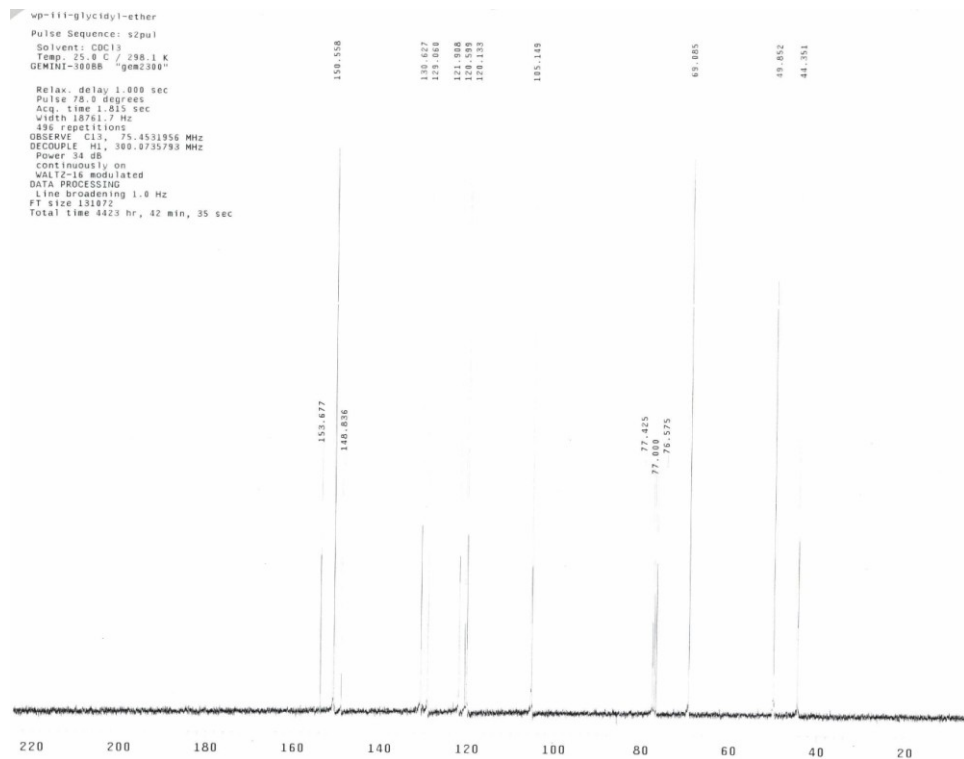
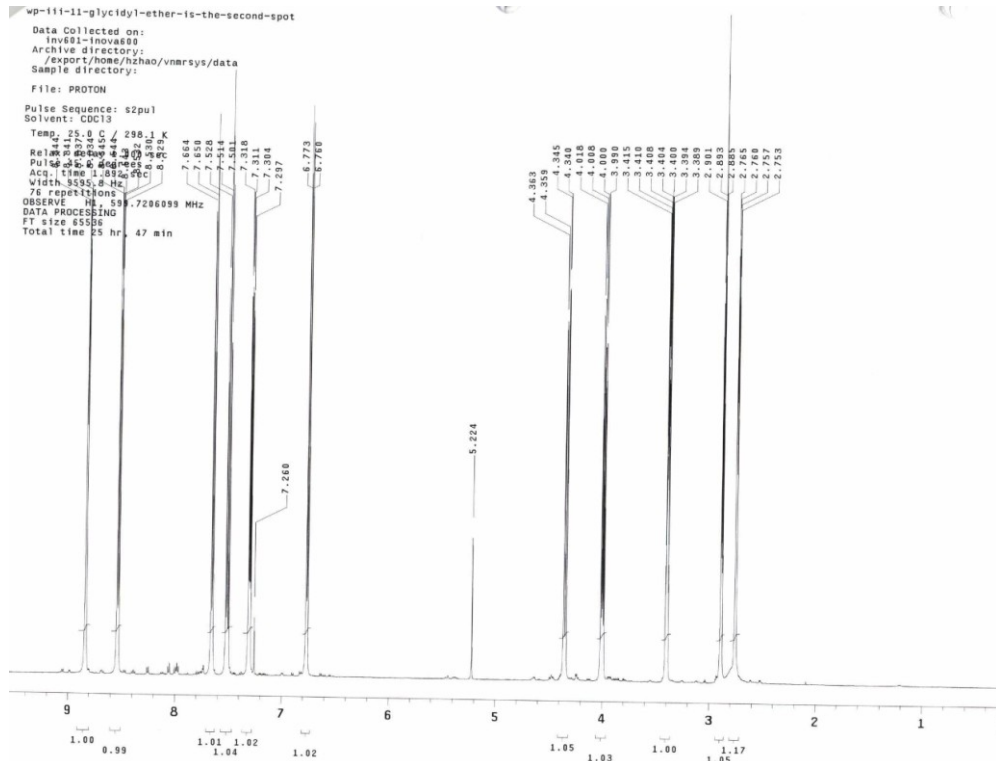


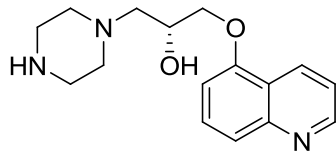
wp-111-10-NS-glycidol-pure-c13
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-14-07
 INOVA-500 "inv500"
 Relax. delay 1.000 sec
 Pulse 86.4 degrees
 Acq. time 1.380 sec
 Width 25000.0 Hz
 134 repetitions
 OBSERVE C13, 125.6985384 MHz
 DECOUPLE H1, 499.8963817 MHz
 Power 45 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 0.5 Hz
 FT size 65536
 Total time 205 hr, 16 min, 37 sec



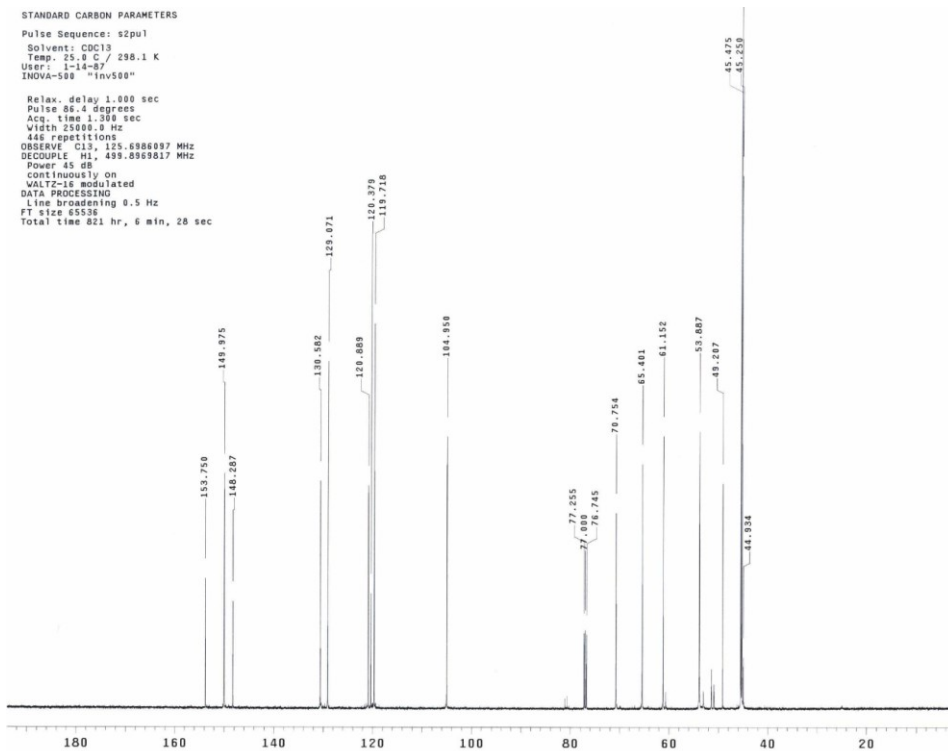
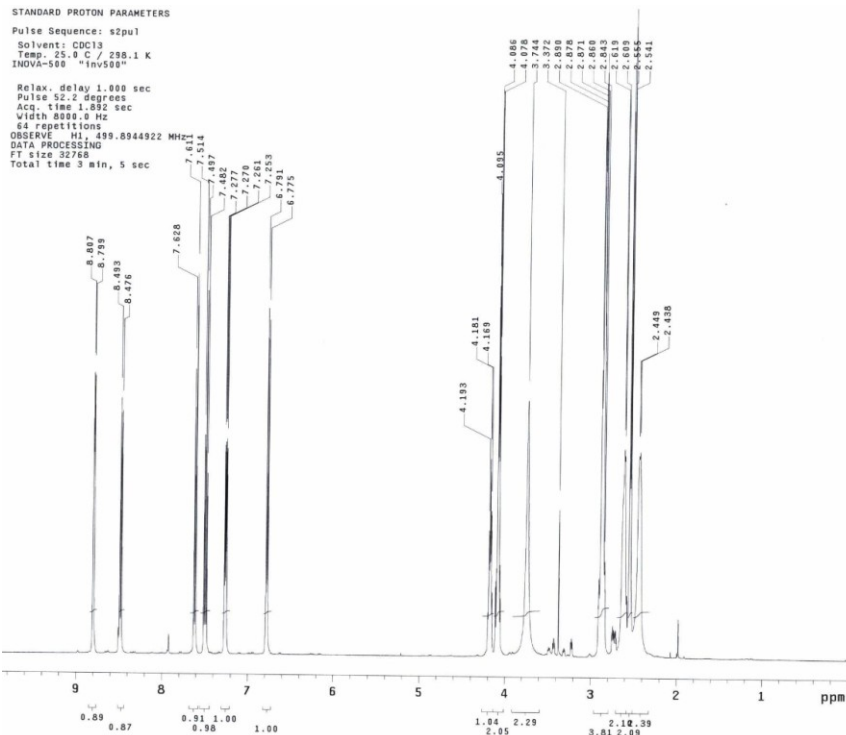


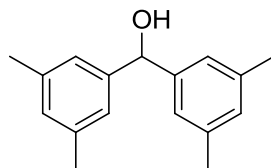
51



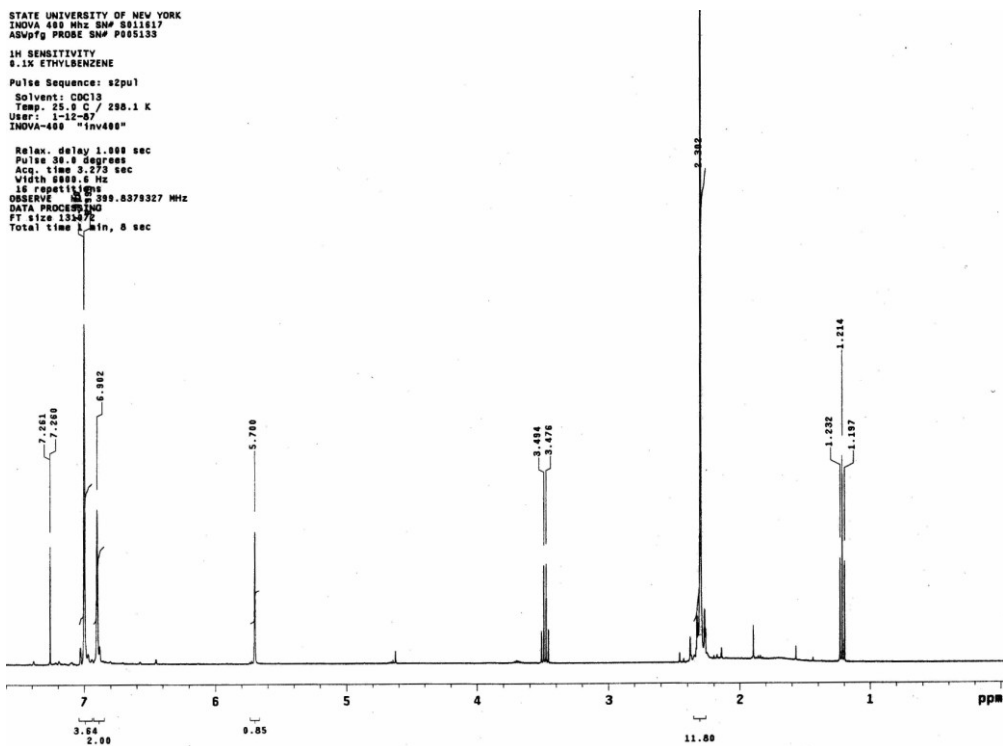
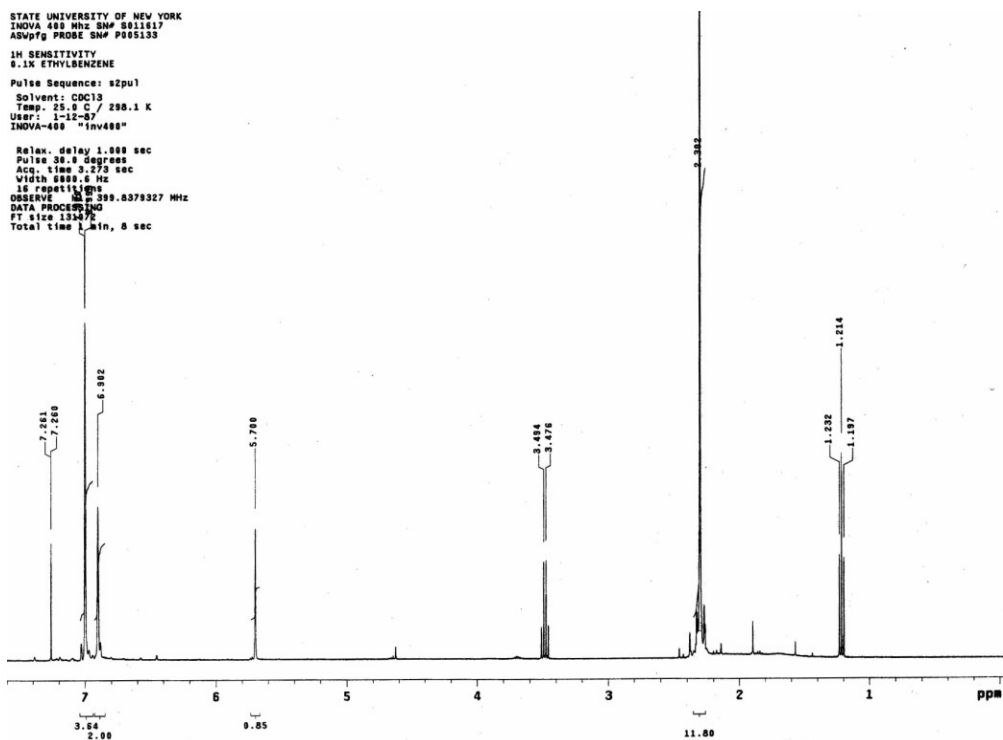


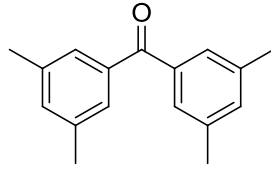
52



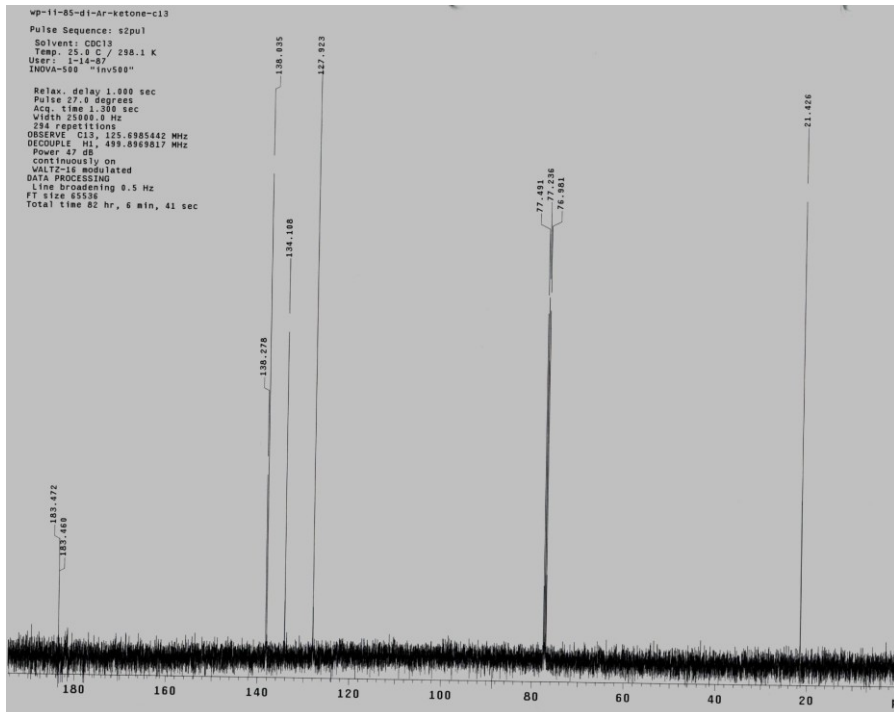
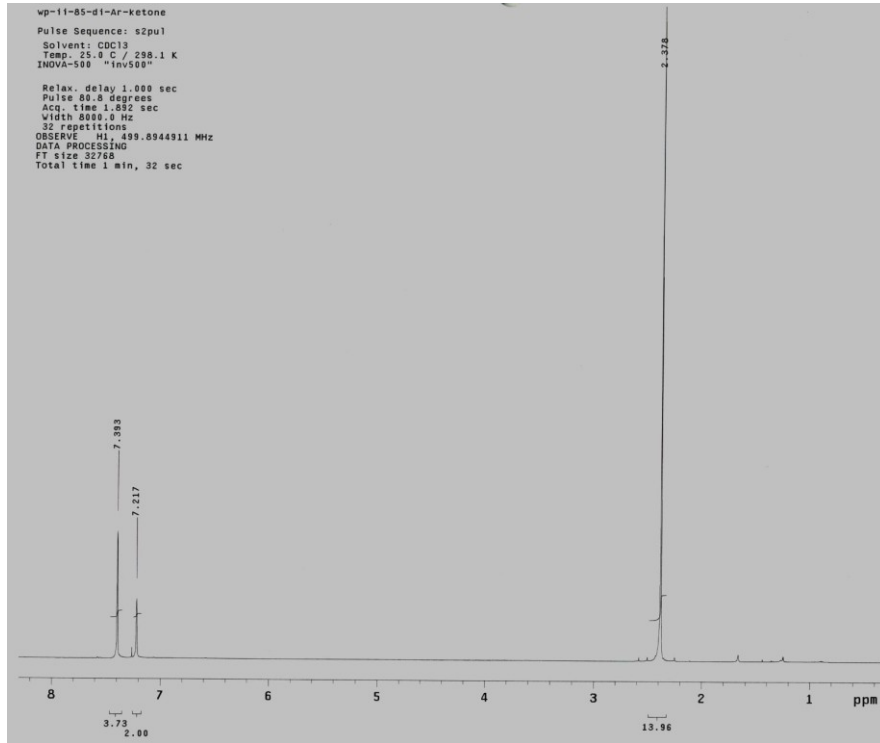


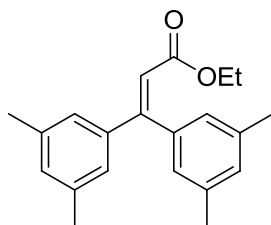
57



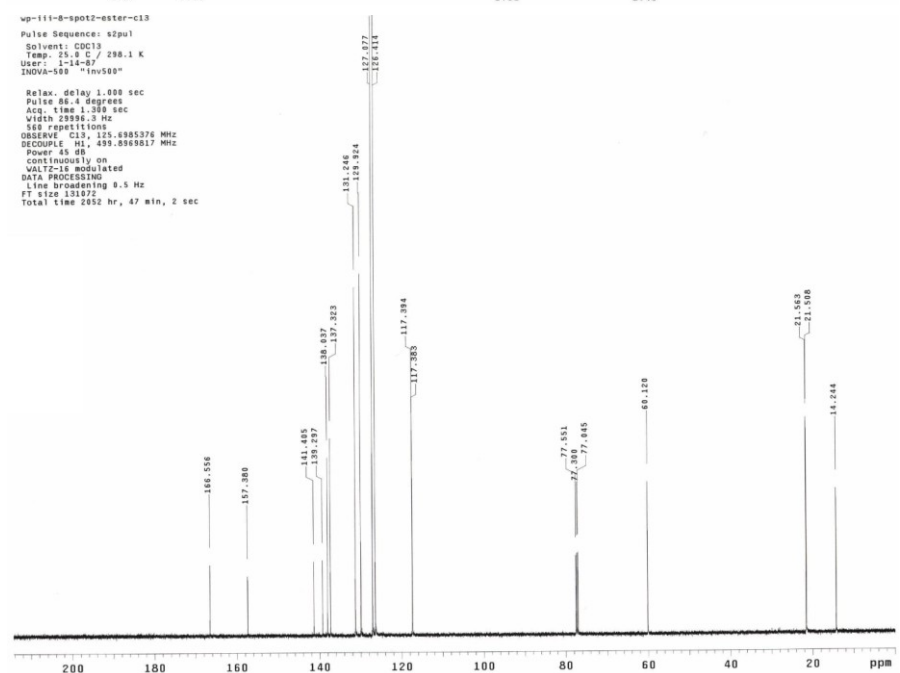
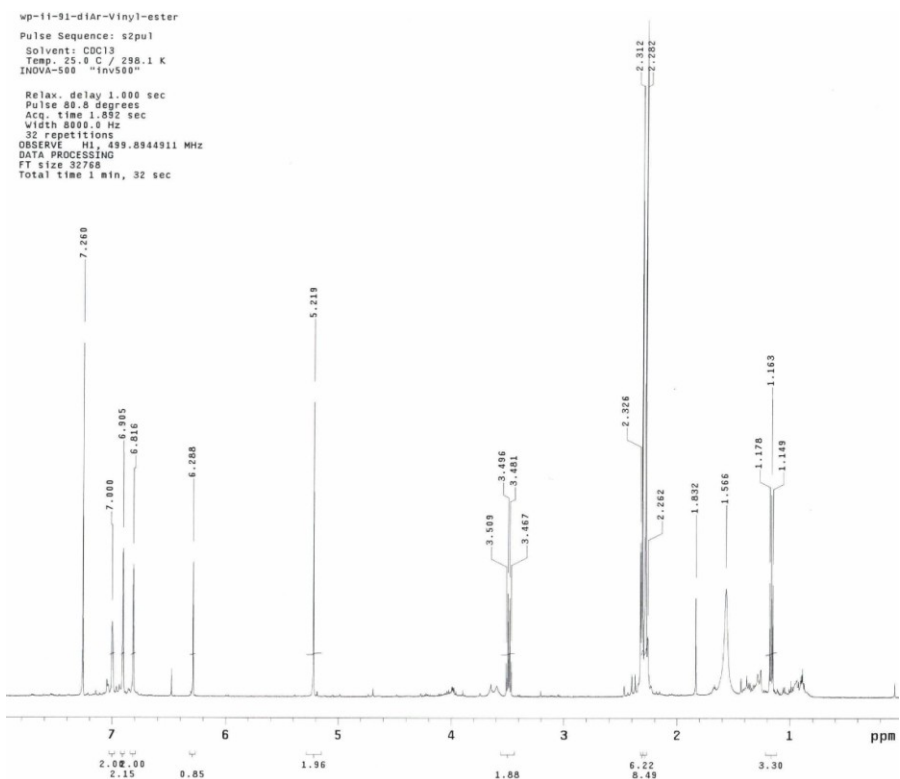


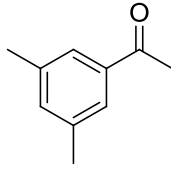
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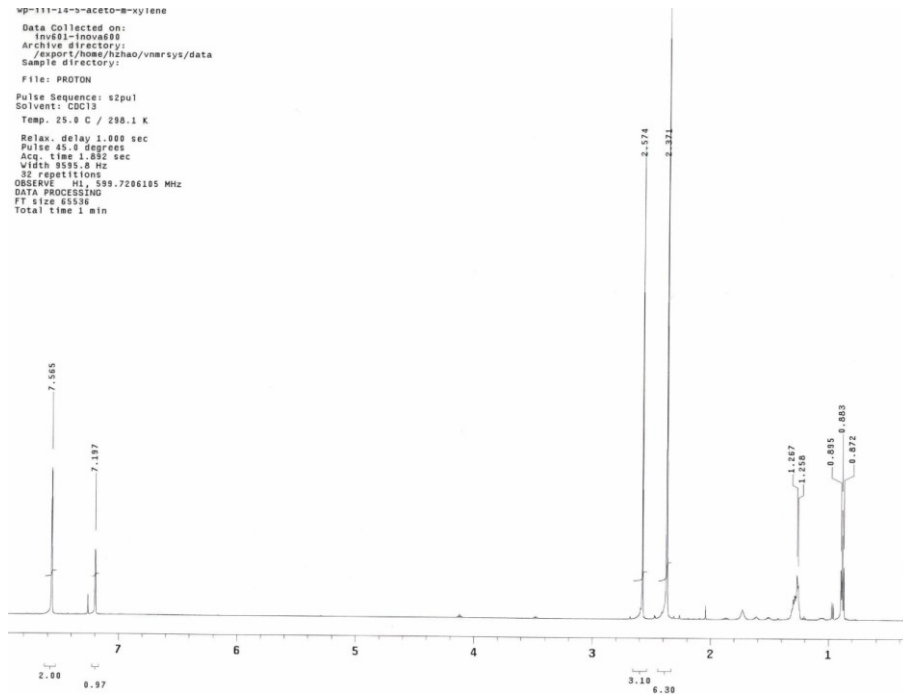


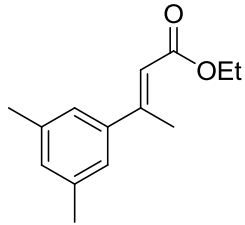
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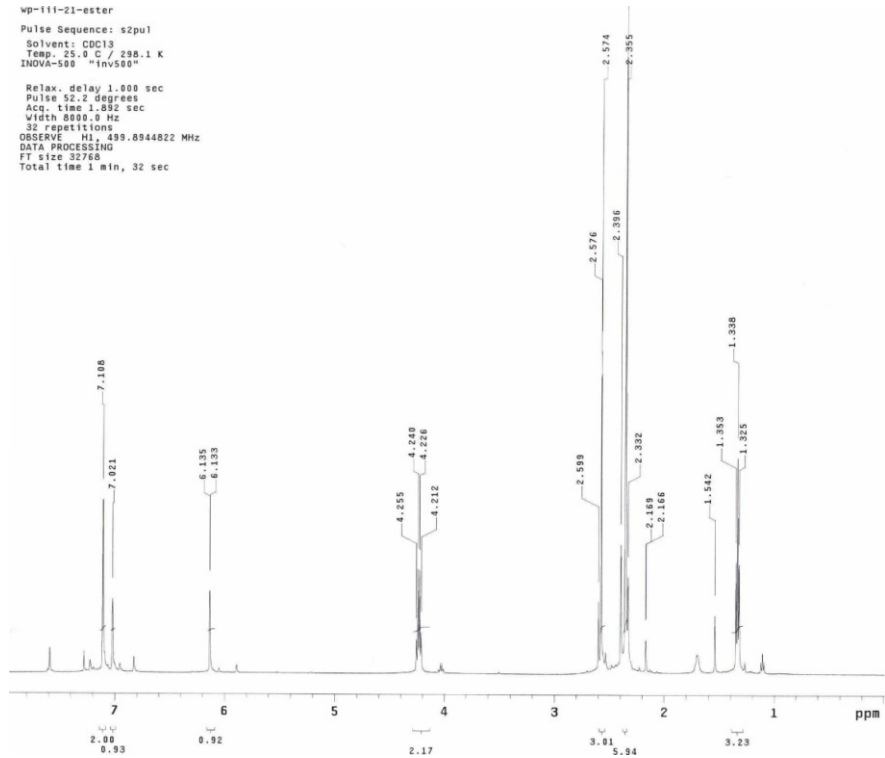
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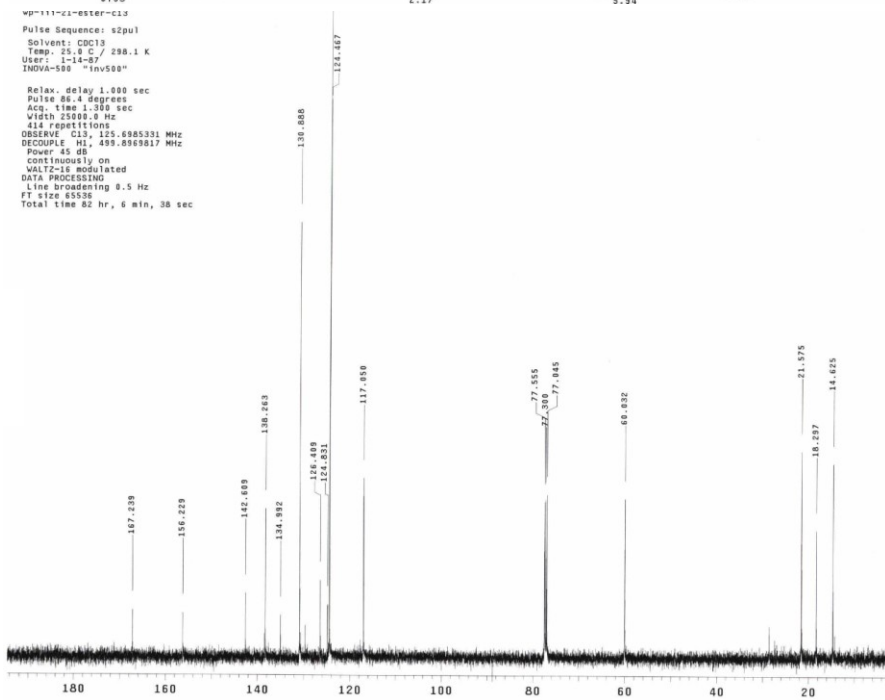


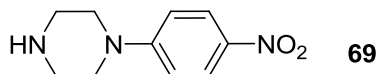
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wp-111-21-ester
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-500 "inv500"
 Relax. delay 1.000 sec
 Pulse 52.2 degrees
 Acq. time 1.892 sec
 Width 8000.0 Hz
 32 repetitions
 OBSERVE H1, 499.8944822 MHz
 DATA PROCESSING
 FT size 32768
 Total time 1 min, 32 sec



wp-111-21-ester-c13
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: j-14-87
 INOVA-500 "inv500"
 Relax. delay 1.000 sec
 Pulse 88.4 degrees
 Acq. time 1.308 sec
 Width 25000.0 Hz
 414 repetitions
 OBSERVE C13, 125.6985331 MHz
 DECOUPLE H1, 499.8969017 MHz
 Power 45 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 0.5 Hz
 FT size 65536
 Total time 82 hr, 6 min, 38 sec





STATE UNIVERSITY OF NEW YORK
INOVA 400 Mhz SM# 5811617
ASVpfg PMQBE SN# P085133

1H SENSITIVITY
0.1X ETHYLBENZENE

Pulse Sequence: s2pu1

Solvent: CDCl3

Temp: 25.0 C / 298.1 K

User: 1-12-87

INOVA-400 "inv400"

Relax. delay 1.000 sec

Pulse 39.0 degrees

Acq. time 3.273 sec

Width 6800.6 Hz

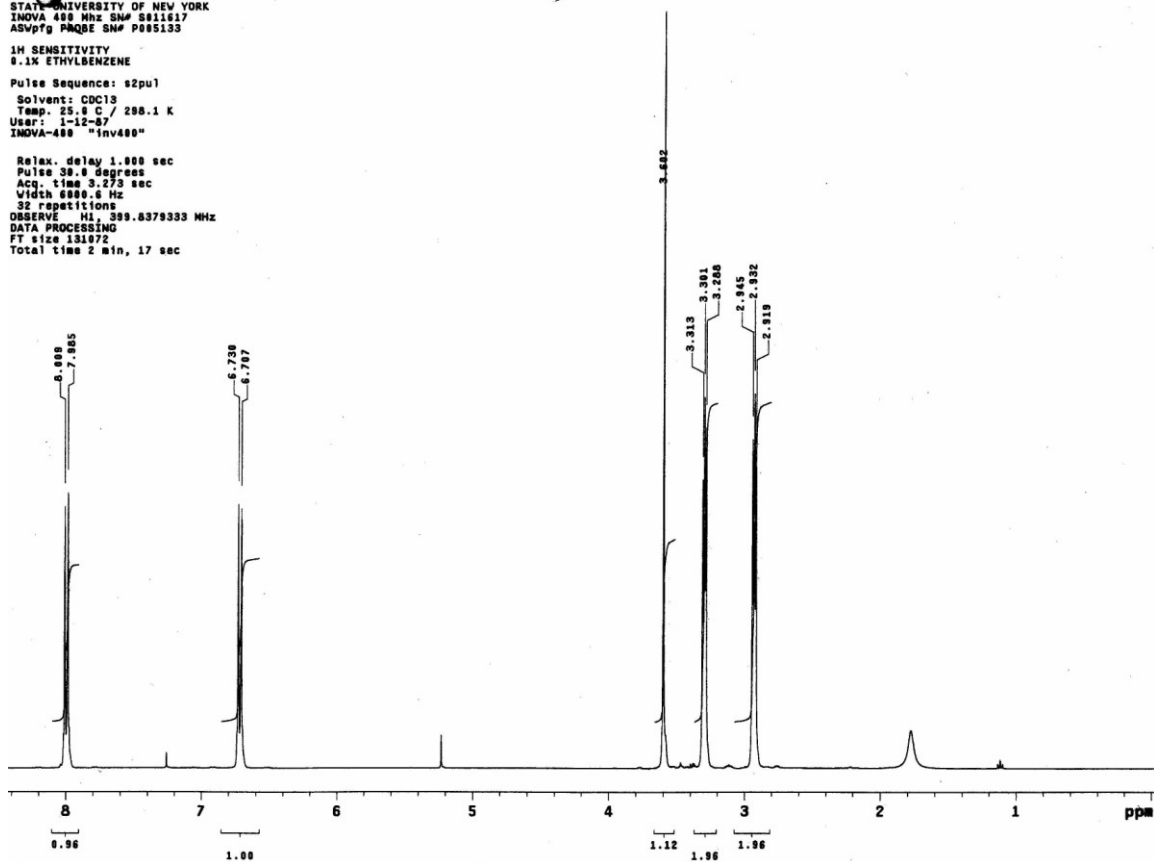
32 repetitions

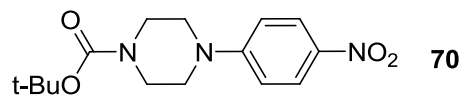
OBSERVE H1, 399.8379333 MHz

DATA PROCESSING

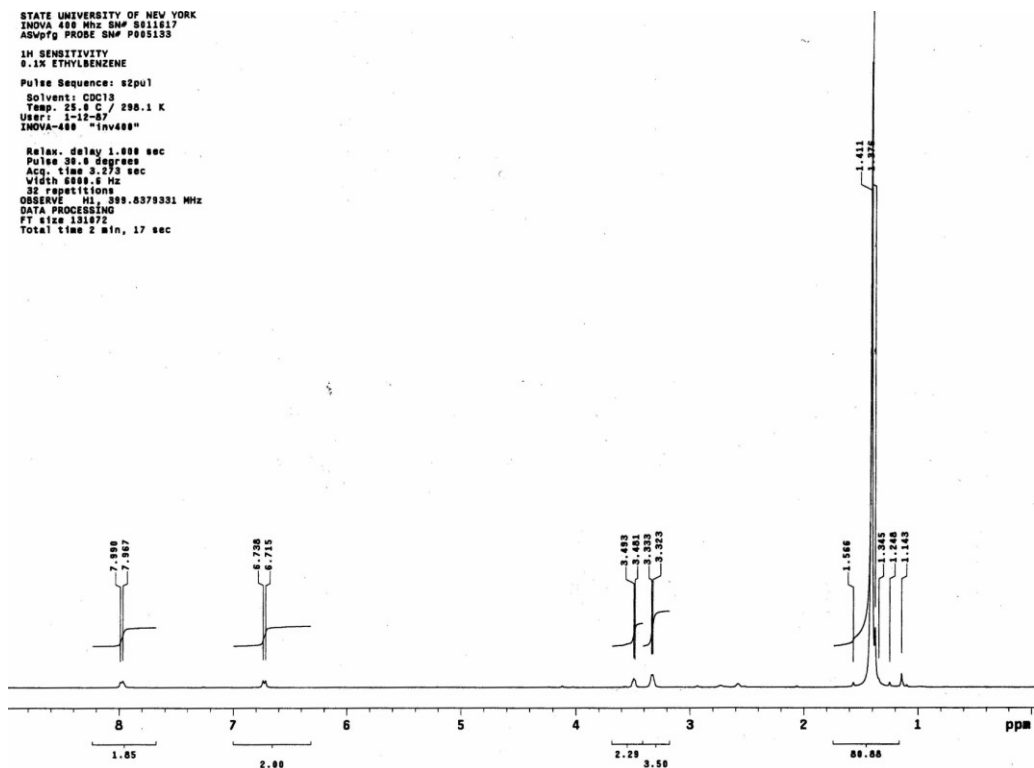
FT size 131072

Total time 2 min, 17 sec



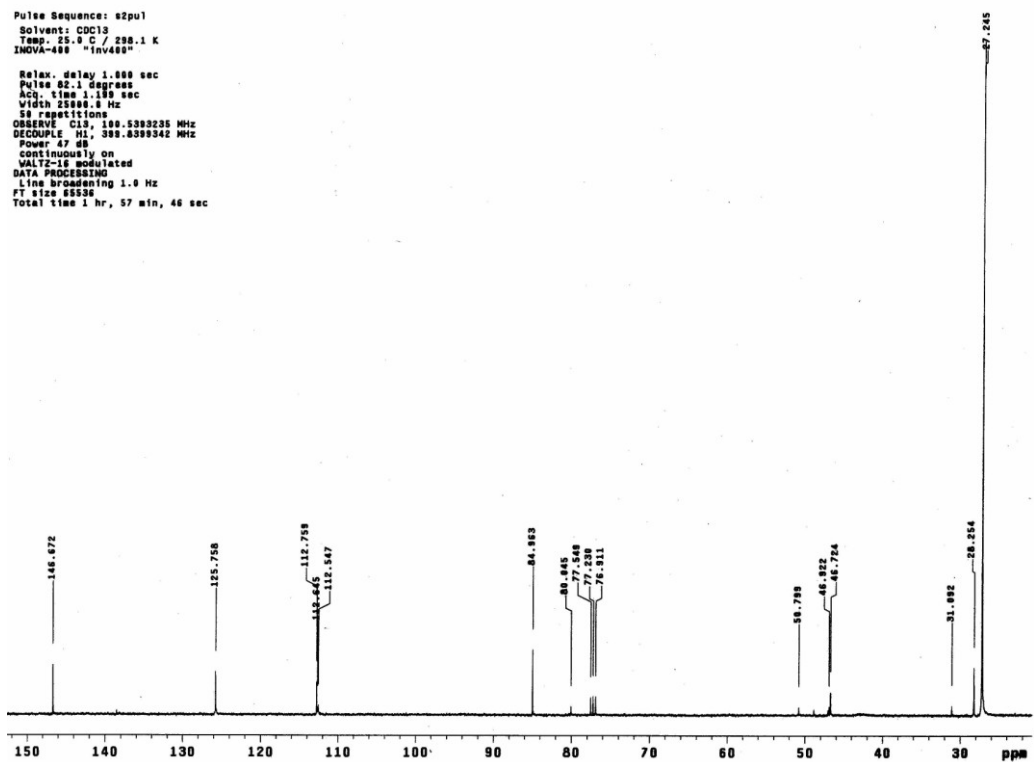


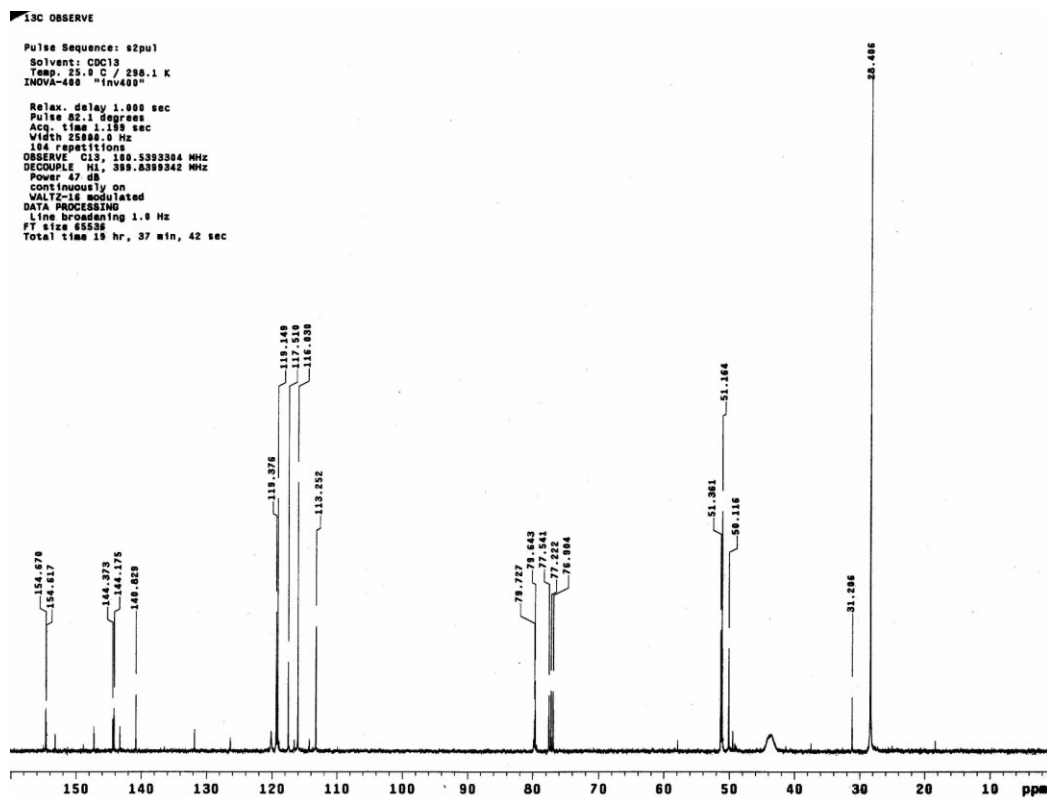
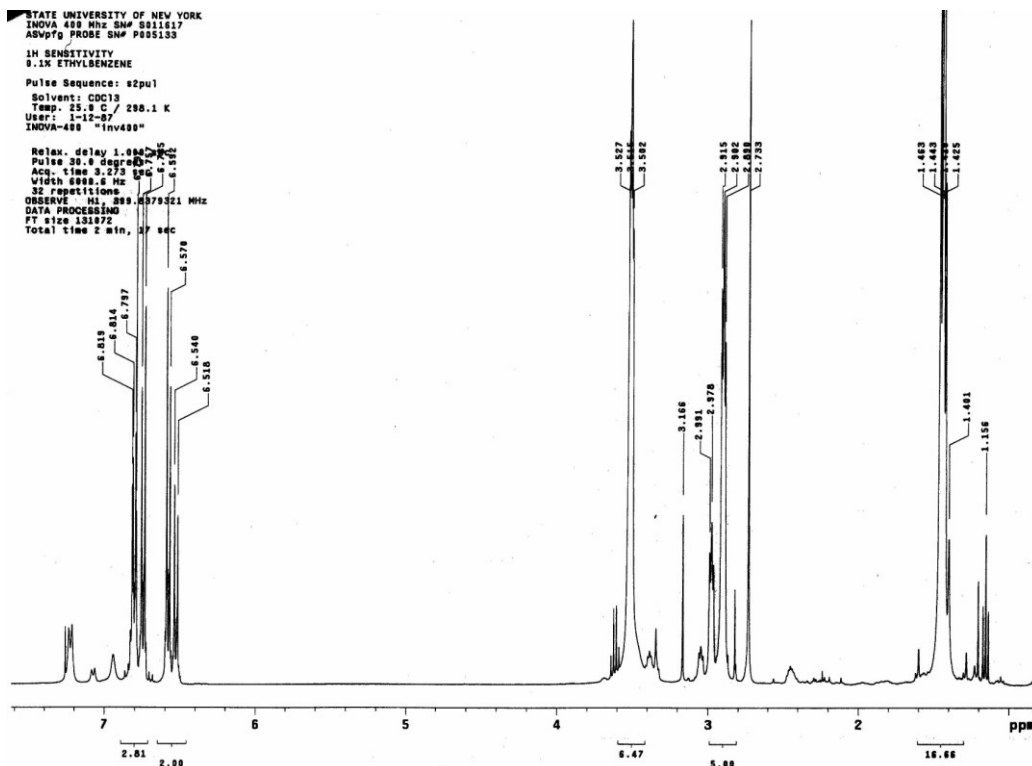
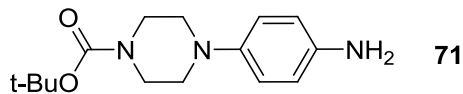
STATE UNIVERSITY OF NEW YORK
 INOVA 400 MHz NMR S611617
 ASVpfg PROBE S6P P05133
 1H SENSITIVITY
 0.1X ETHYLBENZENE
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-87
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 30.0 degrees
 Acq. time 3.273 sec
 Width 6000.0 Hz
 32 repetitions
 OBSERVE H1, 399.6379331 MHz
 DATA PROCESSING
 FT size 131875
 Total time 2 min, 17 sec

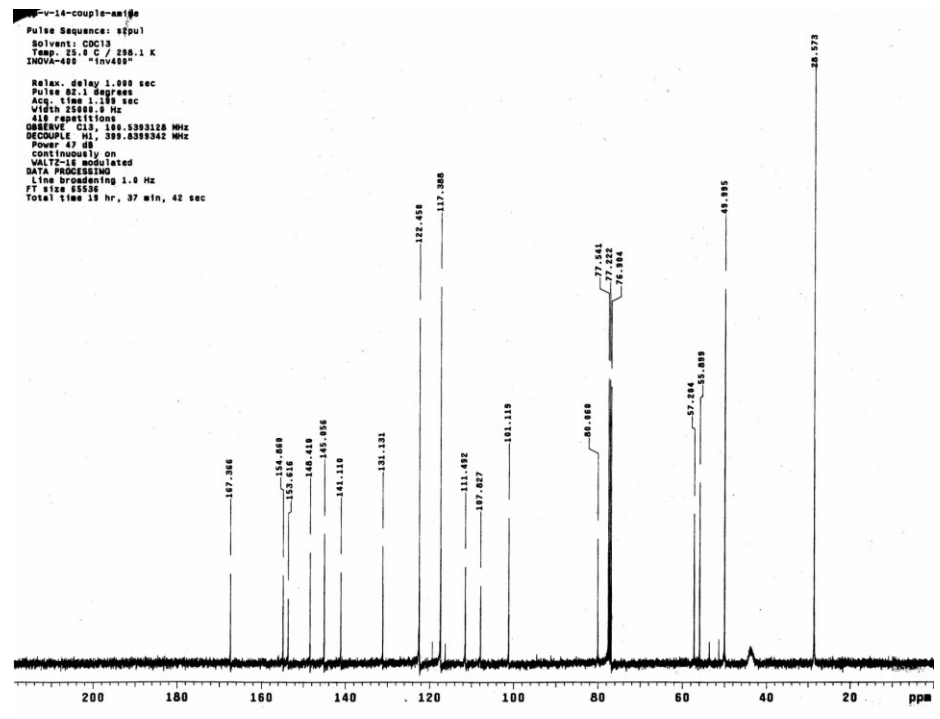
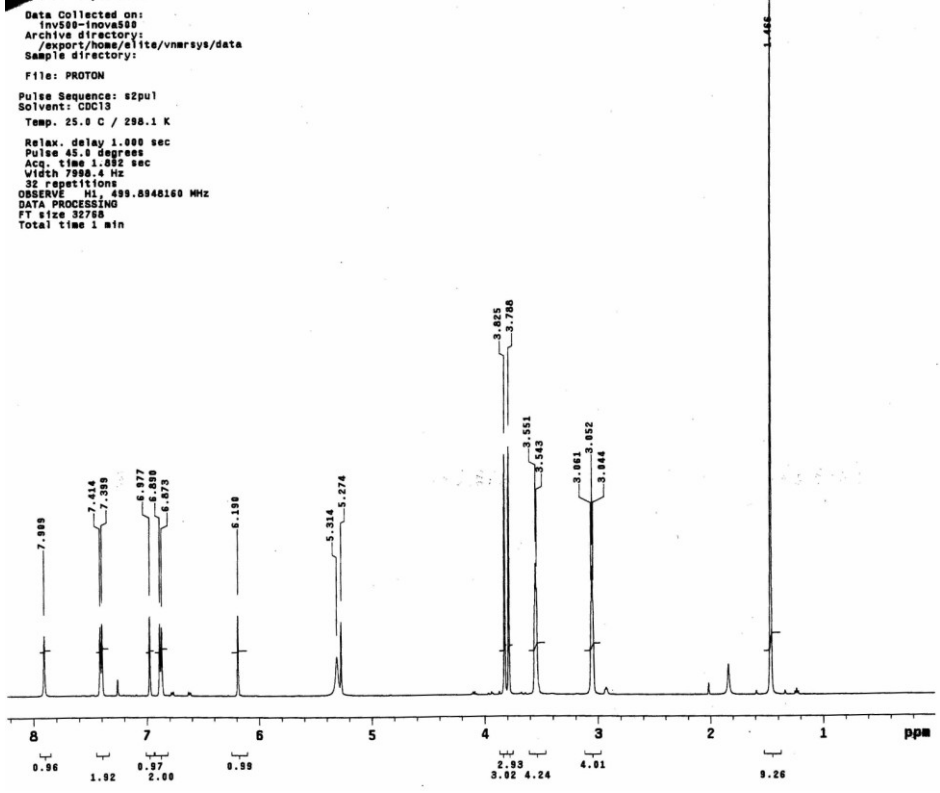
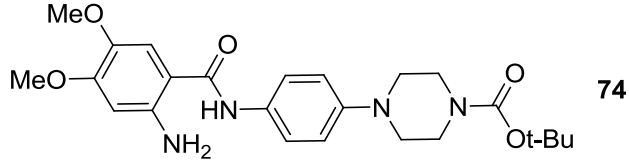


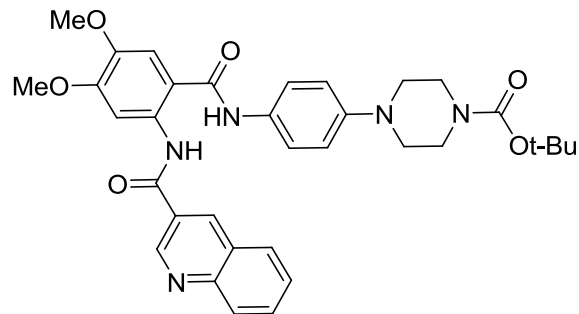
13C OBSERVE

Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.189 sec
 Width 25900.0 Hz
 50 repetitions
 OBSERVE C13, 100.5393235 MHz
 DECOUPLE H1, 399.6379331 MHz
 Power 47 dB
 Continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65538
 Total time 1 hr, 57 min, 46 sec

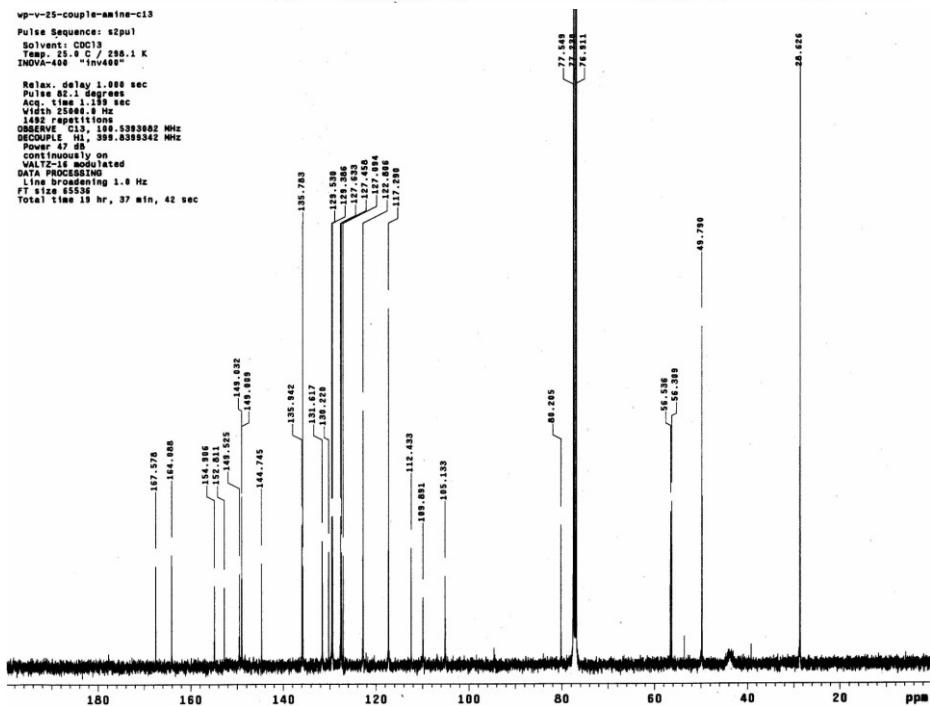
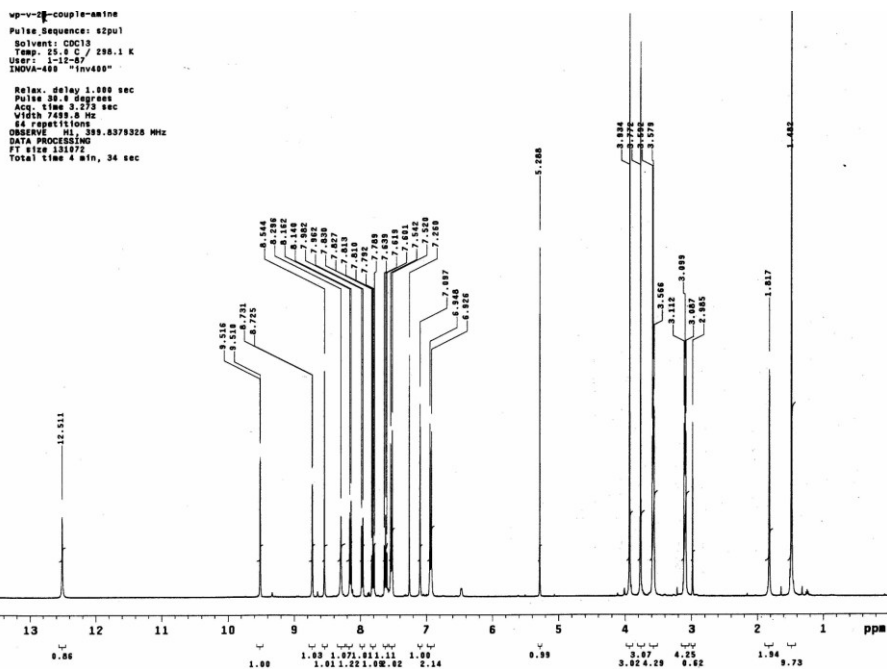


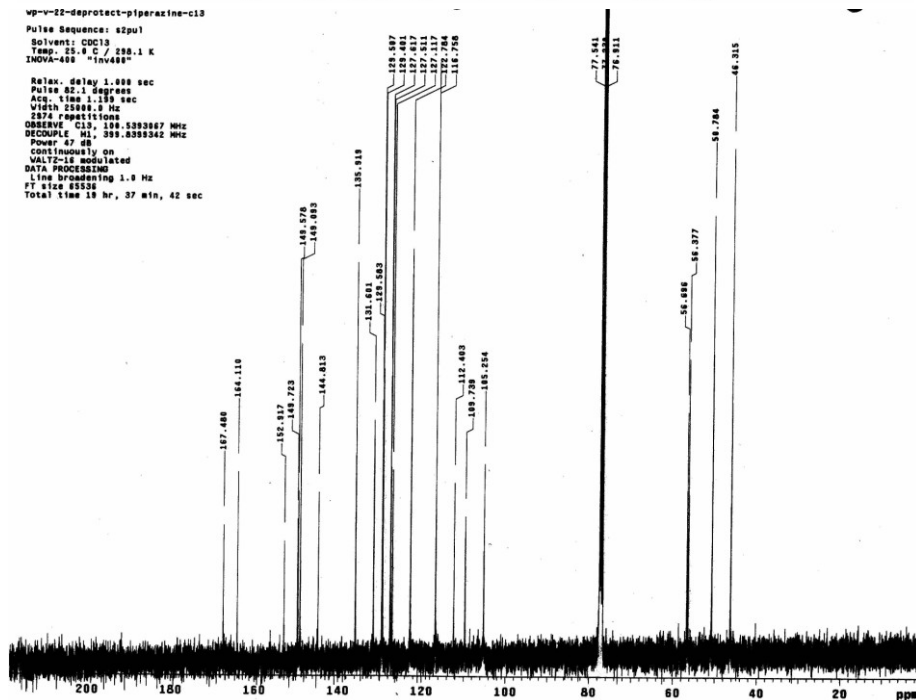
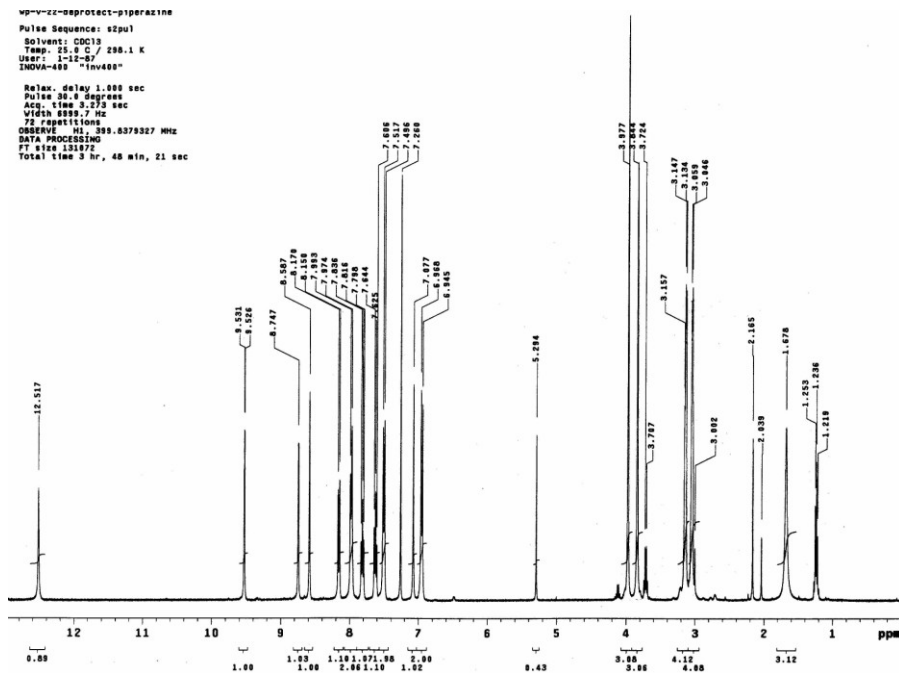
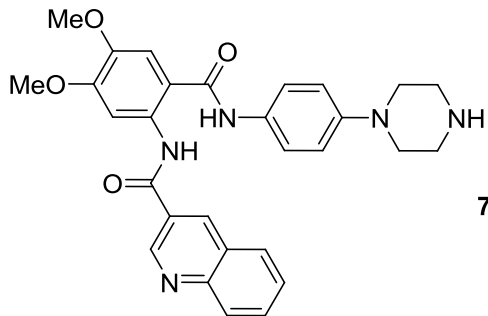


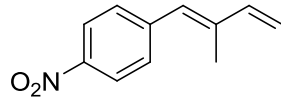




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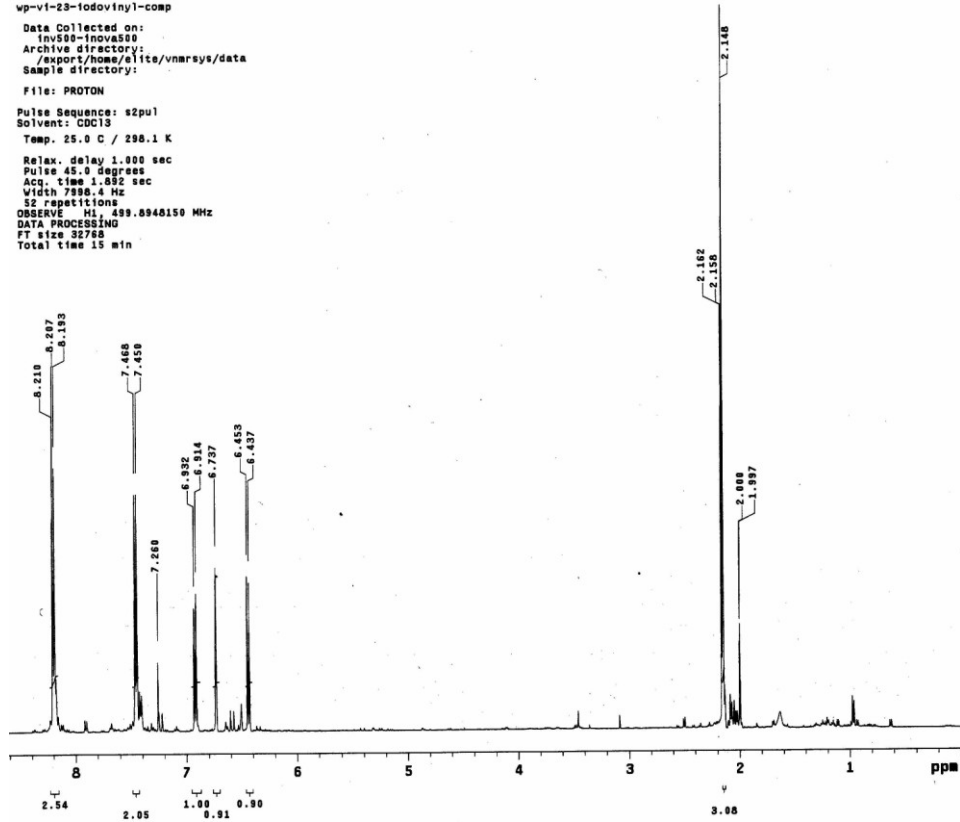




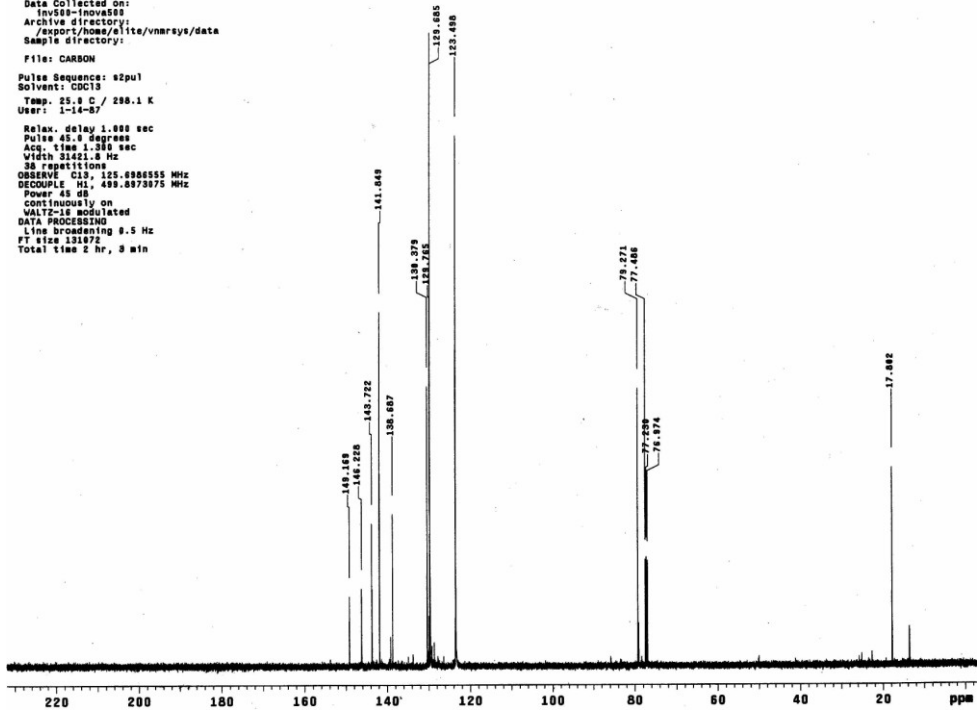


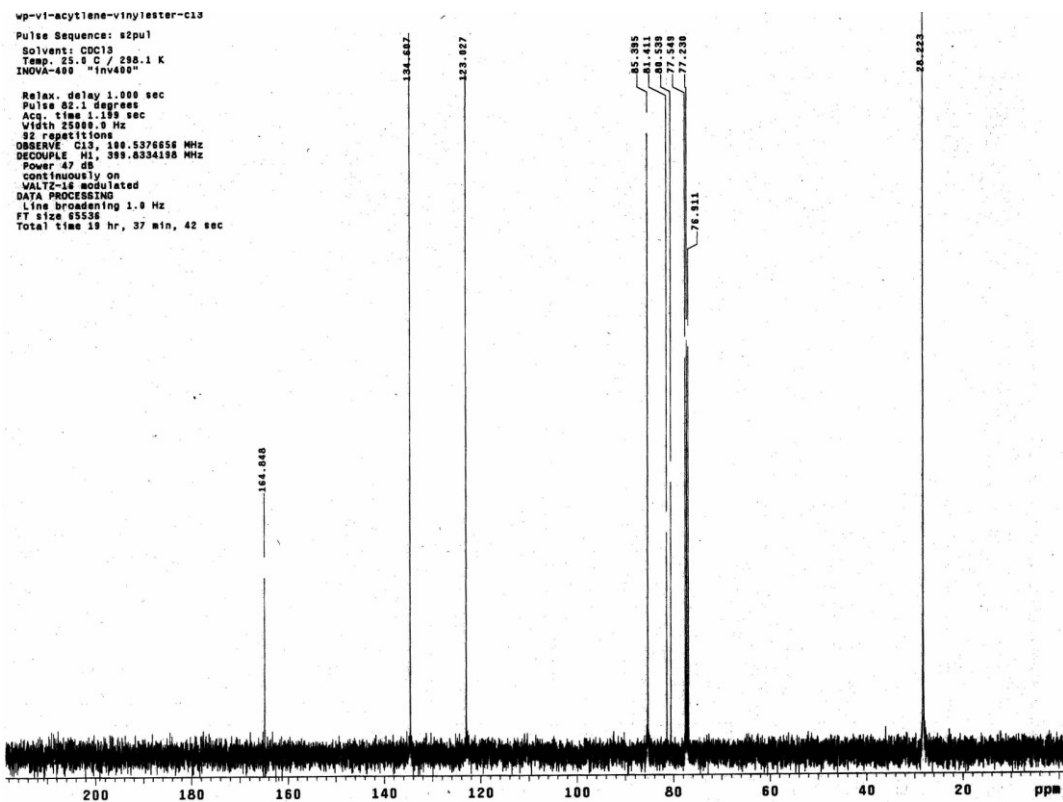
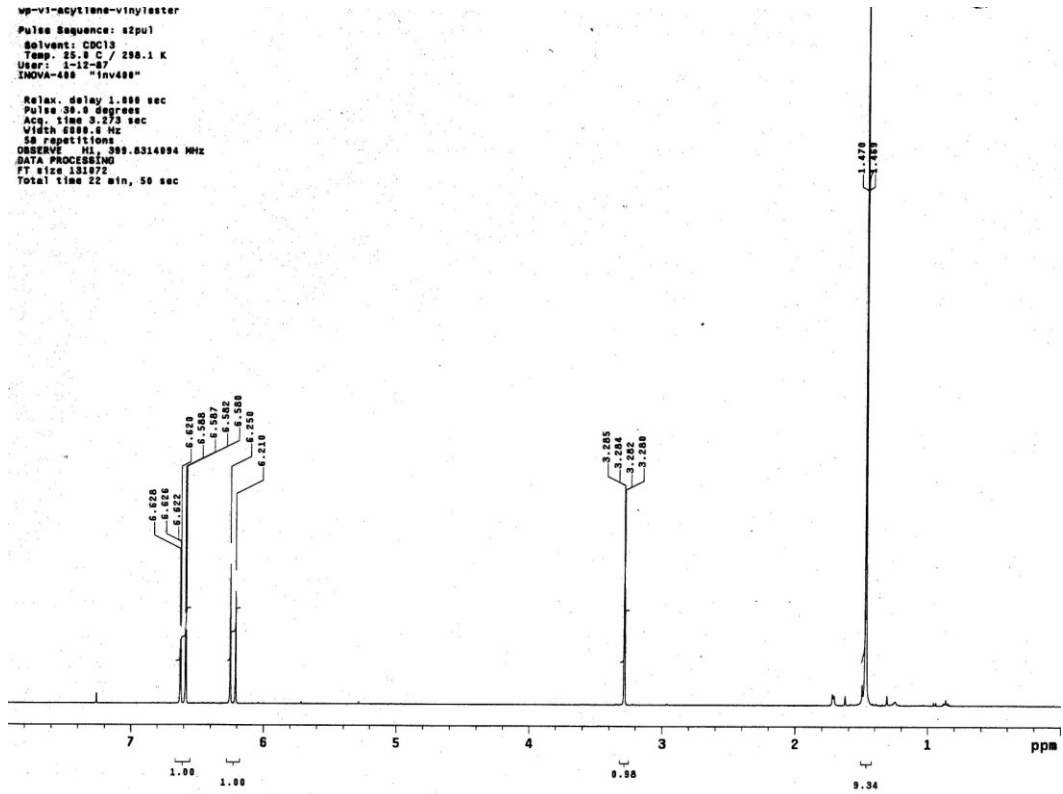
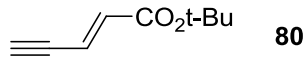
79

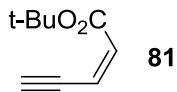
wp-vi-23-iodovinyl-comp
 Data Collected on:
 inv500-inova500
 Archive directory:
 /export/home/elite/vnmrsys/data
 Sample directory:
 File: PROTON
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.852 sec
 Width 7996.4 Hz
 52 repetitions
 OBSERVE H1, 499.8948150 MHz
 DATA PROCESSING
 FT size 32768
 Total time 15 min



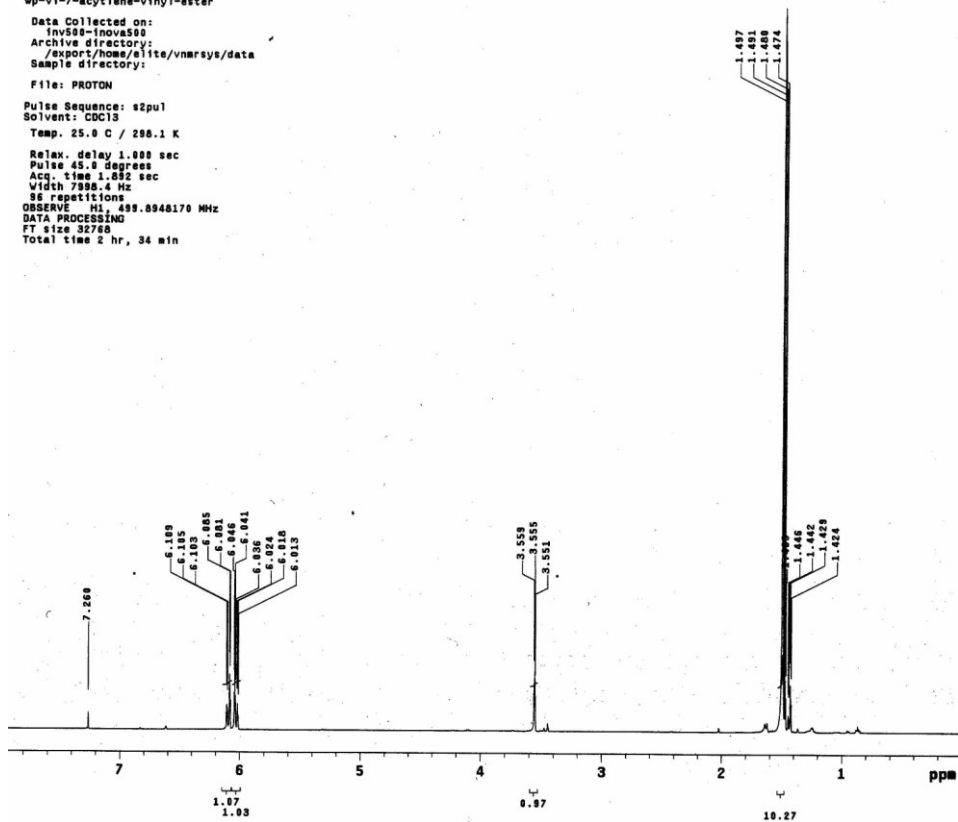
wp-vi-2-cl3-iododiene
 Data Collected on:
 inv500-inova500
 Archive directory:
 /export/home/elite/vnmrsys/data
 Sample directory:
 File: CARBON
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 User: 1-14-87
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.318 sec
 Width 31421.8 Hz
 38 repetitions
 OBSERVE C13, 125.8986555 MHz
 DECOUPLE H1, 499.8973875 MHz
 Power 48 dB
 continuously on
 WALTZ-16 Modulated
 DATA PROCESSING
 Line Broadening 9.5 Hz
 FT size 131872
 Total time 2 hr, 3 min





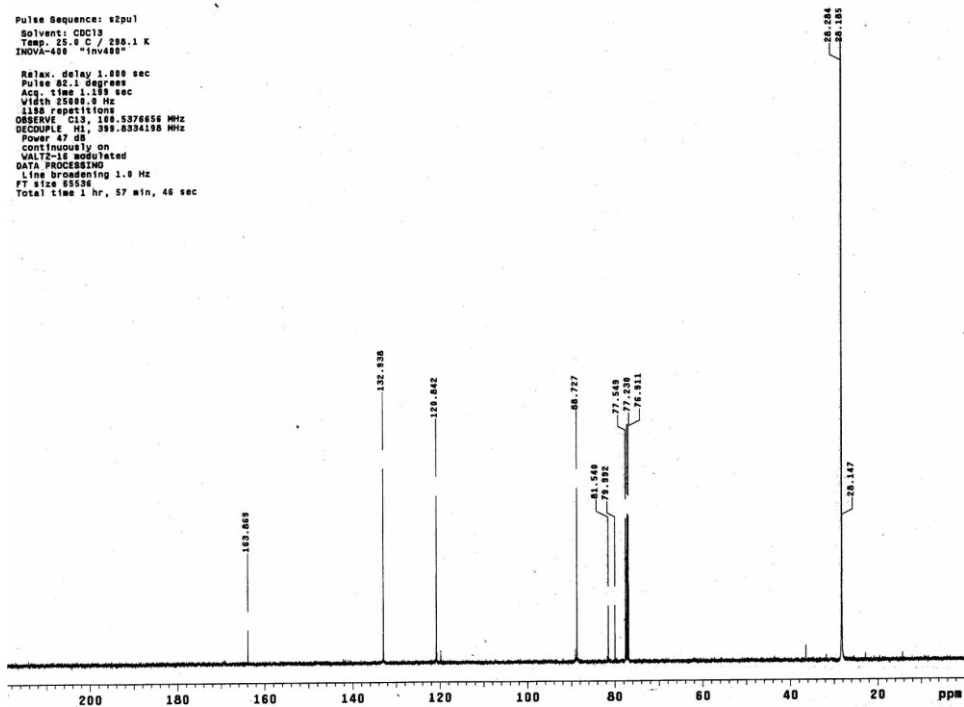


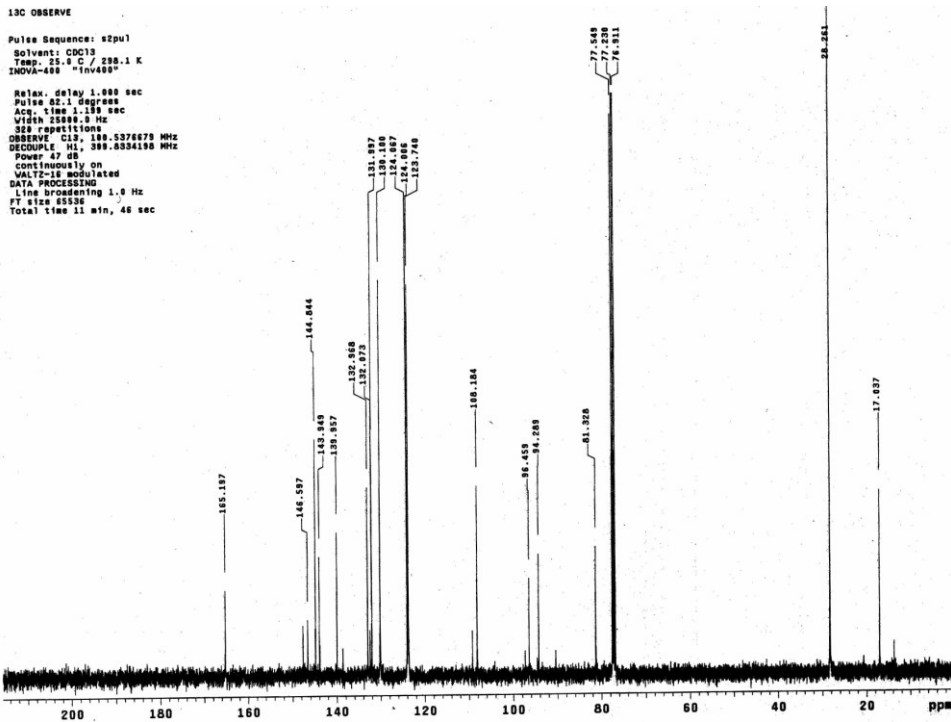
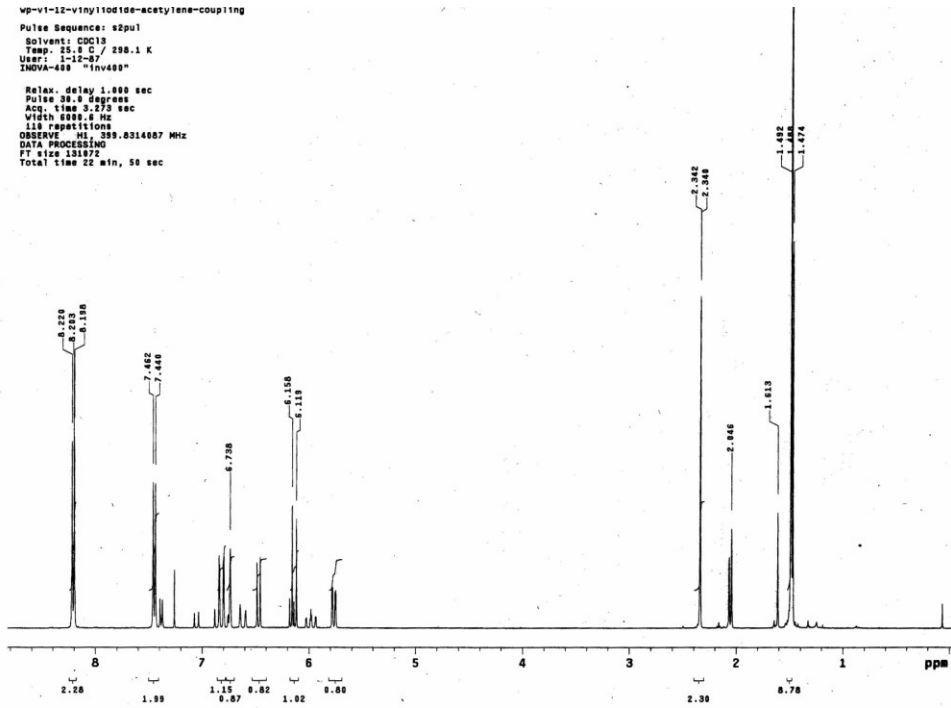
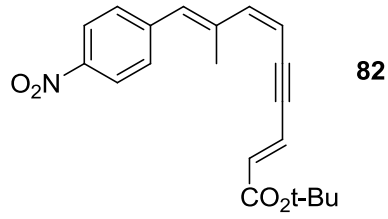
wp-vi-7-acrylene-vinyl-ester
 Data Collected on:
 Inv589-Inv4588
 Archive directory:
 /export/home/elite/vnarsys/data
 Sample directory:
 File: PROTON
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.032 sec
 Width 7098.4 Hz
 86 repetitions
 OBSERVE HI, 499.8948170 MHz
 DATA PROCESSING
 FT size 32788
 Total time 2 hr, 34 min

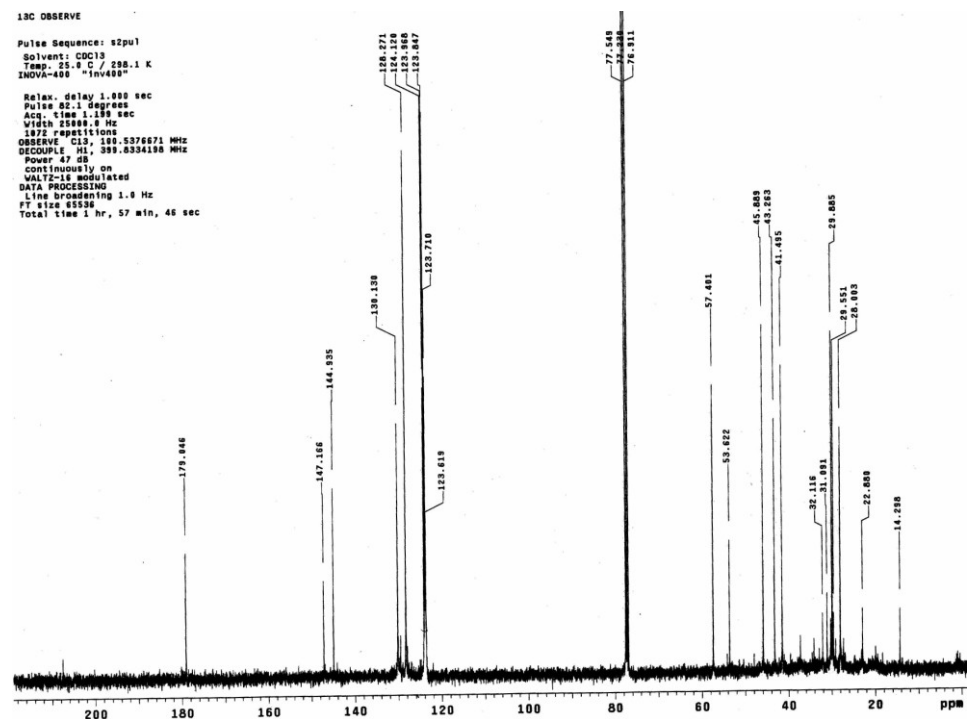
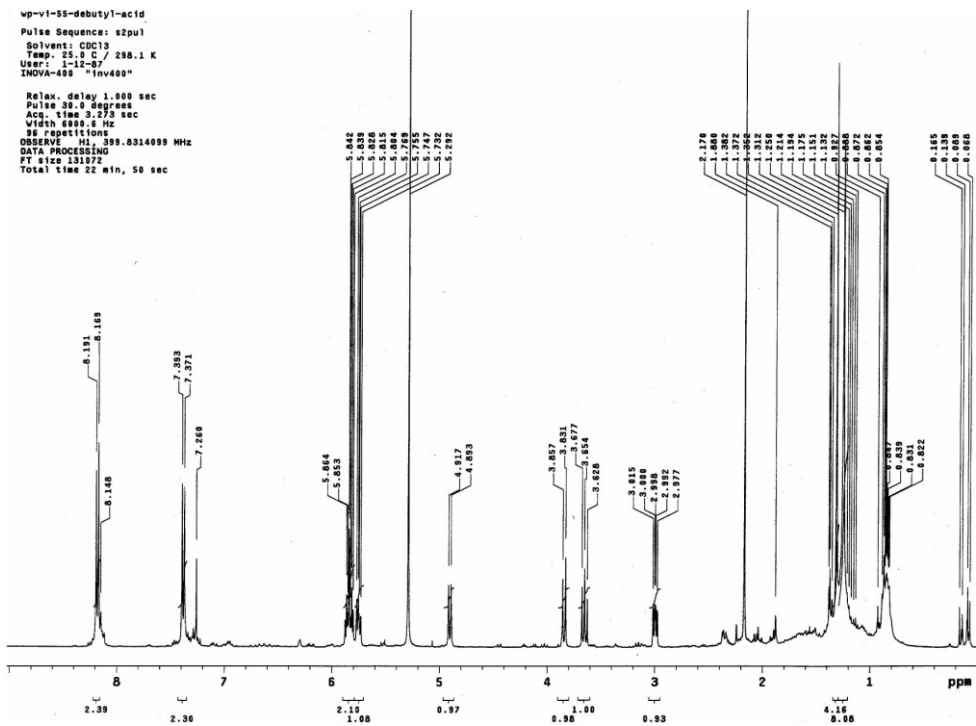
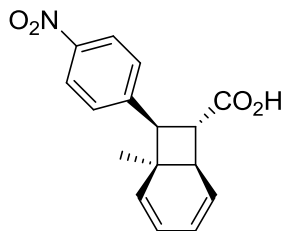


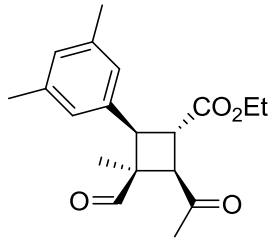
13C UNDECODE

Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 INOVA-400 "Inv488"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.189 sec
 Width 25900.0 Hz
 1338 repetitions
 OBSERVE CH, 100.5376656 MHz
 DECOUPLE HI, 399.8394198 MHz
 Power 47 dB
 continuously on
 WALTZ16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 81538
 Total time 1 hr, 57 min, 46 sec



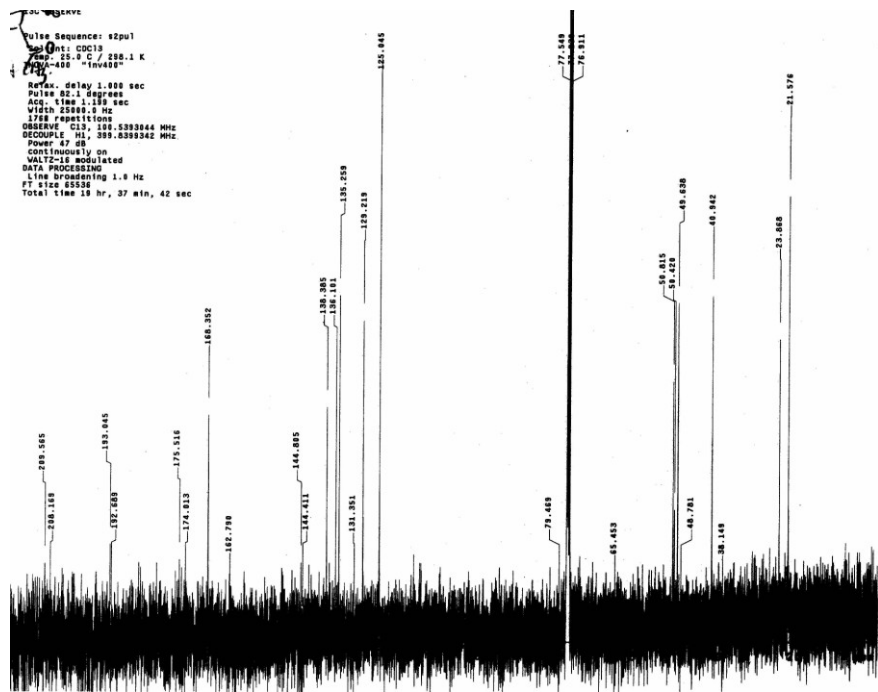
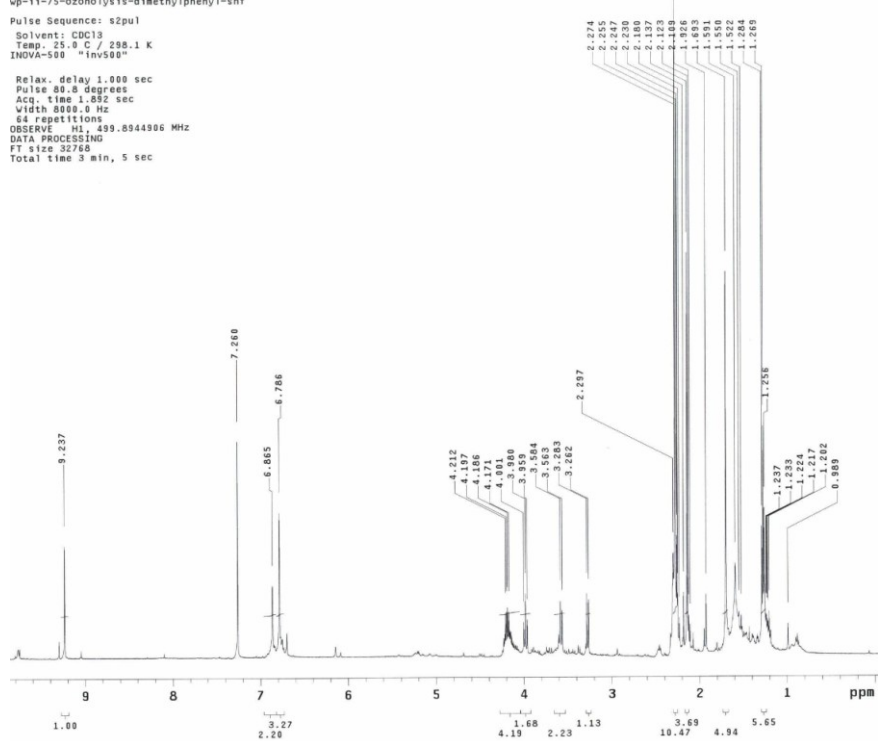


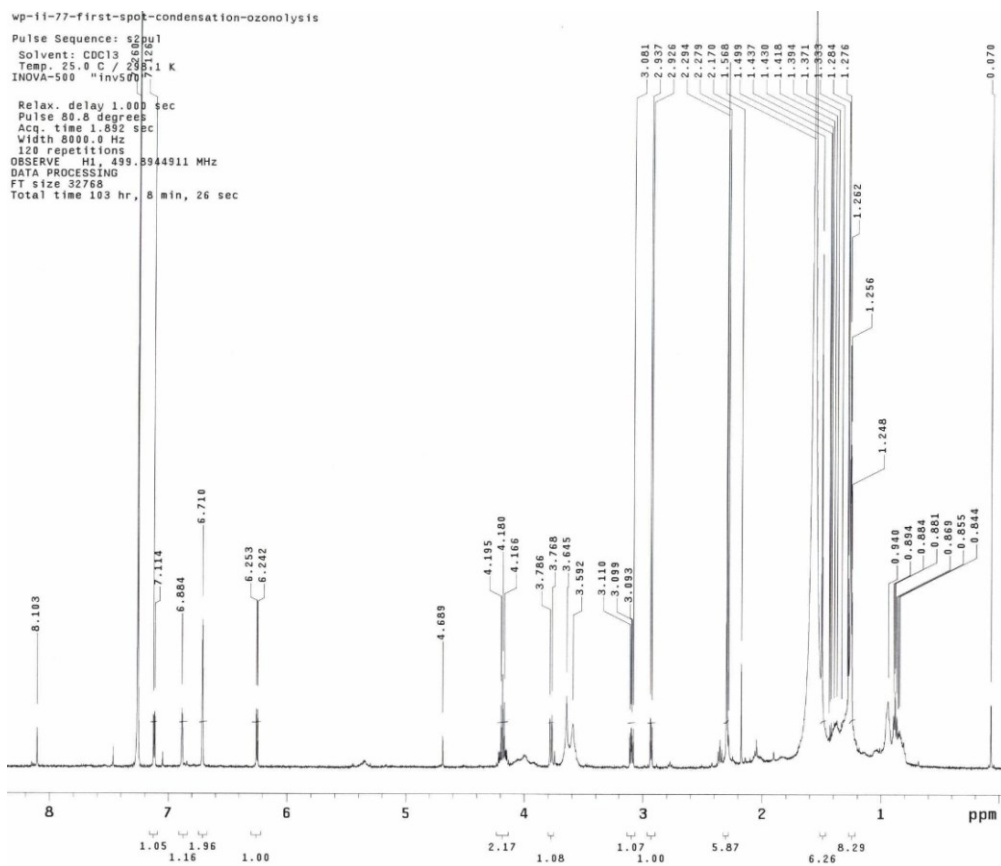
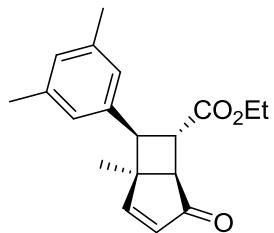


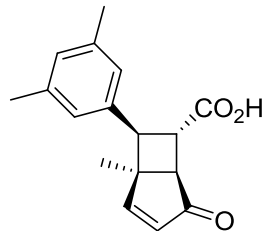


wp-11-73-ozonolysis-1,2-dimethylphenyl-snr
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-500 "inv500"

Relax. delay 1.000 sec
 Pulse 80.8 degrees
 Acq. time 1.582 sec
 Width 8000.0 Hz
 #s repetitions
 OBSERVE H1, 499.8944966 MHz
 DATA PROCESSING
 FT size 32768
 Total time 3 min, 5 sec

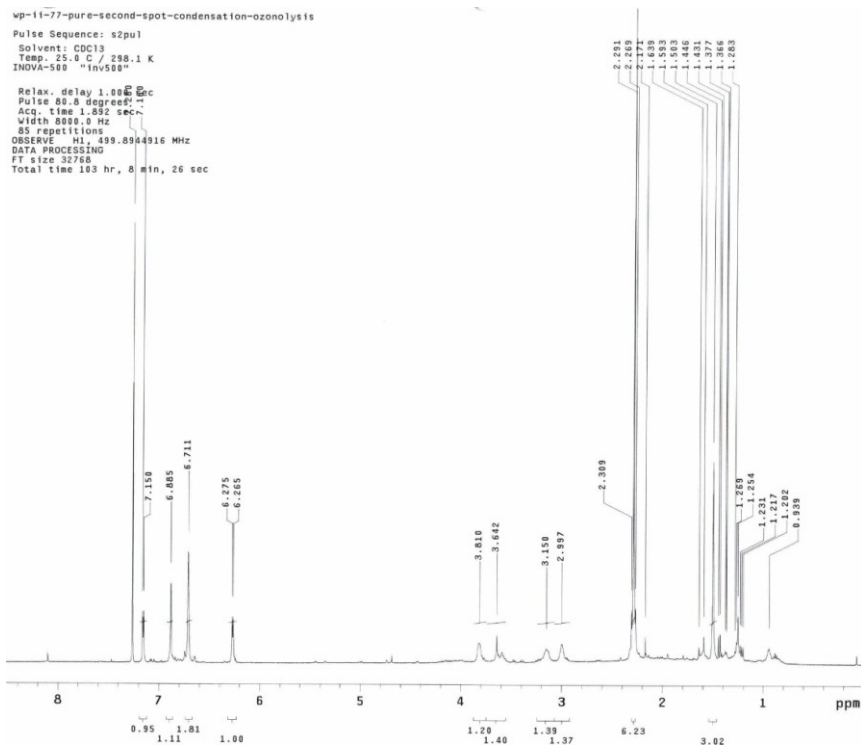




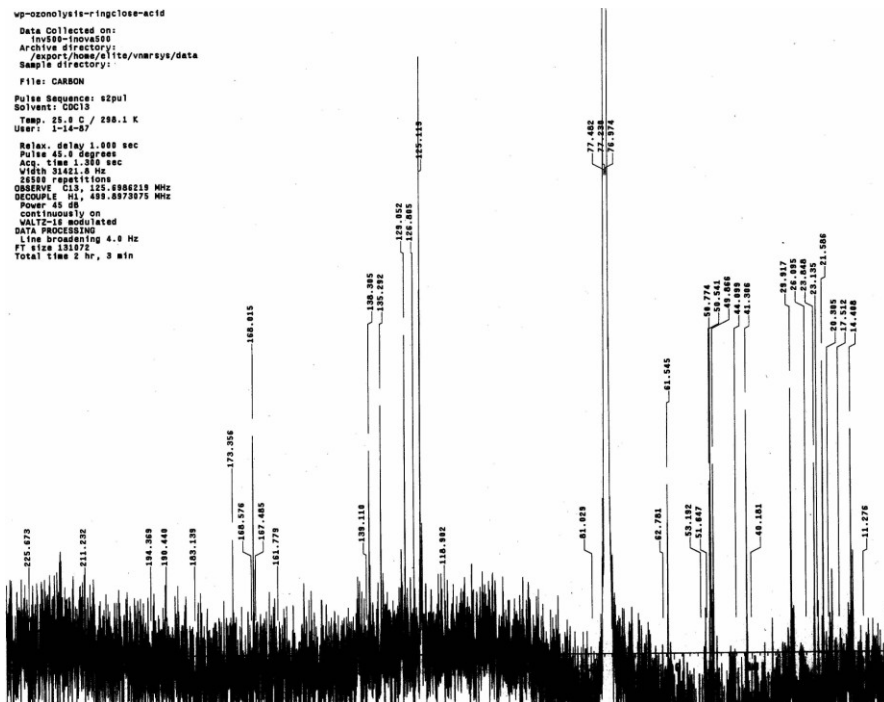


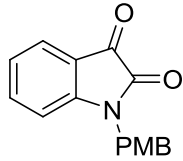
95

wp-11-77-pure-second-spot-condensation-ozonolysis
Pulse Sequence: s2pul
Solvent: CDCl₃
Temp: 25.0 C / 298.1 K
INOVA-500 "1h50p"
Relax. delay 1.000 sec
Pulse 88.8 degrees
Acq. time 1.892 sec
Width 8880.0 Hz
85 repetitions
OBSERVE H1, 499.8944916 MHz
DATA PROCESSING
FT size 32768
Total time 103 hr, 0 min, 26 sec

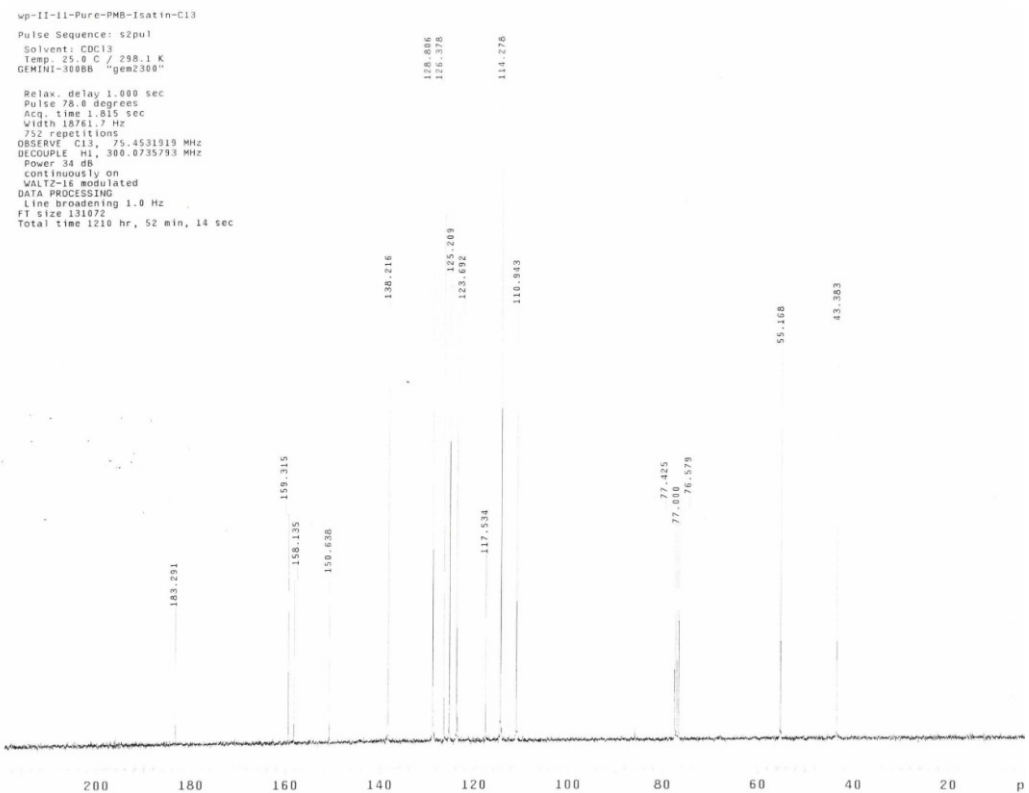
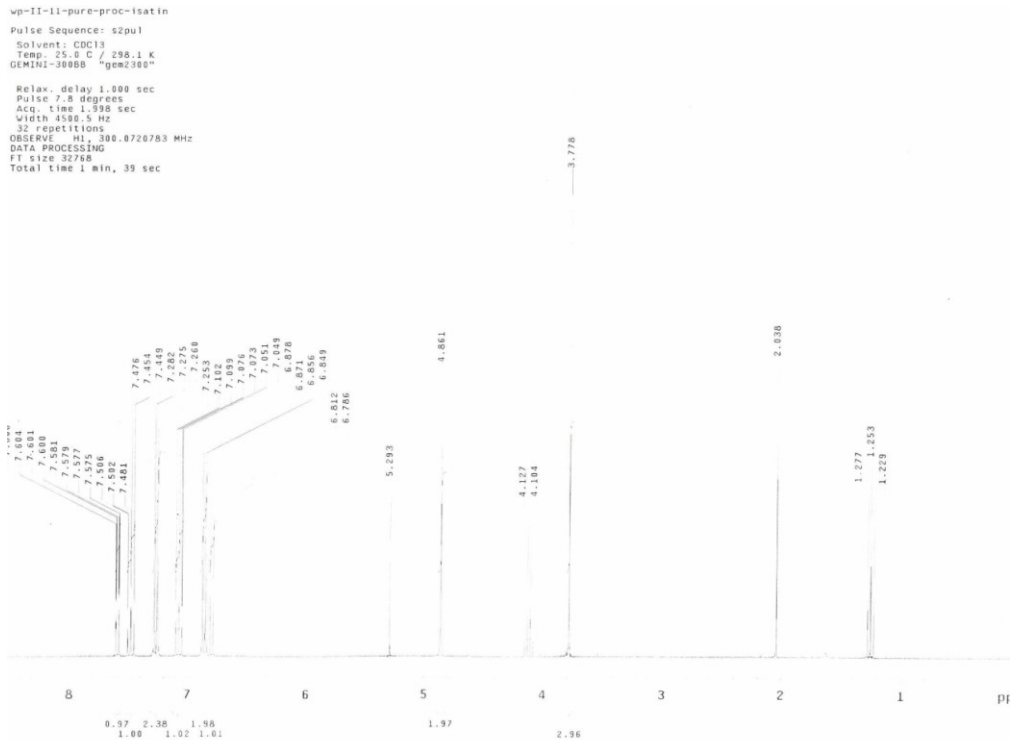


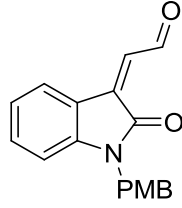
wp-ozonolysis-r-1ngc1ose-acid
Data Collected on:
Invs00-Inova500
Archive directory:
exp001/home/vlita/vmrays/data
Sample directory:
File: CARBON
Pulse Sequence: s2pul
Solvent: CDCl₃
Temp: 25.0 C / 298.1 K
User: 1-14-07
Relax. delay 1.000 sec
Pulse 45.0 degrees
Acq. time 1.350 sec
Width 31822.0 Hz
28180 repetitions
OBSERVE C13, 125.6982218 MHz
DECOUPLE H1, 499.8973076 MHz
Power 45
continuously on
MAGNETIC modulation
DATA PROCESSING
Line broadening 4.0 Hz
FT size 181872
Total time 2 hr, 3 min





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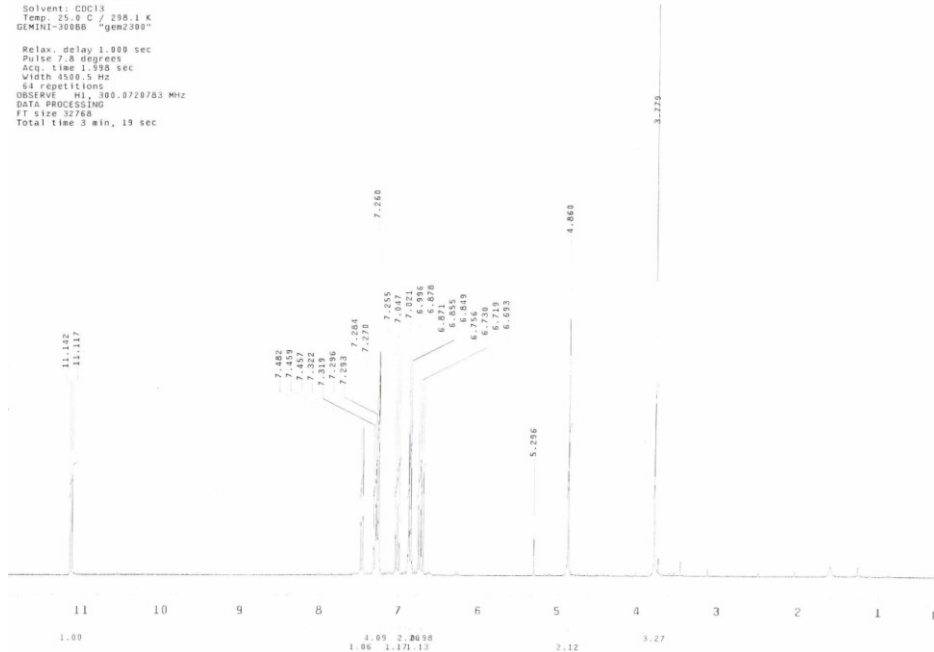




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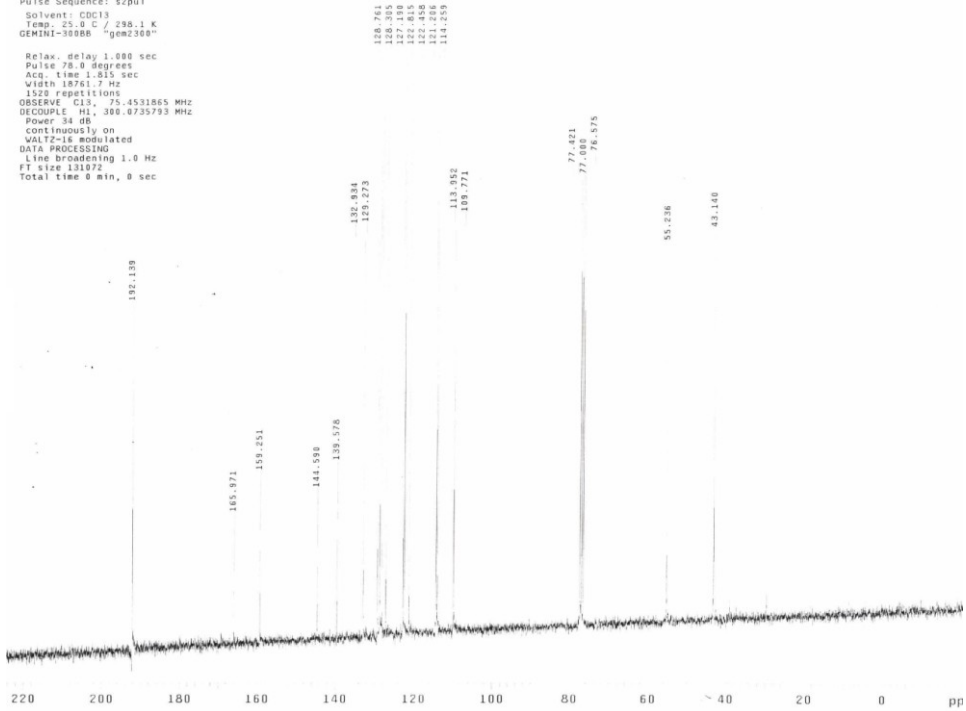
wp-II-12-Pure-pmb-aldehyde
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 GEMINI-300BB "gme230"

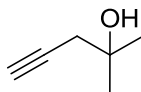
Relax. delay 1.000 sec
 Pulse 7.0 degrees
 Acq. time 1.990 sec
 Width 6500.5 Hz
 64 repetitions
 OBSERVE HI, 300.0720763 MHz
 DATA PROCESSING
 FT size 32768
 Total time 3 min, 19 sec



wp-II-pure-pmb-aldehyde-C13
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 GEMINI-500BB "gme2300"

Relax. delay 1.000 sec
 Pulse 70.0 degrees
 Acq. time 1.815 sec
 Width 18761.7 Hz
 1520 repetitions
 OBSERVE C13, 75.4531865 MHz
 DECOUPLE HI, 300.0735793 MHz
 Power 34 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 131072
 Total time 0 min, 0 sec

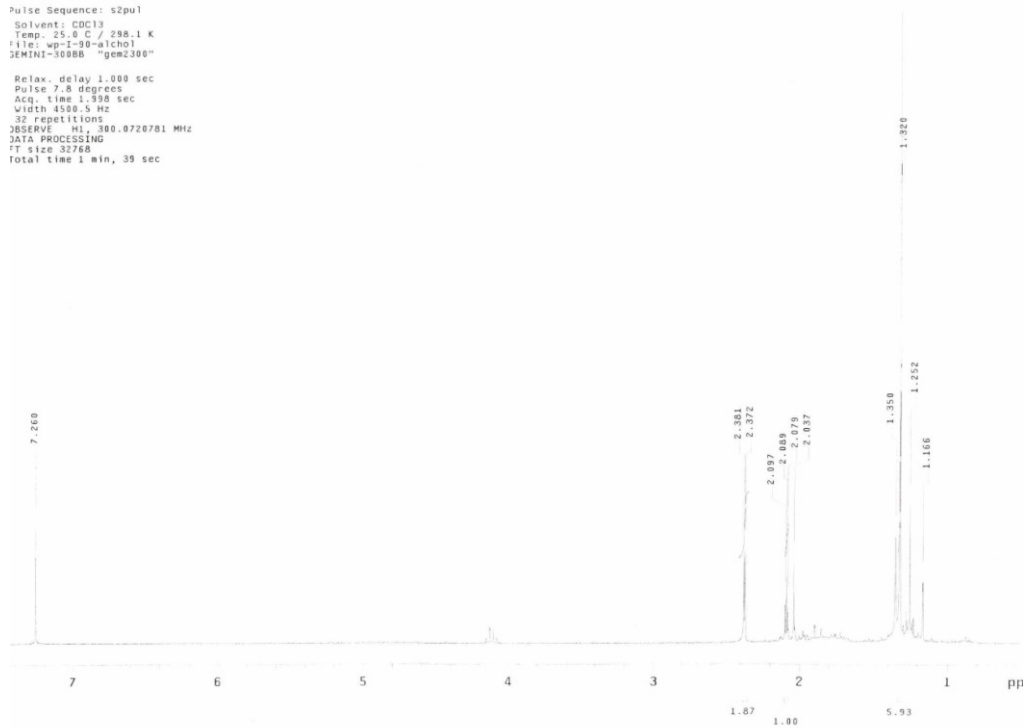


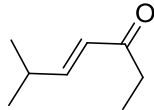


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wp-I-99-alcohol
Pulse Sequence: s2pu1
Solvent: CDCl3
Temp: 25.0 C / 298.1 K
File: wp-I-99-alcohol
JEMINI-30088 "gem2308"

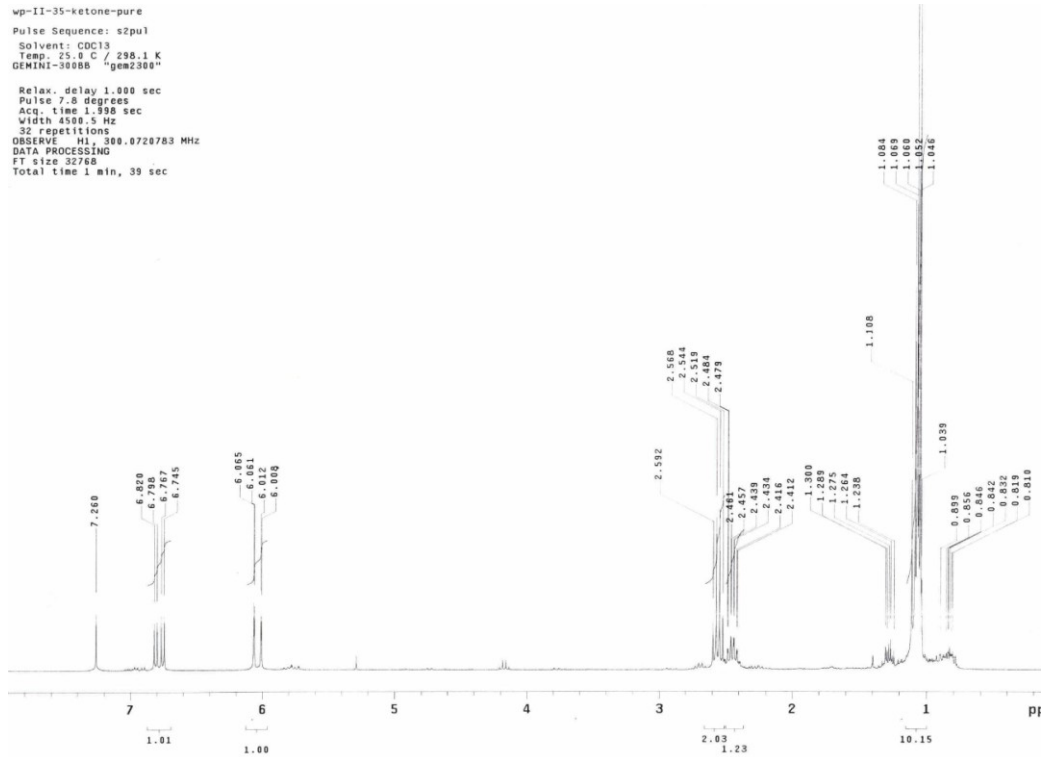
Relax. delay 1.000 sec
Pulse 7.8 degrees
Acq. time 1.998 sec
Width 4500.5 Hz
32 repetitions
OBSERVE: H1, 300.0720701 MHz
DATA PROCESSING
F1 size 32768
Total time 1 min, 39 sec





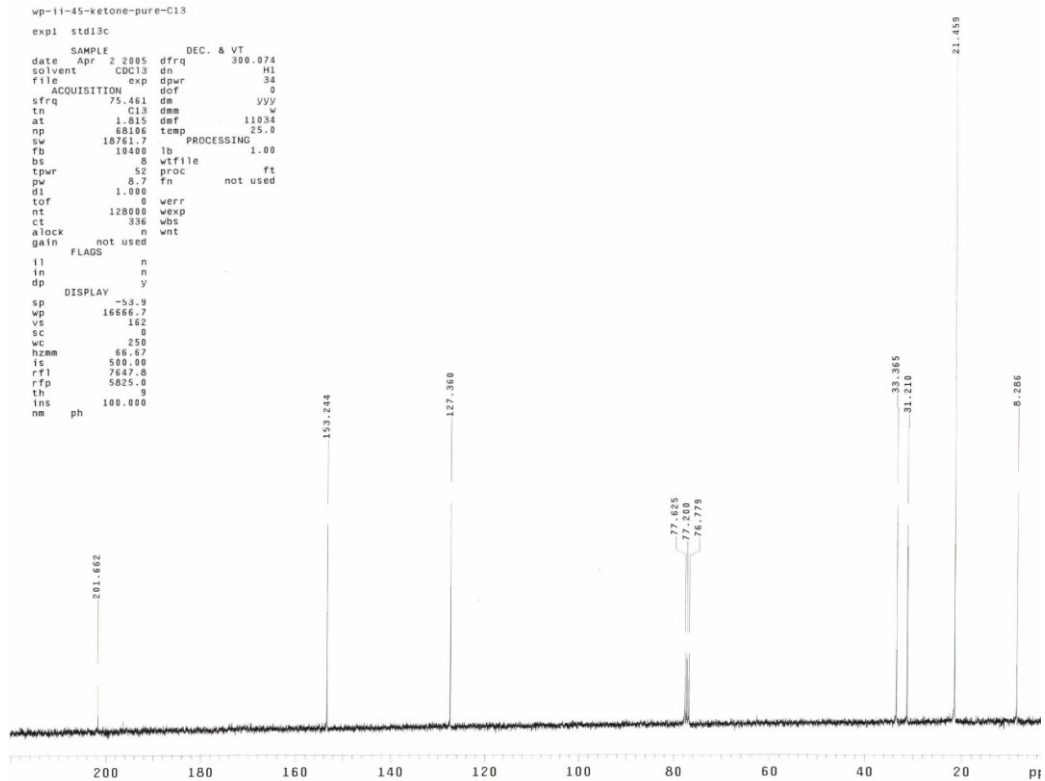
115

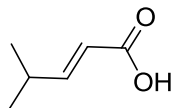
wp-II-35-ketone-pure
Pulse Sequence: s2pul
Solvent: CDCl3
Temp: 25.0 C / 298.1 K
DEMNI-300B "gem2300"
Relax. delay 1.000 sec
Pulse 7.8 degrees
Acq. time 1.998 sec
Width 4500.5 Hz
32 repetitions
OBSERVE H1: 300.0720783 MHz
DATA PROCESSING
FT size 32768
Total time 1 min, 39 sec



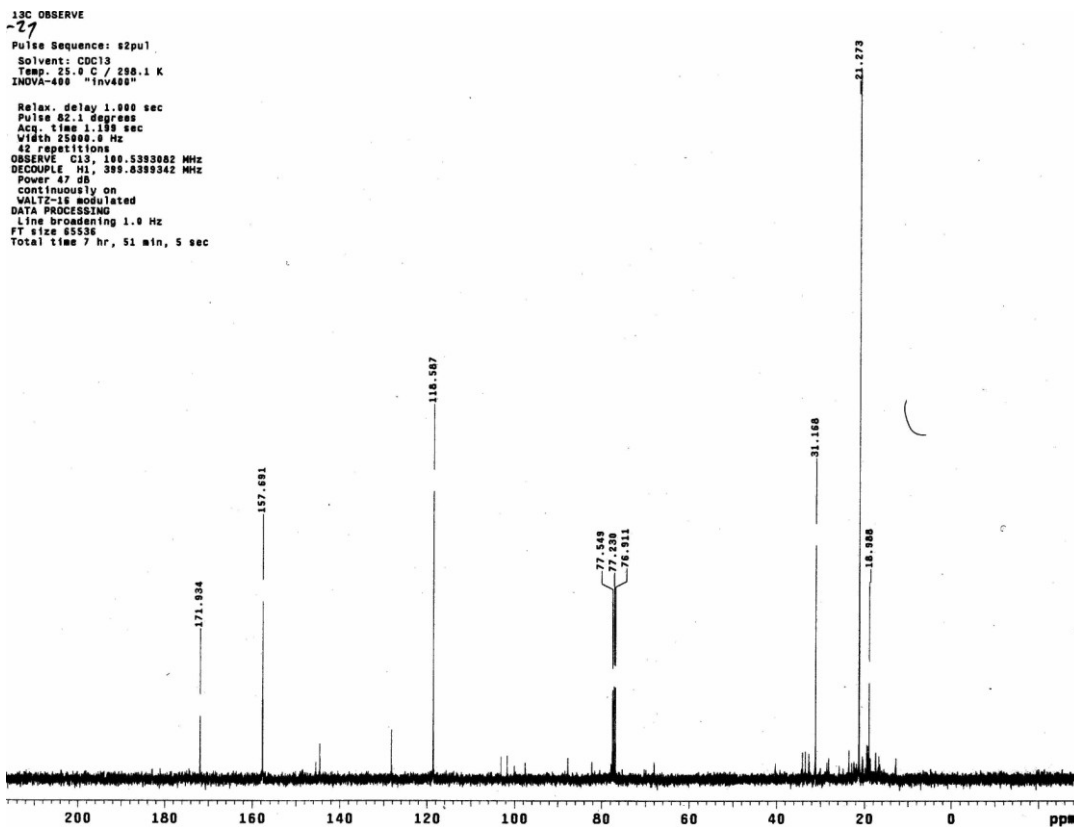
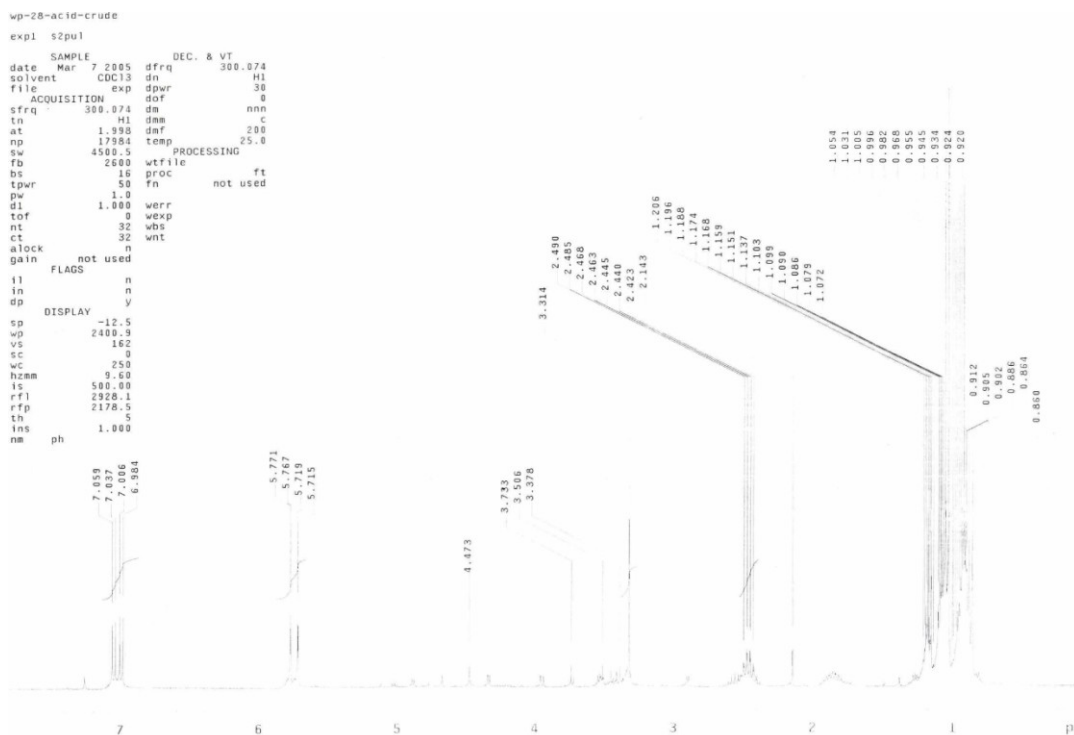
wp-II-45-ketone-pure-Cl3

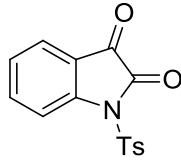
expi std13c
SAMPLE DEC. & VT
date Apr 2 2005 dfrq 300.074
solvent CDCl3 dn H1
file exp dpwr 34
ACQUISITION exp dof 0
sfrq 75.461 dn vvy
tn C13 dmm
at 1.815 def 11034
np 68106 temp 25.0
sw 18761.7 PROCESSING 1.00
fb 10400 lb
bs 8 wtfile ft
tpwr 52 proc not used
pw 8.7 fn
d1 1.000
tof 0 werr
nt 128000 wexp
ct 396 wbs
alock n wnt
gain not used
FLAGS n
i1 n
in n
dp DISPLAY y
sp -53.9
wp 16686.7
vs 162
sc 0
wc 250
hzmm 66.67
is 500.00
rfl 7647.8
rfp 5825.0
th 9
ins 100.000
nm ph



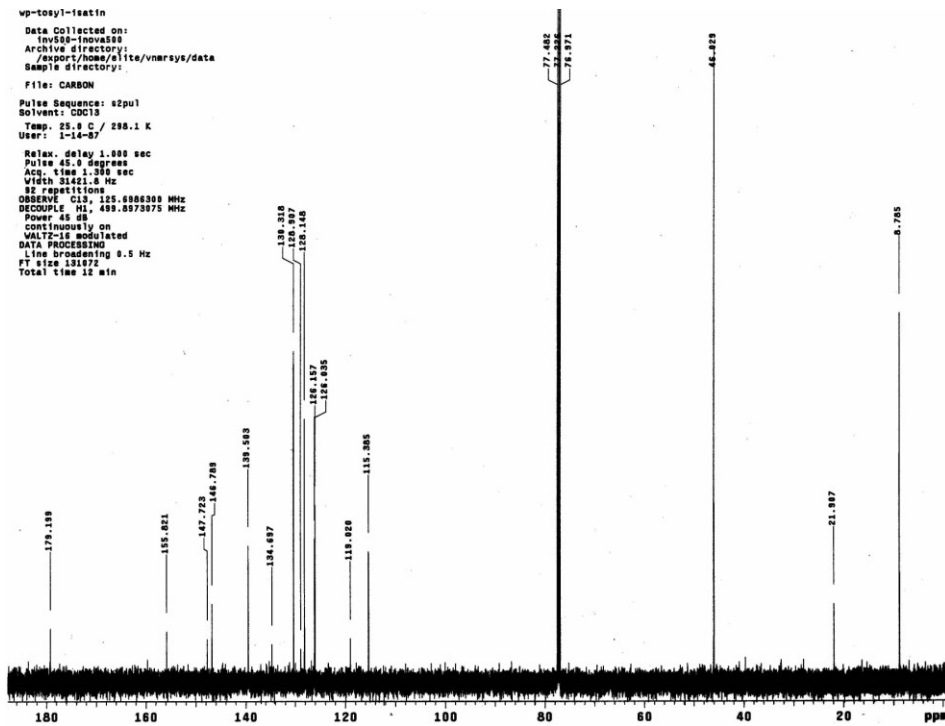
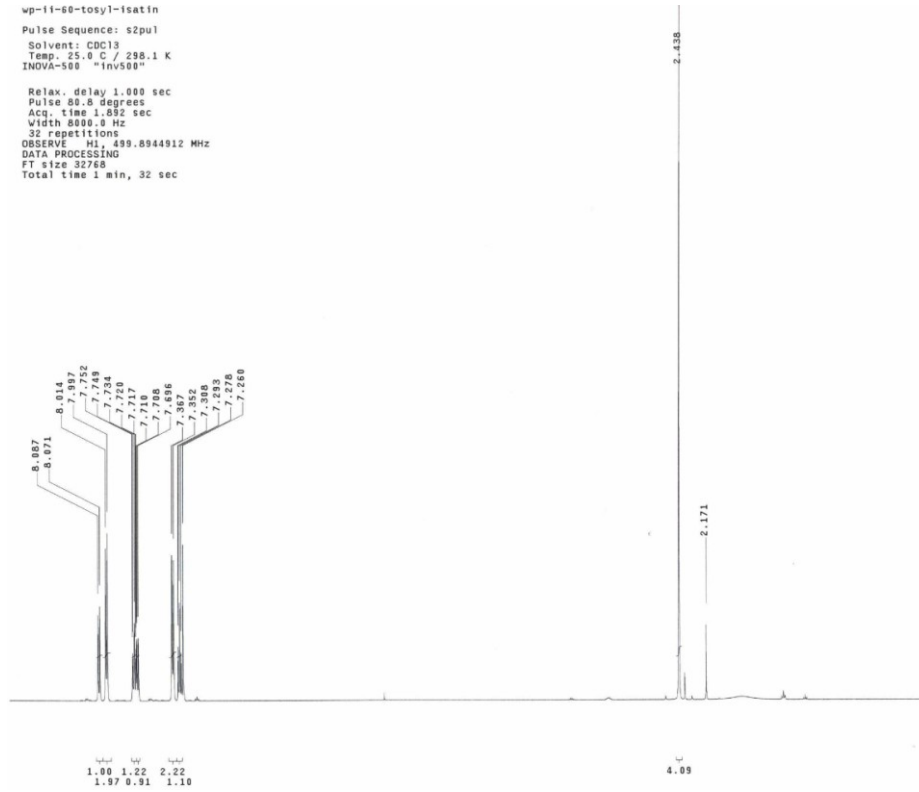


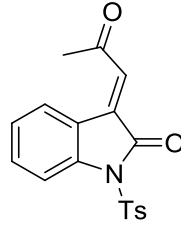
117



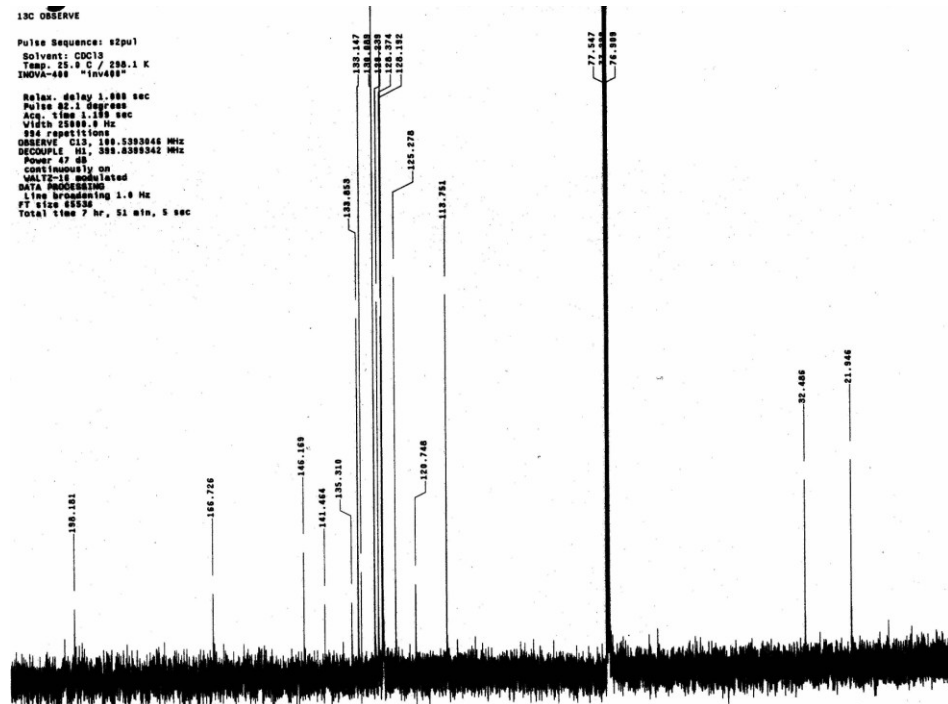
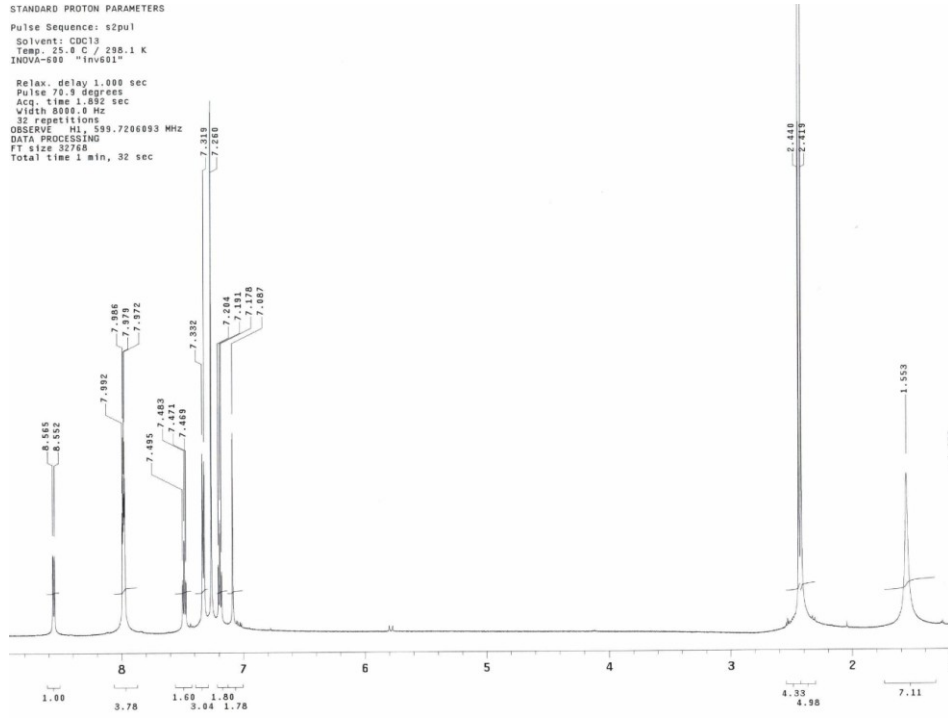


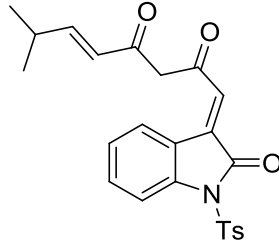
118



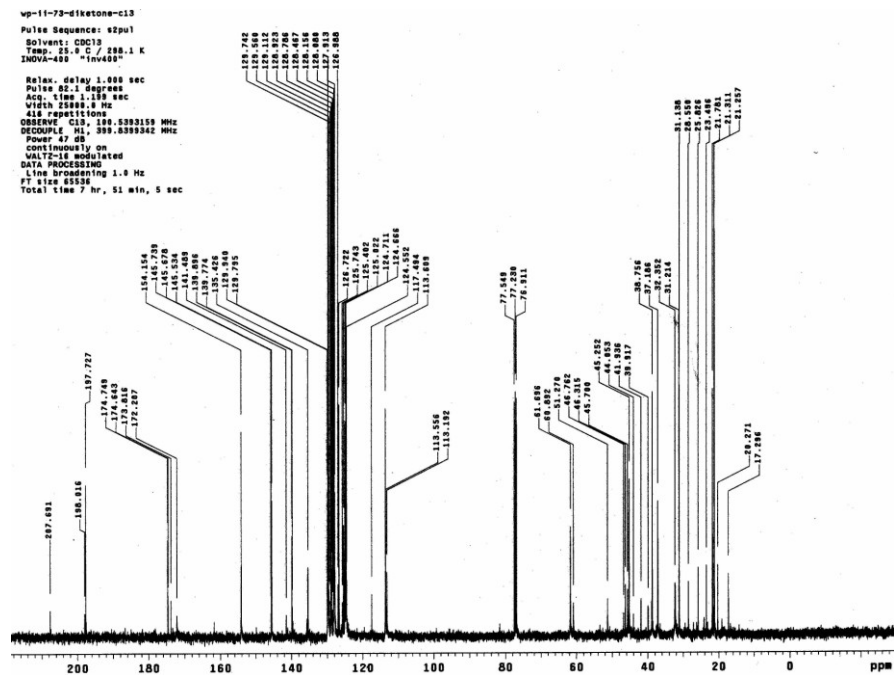
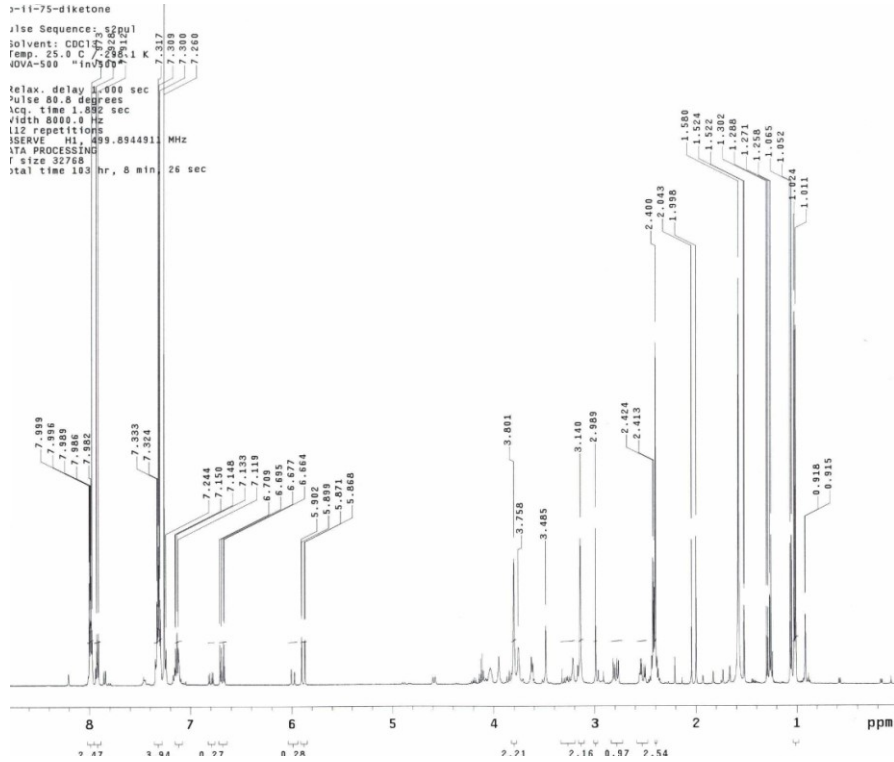


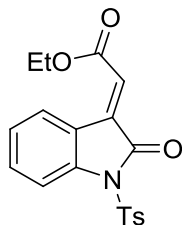
120



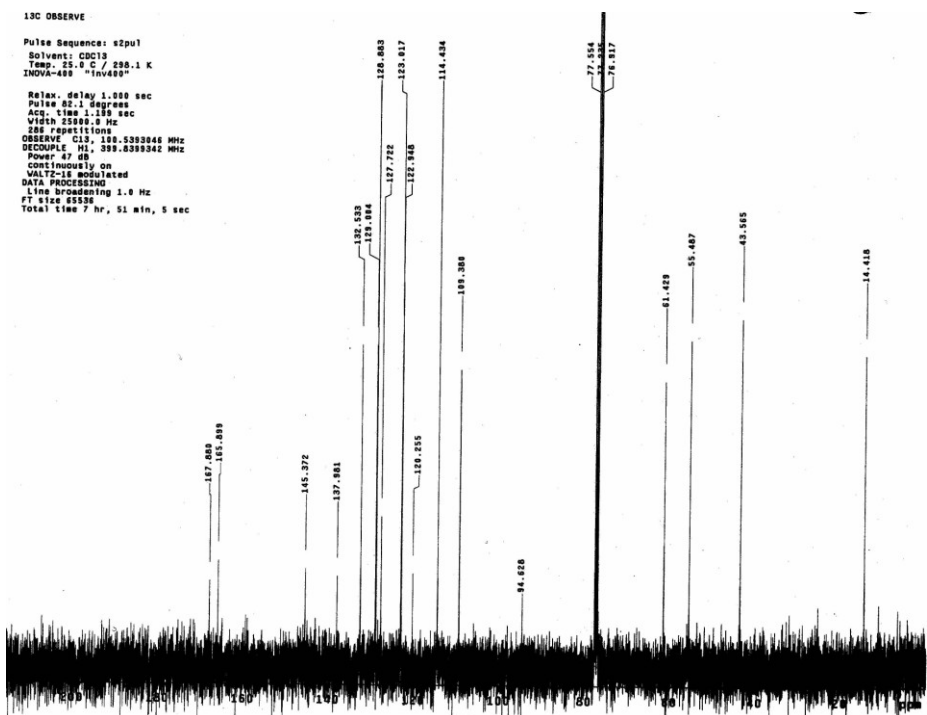
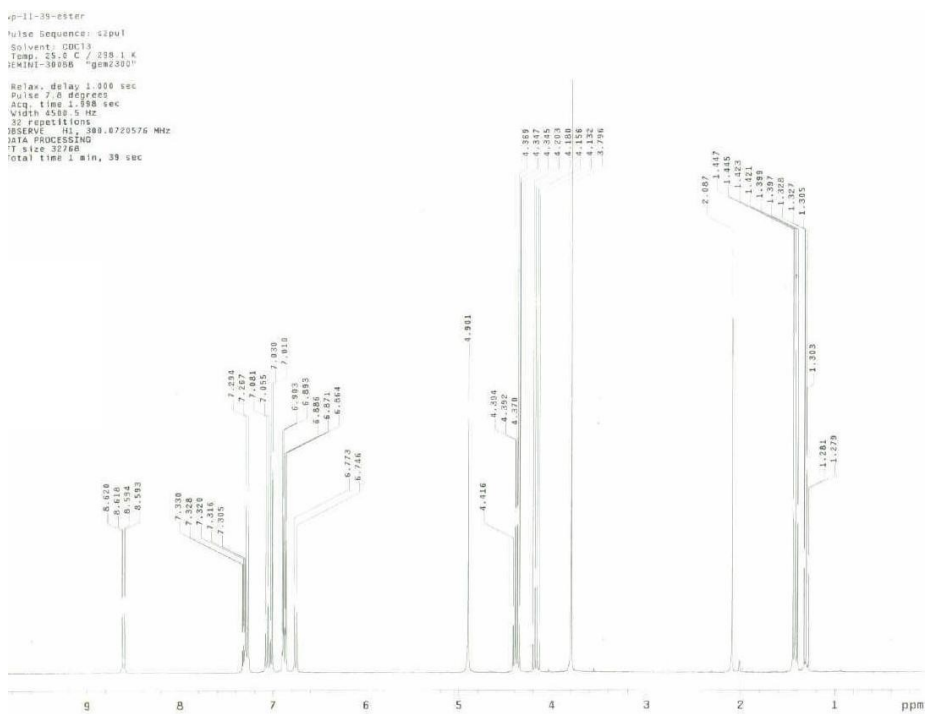


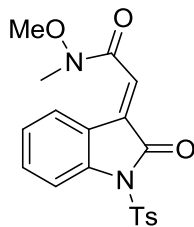
121





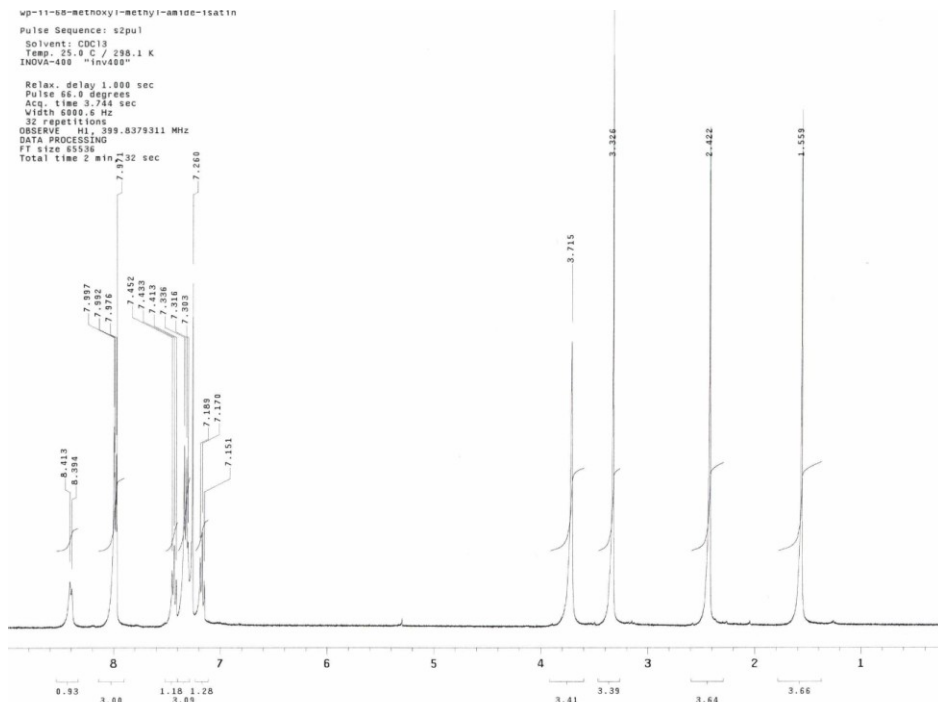
122



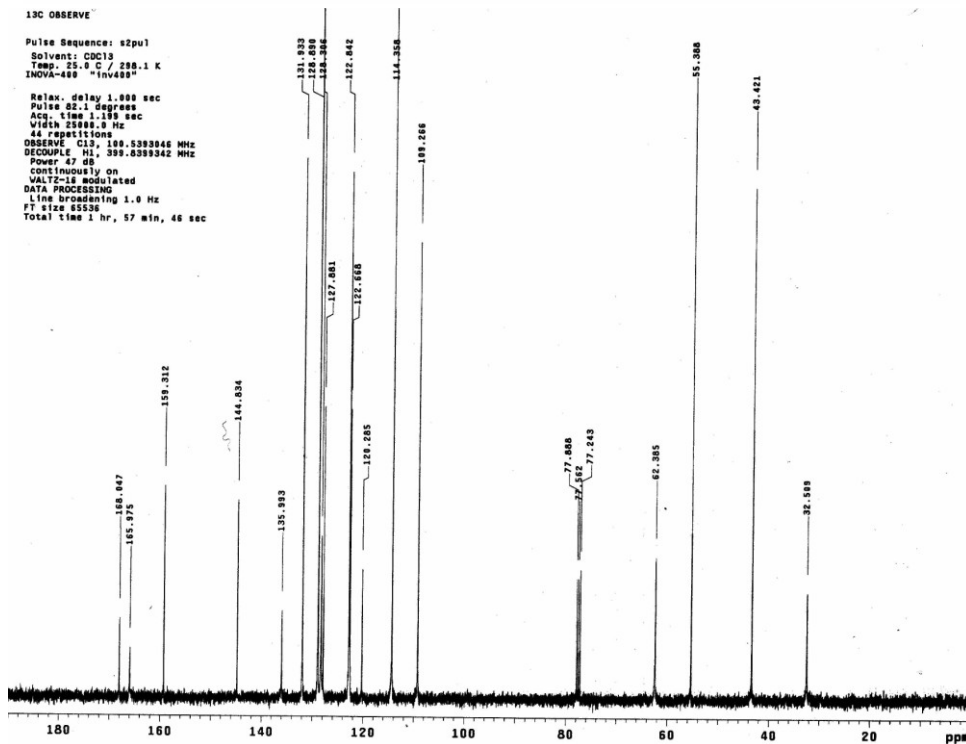


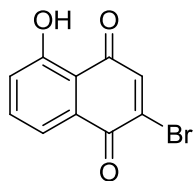
124

wp-11-88-methoxy-methyl-amide-isatin
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 66.0 degrees
 Acq. time 3.744 sec
 Width 6500.0 Hz
 32 repetitions
 OBSERVE H1, 399.8379311 MHz
 DATA PROCESSING
 FT size 65536
 Total time 2 min 32 sec

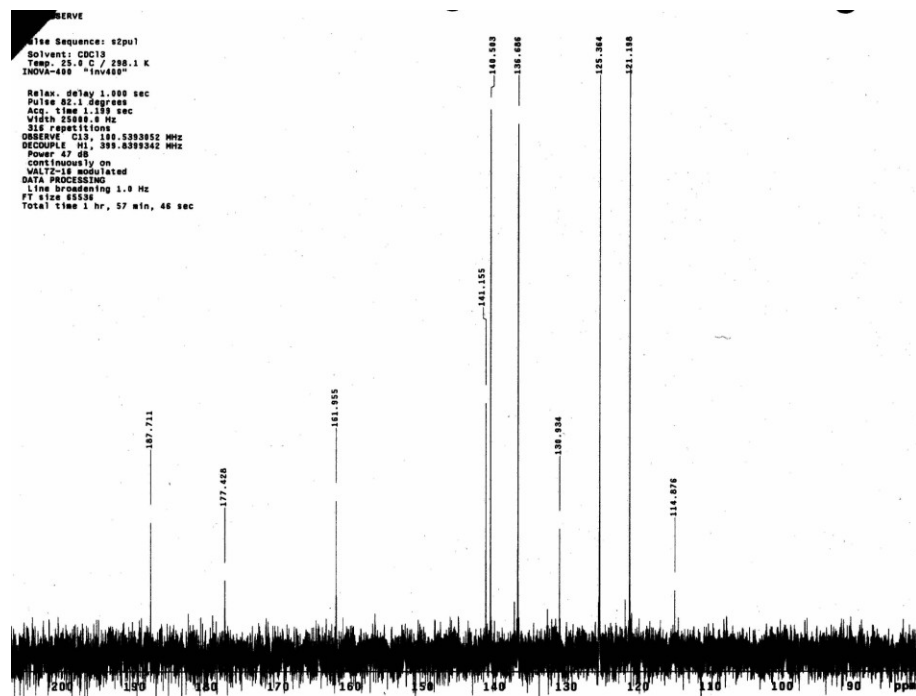
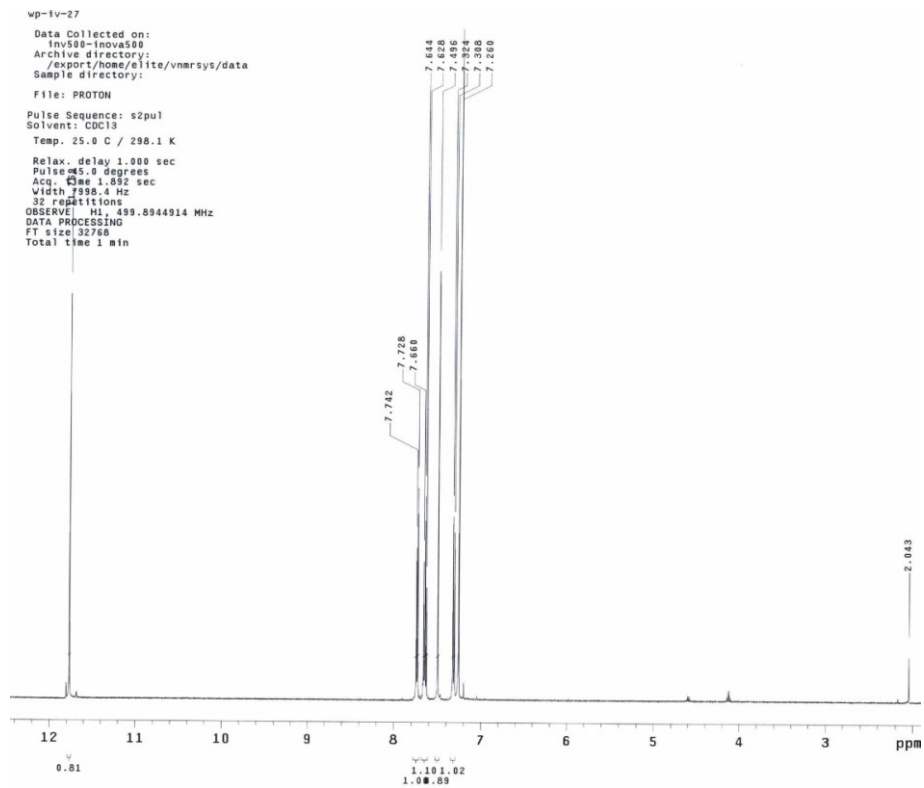


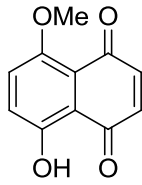
13C OBSERVE
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.138 sec
 Width 25000.0 Hz
 44 repetitions
 OBSERVE C13, 100.5383046 MHz
 DECOUPLE H1, 399.8389342 MHz
 Power 47 dB
 CONTINUOUSLY on
 VALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 1 hr, 57 min, 46 sec



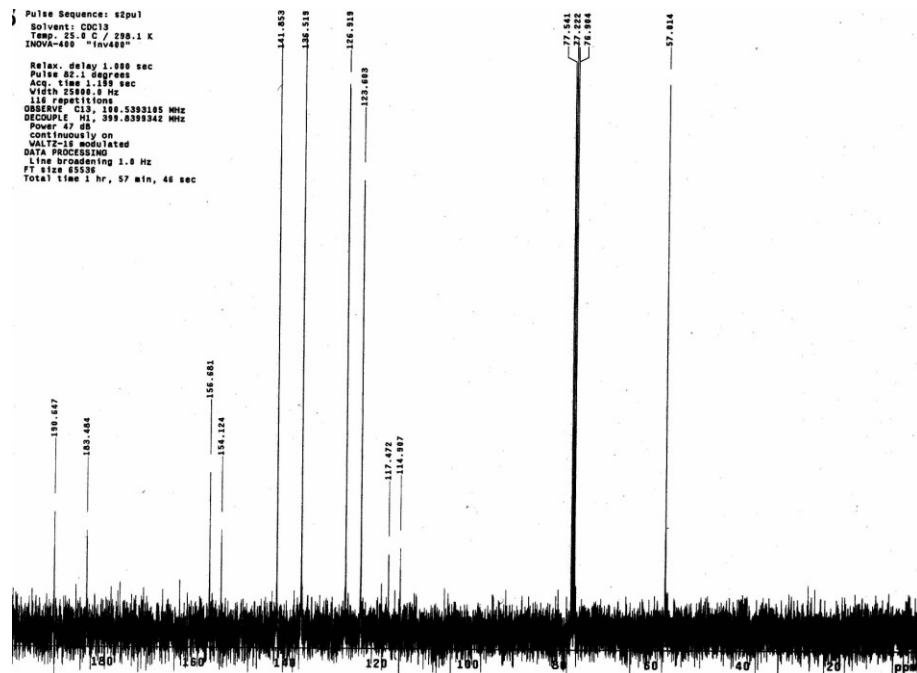
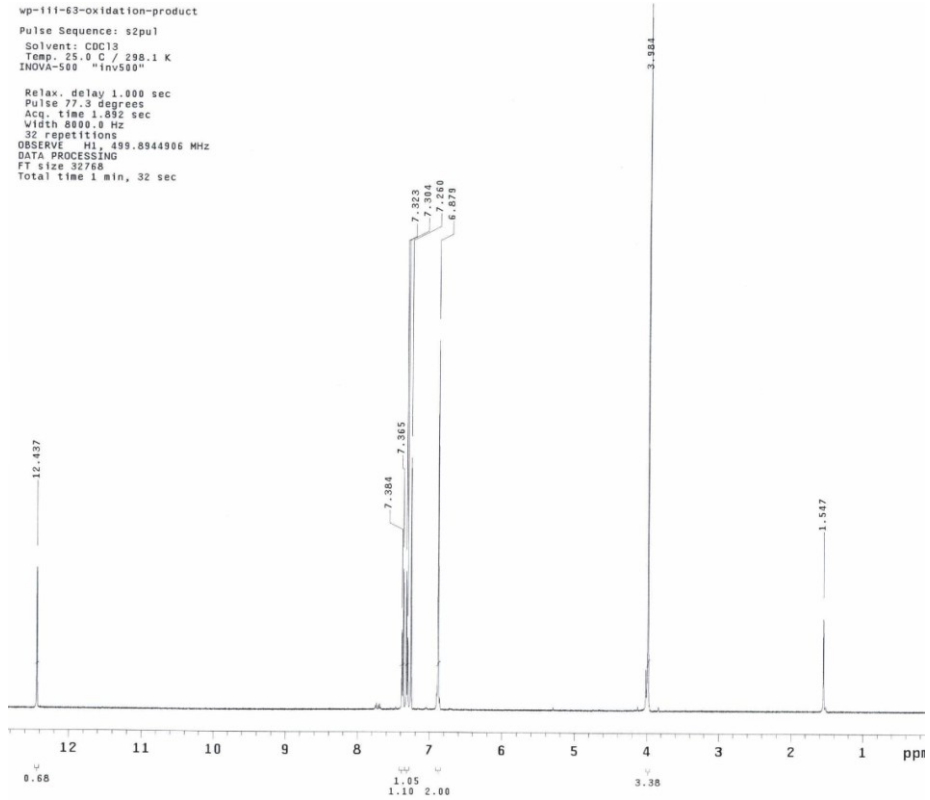


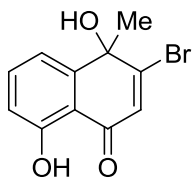
134





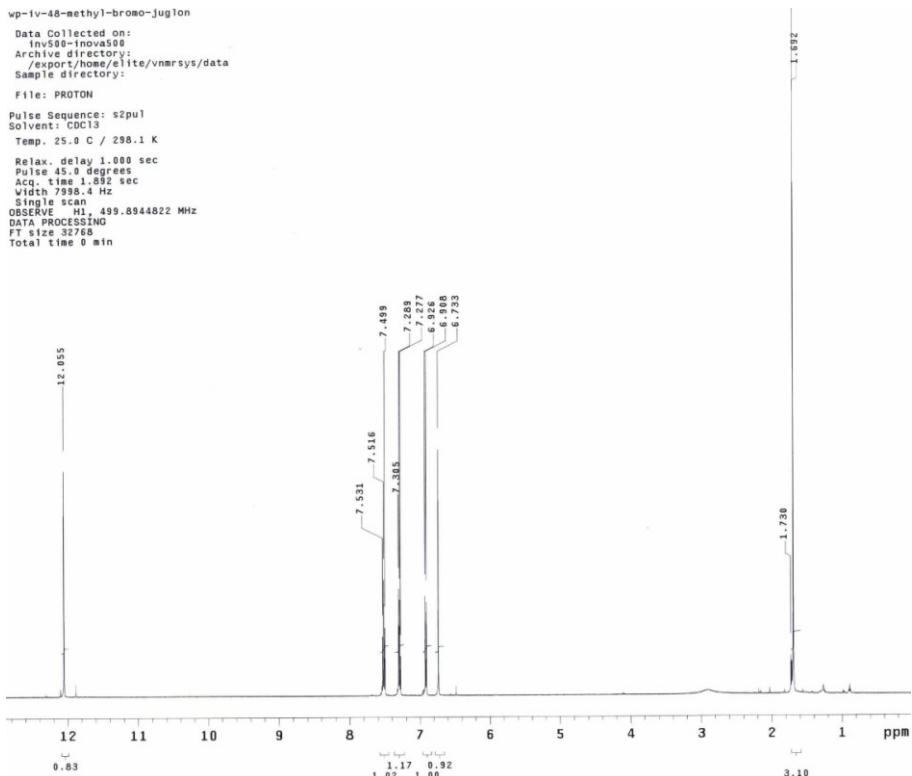
135



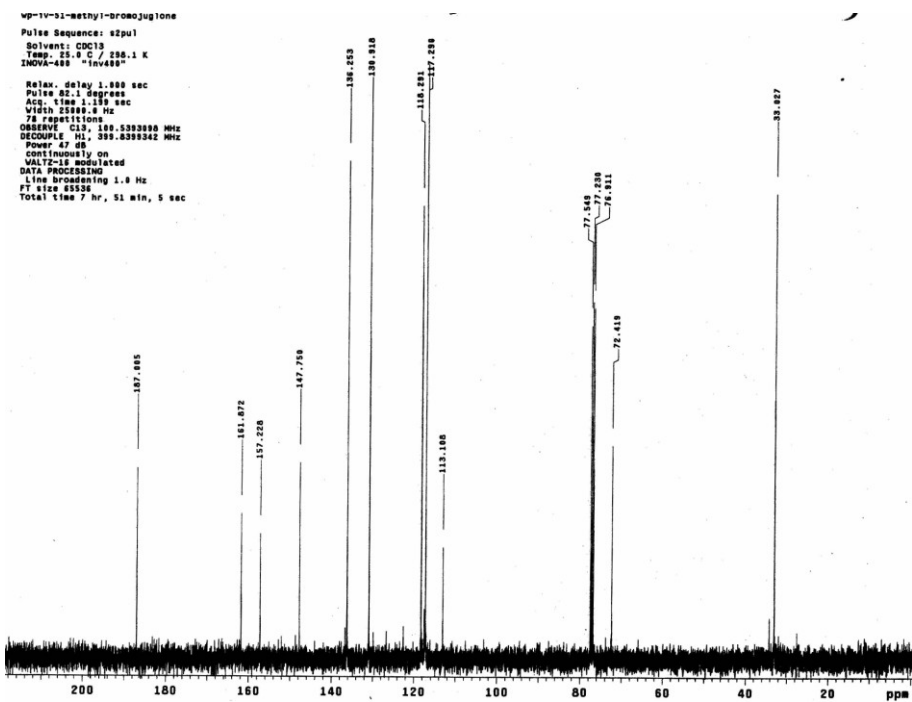


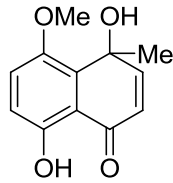
137

wp-1v-48-methyl-bromo-juglon
 Data Collected on:
 Inv588-inova500
 Archive directory:
 /export/home/elite/vnmrsys/data
 Sample directory:
 File: PROTON
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.852 sec
 Width 7998.4 Hz
 Single scan
 OBSERVE H1, 499.8944822 MHz
 DATA PROCESSING
 FT size 32768
 Total time 0 min



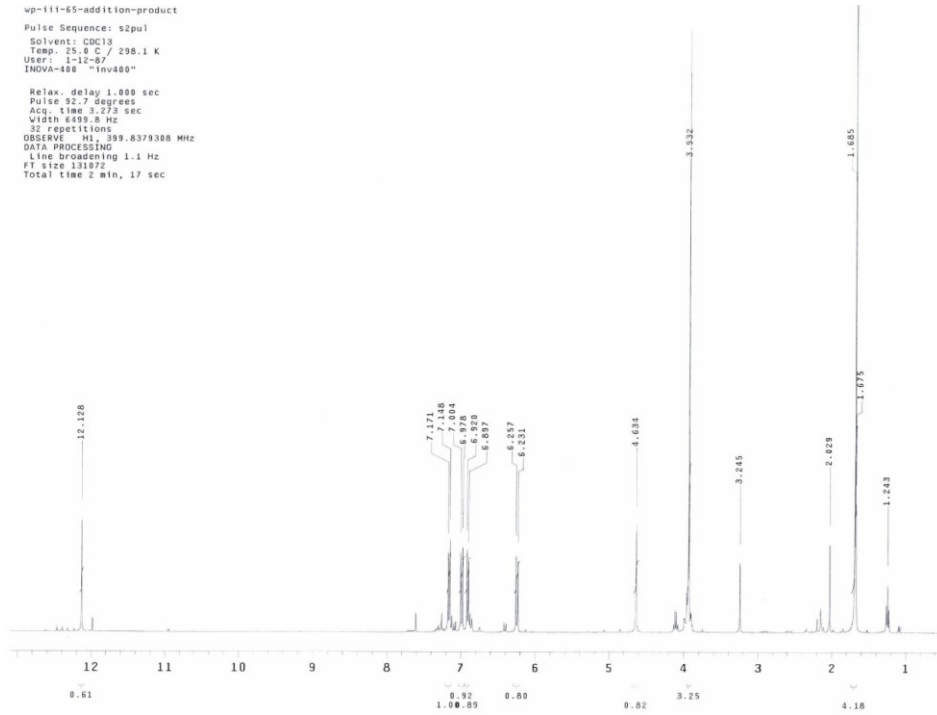
wp-1v-51-methyl-bromo-juglone
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 INOVA-400 "Inv400"
 Relax. delay 1.800 sec
 Pulse 82.1 degrees
 Acq. time 1.159 sec
 Width 2280.0 Hz
 78 repetitions
 OBSERVE C13, 100.628888 MHz
 DECOUPLE H1, 399.809342 MHz
 Power 47 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 7 hr, 51 min, 5 sec



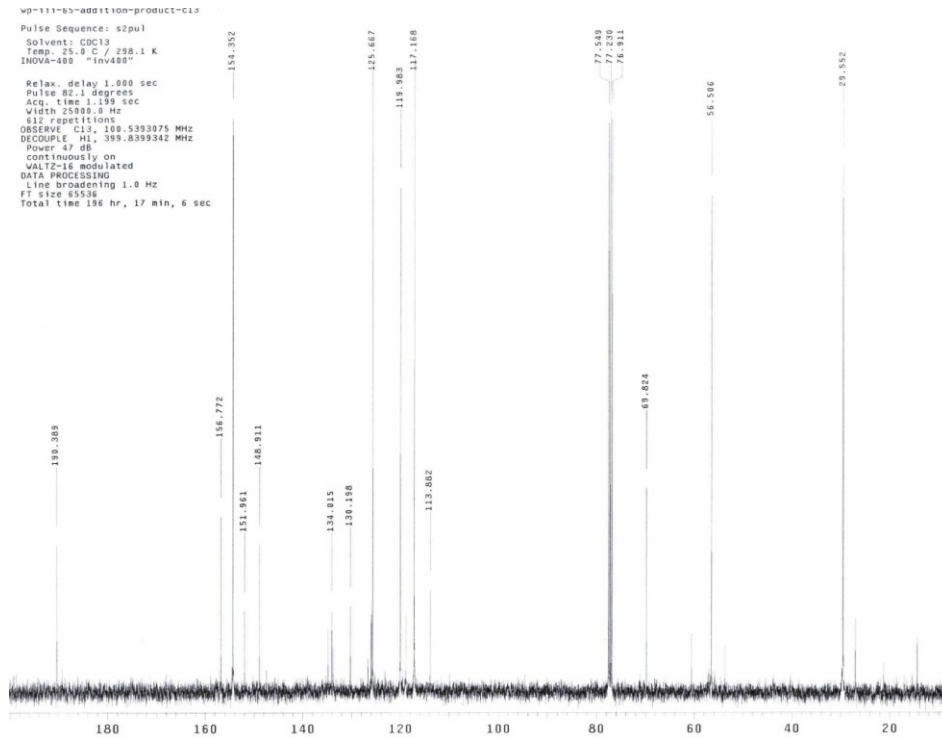


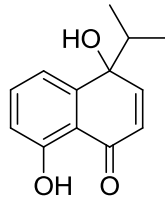
138

wp-111-65-addition-product
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-97
 INOVA-400 "Inv400"
 Relax. delay 1.000 sec
 Pulse 92.7 degrees
 Acq. time 3.273 sec
 Width 6400.0 Hz
 32 repetitions
 OBSERVE H1: 399.8379308 MHz
 DATA PROCESSING
 Line broadening 1.1 Hz
 FT size 131072
 Total time 2 min, 17 sec

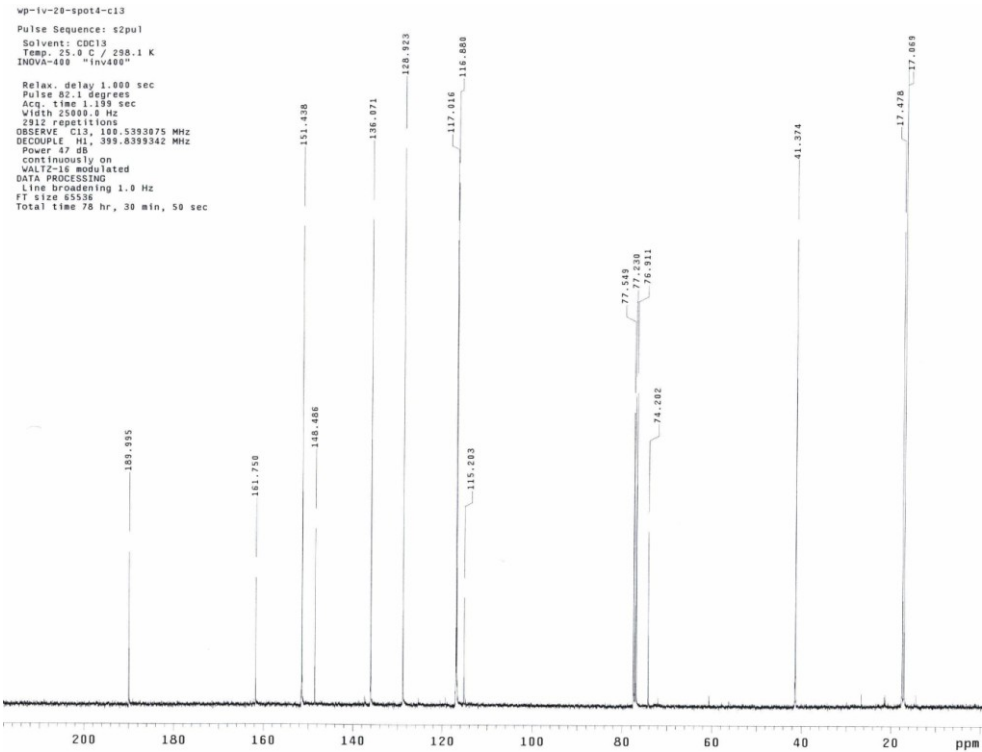
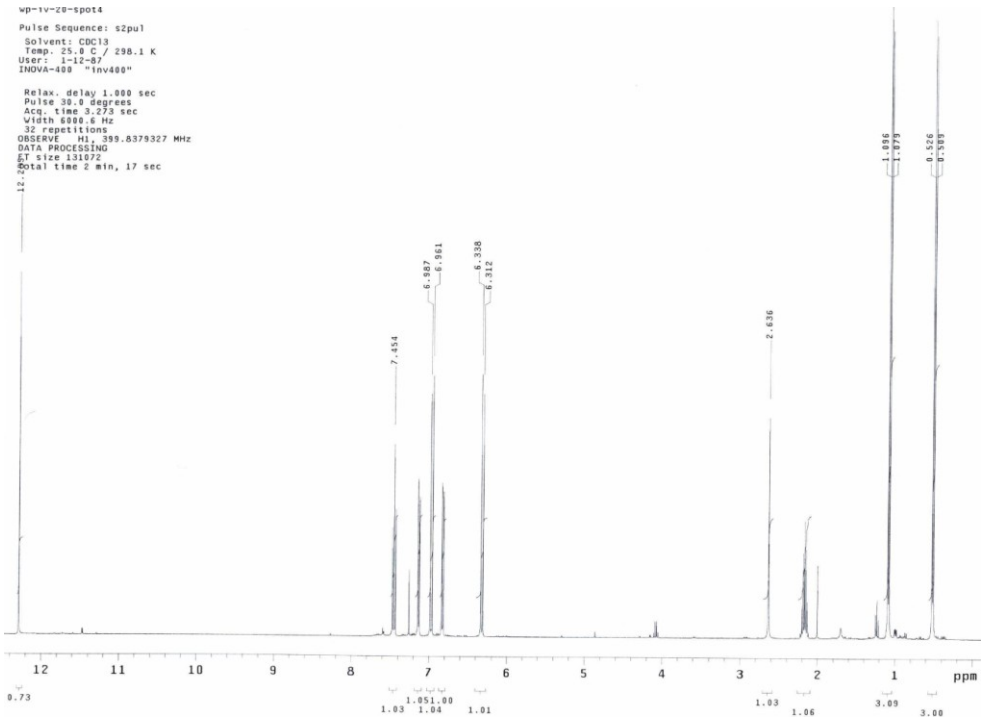


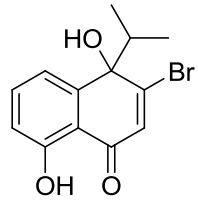
wp-111-65-addition-product-cl3
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-97
 INOVA-400 "Inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.199 sec
 Width 25000.0 Hz
 612 repetitions
 OBSERVE C13: 100.5393075 MHz
 DECOUPLE H1: 399.8393342 MHz
 Power 47 dB
 Continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 45336
 Total time 196 hr, 17 min, 6 sec



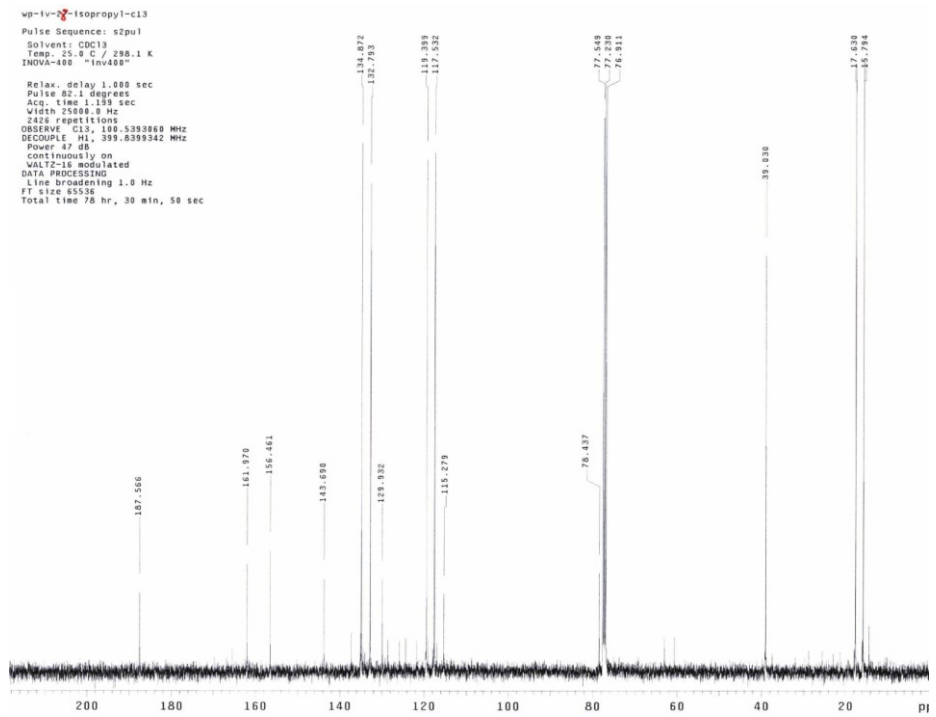
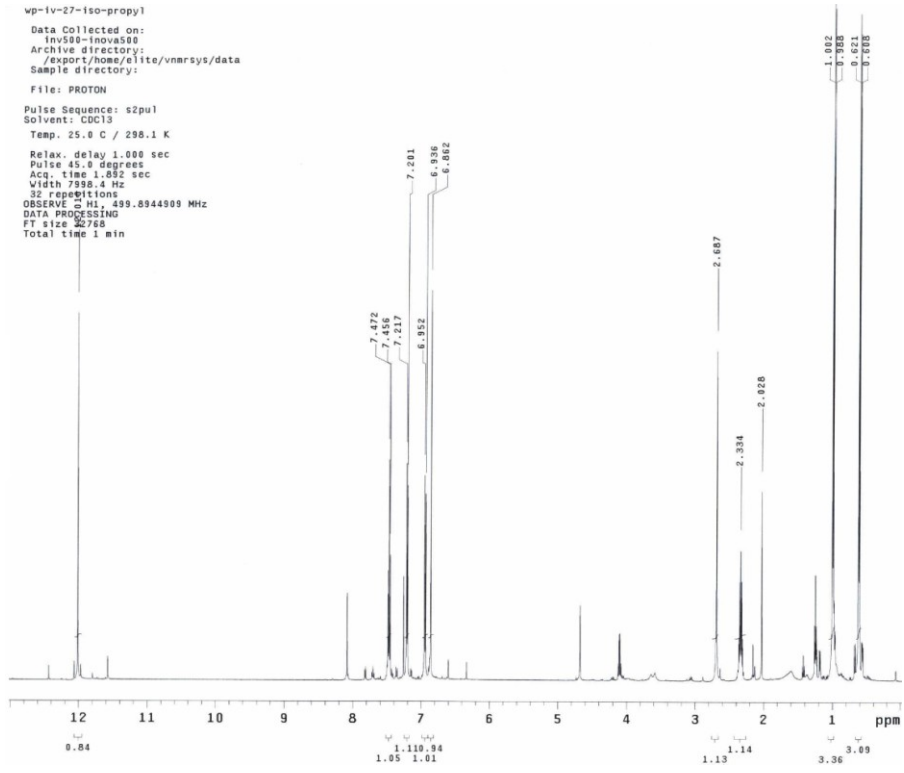


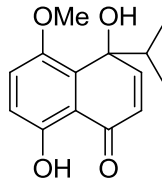
139



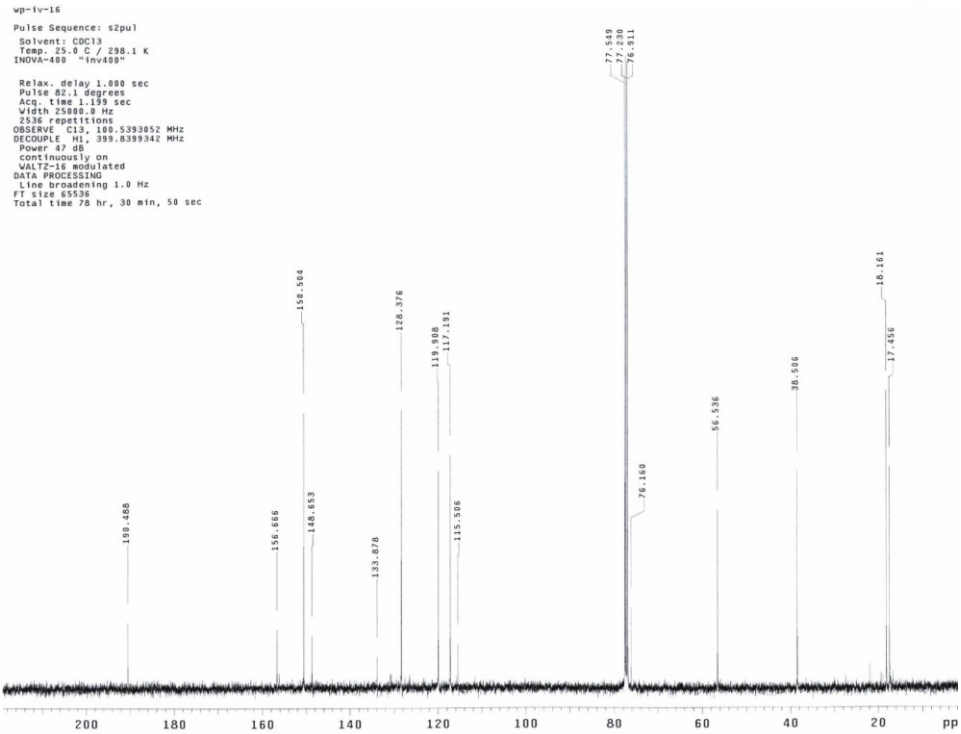
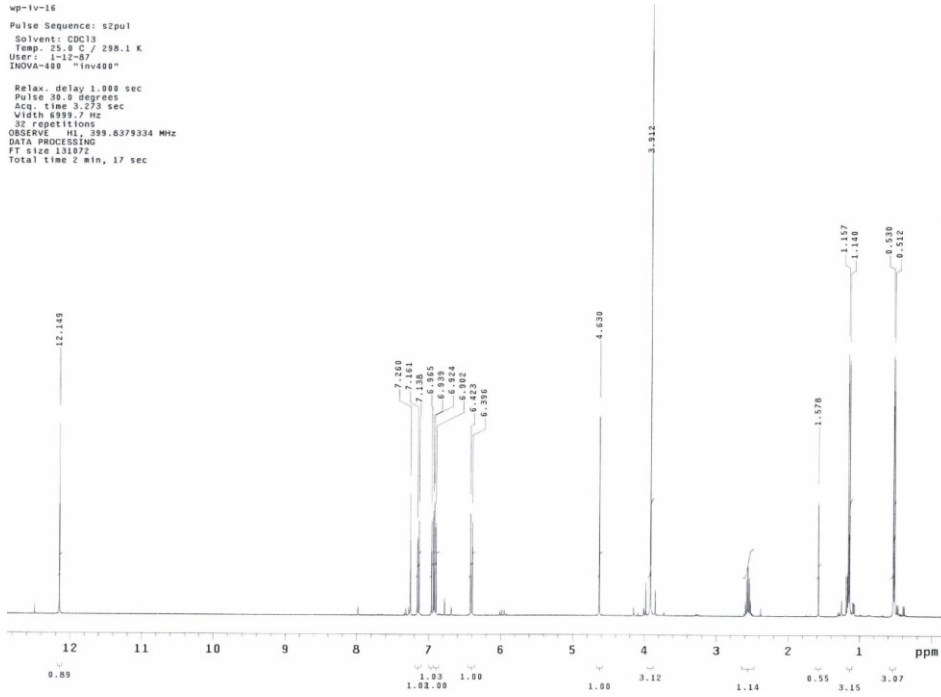


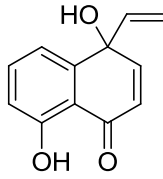
140



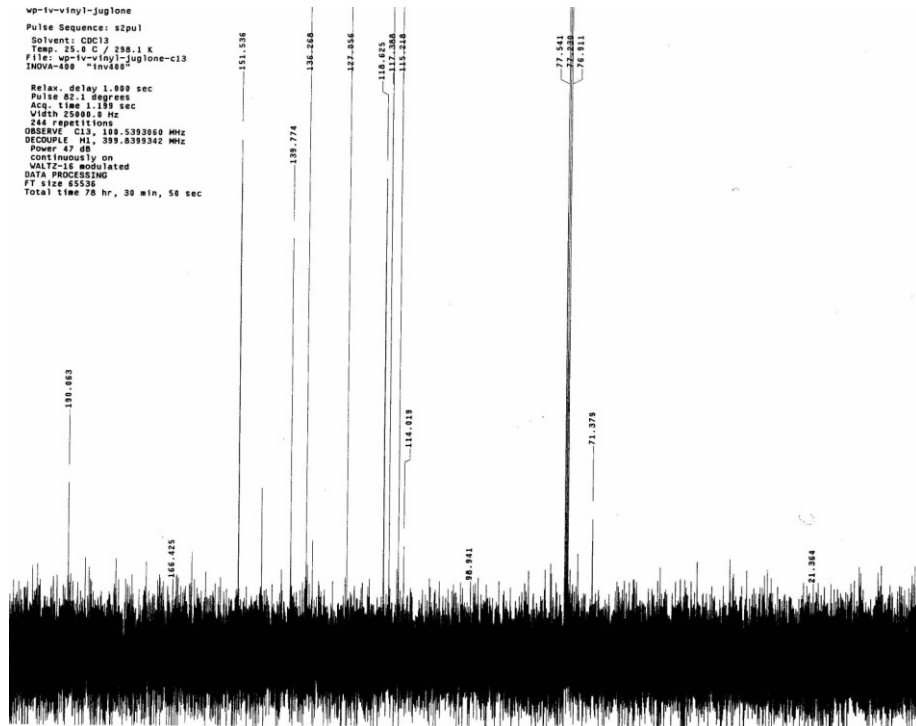
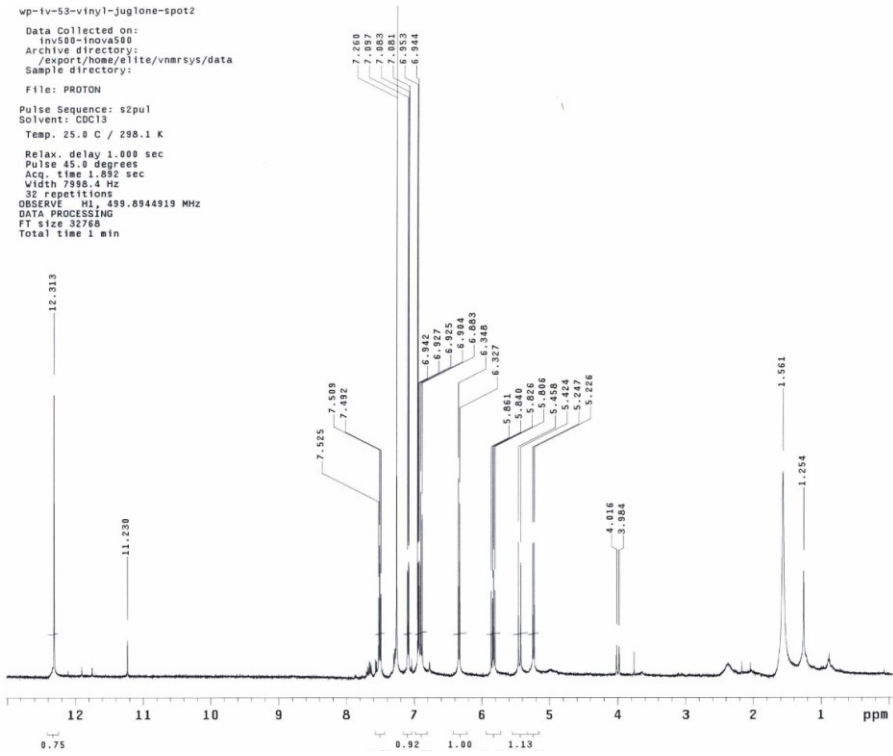


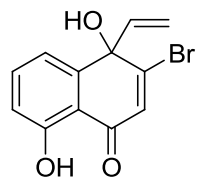
141



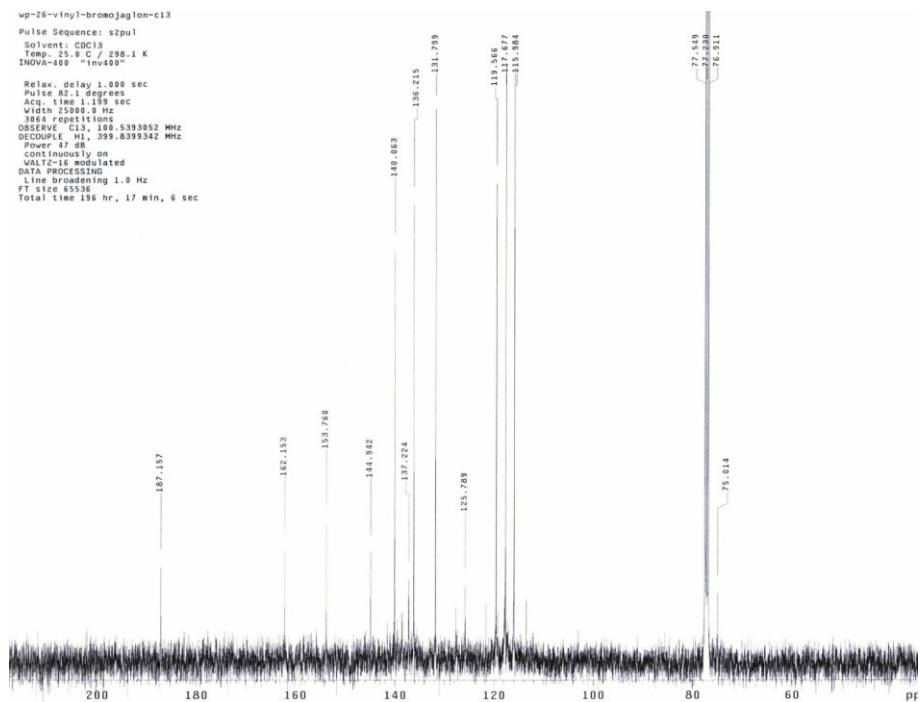
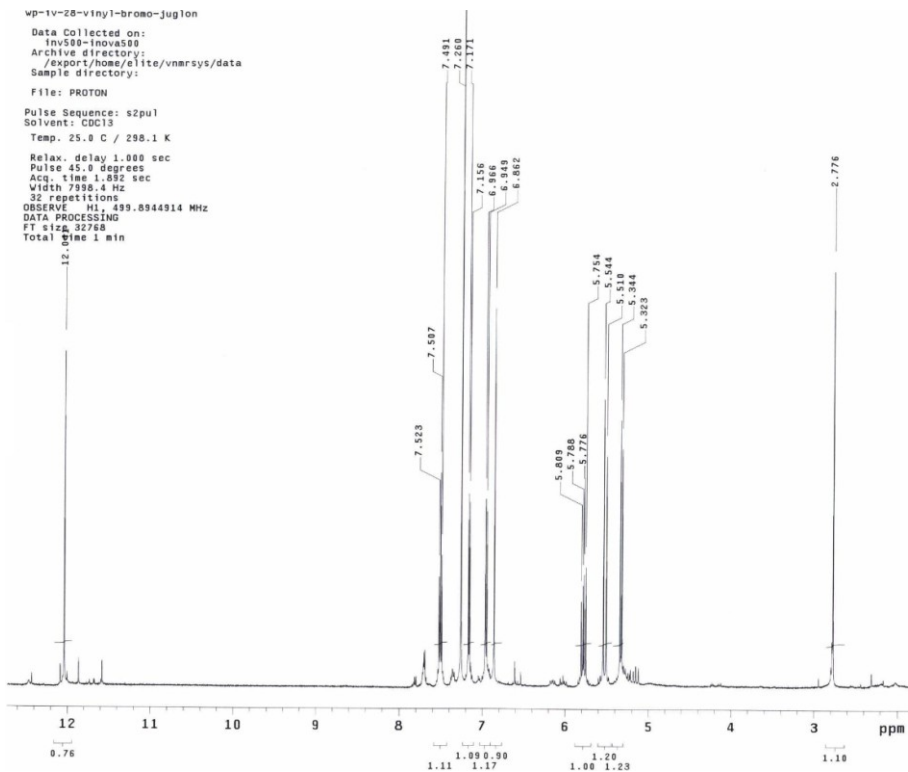


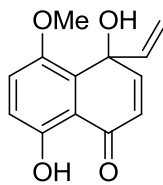
142



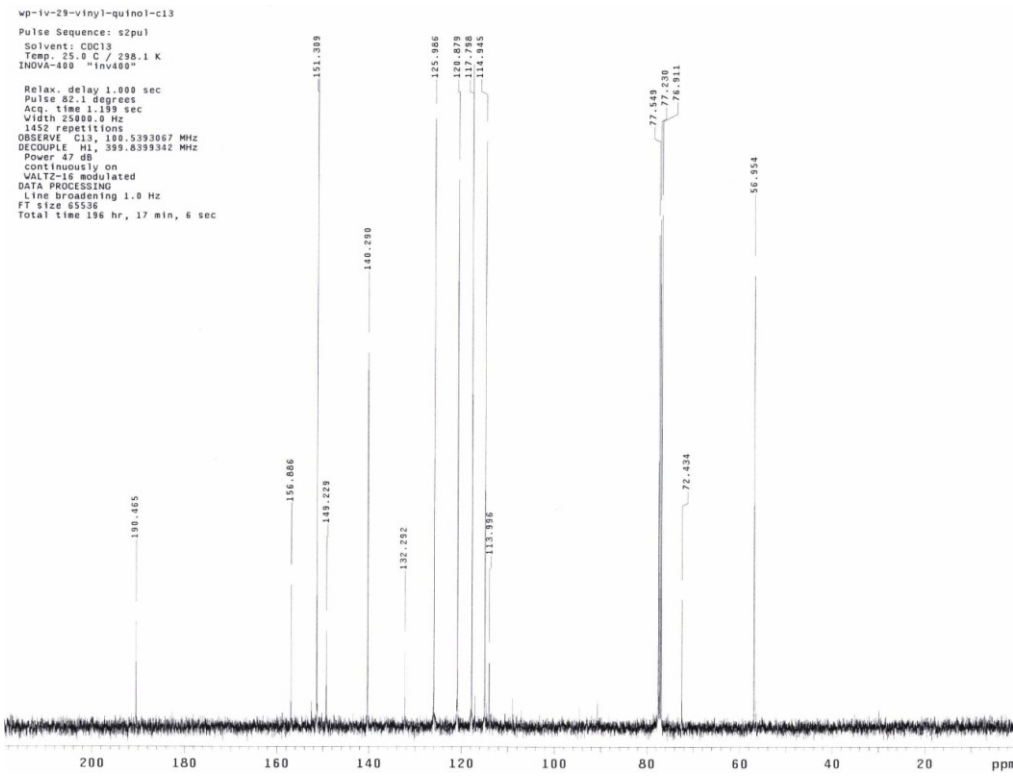
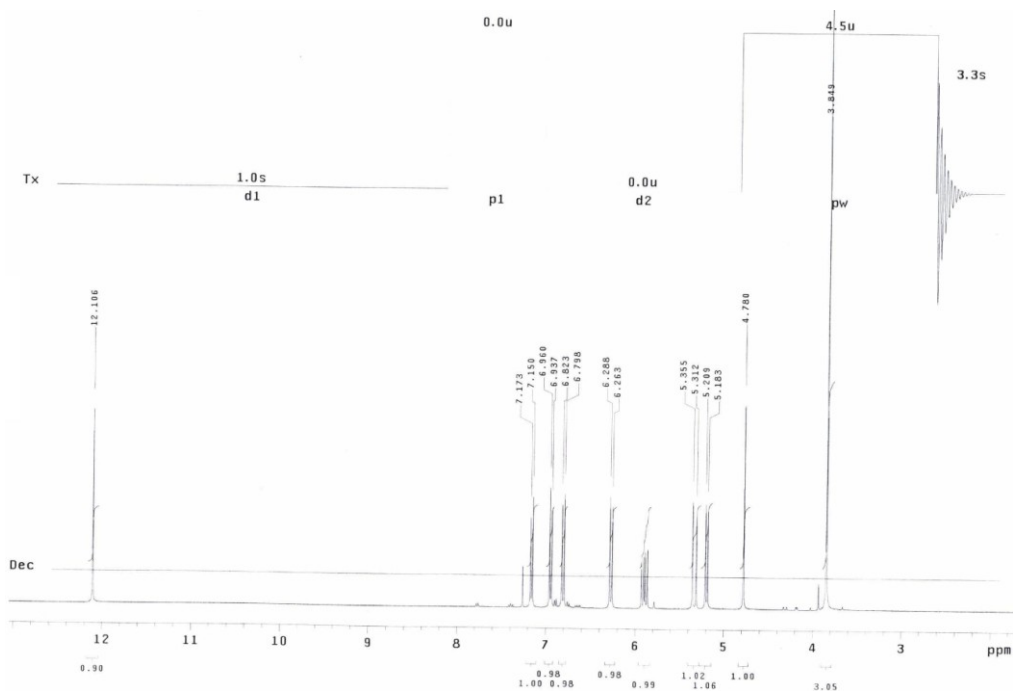


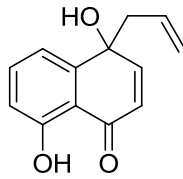
143



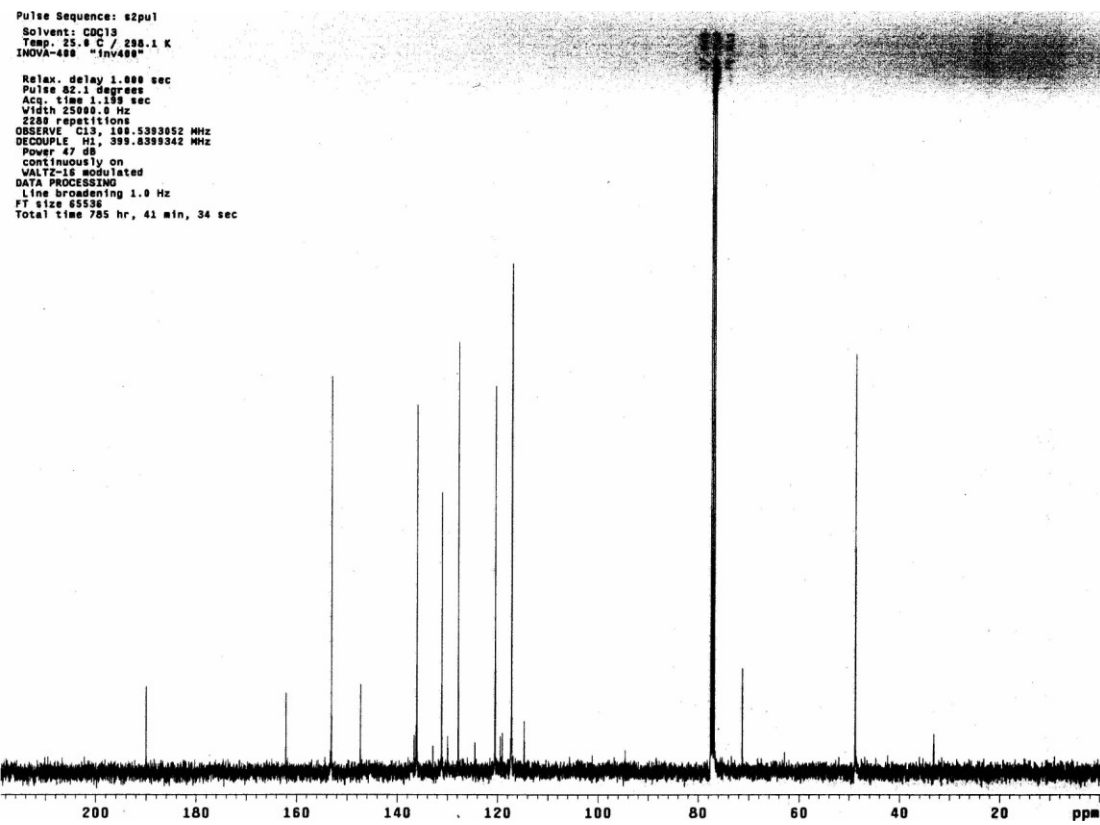
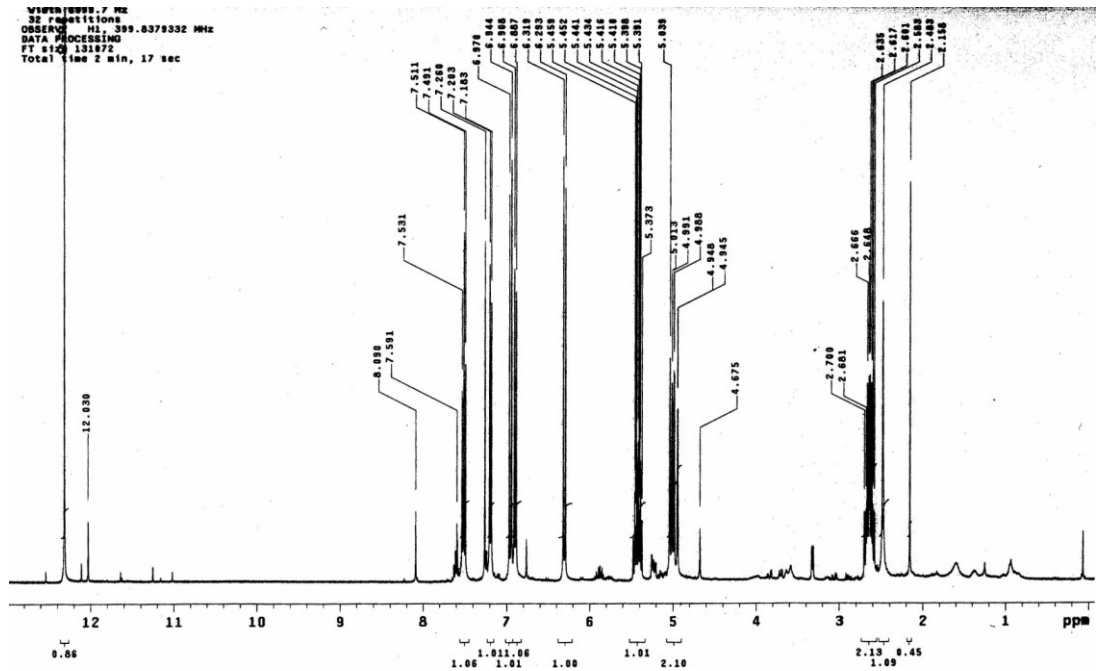


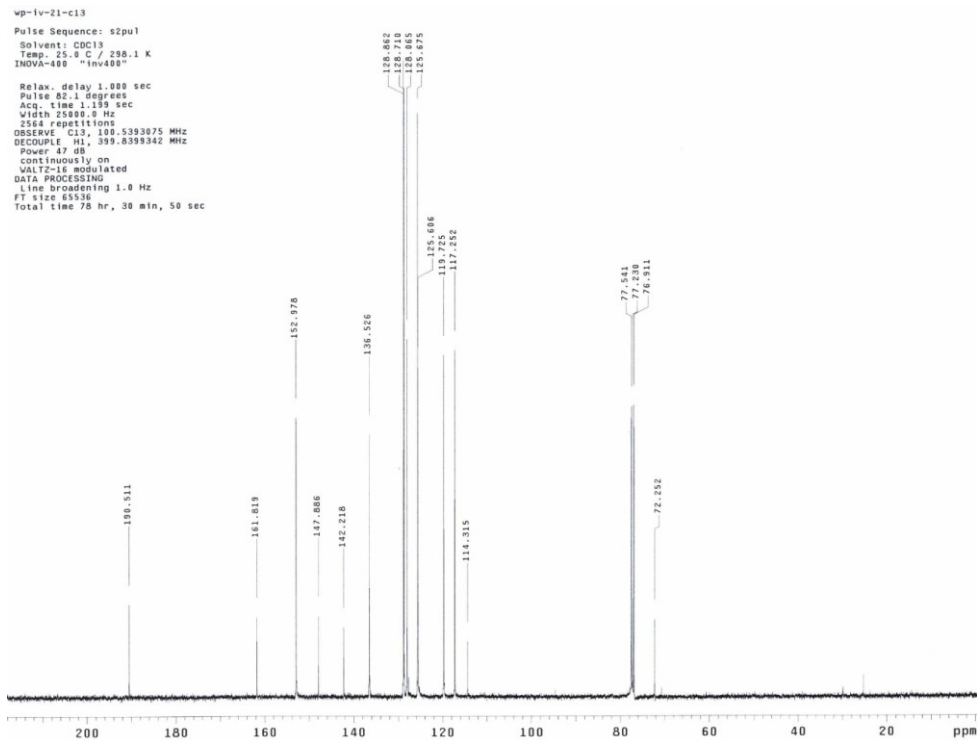
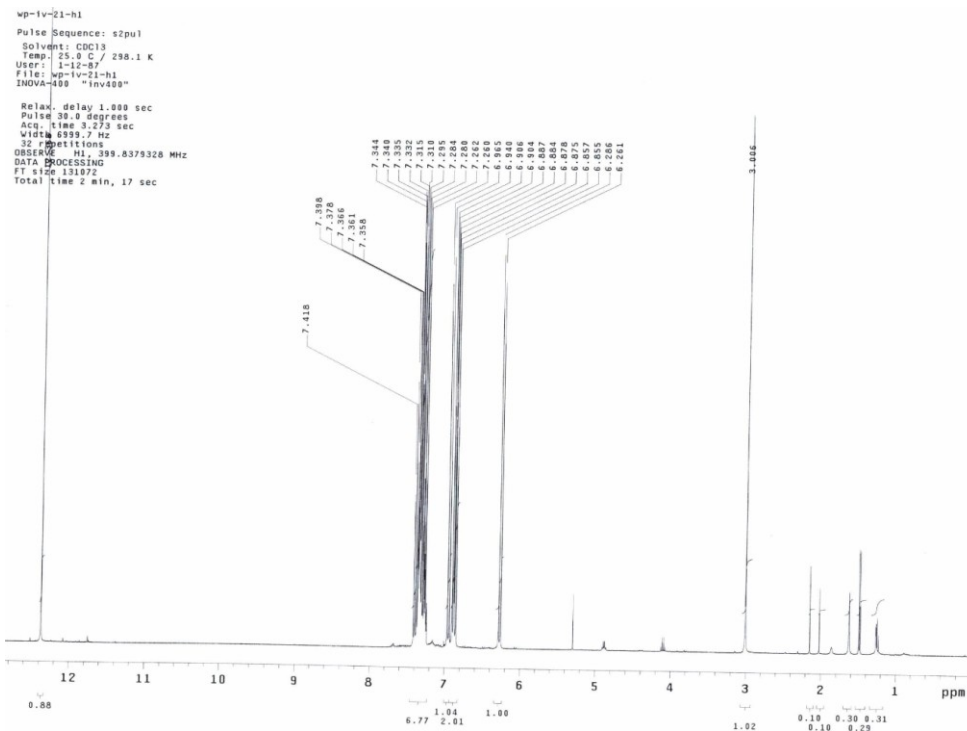
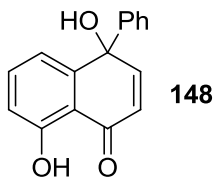
144

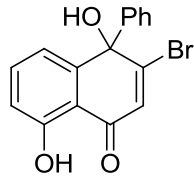




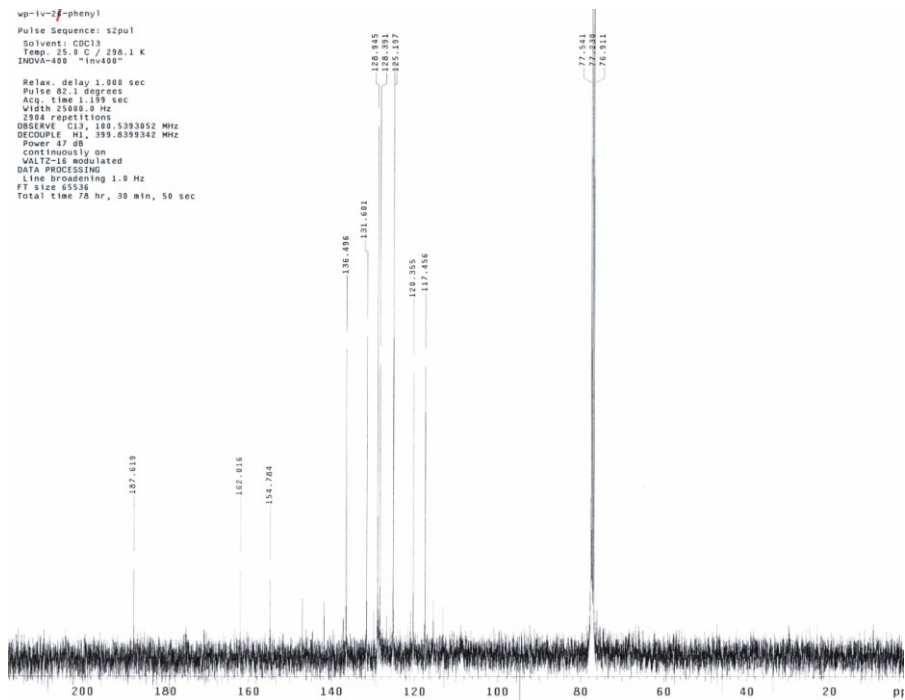
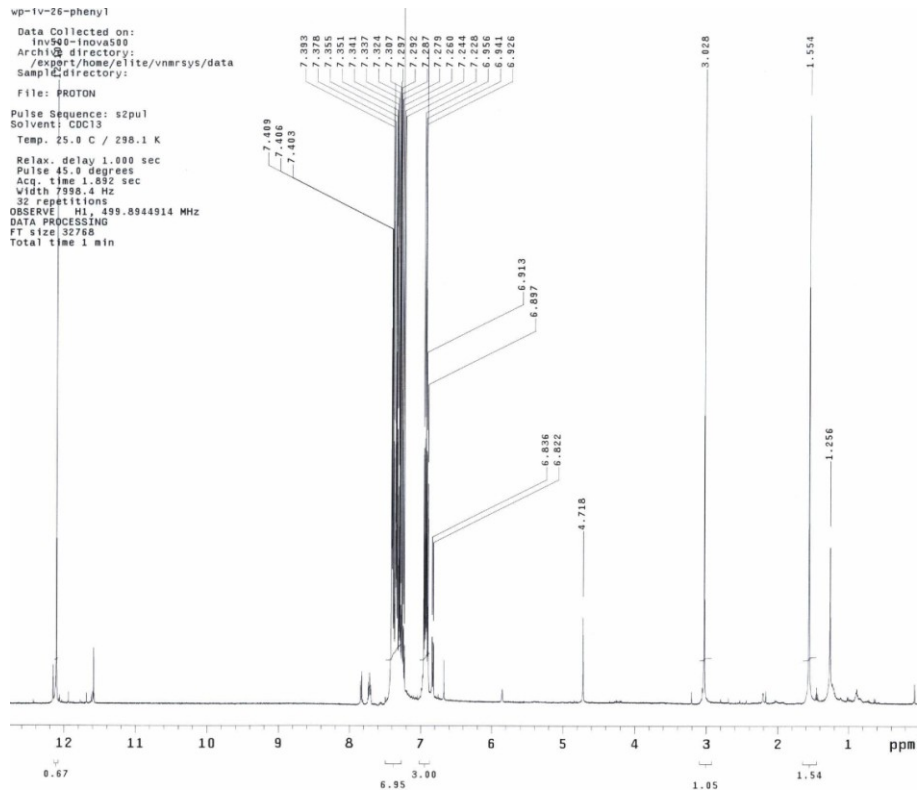
145

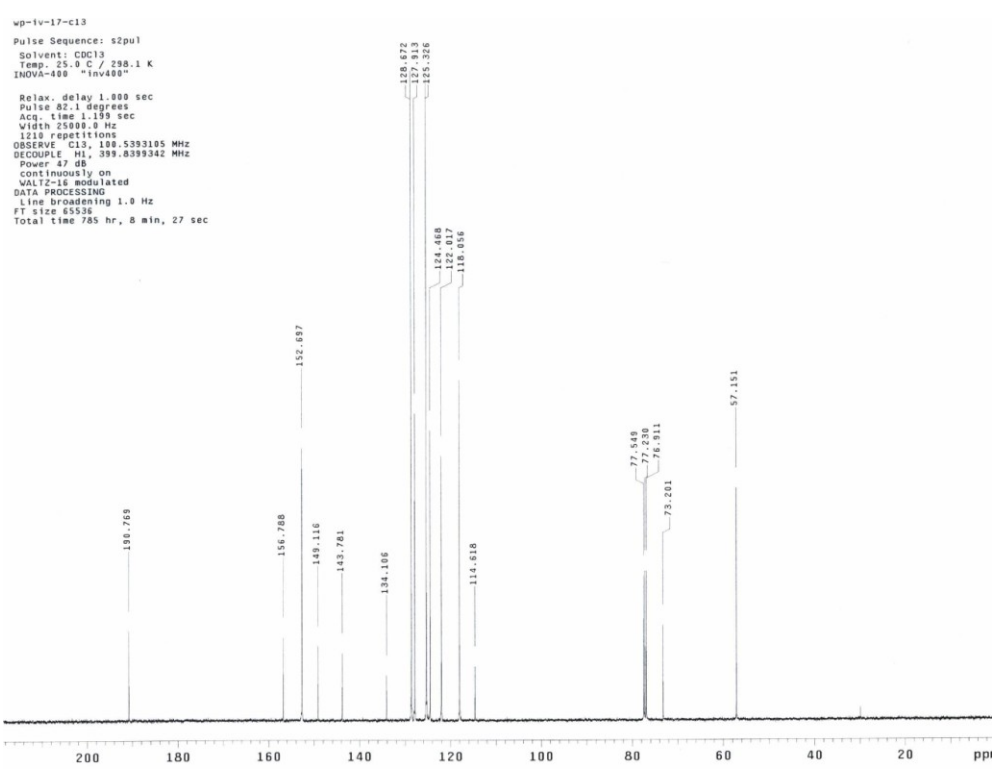
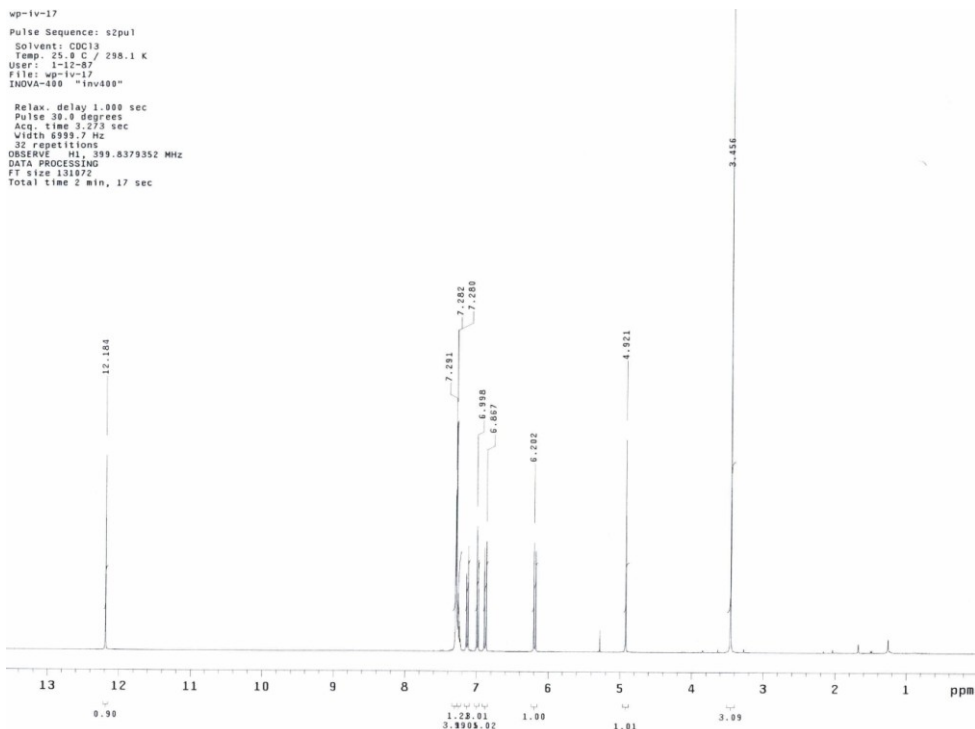
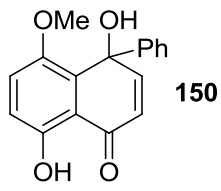


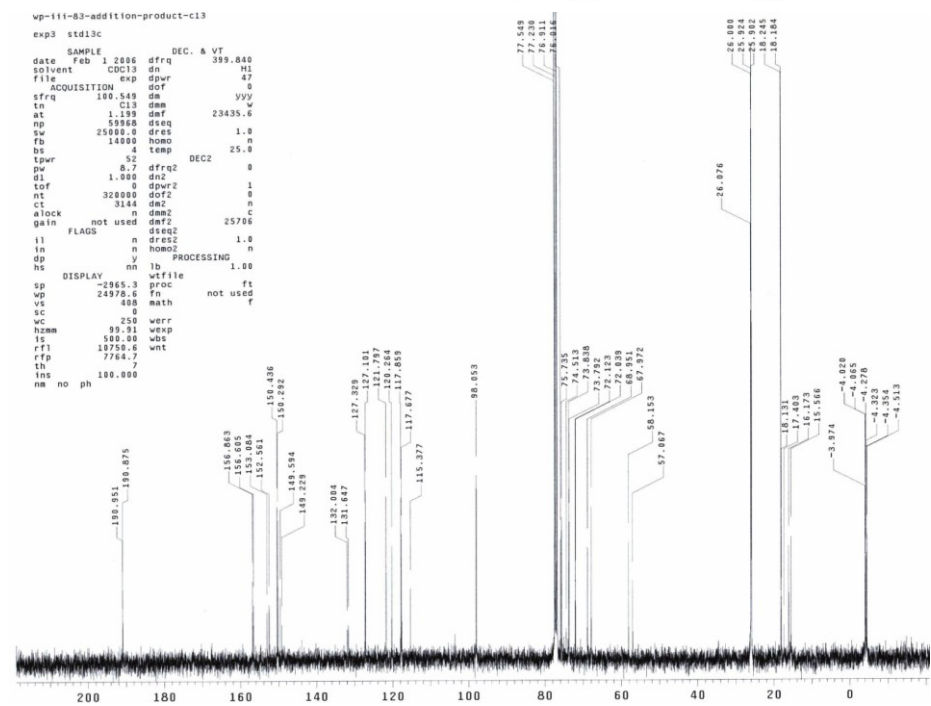
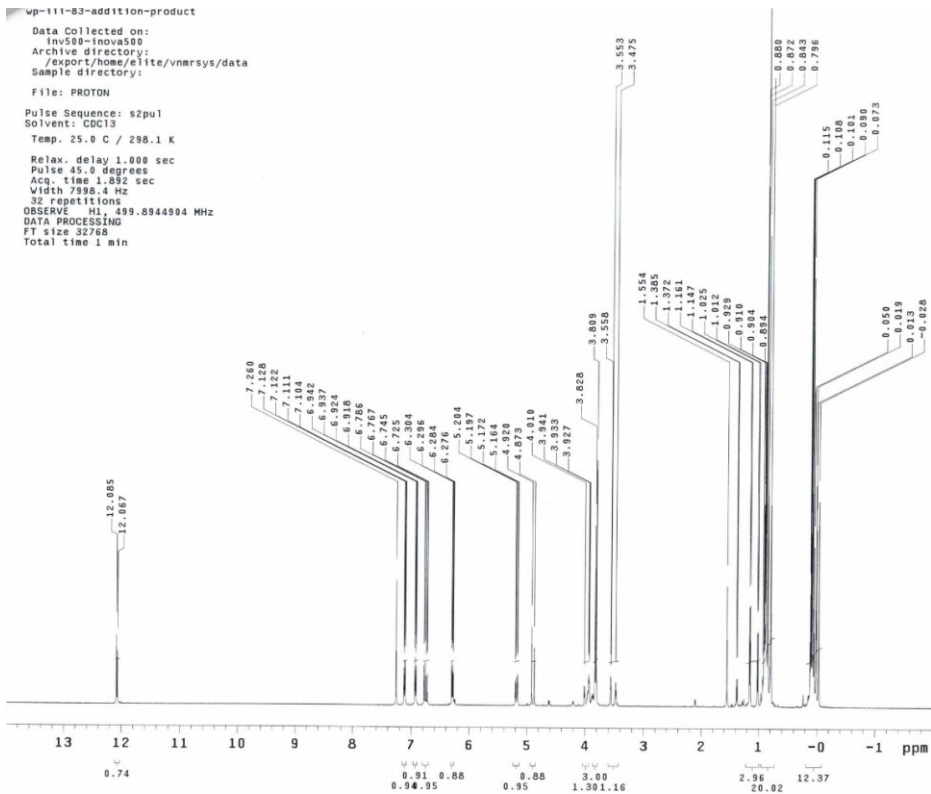
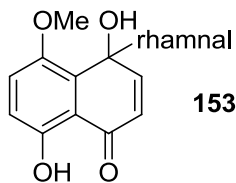


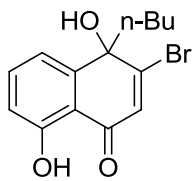


149

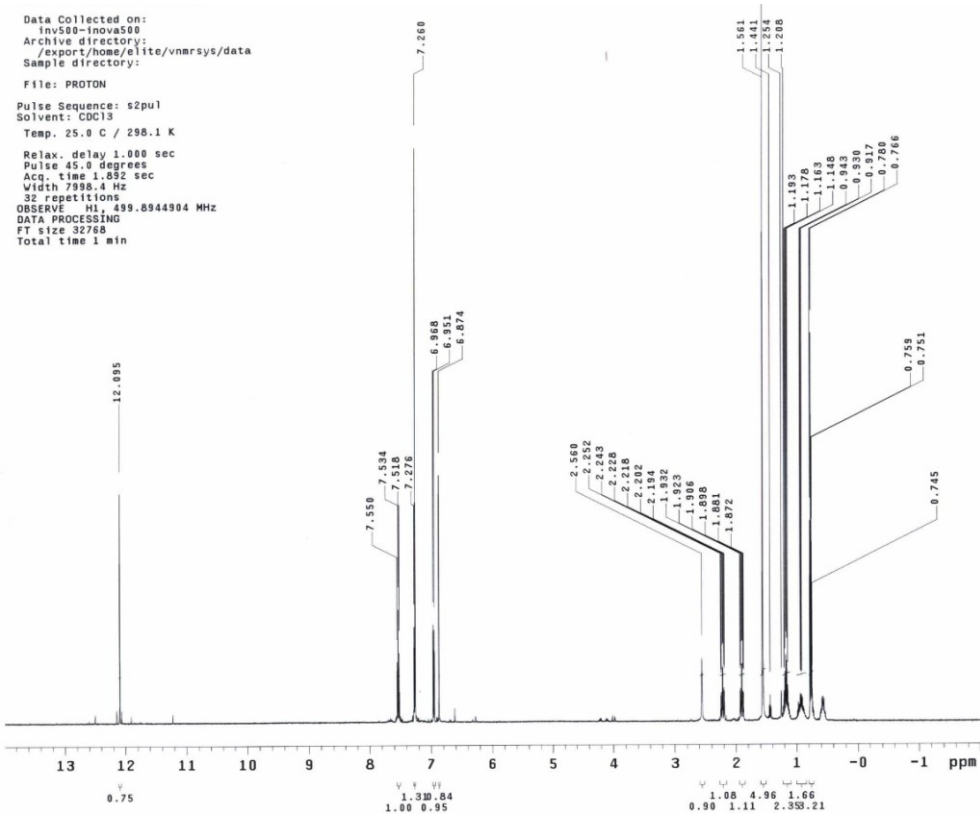


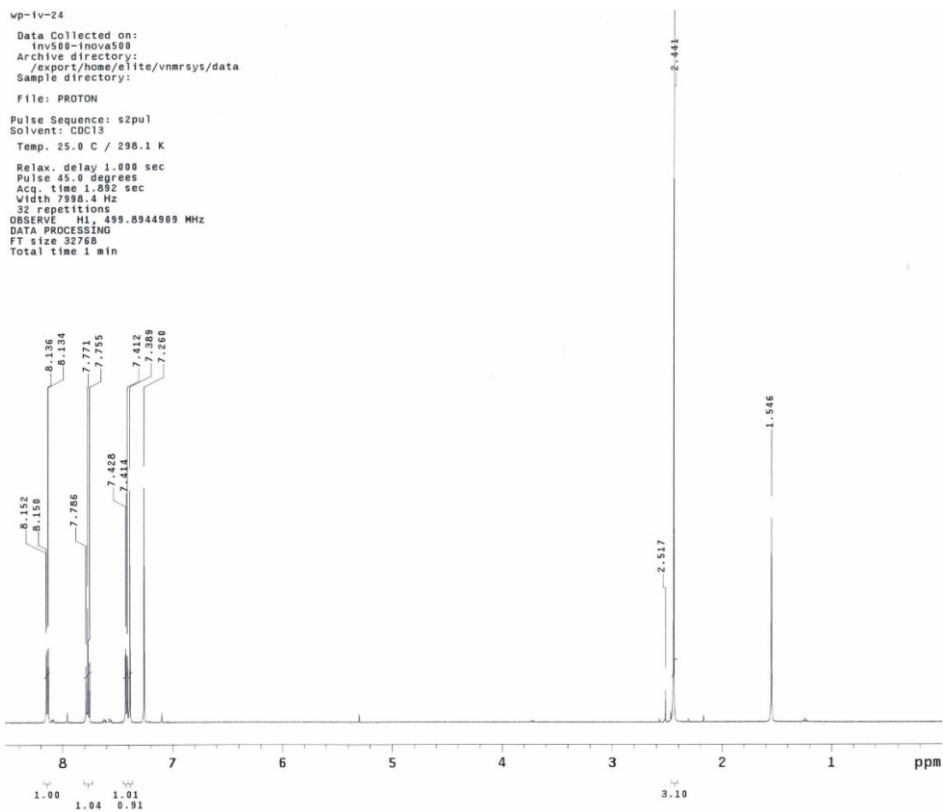
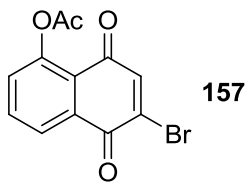


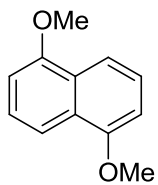




154



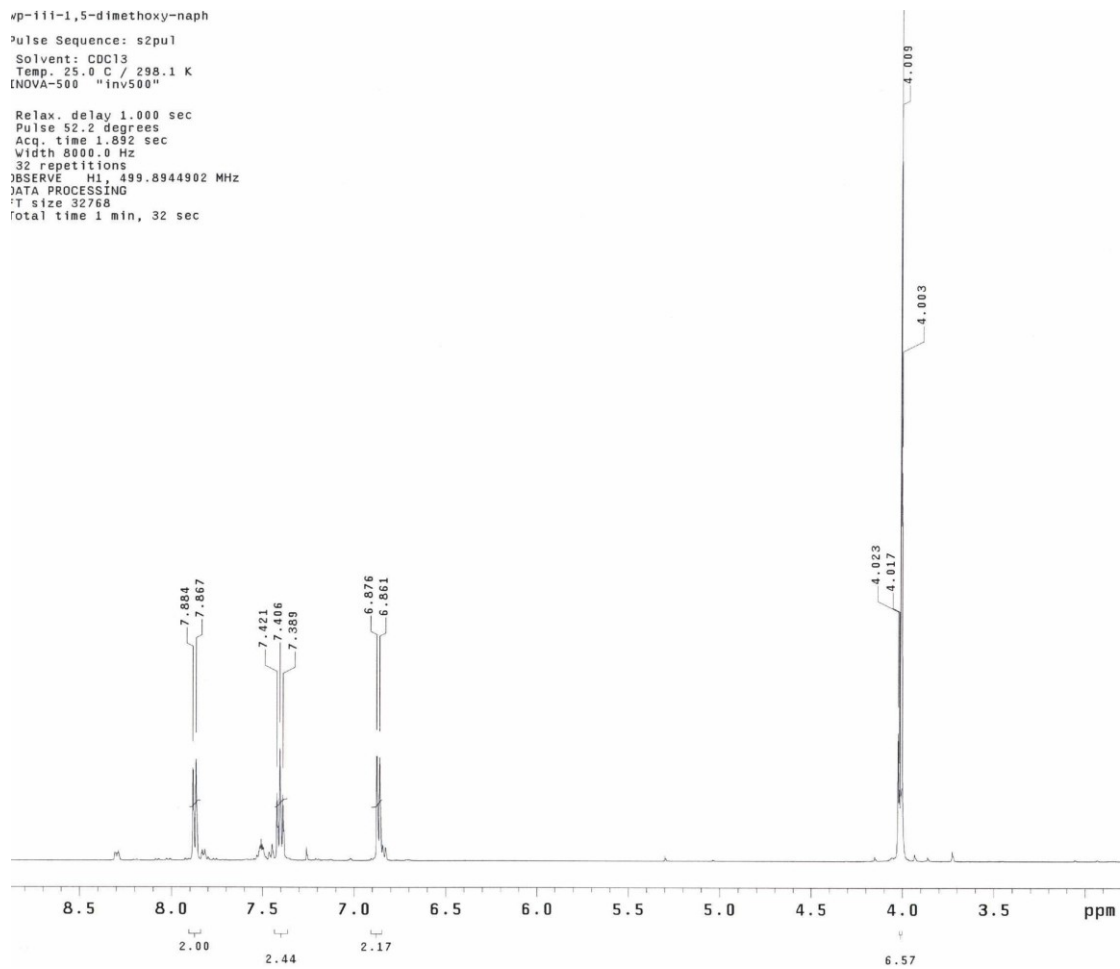


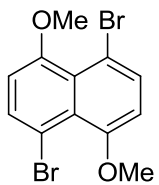


158

mp-111-1,5-dimethoxy-naph
Pulse Sequence: s2pu1
Solvent: CDCl3
Temp. 25.0 C / 298.1 K
INOVA-500 "inv500"

Relax. delay 1.000 sec
Pulse 52.2 degrees
Acq. time 1.832 sec
Width 8000.0 Hz
32 repetitions
OBSERVE H1, 499.8944902 MHz
DATA PROCESSING
F1 size 32768
Total time 1 min, 32 sec

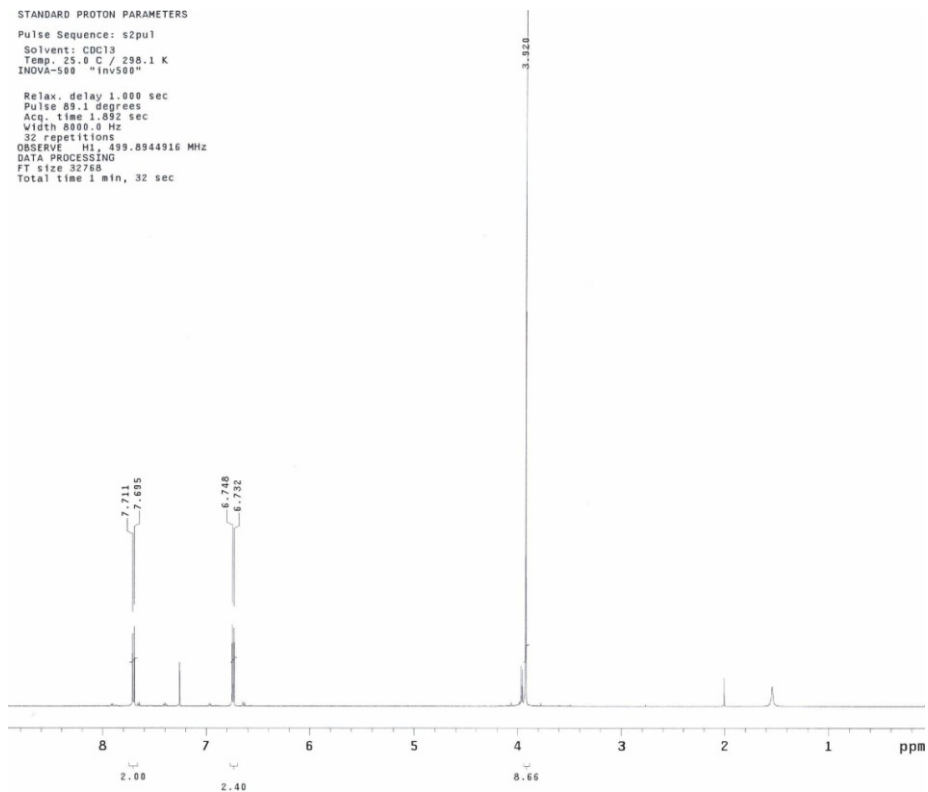




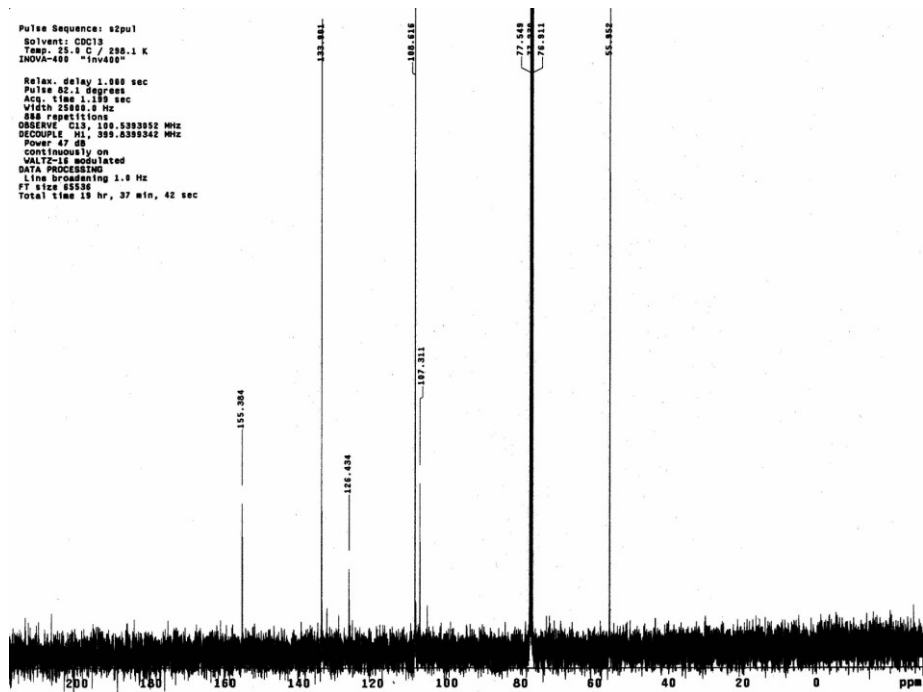
159

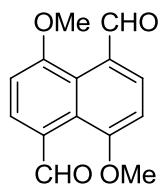
STANDARD PROTON PARAMETERS

Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 INOVA-500 "Inv500"
 Relax. delay 1.000 sec
 Pulse 89.1 degrees
 Acq. time 1.852 sec
 Width 8000.0 Hz
 32 repetitions
 OBSERVE H1, 499.8944916 MHz
 DATA PROCESSING
 FT size 32768
 Total time 1 min, 32 sec

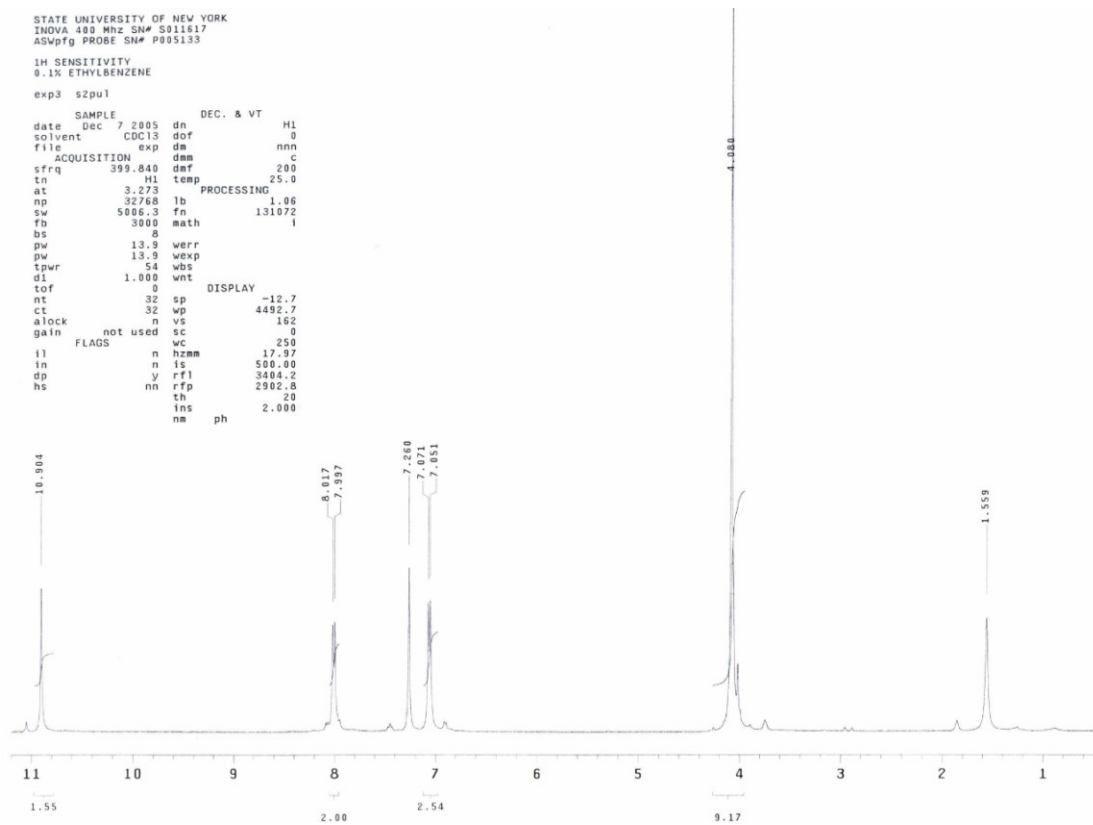


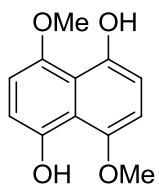
Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 INOVA-400 "Inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.189 sec
 Width 20000.0 Hz
 256 repetitions
 OBSERVE C13, 100.6283852 MHz
 DECUPLE H1, 399.8393842 MHz
 Power 47 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 19 hr, 37 min, 42 sec



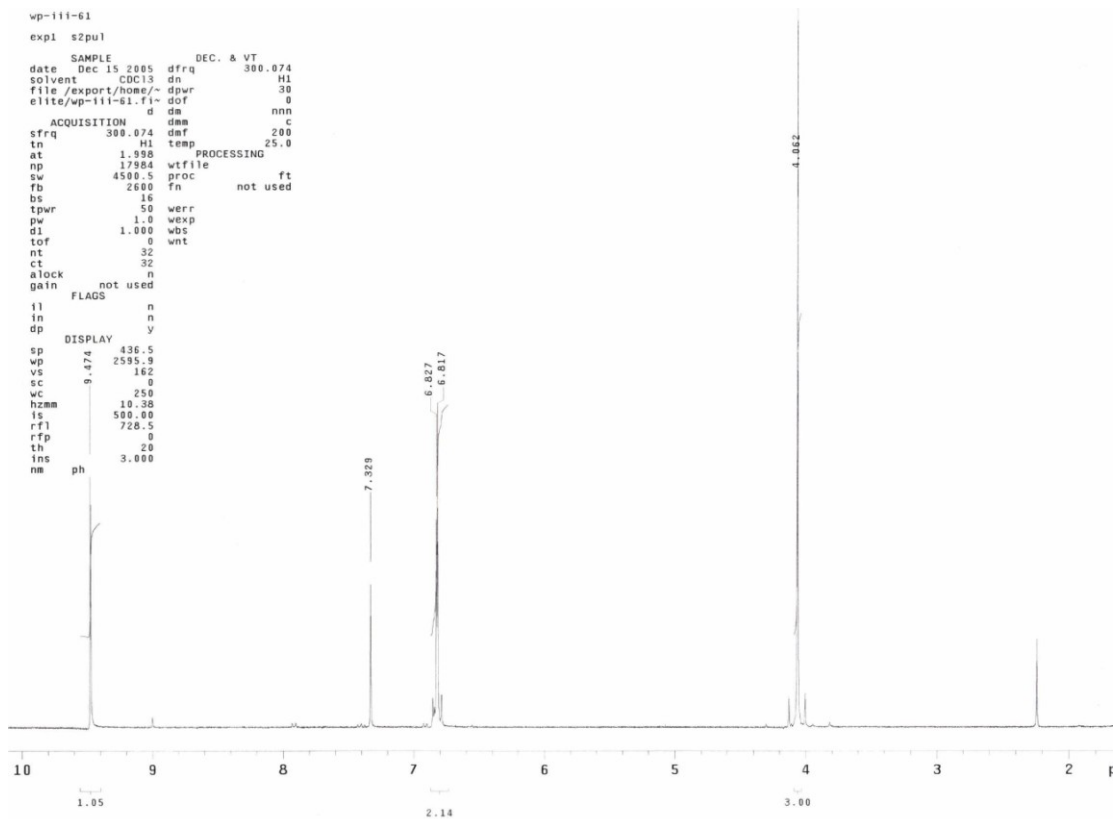


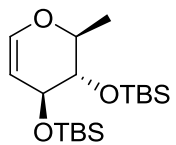
160





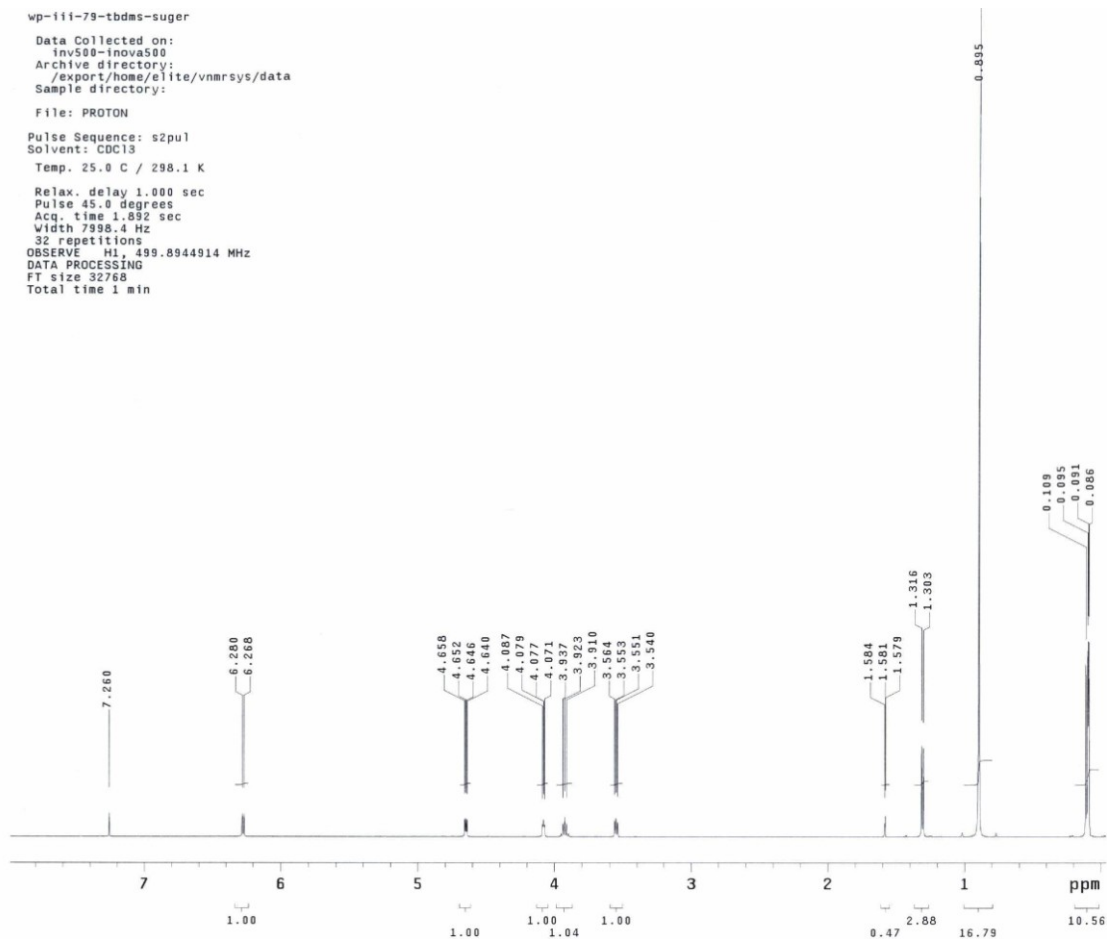
162

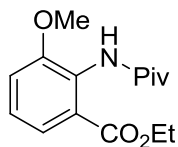




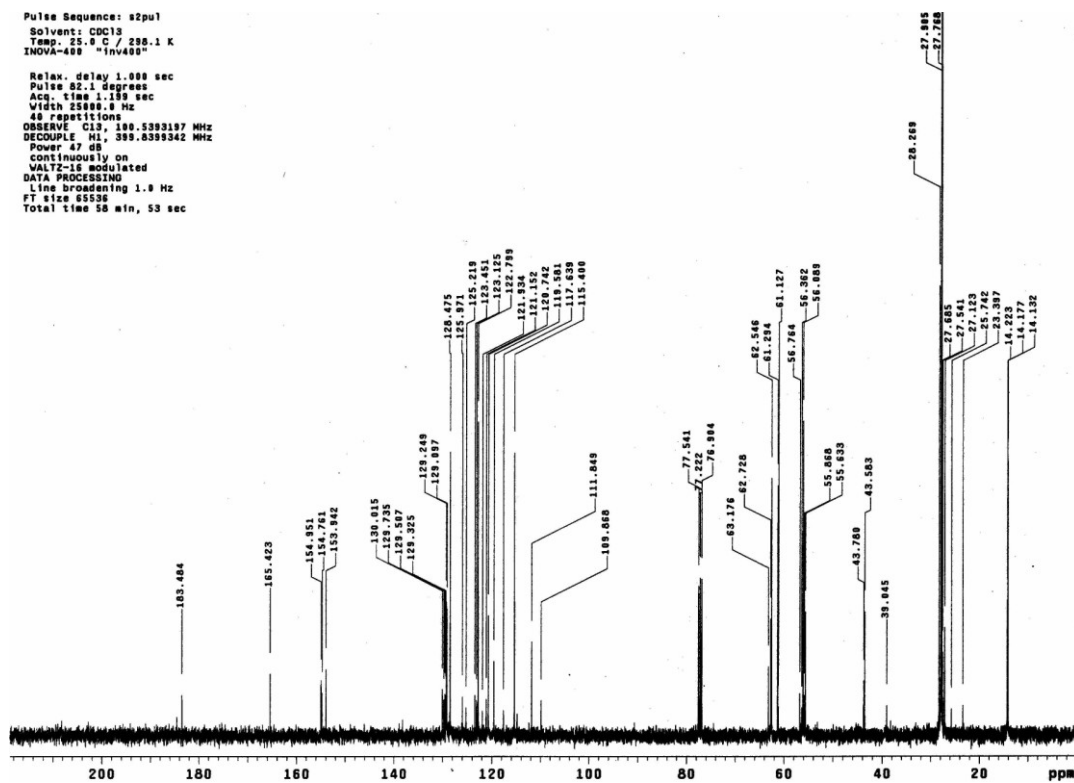
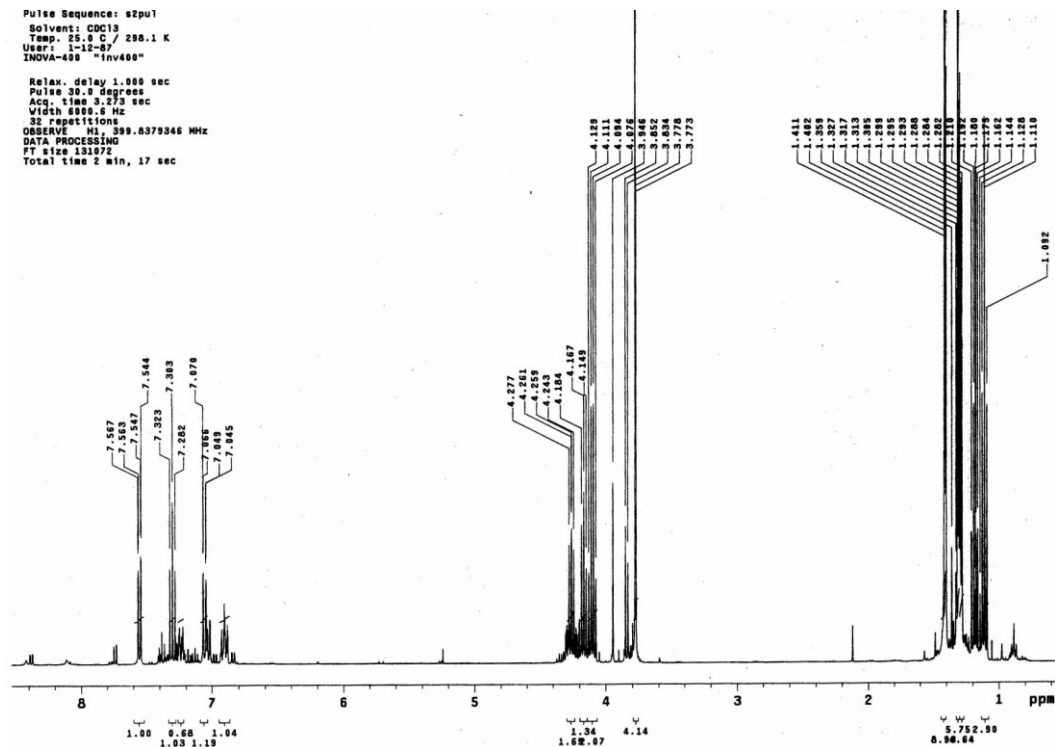
166

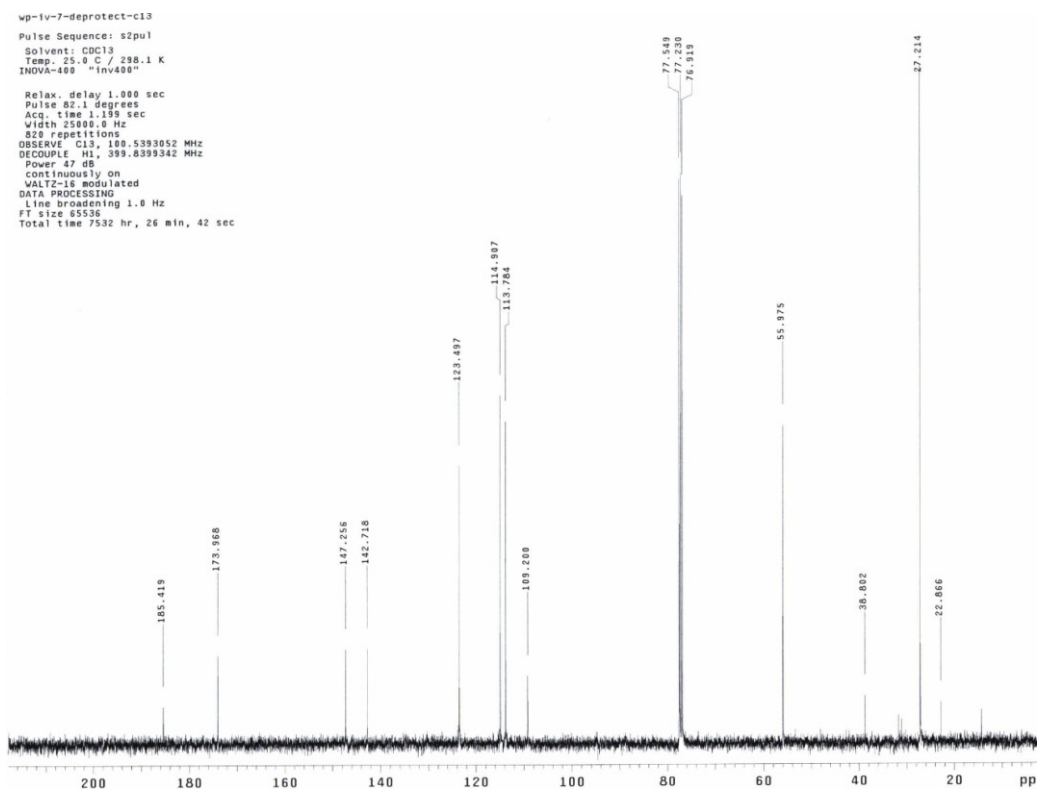
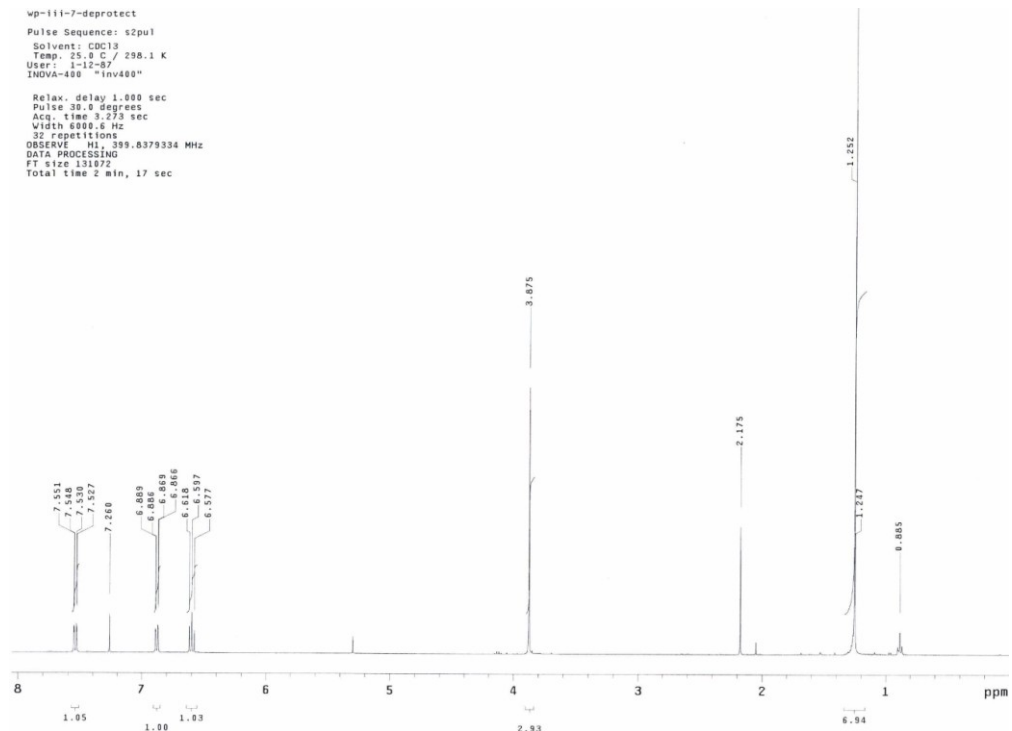
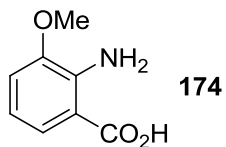
wp-iii-79-tbdms-suger
 Data Collected on:
 inv500-inova500
 Archive directory:
 /export/home/elite/vnmrsys/data
 Sample directory:
 File: PROTON
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.892 sec
 Width 7998.4 Hz
 32 repetitions
 OBSERVE H1, 499.8944914 MHz
 DATA PROCESSING
 FT size 32768
 Total time 1 min

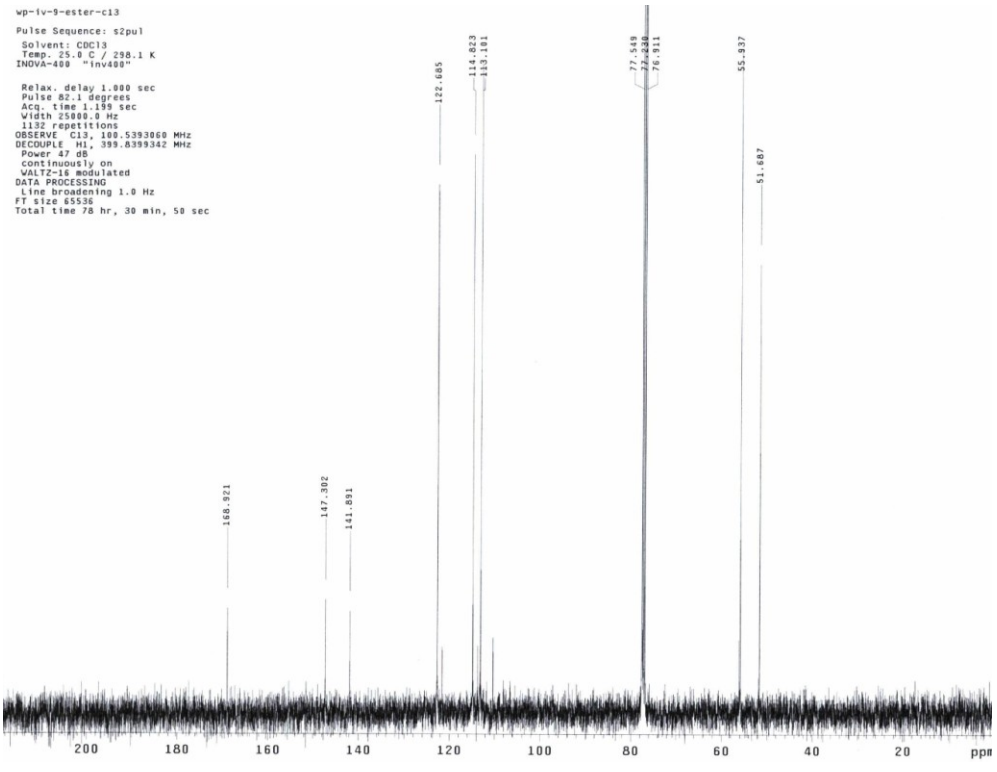
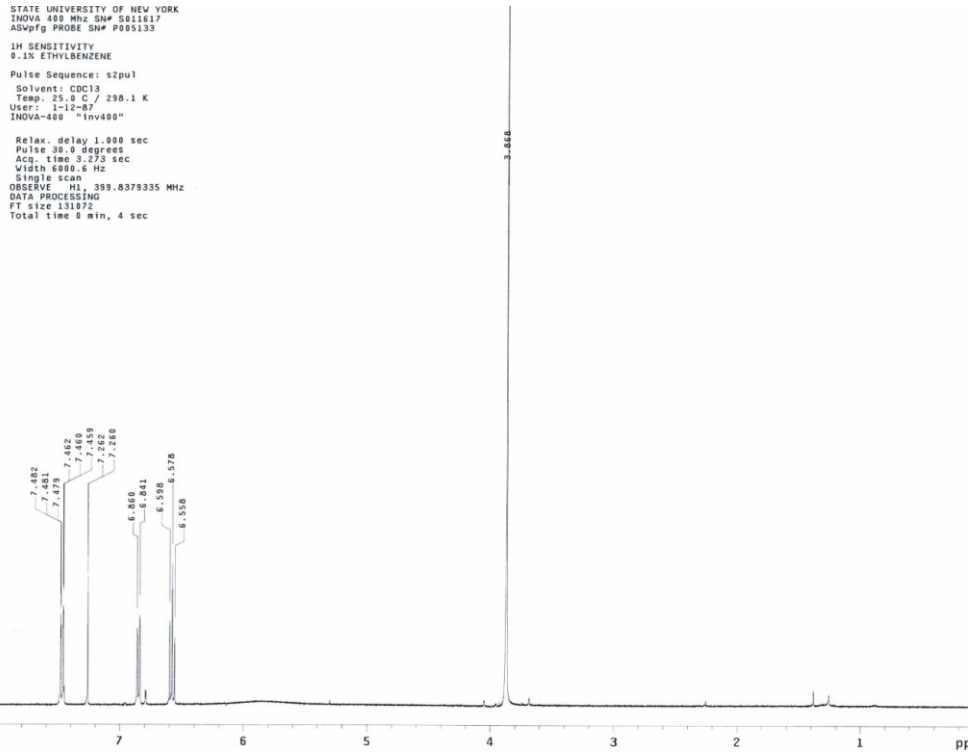
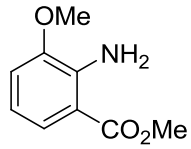


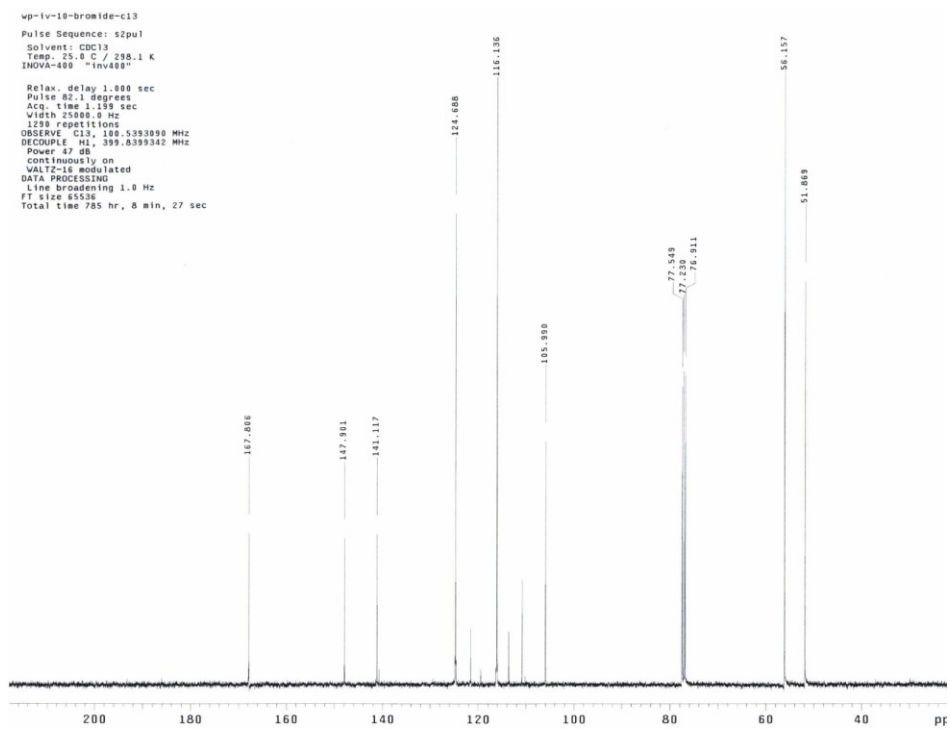
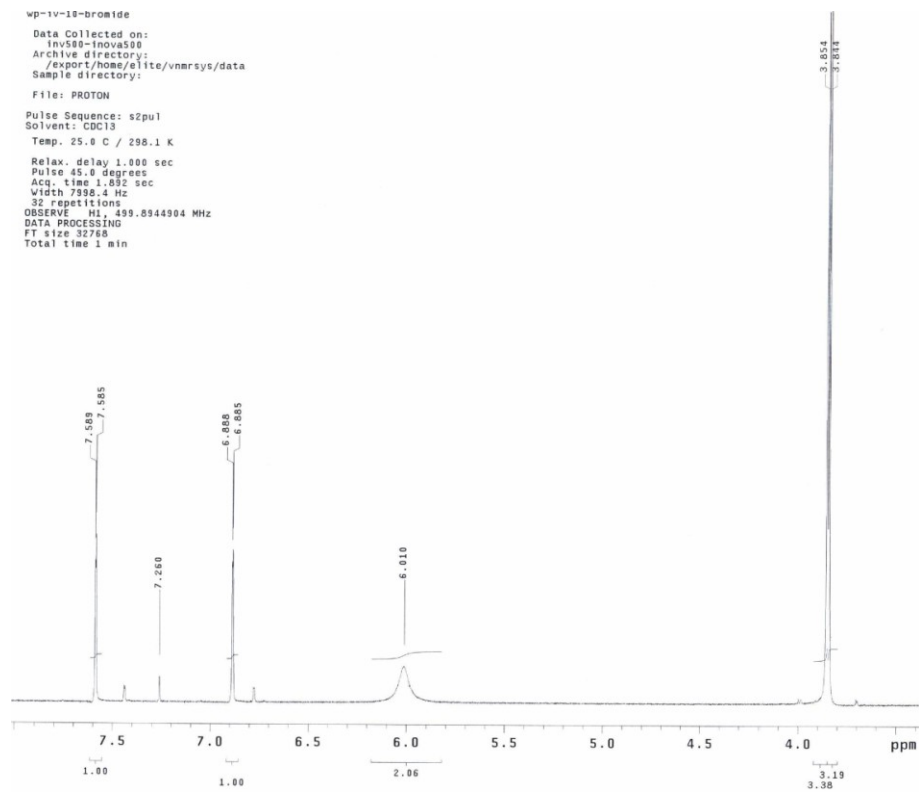
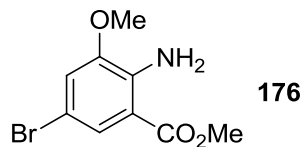


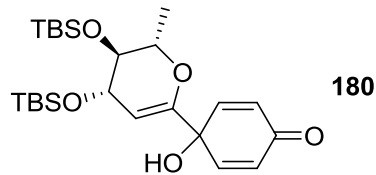
173



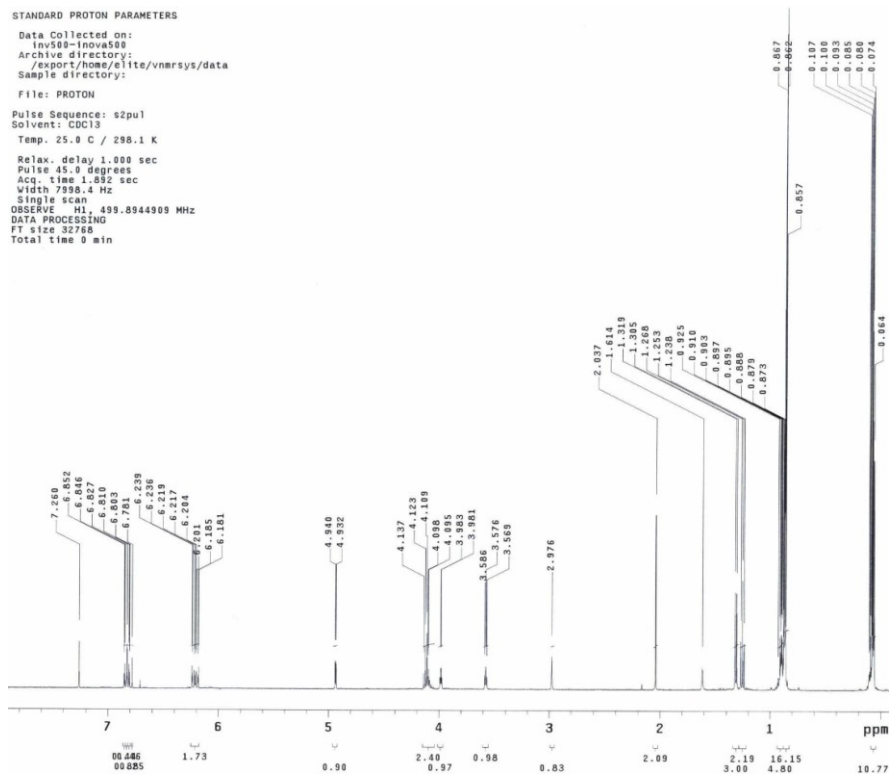




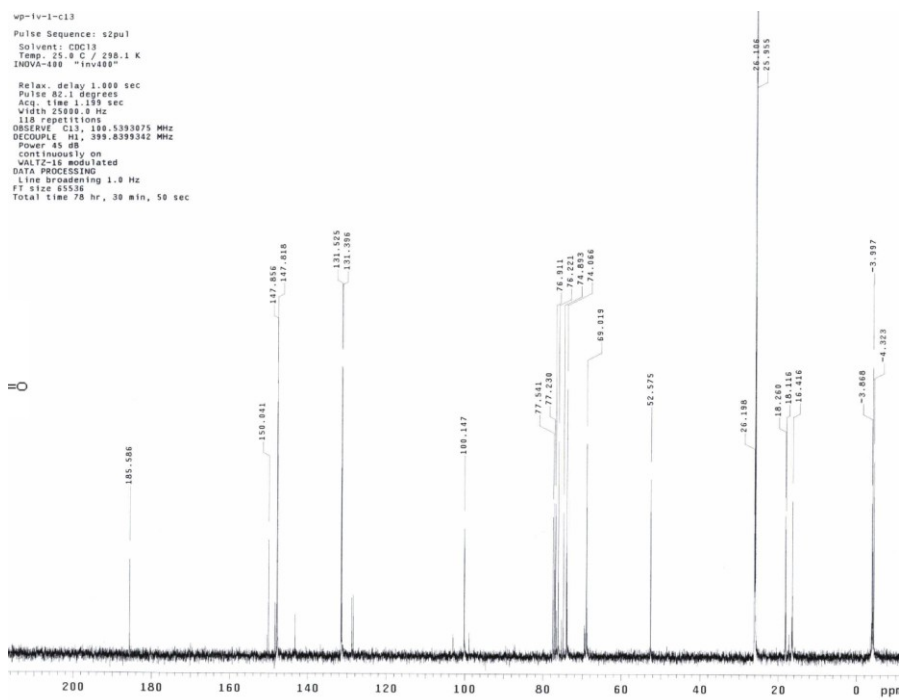


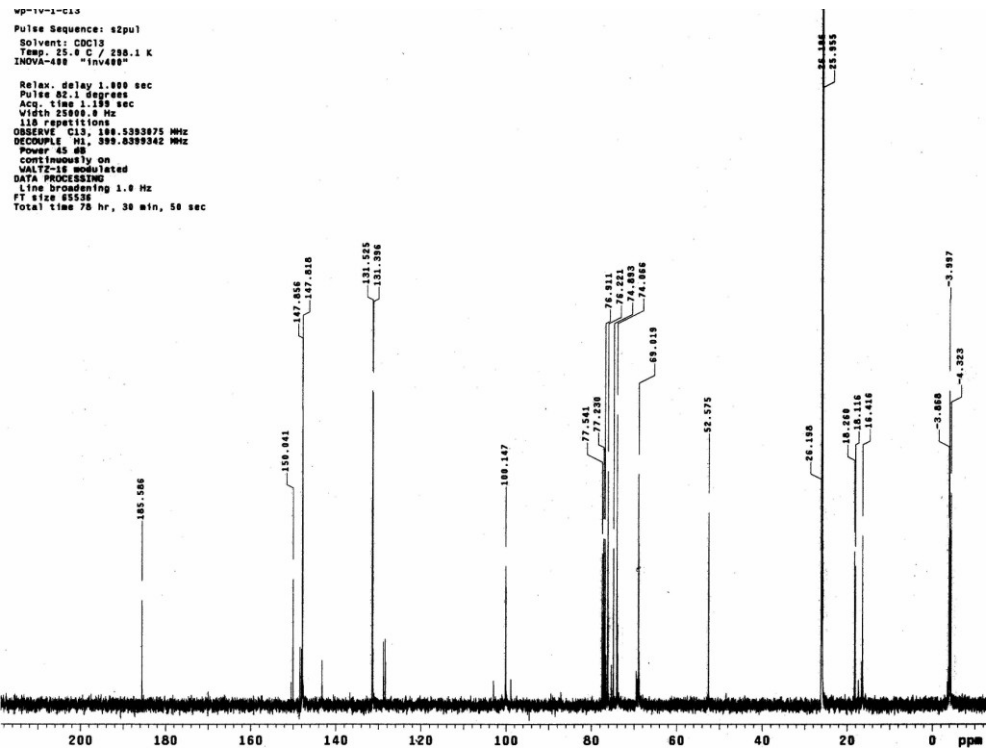
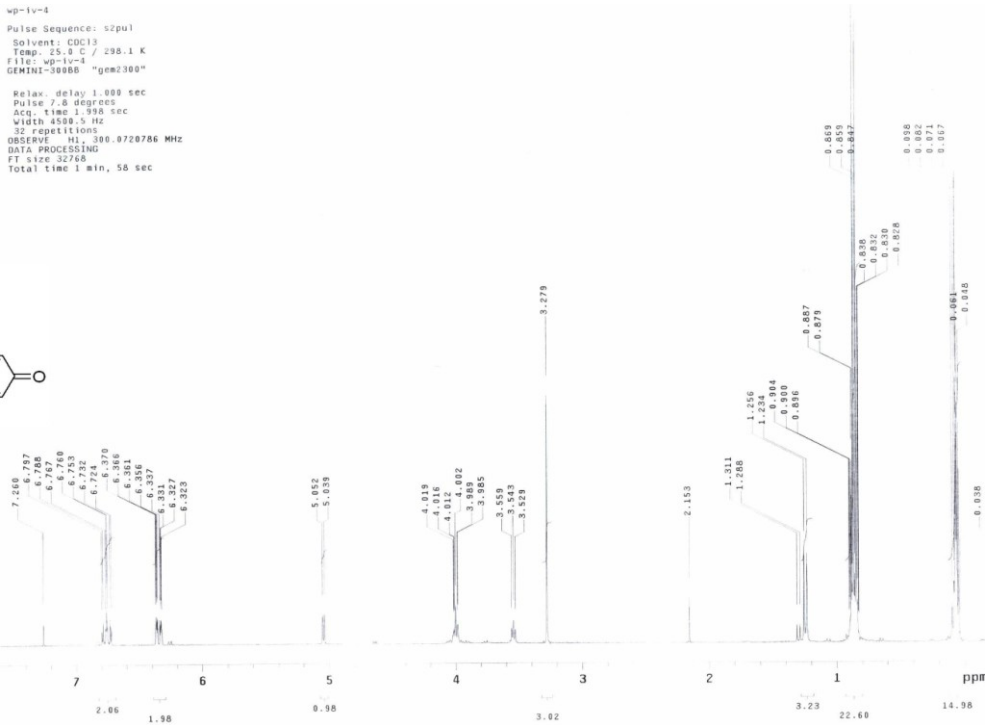
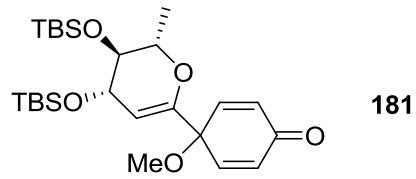


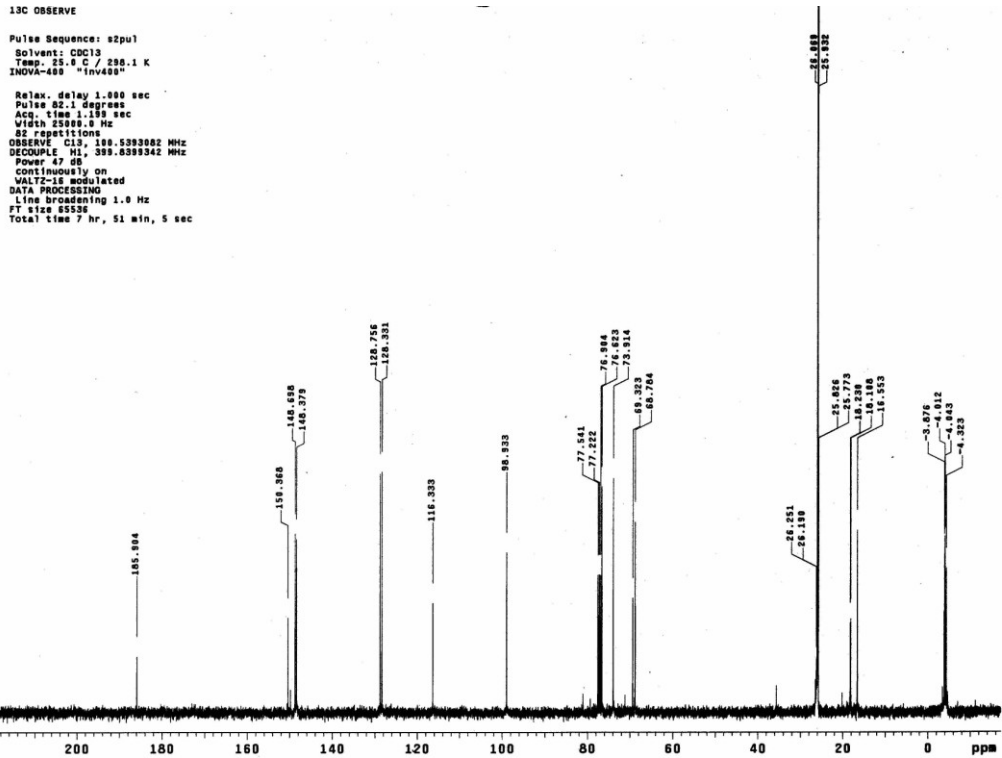
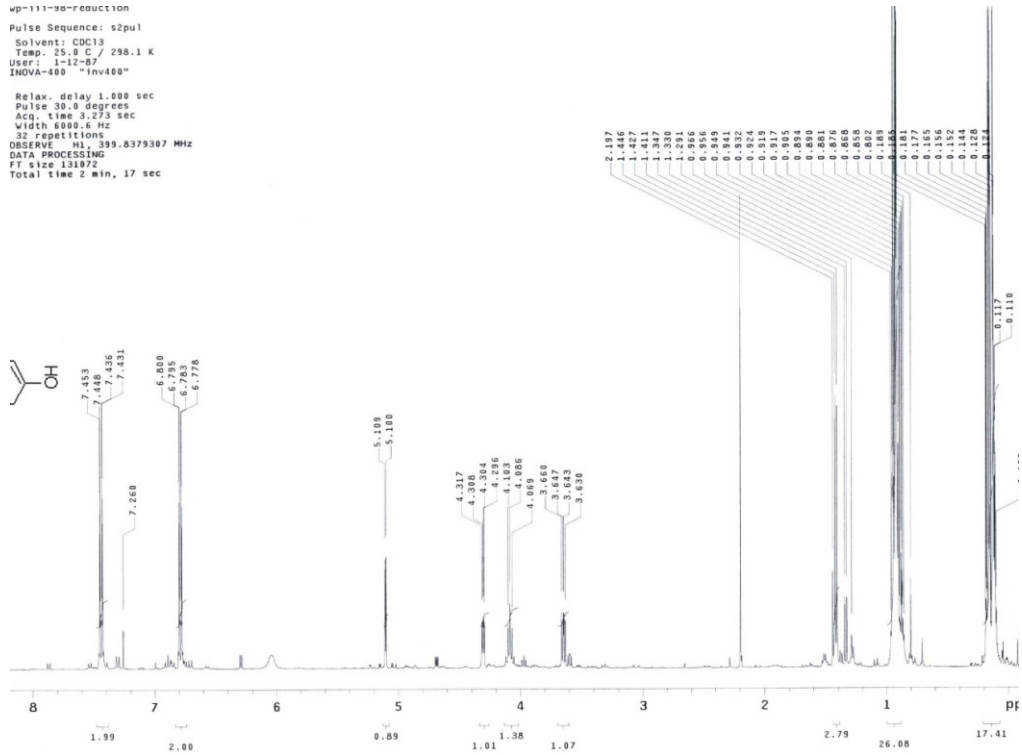
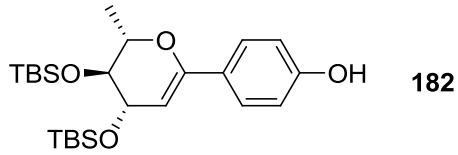
STANDARD PROTON PARAMETERS
 Data Collected on:
 inv568-inova508
 Archive directory:
 /export/home/elite/vnmrsys/data
 Sample directory:
 File: PROTON
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.892 sec
 Width 7998.4 Hz
 Single scan
 OBSERVE H1, 499.8944909 MHz
 DATA PROCESSING
 FT size 32768
 Total time 0 min

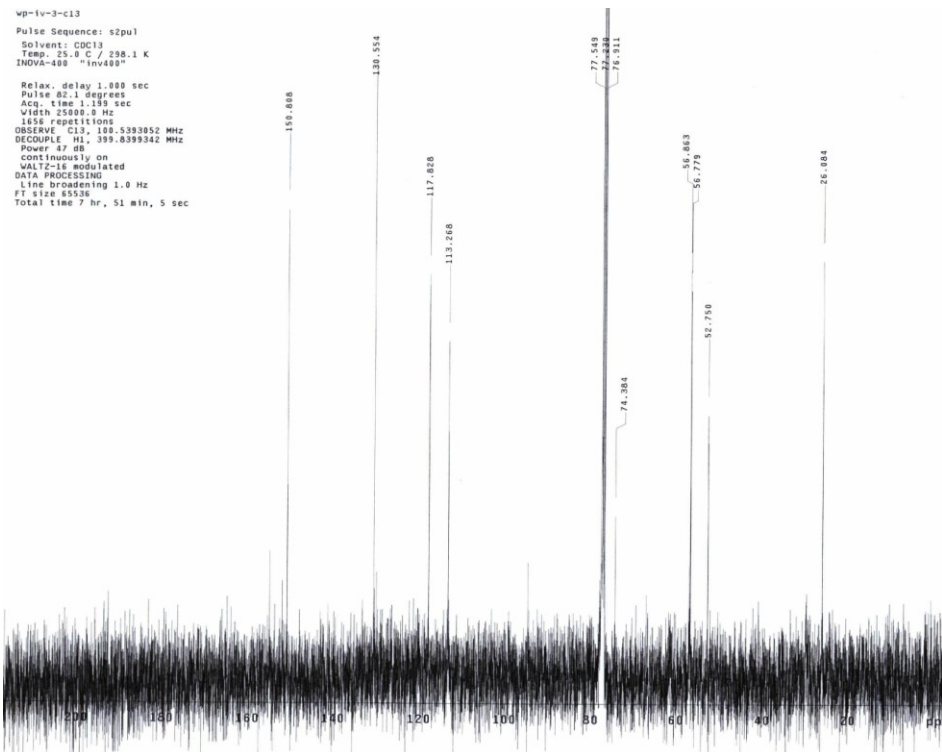
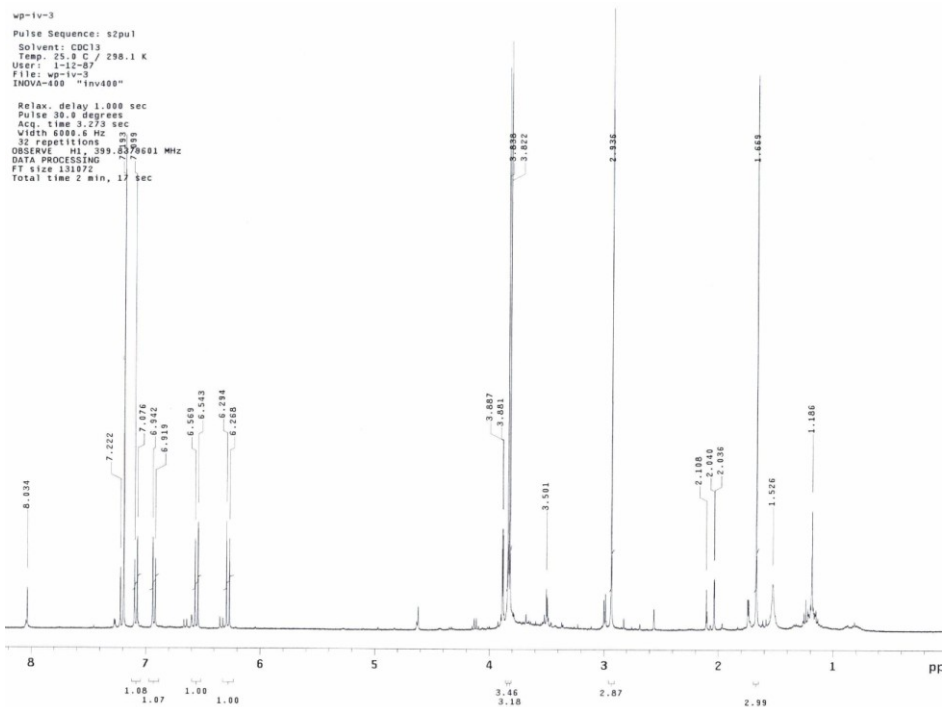
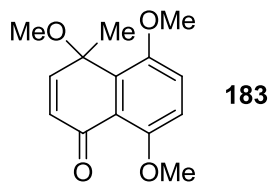


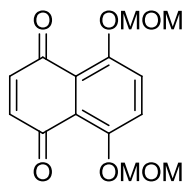
wp-1v-1-c13
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.189 sec
 Width 5588.0 Hz
 118 repetitions
 OBSERVE C13, 100.625075 MHz
 DECOUPLE H1, 399.8399342 MHz
 Power 4.00 W
 Continuously on
 VOLTAGE modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 8536
 Total time 78 hr, 30 min, 50 sec



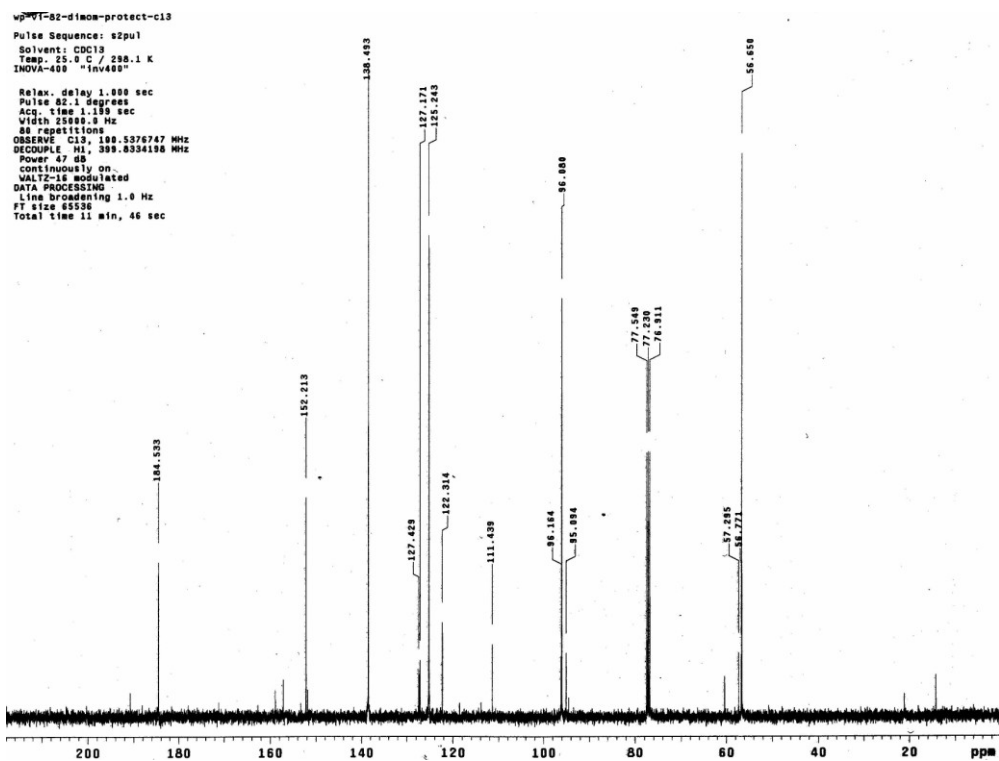
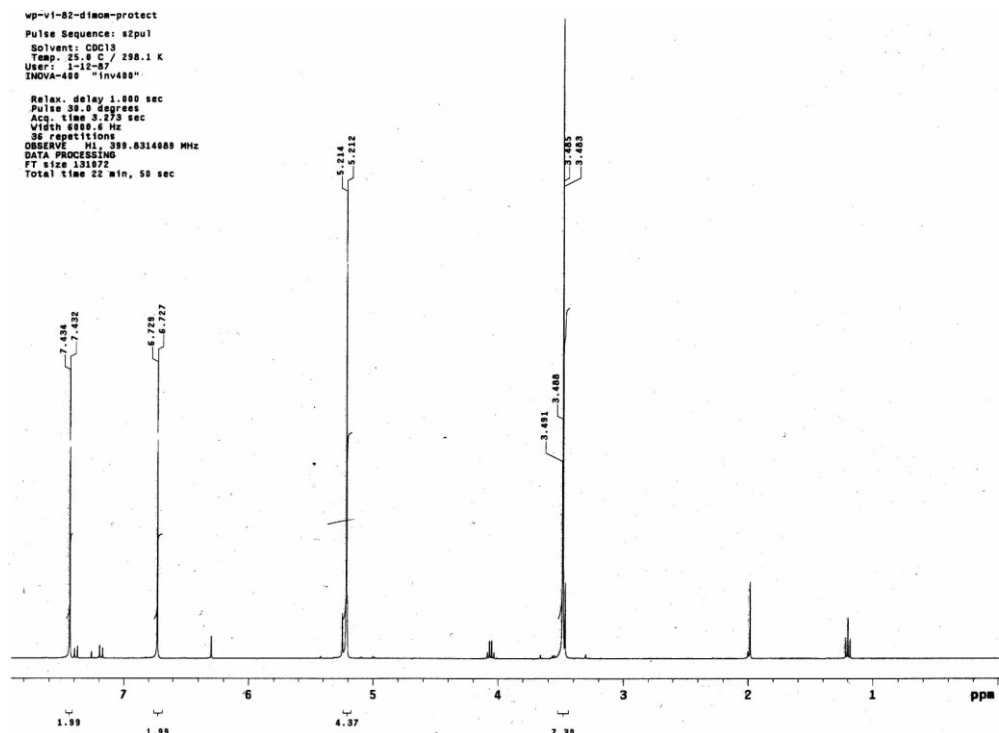


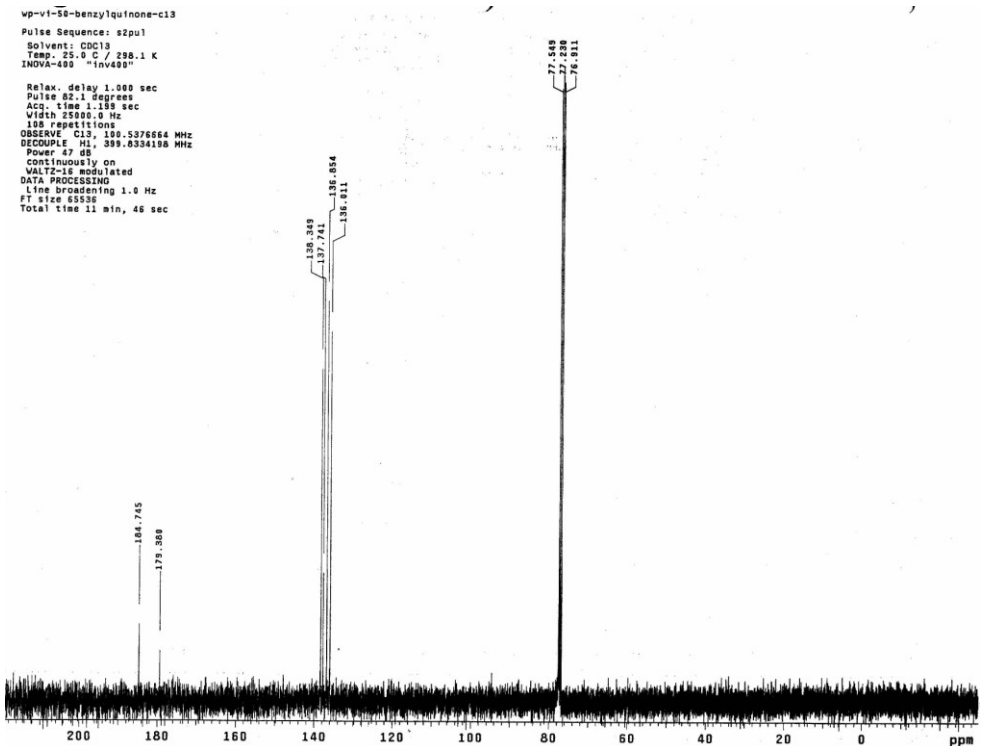
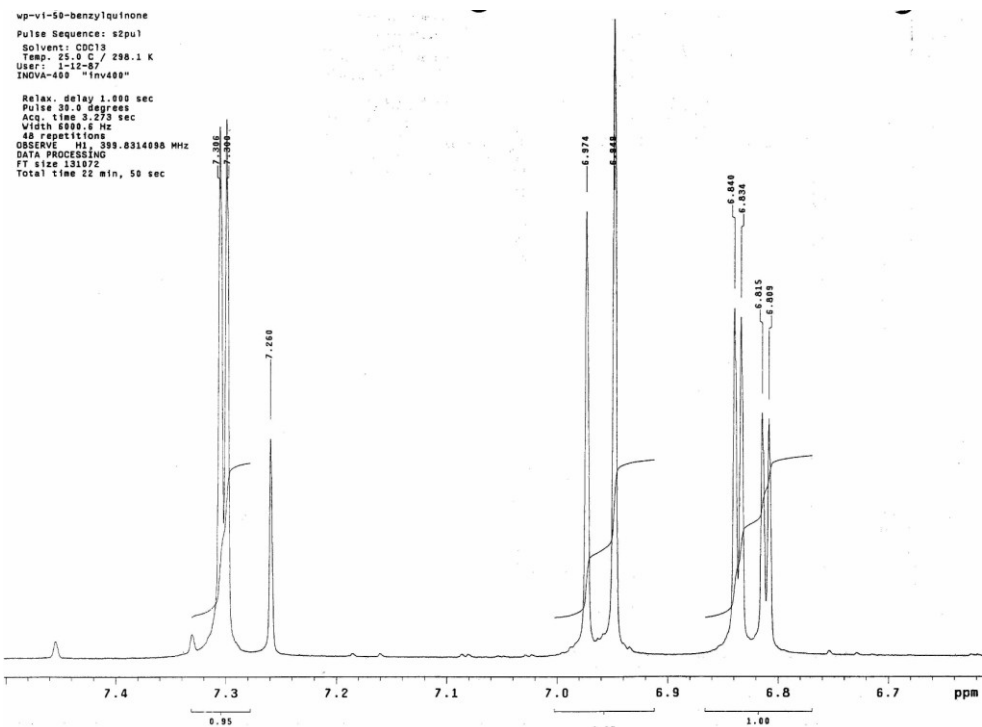
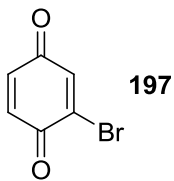


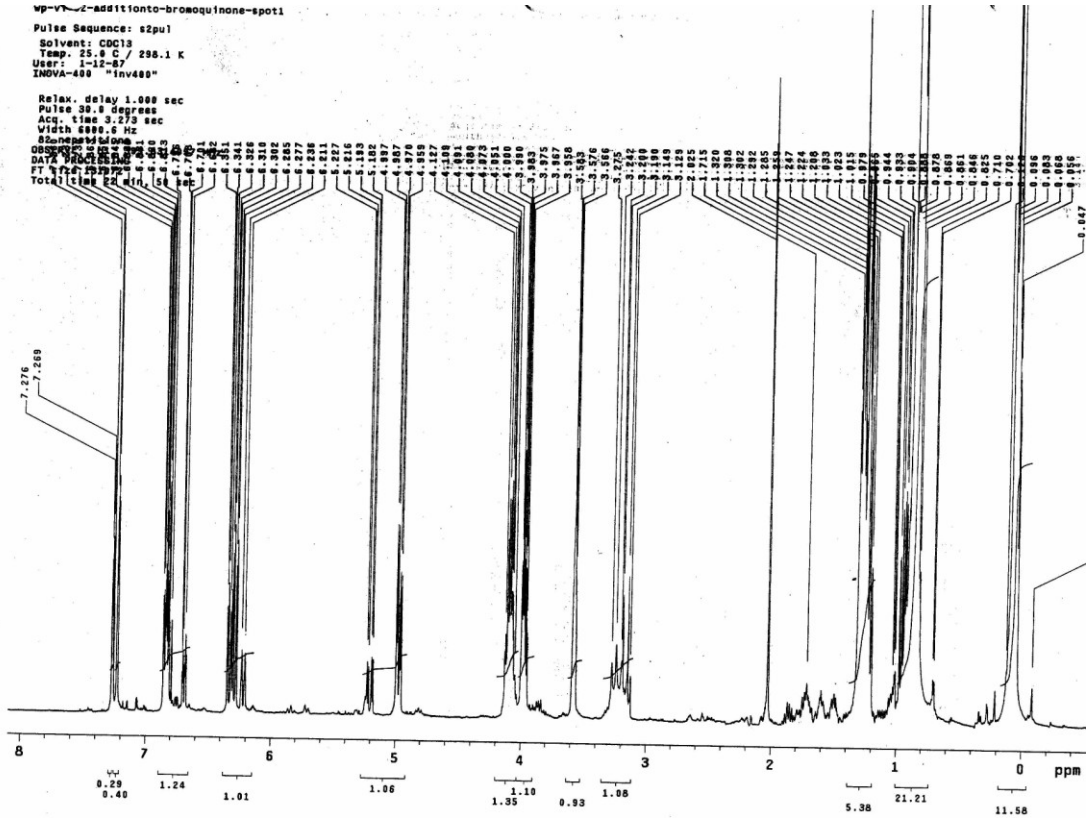
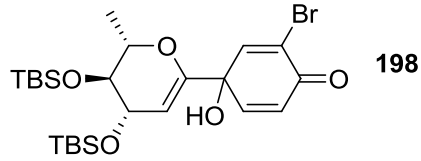


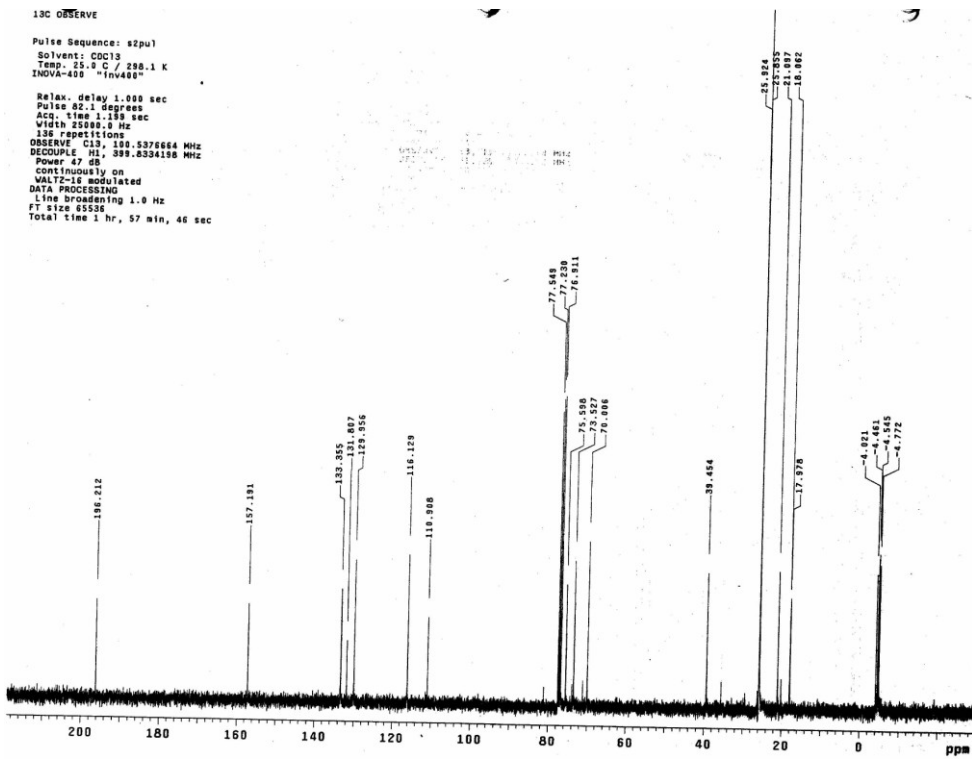
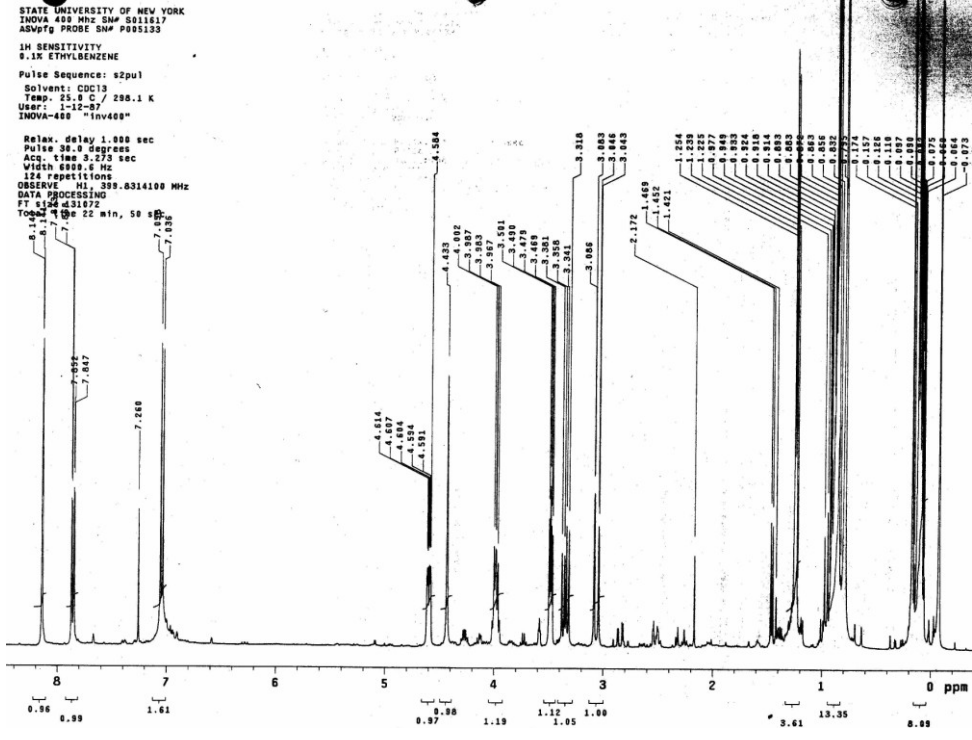
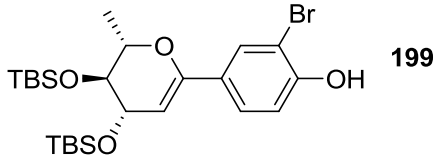


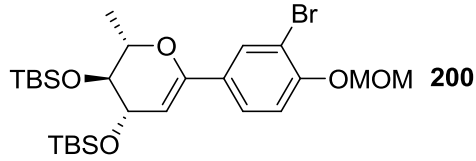
186



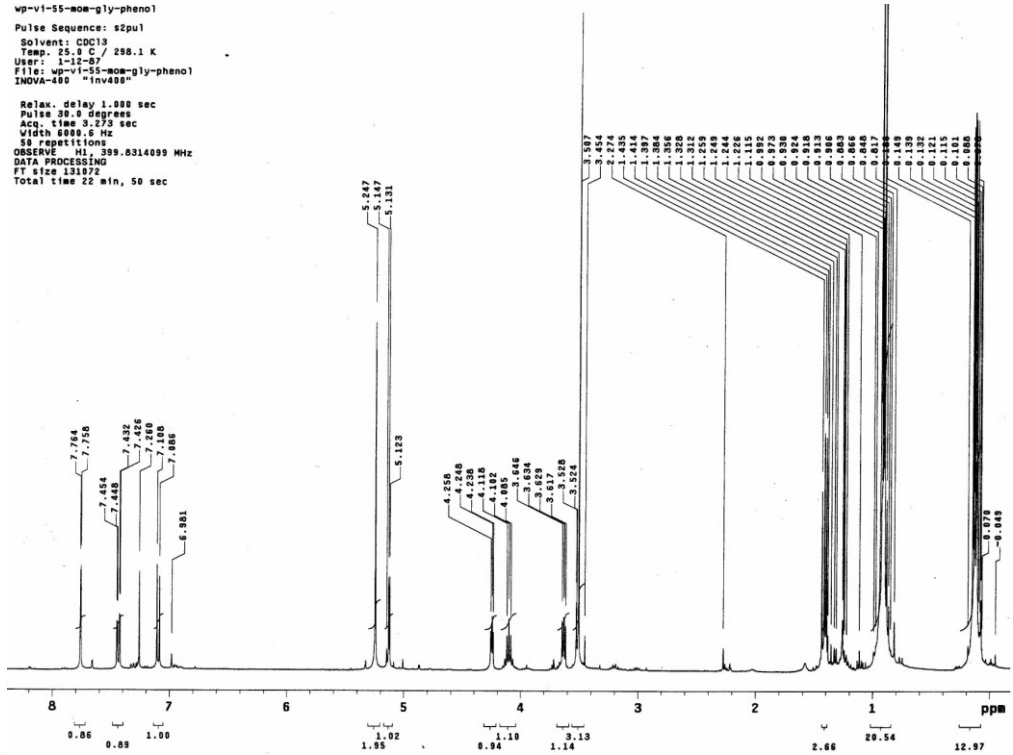




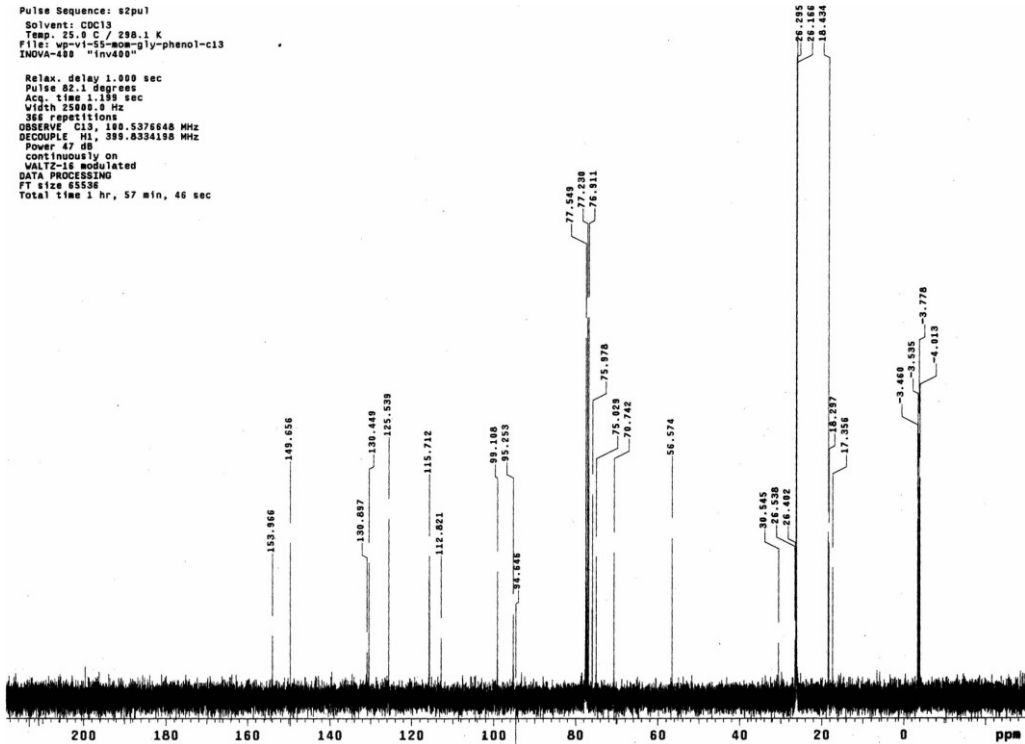


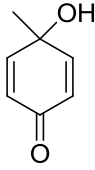


vp-vi-55-mom-gly-phenol
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-07
 File: vp-vi-55-mom-gly-phenol
 INOVA-400 "inv400"



Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 File: vp-vi-55-mom-gly-phenol-c13
 INOVA-400 "inv400"

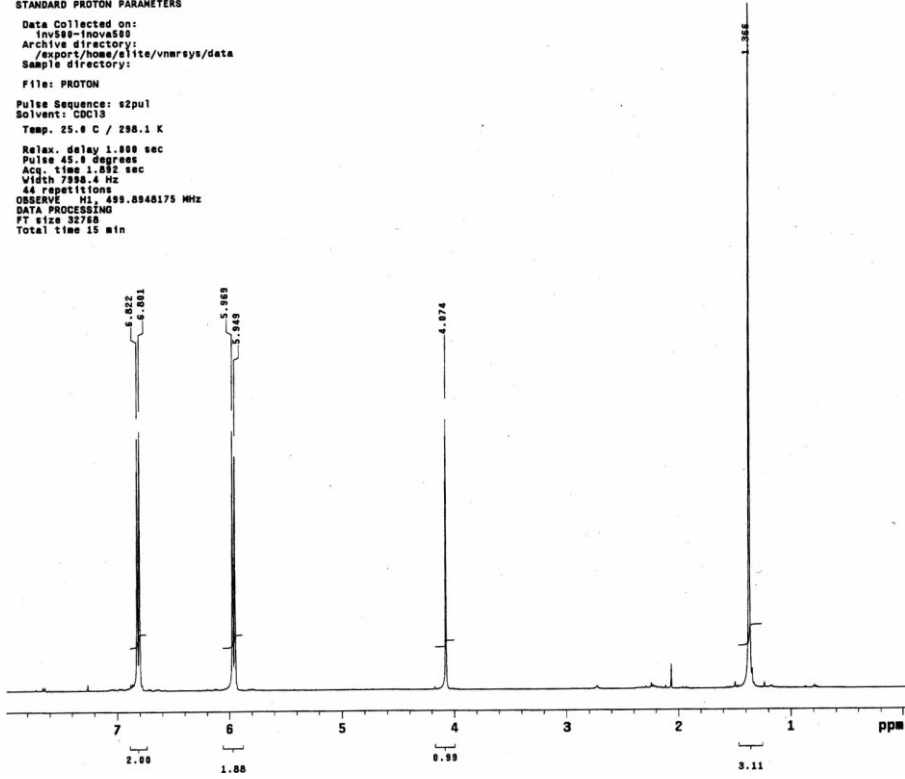




202

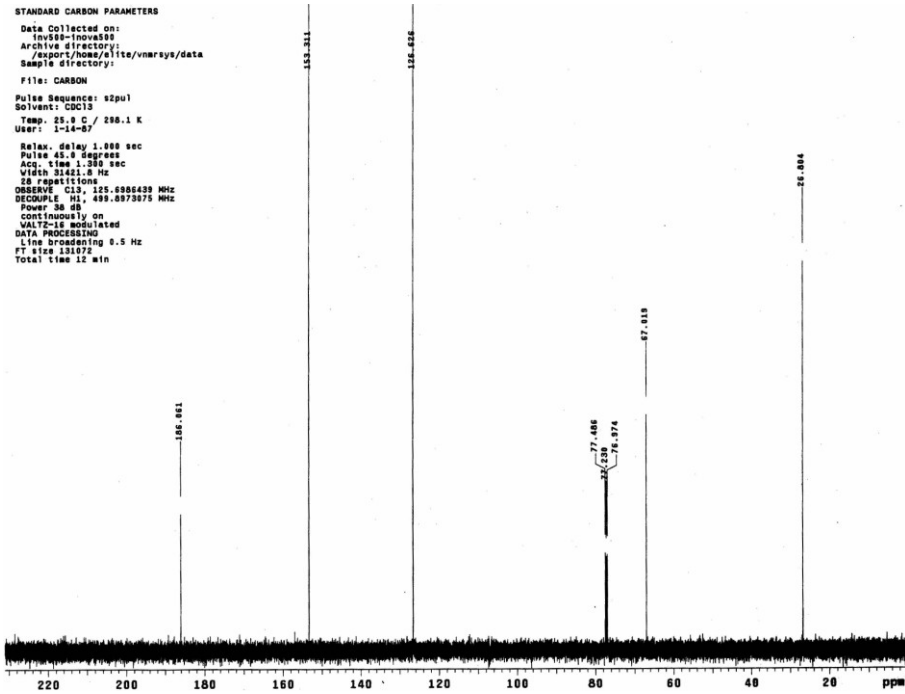
STANDARD PROTON PARAMETERS

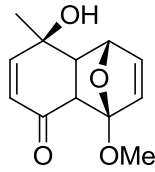
Data Collected on: inv500-inova500
 Archive directory: /export/home/elite/vnmrsys/data
 Sample directory:
 File: PROTON
 Pulse Sequence: e2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 Relax. delay 1.800 sec
 Pulse 45.0 degrees
 Acq. time 1.832 sec
 Width 7890.4 Hz
 44 repetitions
 OBSERVE H1, 499.8948175 MHz
 DATA PROCESSING
 FT size 32788
 Total time 15 min



STANDARD CARBON PARAMETERS

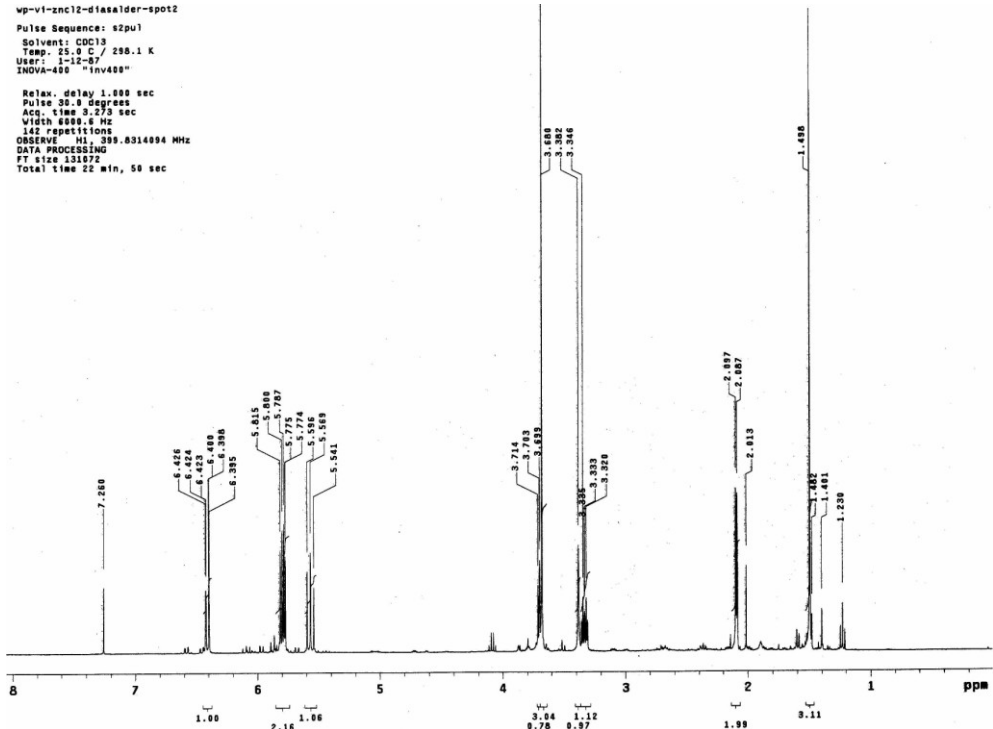
Data Collected on: inv500-inova500
 Archive directory: /export/home/elite/vnmrsys/data
 Sample directory:
 File: CARBON
 Pulse Sequence: e2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: j-l-r-87
 Relax. delay 1.800 sec
 Pulse 45.0 degrees
 Acq. time 1.300 sec
 Width 31421.8 Hz
 20 repetitions
 OBSERVE C13, 125.8966439 MHz
 DECOUPLE H1, 499.8973075 MHz
 Power 30 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 0.5 Hz
 FT size 131972
 Total time 12 min



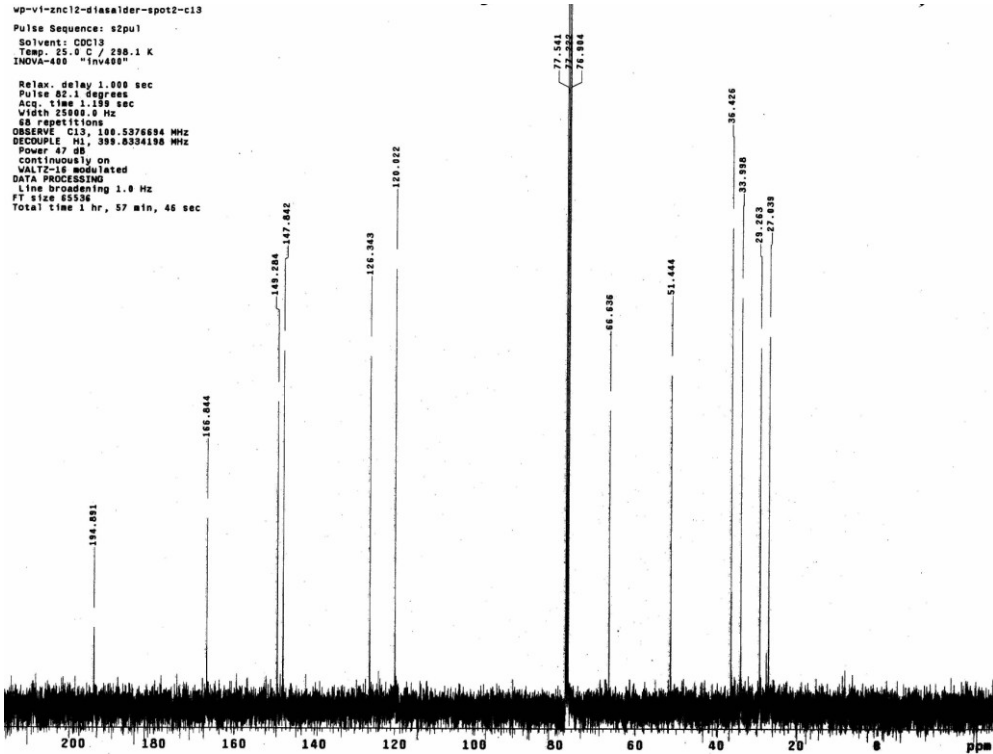


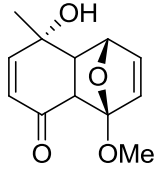
203

wp-vi-znc12-dissalder-spot2
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: j-12-07
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 35.0 degrees
 Acq. time 3.273 sec
 Width 6500.0 Hz
 142 repetitions
 OBSERVE H1, 399.8314094 MHz
 DATA PROCESSING
 FT size 131072
 Total time 22 min, 58 sec



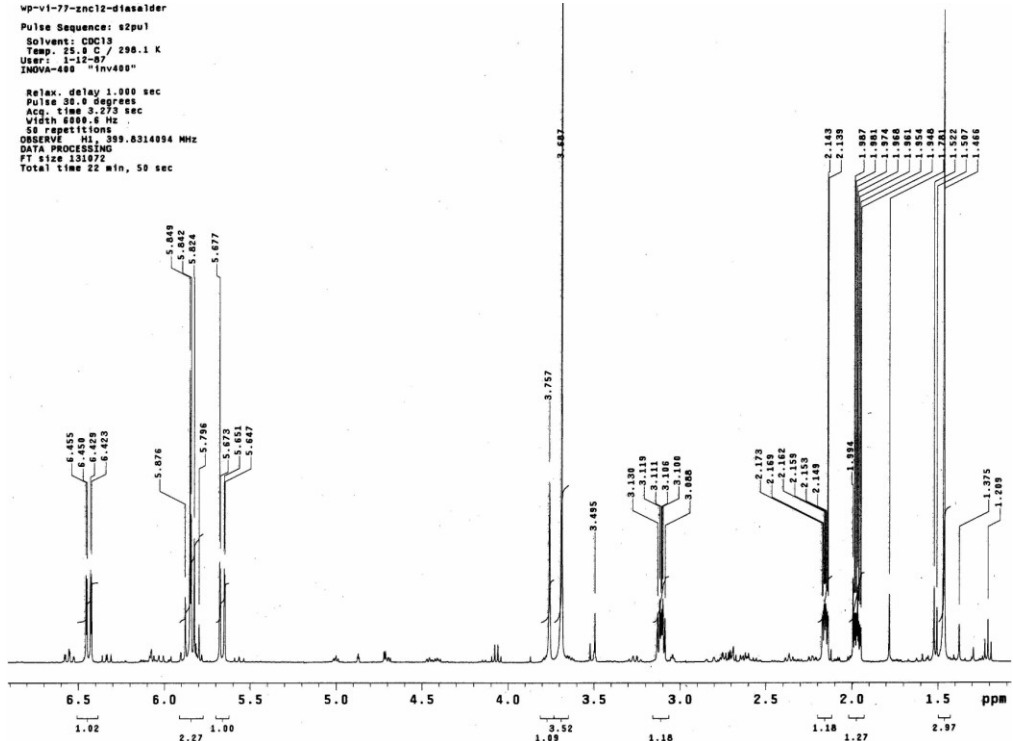
wp-vi-znc12-dissalder-spot2-c13
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: j-12-07
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.139 sec
 Width 25800.0 Hz
 88 repetitions
 OBSERVE C13, 100.5376694 MHz
 DECOUPLE H1, 399.8334198 MHz
 Power 47.08
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 1 hr, 57 min, 46 sec



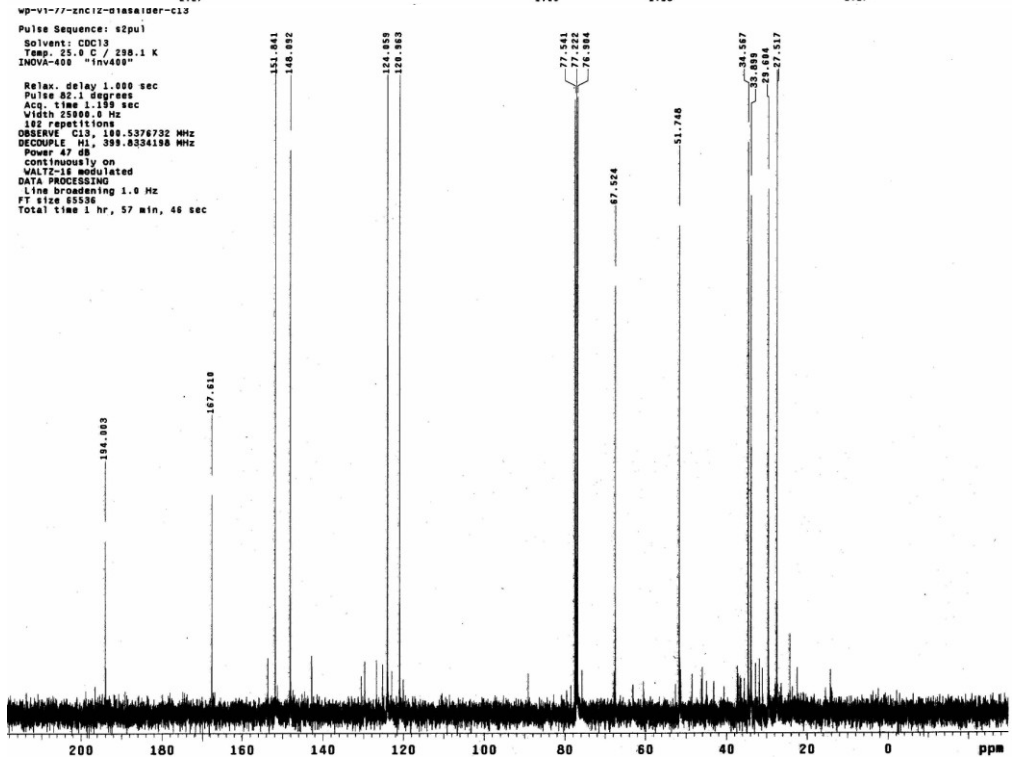


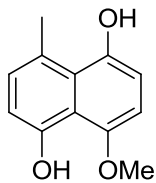
204

wp-vi-77-znc12-diasalder
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-97
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 30.0 degrees
 Acq. time 9.273 sec
 Width 6000.0 Hz
 50 repetitions
 OBSERVE H1 399.8314094 MHz
 DATA PROCESSING
 FT size 131072
 Total time 22 min, 50 sec



wp-vi-77-znc12-diasalder-c13
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-97
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.139 sec
 Width 25000.0 Hz
 102 repetitions
 OBSERVE C13 100.5376732 MHz
 DECOUPLE H1 399.8334198 MHz
 Power 47 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 85308
 Total time 1 hr, 57 min, 46 sec





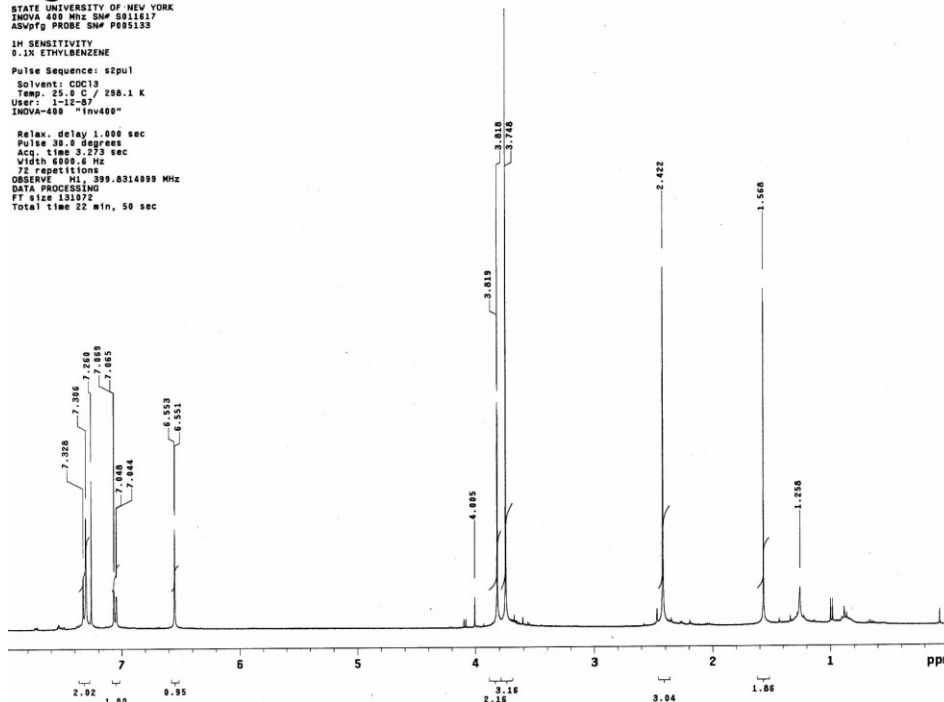
205

STATE UNIVERSITY OF NEW YORK
 INOVA 400 MHz S/NP 031117
 ASXDFG PROBE S/NP P055133

1H SENSITIVITY
 0.1X ETHYLBENZENE

Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-07
 INOVA-400 "Inv400"

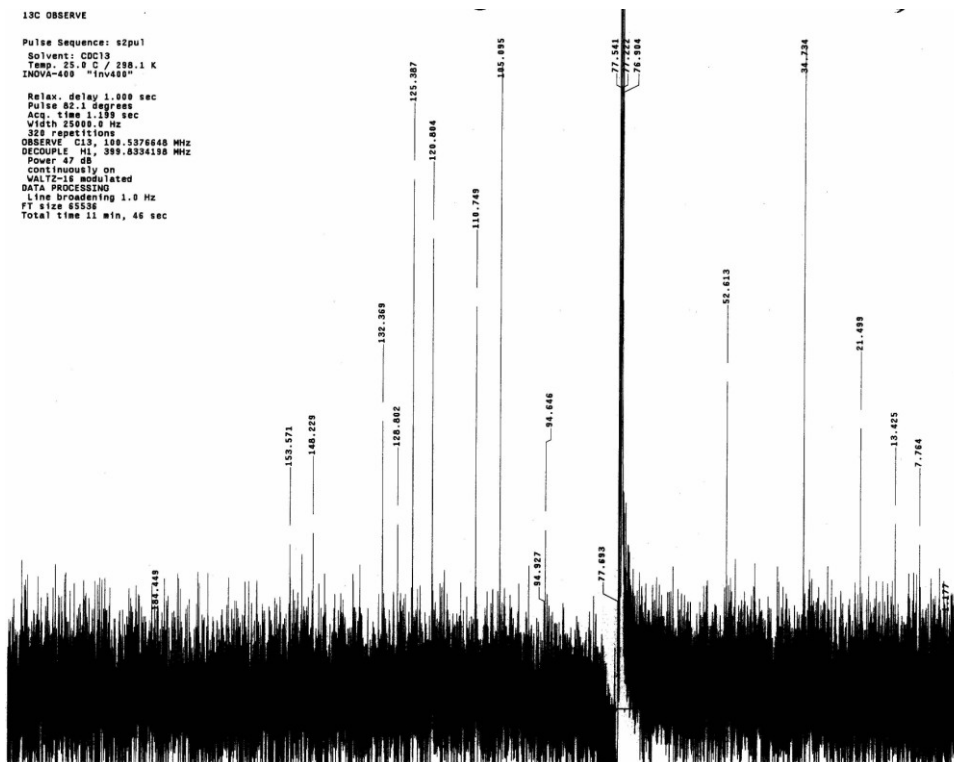
Relax. delay 1.000 sec
 Pulse 30.0 degrees
 Acq. time 3.273 sec
 Width 8080.0 Hz
 F2 repetitions
 OBSERVE H1, 399.8314999 MHz
 DATA PROCESSING
 FT size 131072
 Total time 22 min, 50 sec

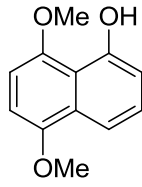


13C OBSERVE

Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-400 "Inv400"

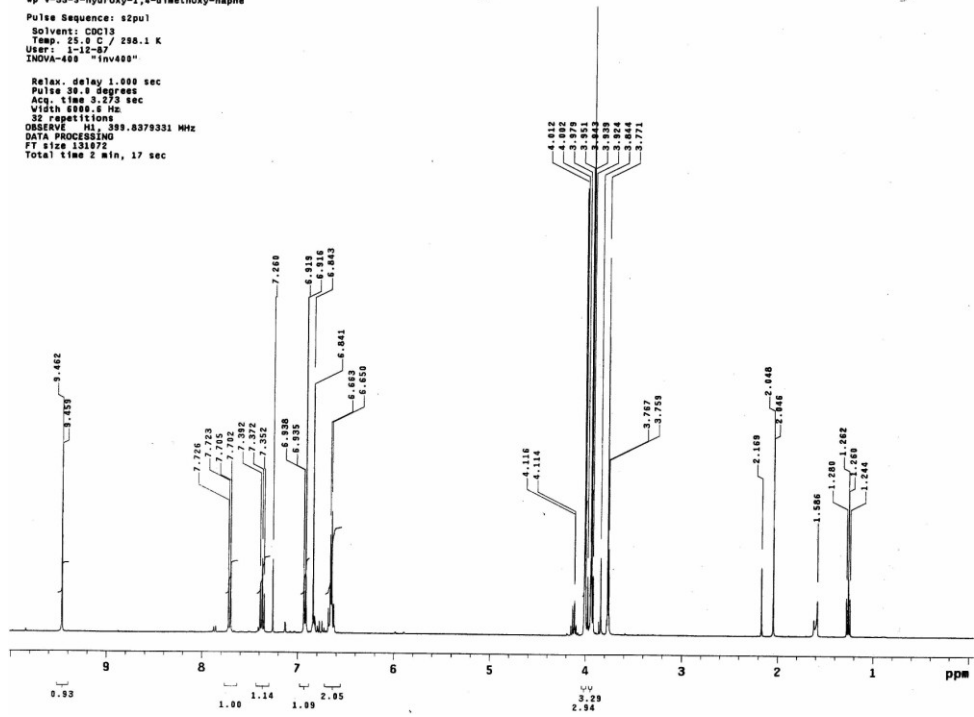
Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.139 sec
 Width 25000.0 Hz
 F2 repetitions
 OBSERVE C13, 100.5376648 MHz
 DECOUPLE H1, 399.8334198 MHz
 Power 47 dB
 continuously on
 VOLT-15 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 85536
 Total time 11 min, 46 sec

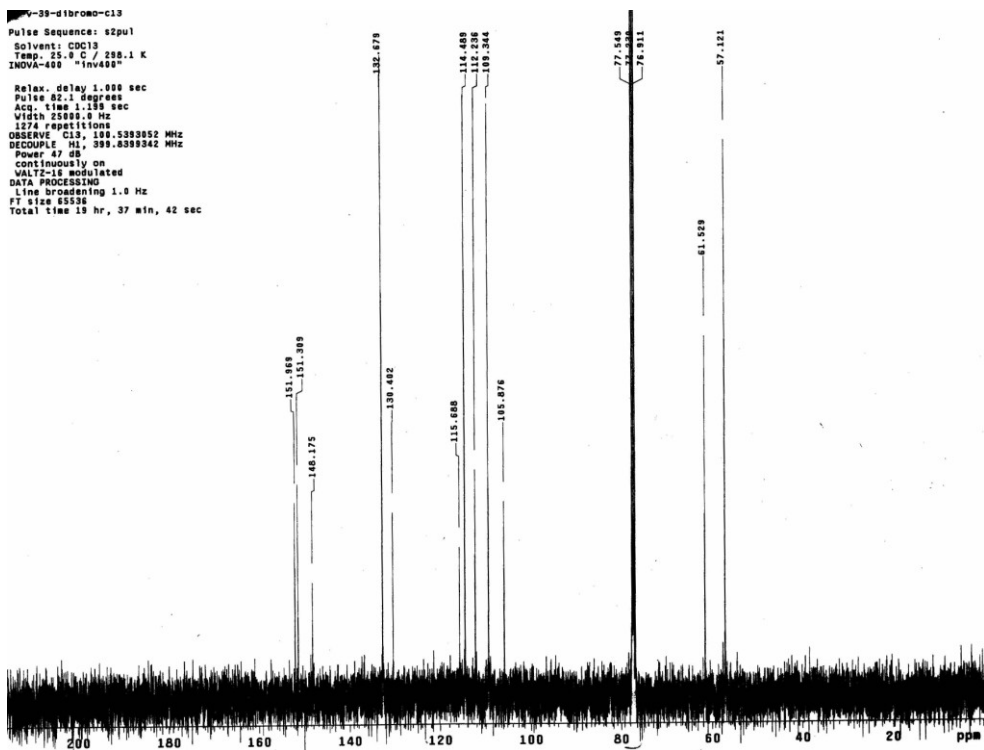
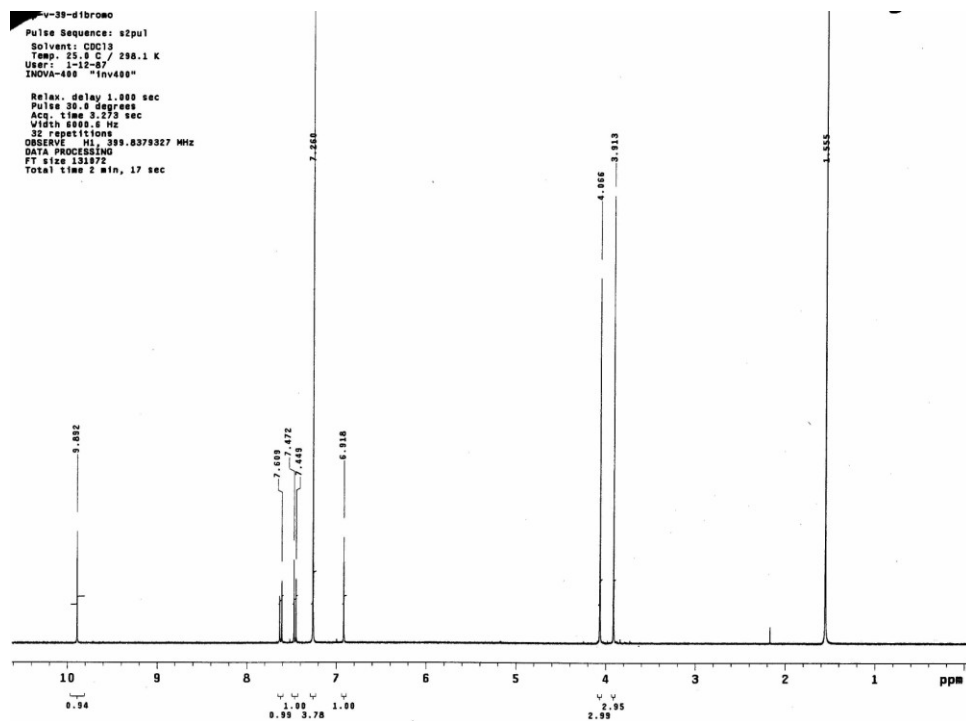
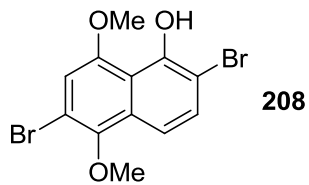


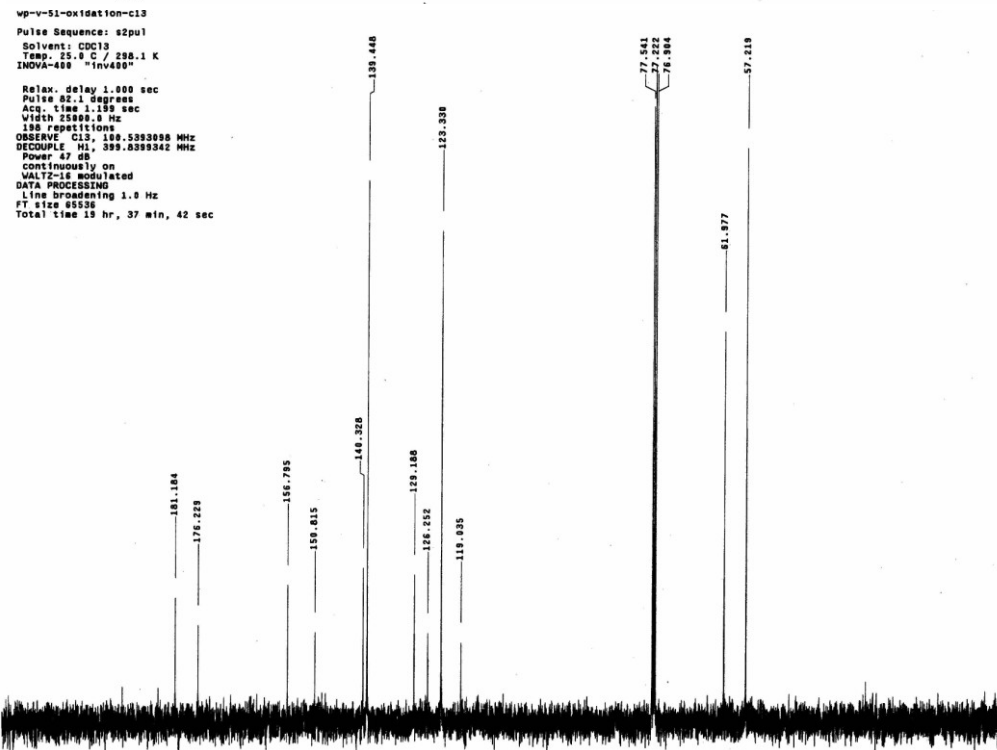
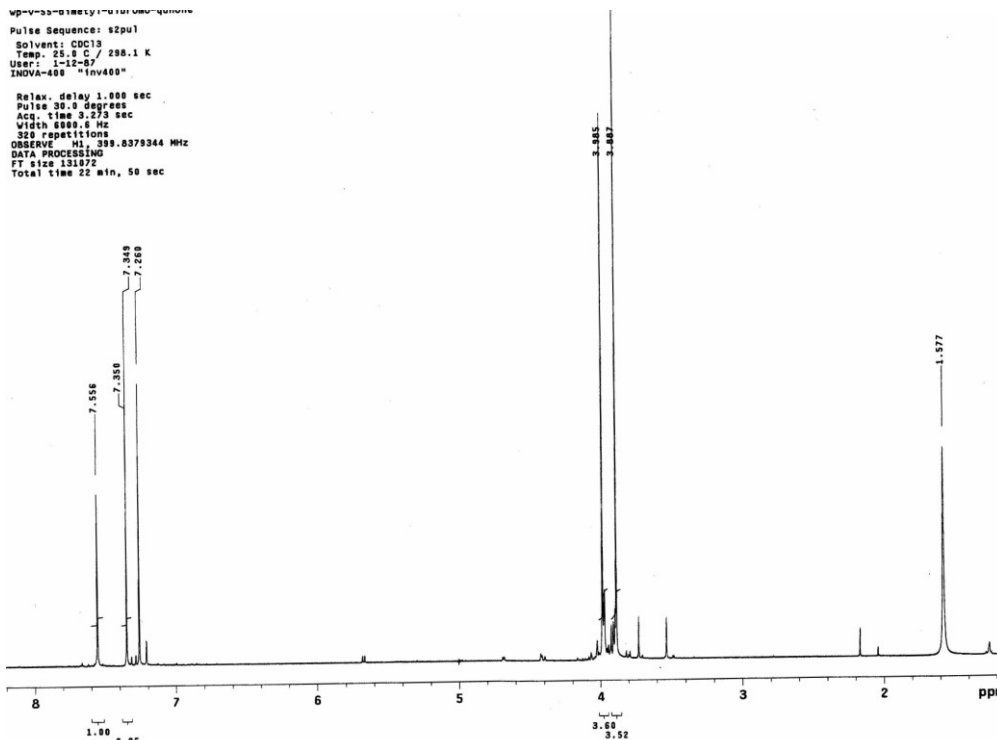
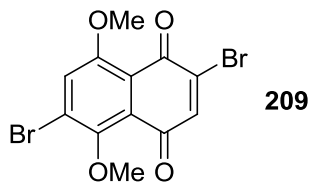


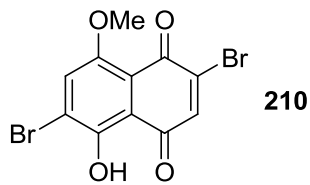
207

VP-V-33*5-hydroxy-1,4-dimethoxy-naphe
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-87
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 30.0 degree
 Acq. time 3.273 sec
 Width 6300.0 Hz
 32 repetitions
 OBSERVE H1 399.8379331 MHz
 DATA PROCESSING
 FT size 131072
 Total time 2 min, 17 sec

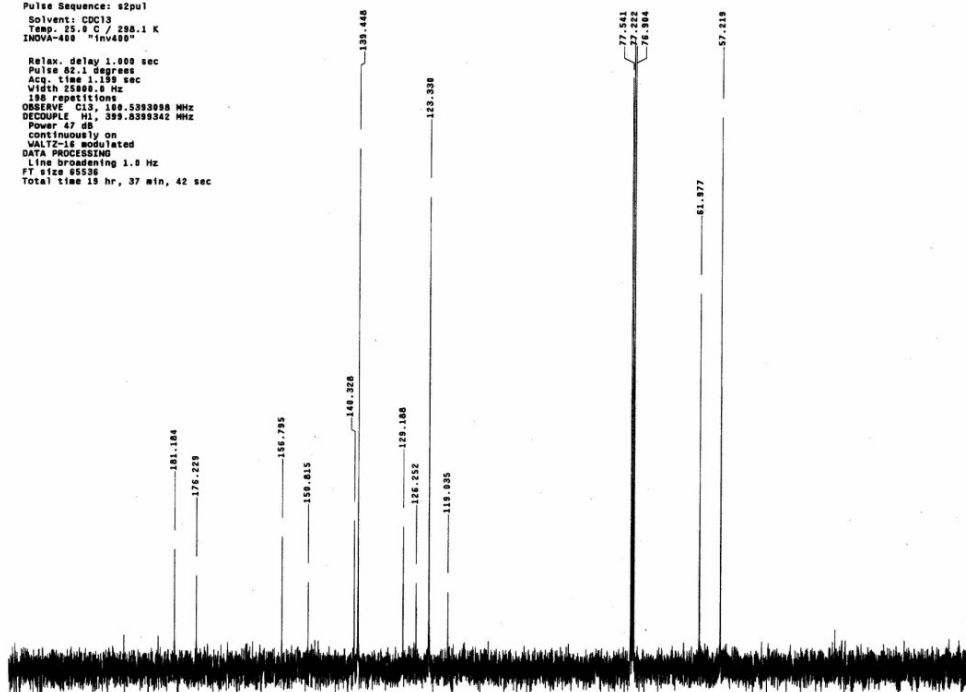






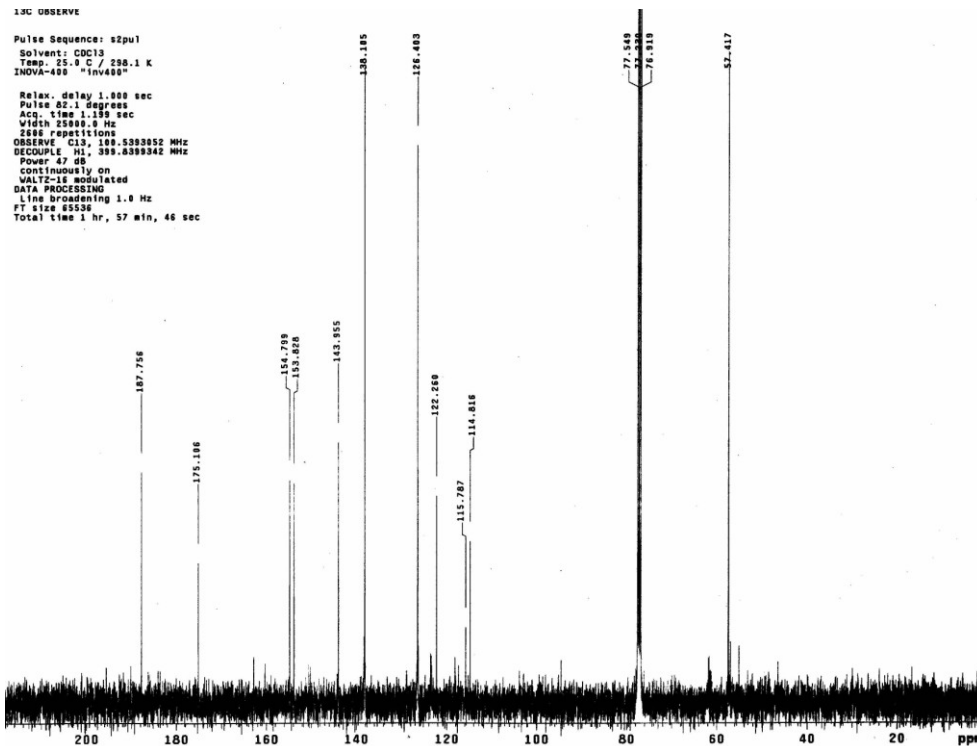


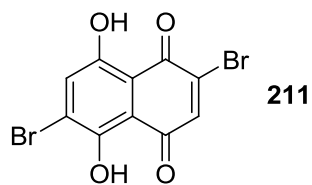
wp-v-51-oxidation-c13
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.1 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.199 sec
 Width 25800.0 Hz
 130 Repetitions
 OBSERVE C13, 100.6393098 MHz
 DECOUPLE H1, 399.8393042 MHz
 Power 47 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 19 hr, 37 min, 42 sec



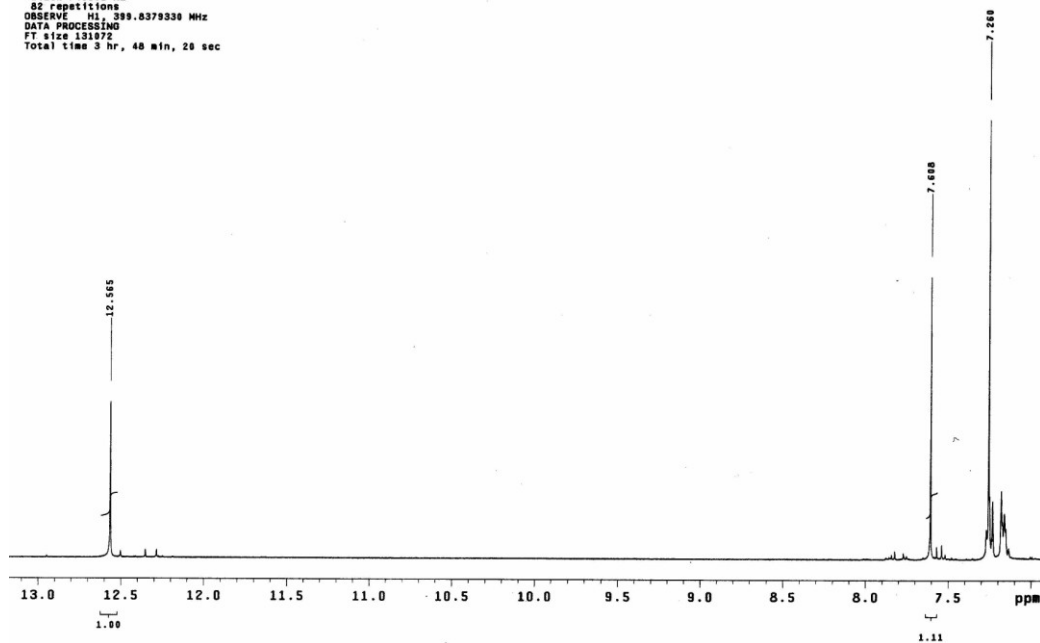
13C OBSERVE

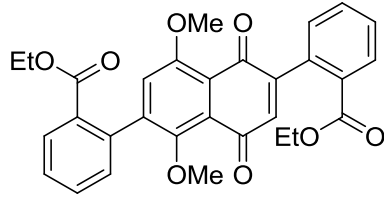
Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.199 sec
 Width 25800.0 Hz
 2686 repetitions
 OBSERVE C13, 100.6393052 MHz
 DECOUPLE H1, 399.8393042 MHz
 Power 47 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 1 hr, 57 min, 46 sec





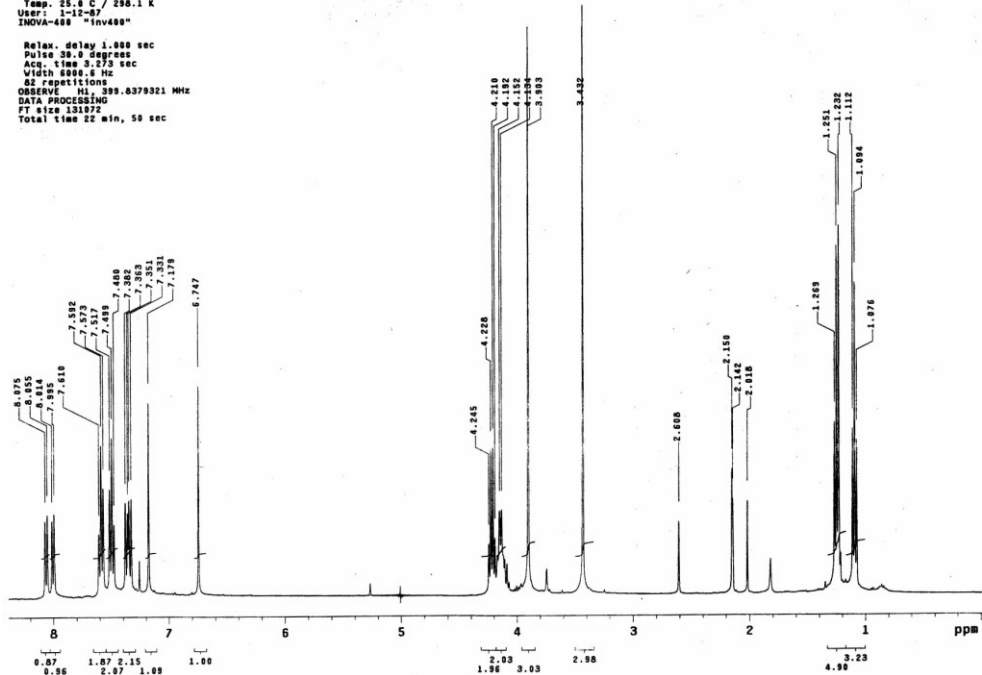
wp-v-45-demethylation
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-87
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 30.0 degrees
 Acq. time 3.273 sec
 Width 7499.0 Hz
 65 repetitions
 OBSERVE H1, 399.6379330 MHz
 DATA PROCESSING
 FT size 131972
 Total time 3 hr, 48 min, 20 sec



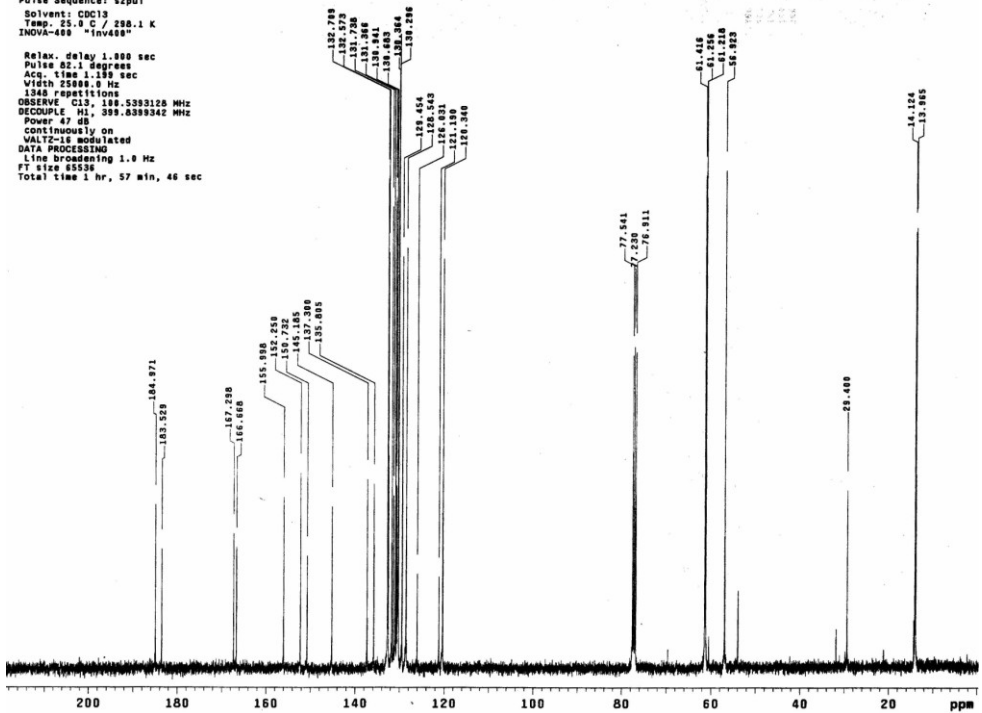


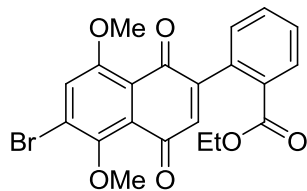
213

wp-v-56-borane-coupling-spot2
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-87
 INOVA-400 "Inv400"
 Relax. delay 1.000 sec
 Pulse 50.0 degrees
 Acq. time 3.273 sec
 Width 6000.0 Hz
 62 repetitions
 OBSERVE H1: 399.8379321 MHz
 DATA PROCESSING
 FT size 131972
 Total time 22 min, 59 sec



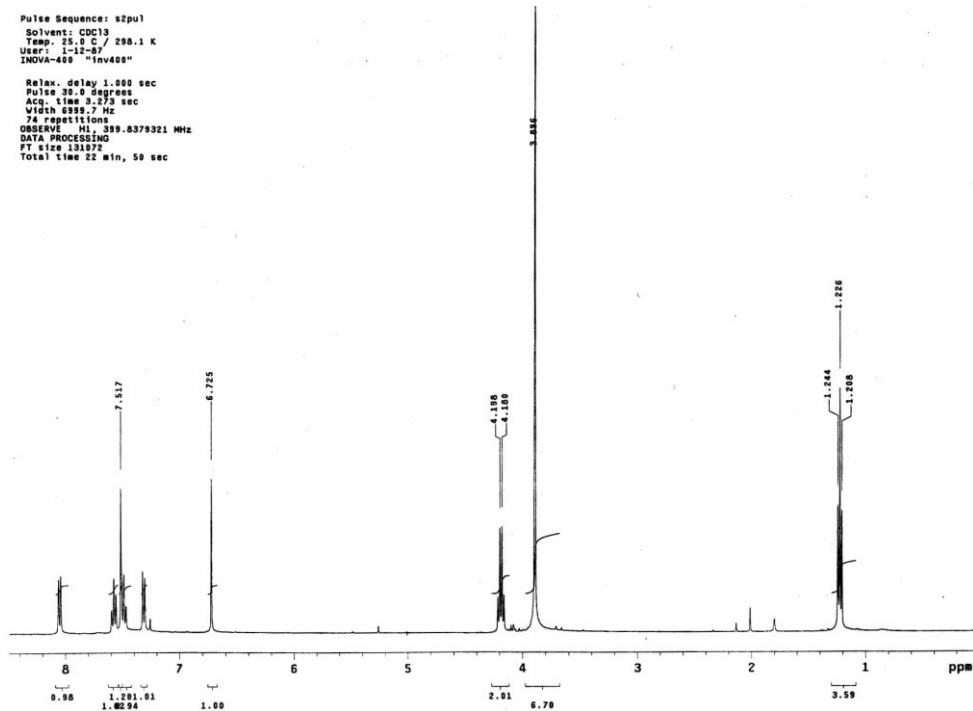
wp-v-56-borane-coupling-spot2-c13
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-400 "Inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.189 sec
 Width 25000.0 Hz
 1340 repetitions
 OBSERVE C13: 100.5393126 MHz
 DECOUPLE H1: 399.8393342 MHz
 Power 47 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 45538
 Total time 1 hr, 57 min, 46 sec



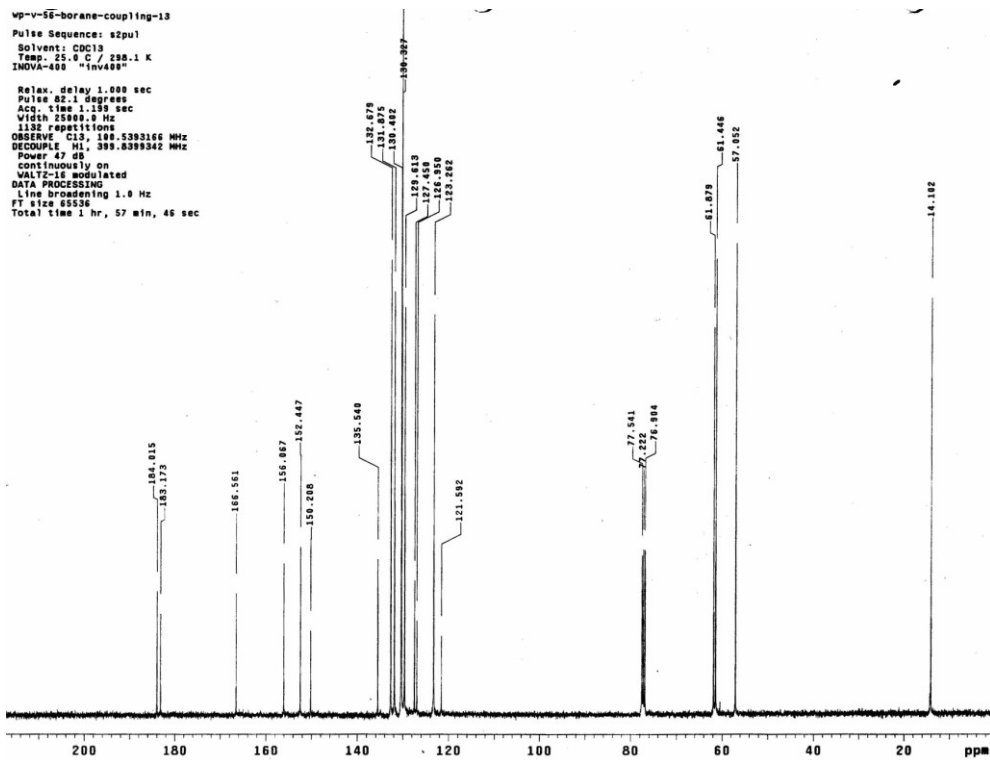


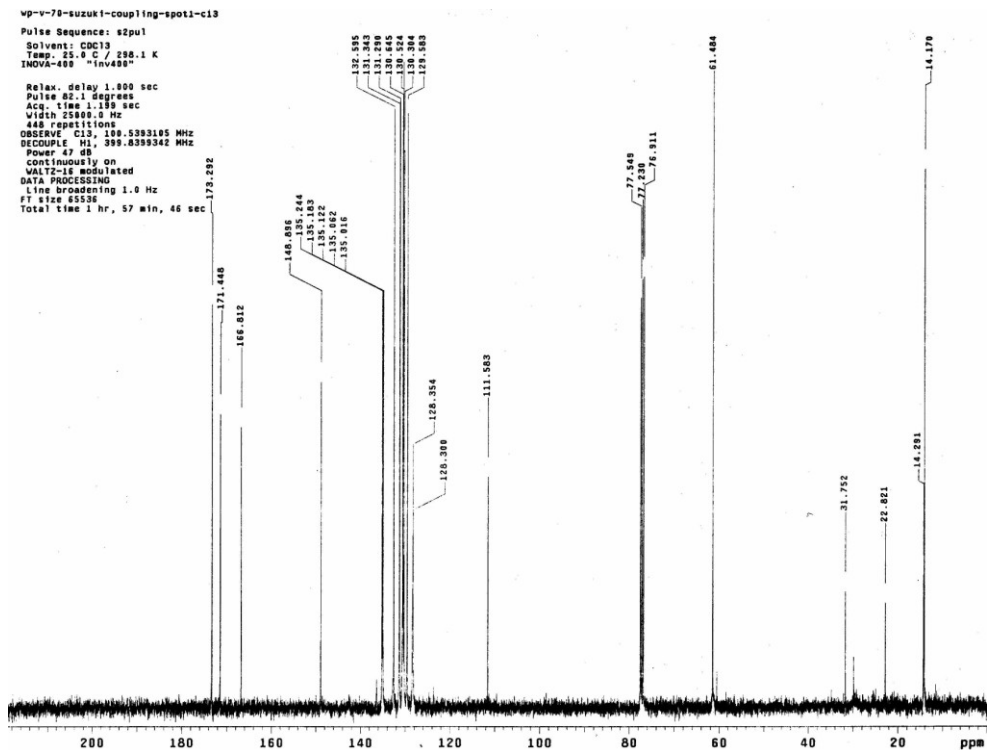
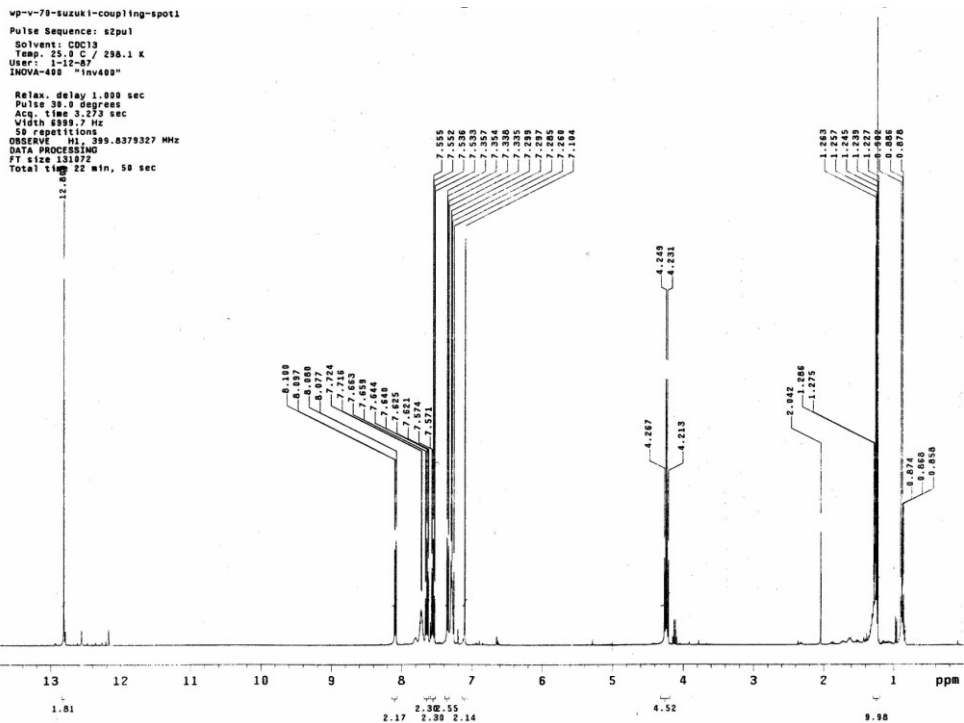
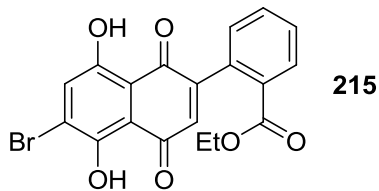
214

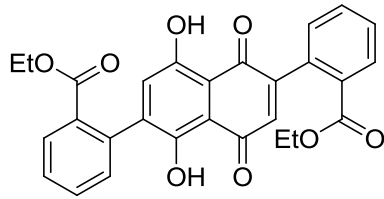
Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-97
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 39.0 degrees
 Acq. time 9.279 sec
 Width 6999.7 Hz
 74 repetitions
 OBSERVE H1, 399.8379321 MHz
 DATA PROCESSING
 FT size 133872
 Total time 22 min, 50 sec



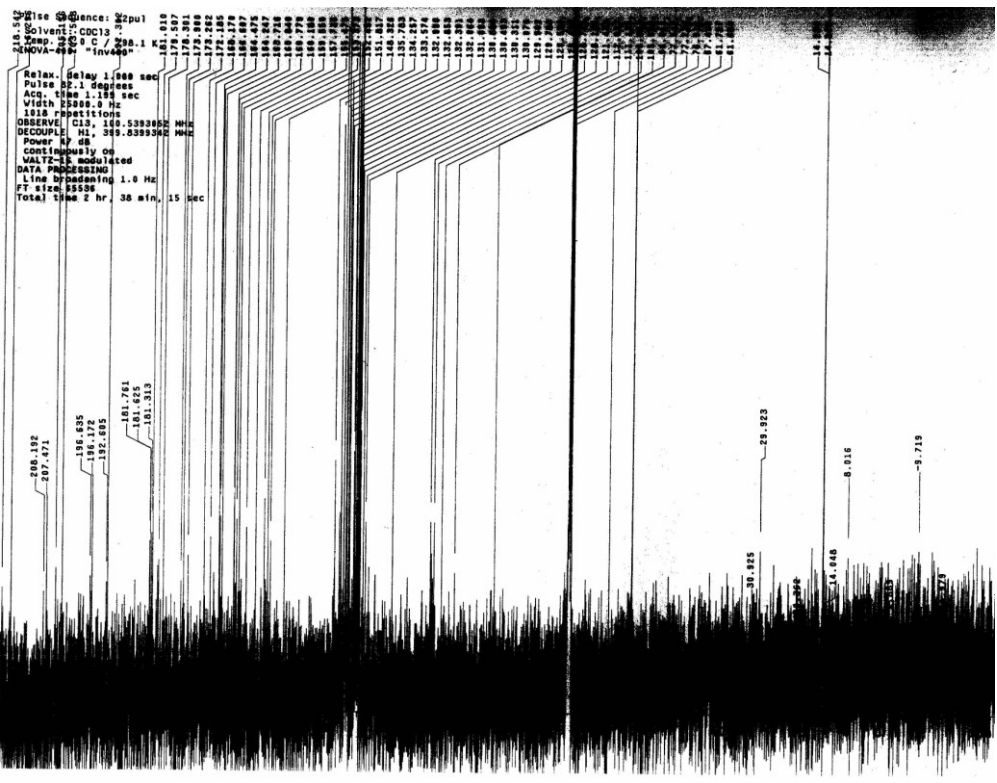
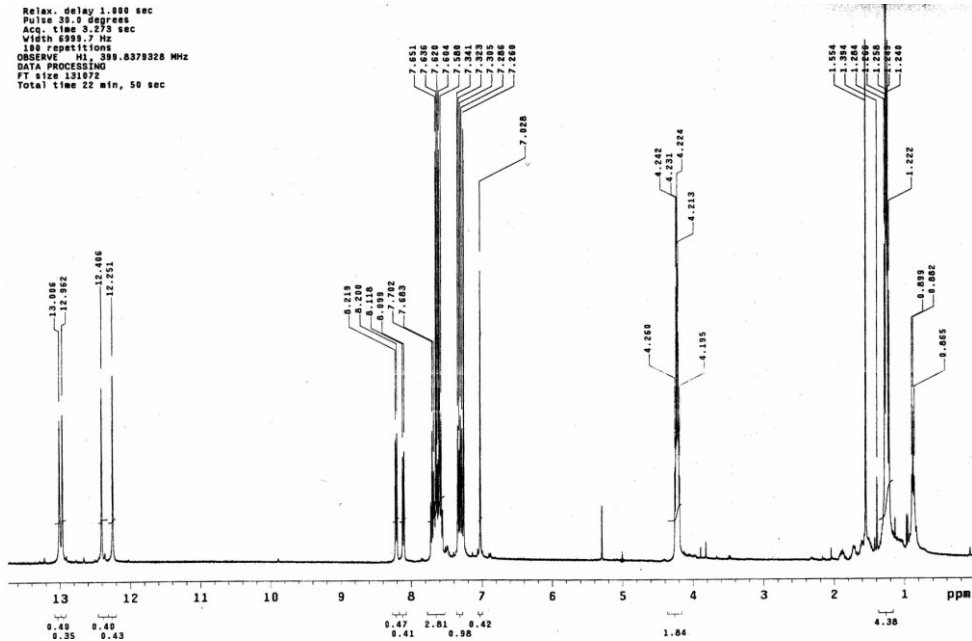
WP-V-56-borane-coupling-13
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 92.1 degrees
 Acq. time 1.139 sec
 Width 25809.0 Hz
 1132 repetitions
 OBSERVE C13, 100.5393166 MHz
 DECOUPLE H1, 399.8399342 MHz
 Power 47 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 1 hr, 57 min, 46 sec

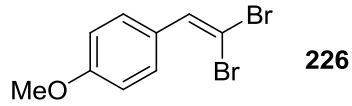




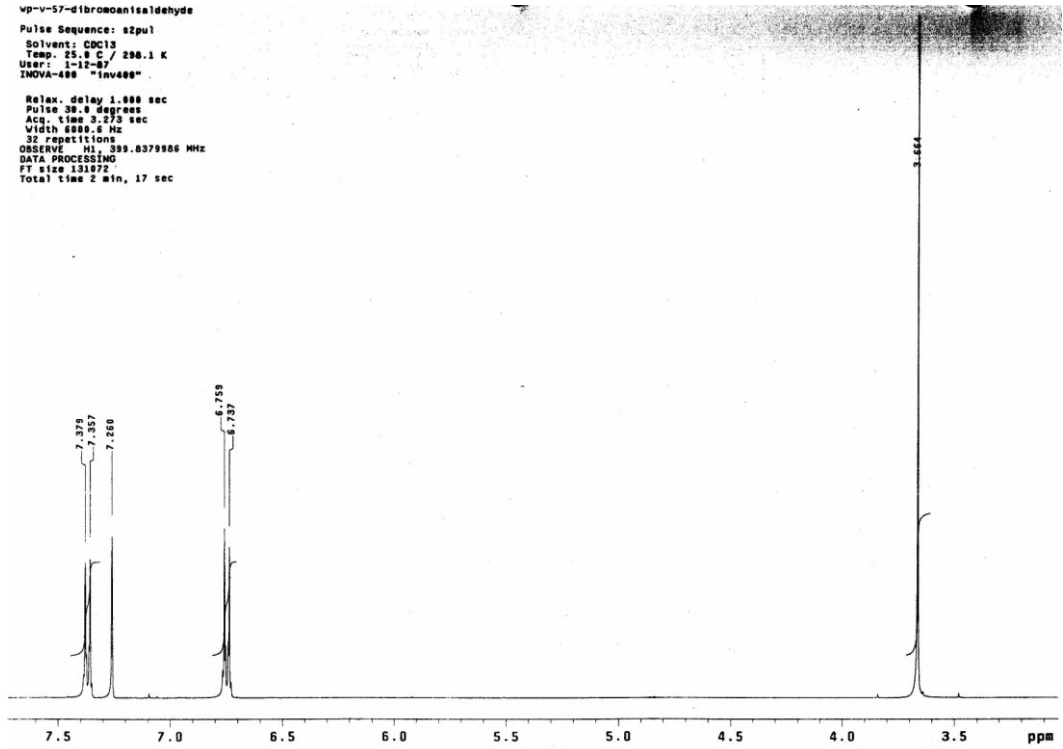


216

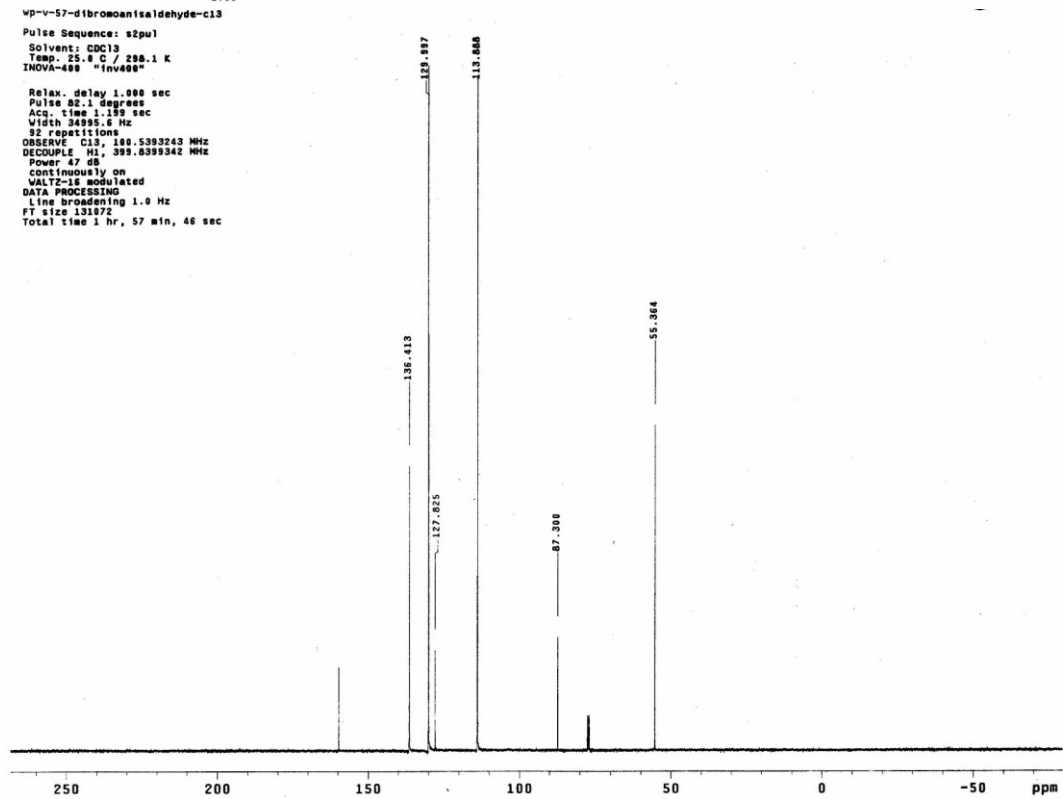


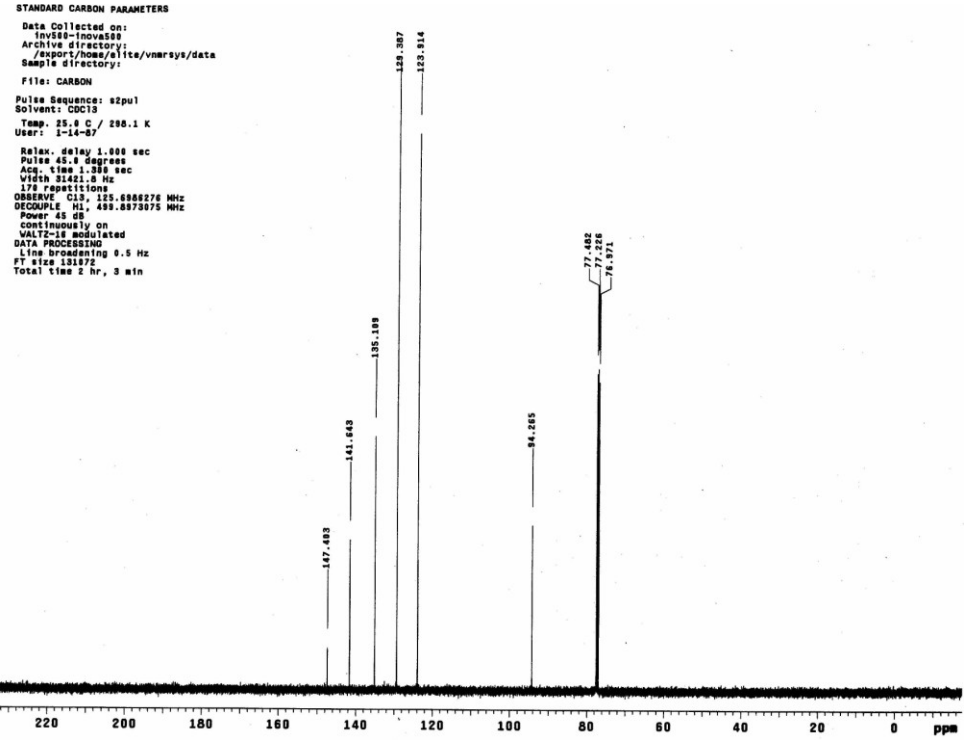
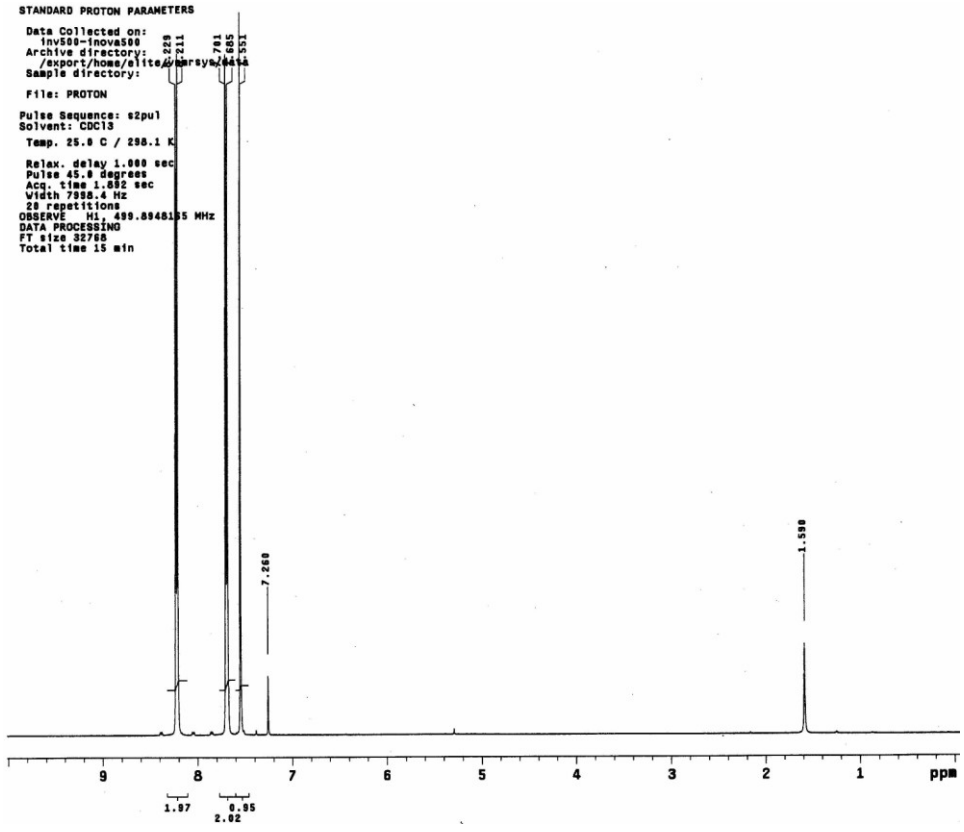
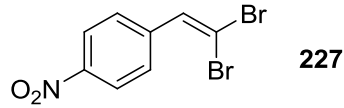


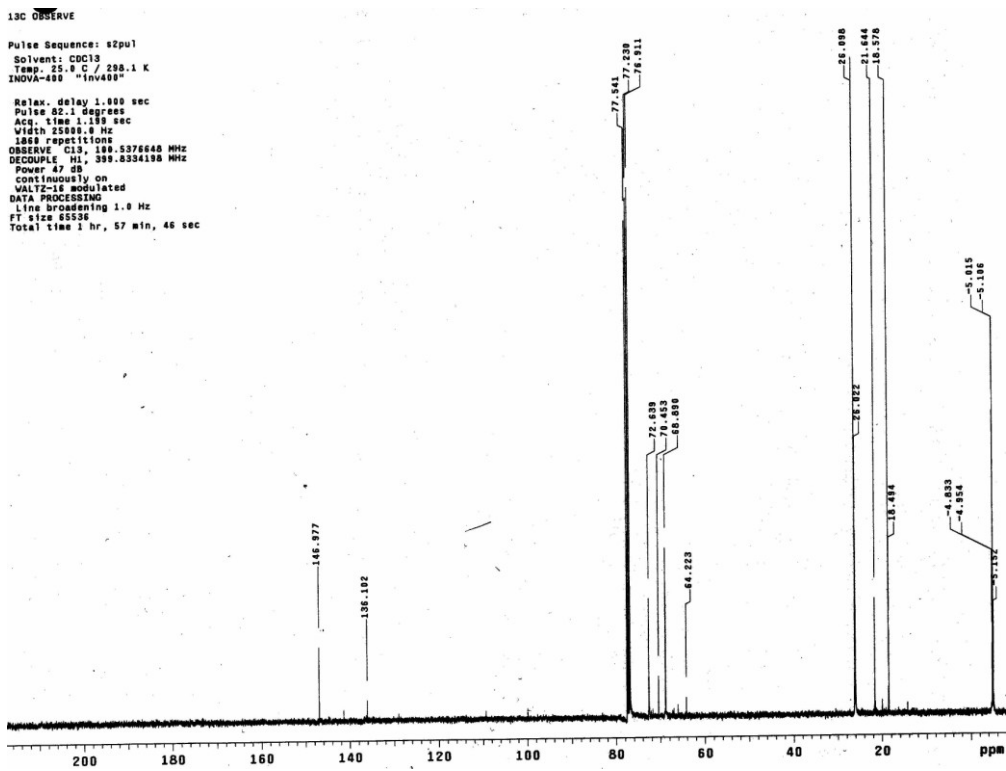
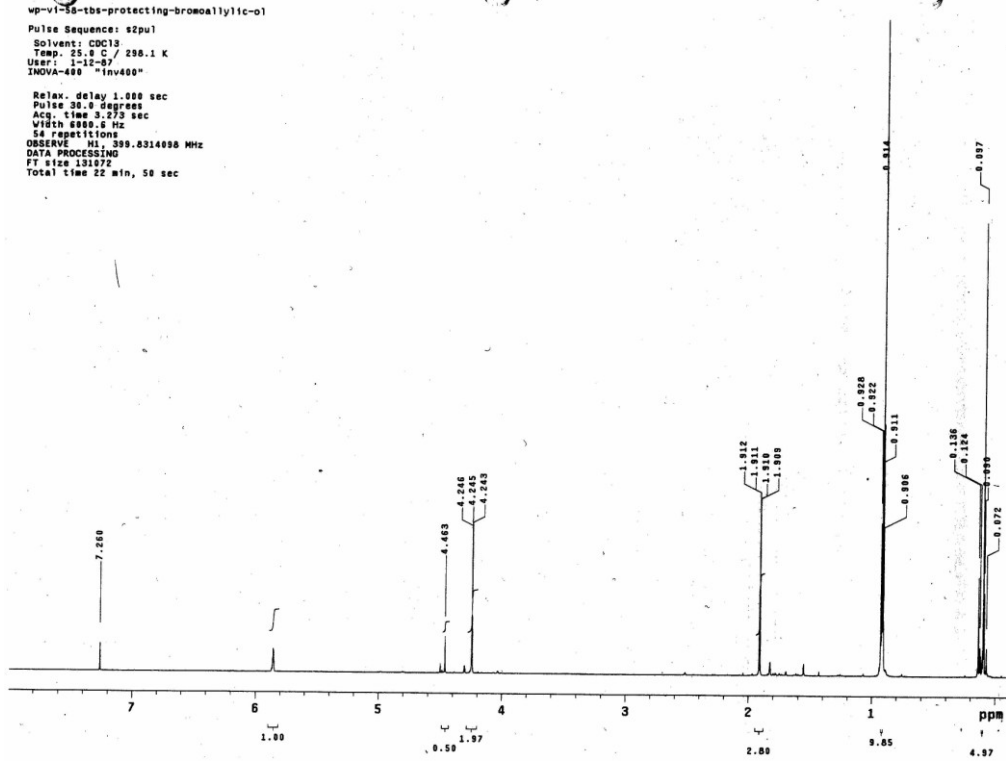
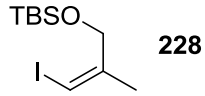
wp-v-57-dibromoanisaldehyde
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: 1-12-87
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 30.0 degrees
 Acq. time 3.273 sec
 Width 6000.0 Hz
 32 repetitions
 OBSERVE H1, 399.8379986 MHz
 DATA PROCESSING
 FT size 131072
 Total time 2 min, 17 sec

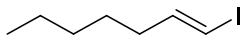


wp-v-57-dibromoanisaldehyde-c13
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 32.1 degrees
 Acq. time 1.199 sec
 Width 34995.0 Hz
 32 repetitions
 OBSERVE C13, 100.5393243 MHz
 DECOUPLE H1, 399.8393342 MHz
 Power 47.00
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 131072
 Total time 1 hr, 57 min, 46 sec









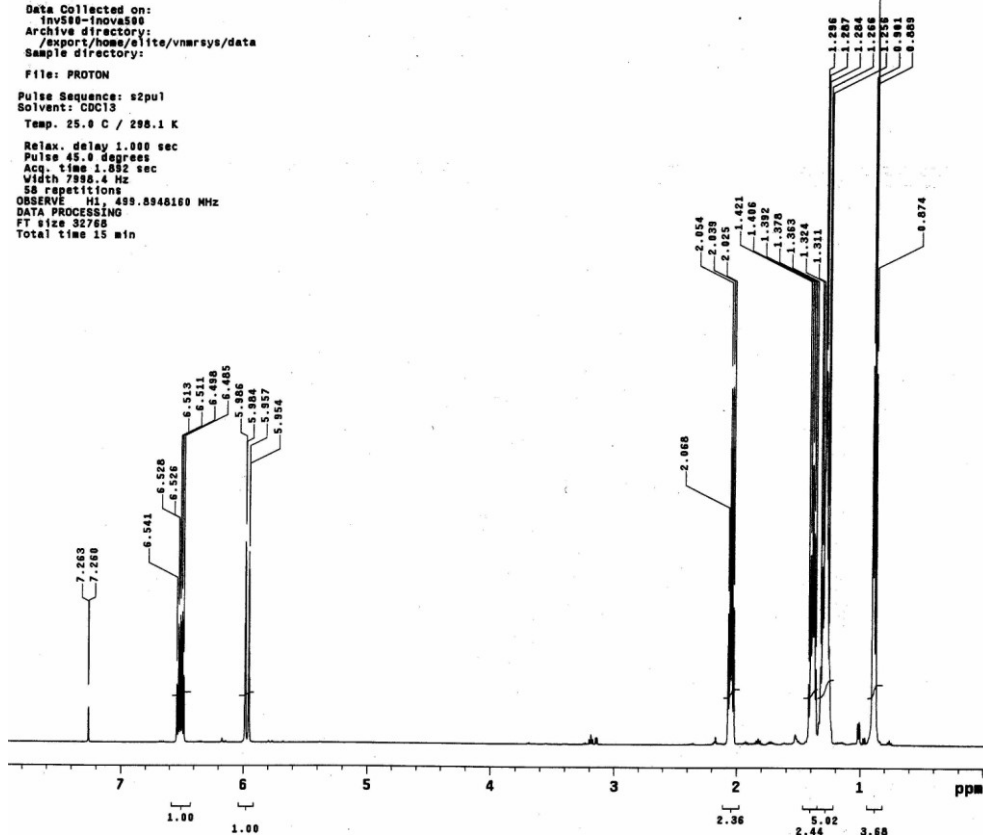
230

STANDARD PROTON PARAMETERS

Data Collected on: inv500-inoxa500
 Archive directory: /export/home/elite/vnmrsys/data
 Sample directory:

File: PROTON
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K

Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.052 sec
 Width 7888.4 Hz
 58 repetitions
 OBSERVE H1, 499.8948160 MHz
 DATA PROCESSING
 FT size 32768
 Total time 15 min

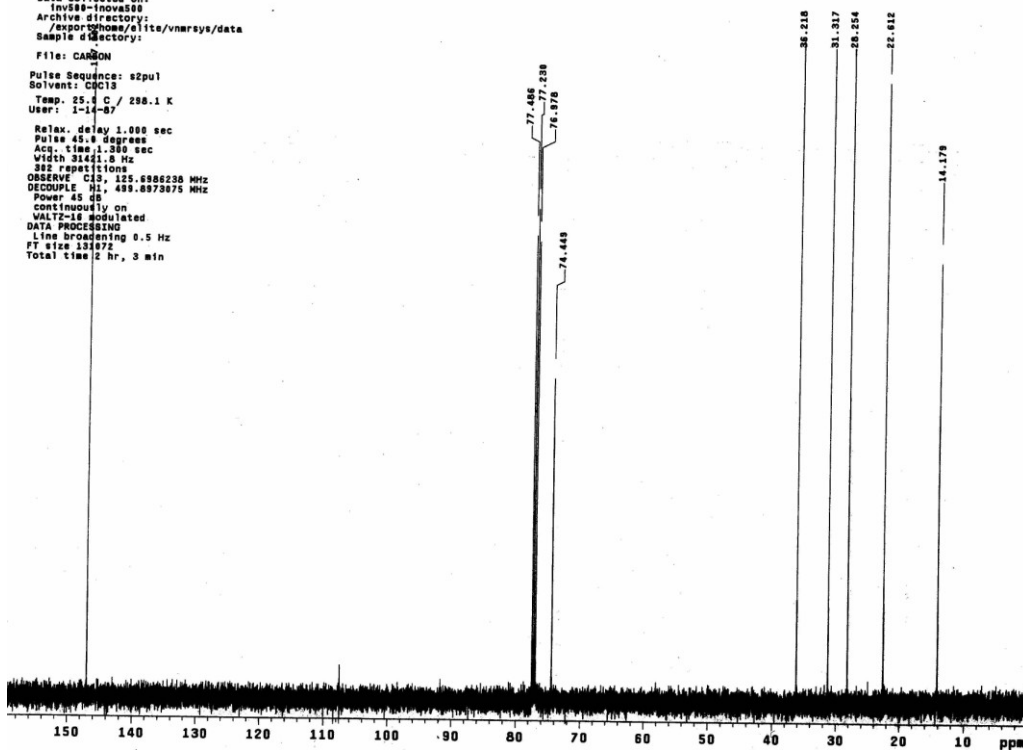


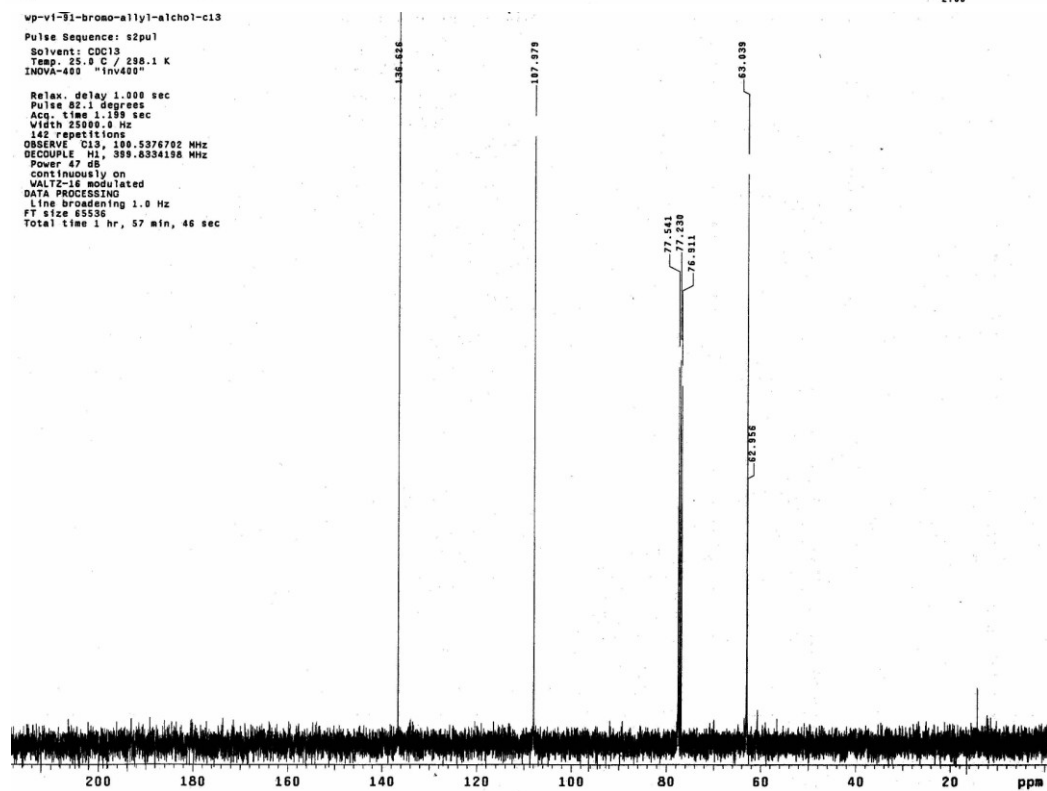
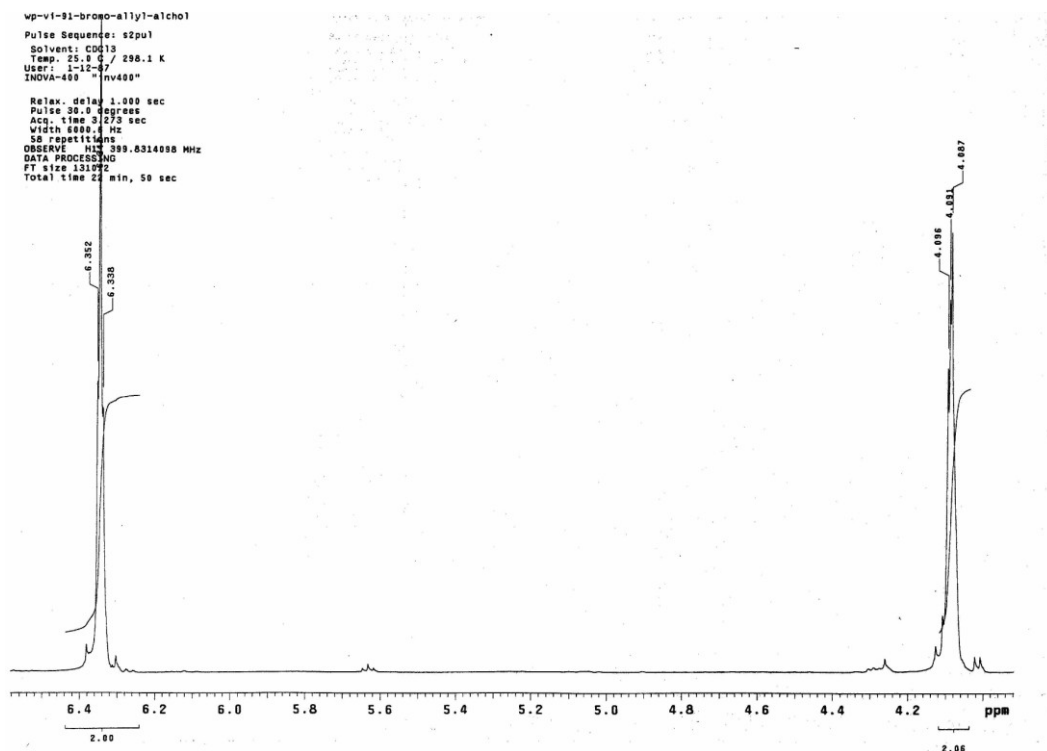
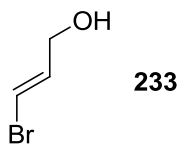
STANDARD CARBON PARAMETERS

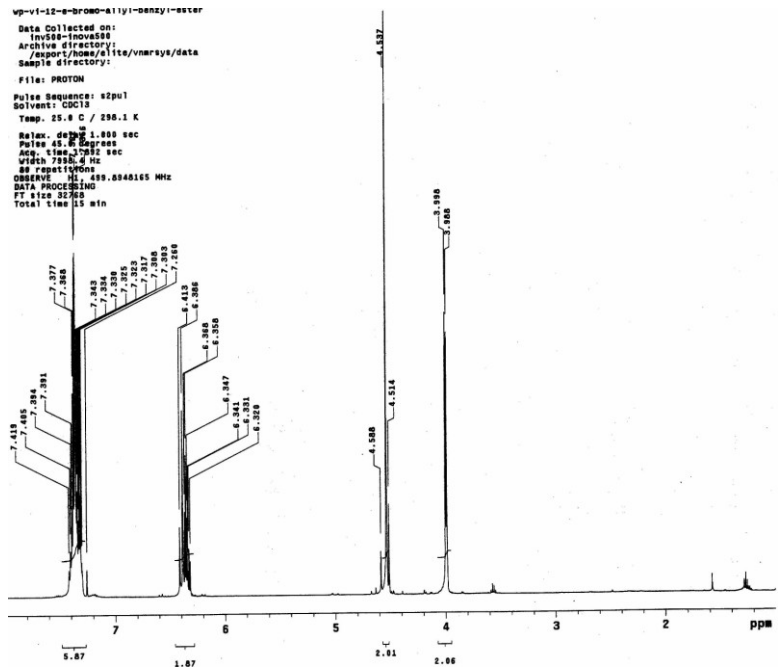
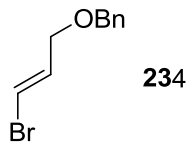
Data Collected on: inv125-inoxa500
 Archive directory: /export/home/elite/vnmrsys/data
 Sample directory:

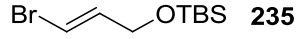
File: CARBON
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 User: i-14-07

Relax. delay 1.000 sec
 Pulse 45.0 degree
 Acq. time 1.380 sec
 Width 31421.6 Hz
 382 repetitions
 OBSERVE C13, 125.8982338 MHz
 DECOUPLE H1, 499.8973875 MHz
 Power 45 db
 continuously on
 WALTZ-16 Modulated
 DATA PROCESSING
 Line Broadening 0.5 Hz
 FT size 131072
 Total time 2 hr, 3 min

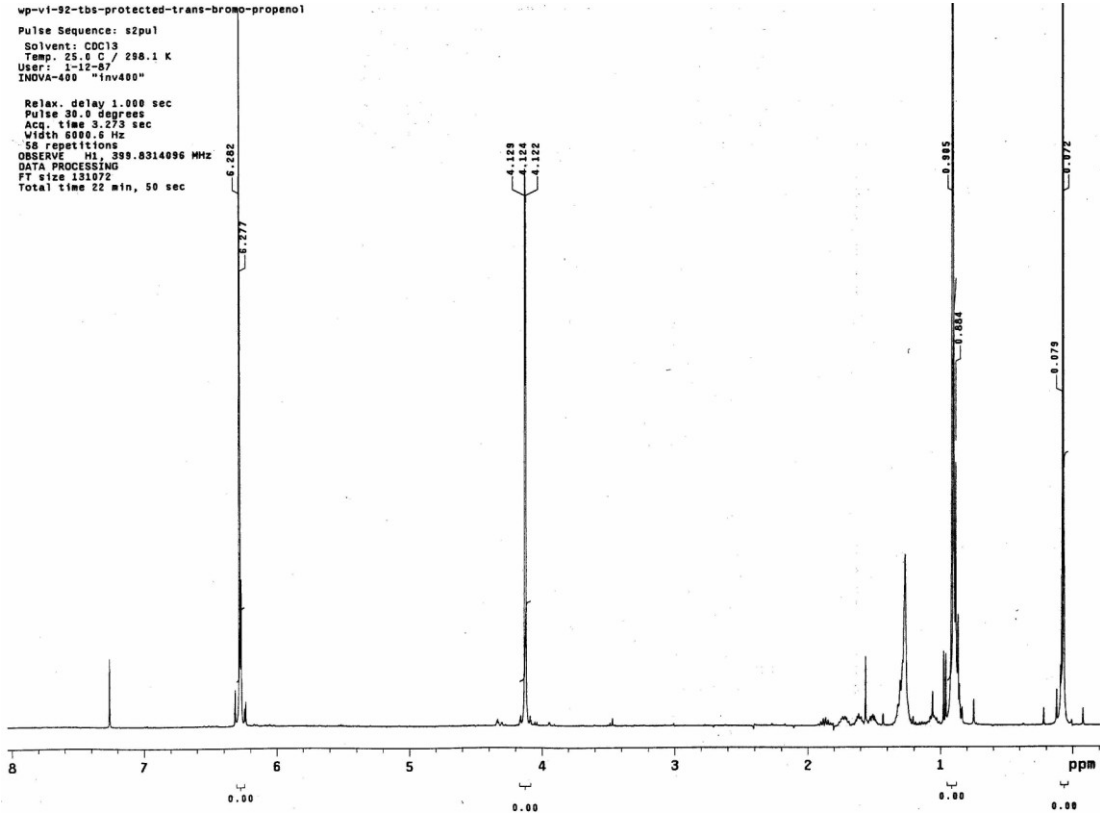






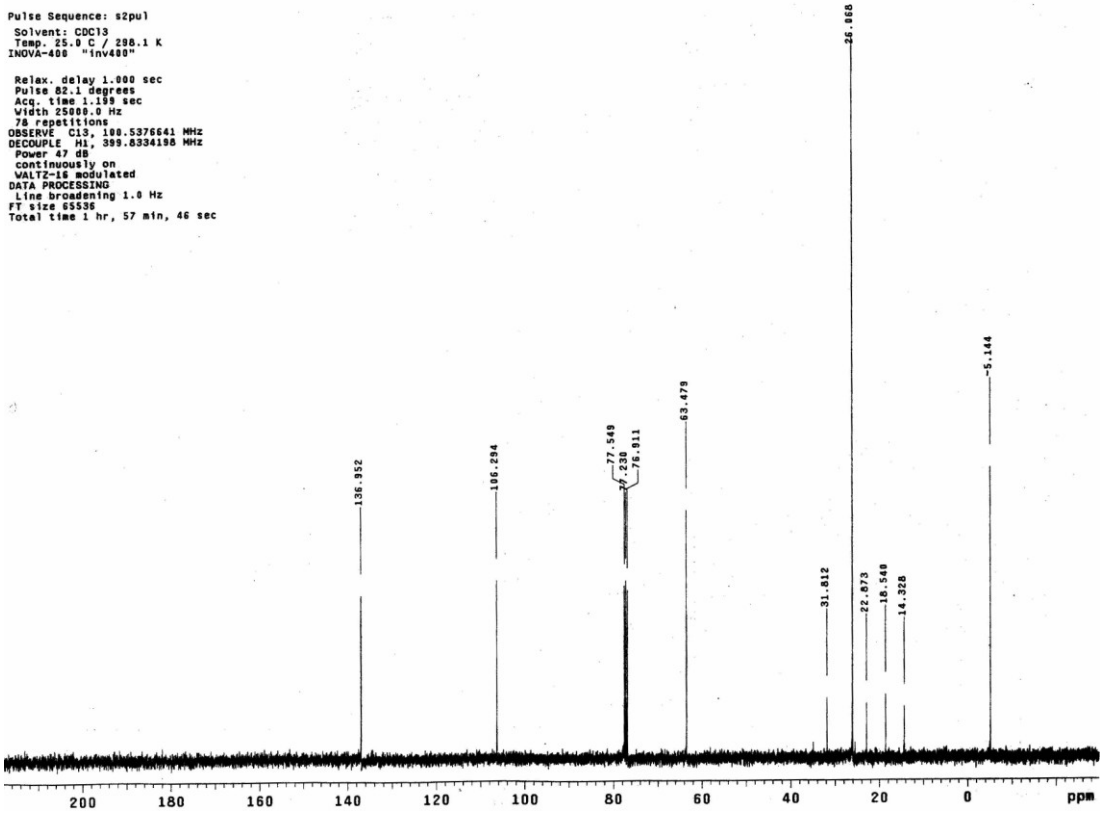


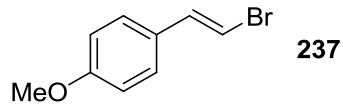
wp-vi-92-tbs-protected-trans-bromo-propenol
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 User: 1-12-87
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 30.0 degrees
 Acq. time 3.273 sec
 Width 6000.6 Hz
 58 repetitions
 OBSERVE H1, 399.8314096 MHz
 DATA PROCESSING
 FT size 131072
 Total time 22 min, 50 sec



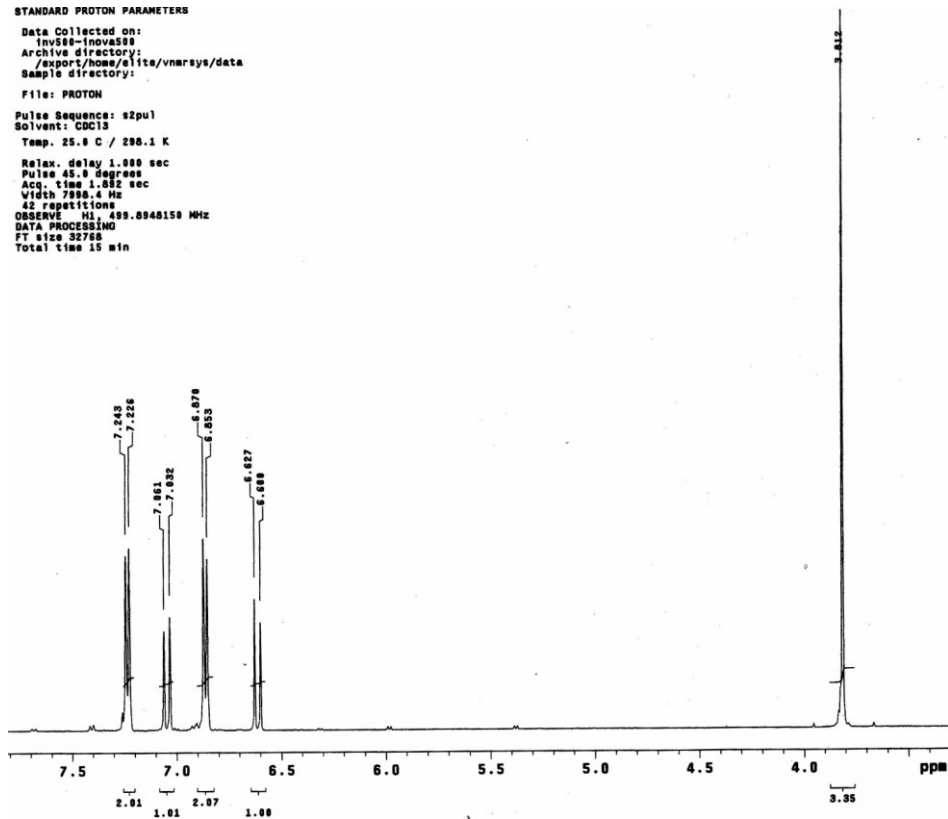
13C OBSERVE

Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 INOVA-400 "inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degrees
 Acq. time 1.199 sec
 Width 25000.0 Hz
 78 repetitions
 OBSERVE C13, 100.5376641 MHz
 DECOUPLE H1, 399.8334188 MHz
 Power 47 dB
 continuously on
 VALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 1 hr, 57 min, 46 sec

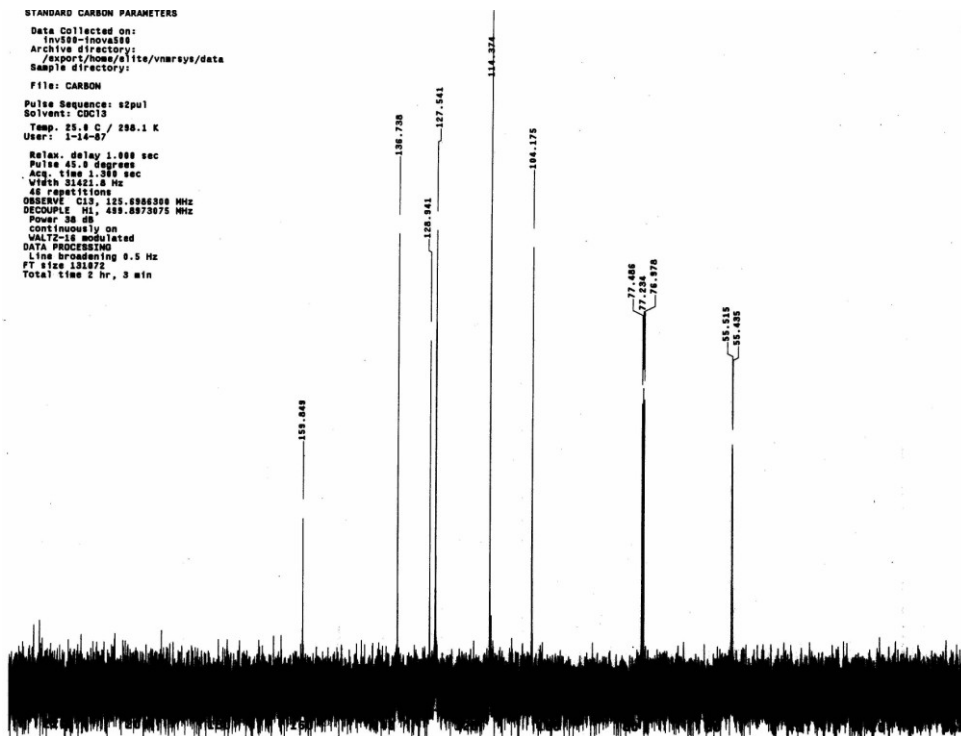


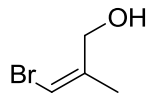


STANDARD PROTON PARAMETERS
 Data Collected on:
 Inv500-Inova500
 Archive directory:
 /export/home/elite/vnmrsys/data
 Sample directory:
 File: PROTON
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.852 sec
 Width 7898.4 Hz
 42 repetitions
 OBSERVE H1, 499.8948150 MHz
 DATA PROCESSING
 FT size 32768
 Total time 15 min

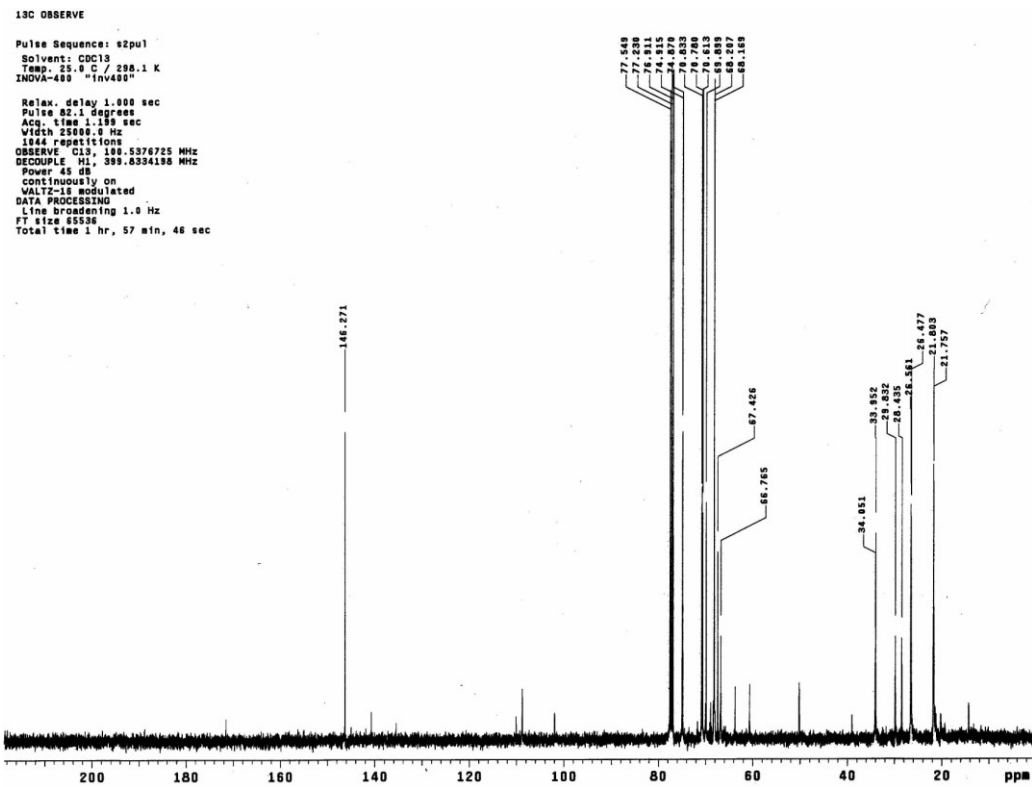
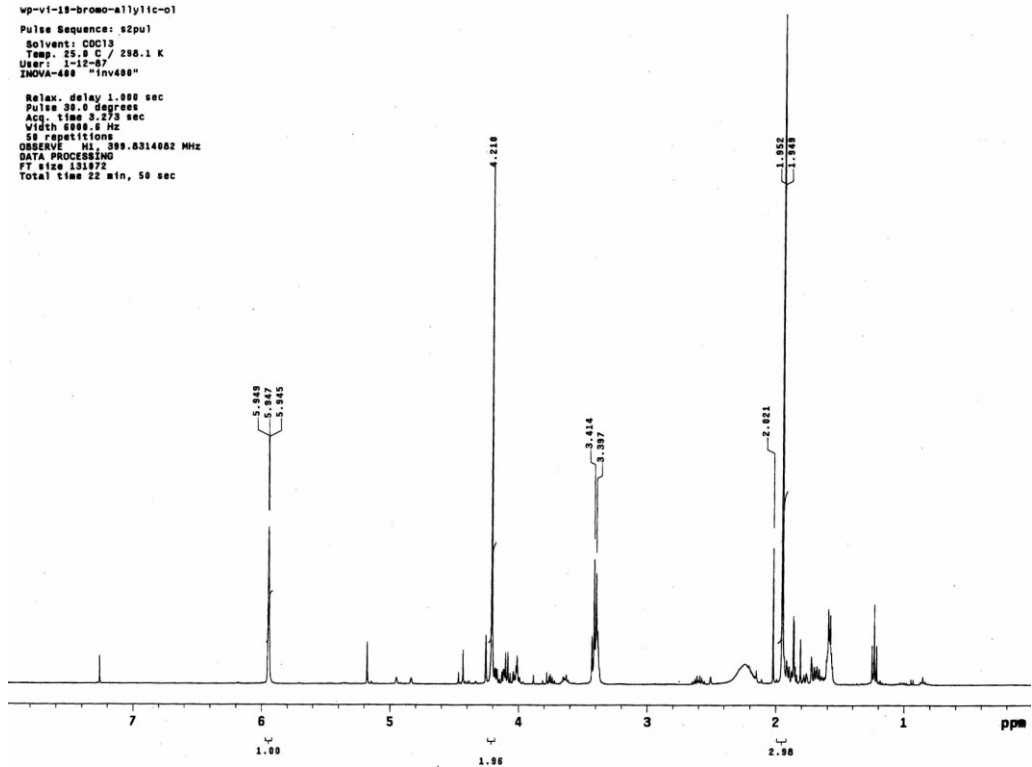


STANDARD CARBON PARAMETERS
 Data Collected on:
 Inv500-Inova500
 Archive directory:
 /export/home/elite/vnmrsys/data
 Sample directory:
 File: CARBON
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 User: 1-14-87
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.385 sec
 Width 31421.8 Hz
 42 repetitions
 OBSERVE C13, 125.6986398 MHz
 DECOUPLE H1, 499.8973075 MHz
 Power 38 dB
 Continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 0.5 Hz
 FT size 131872
 Total time 2 hr, 3 min



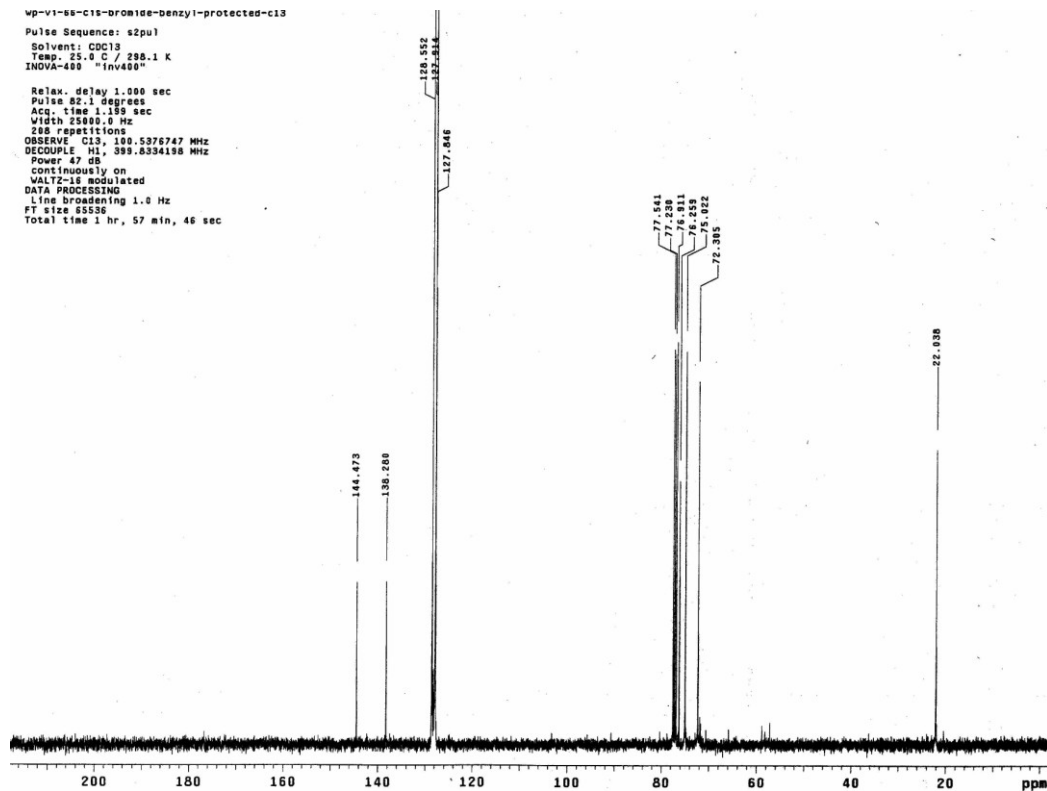
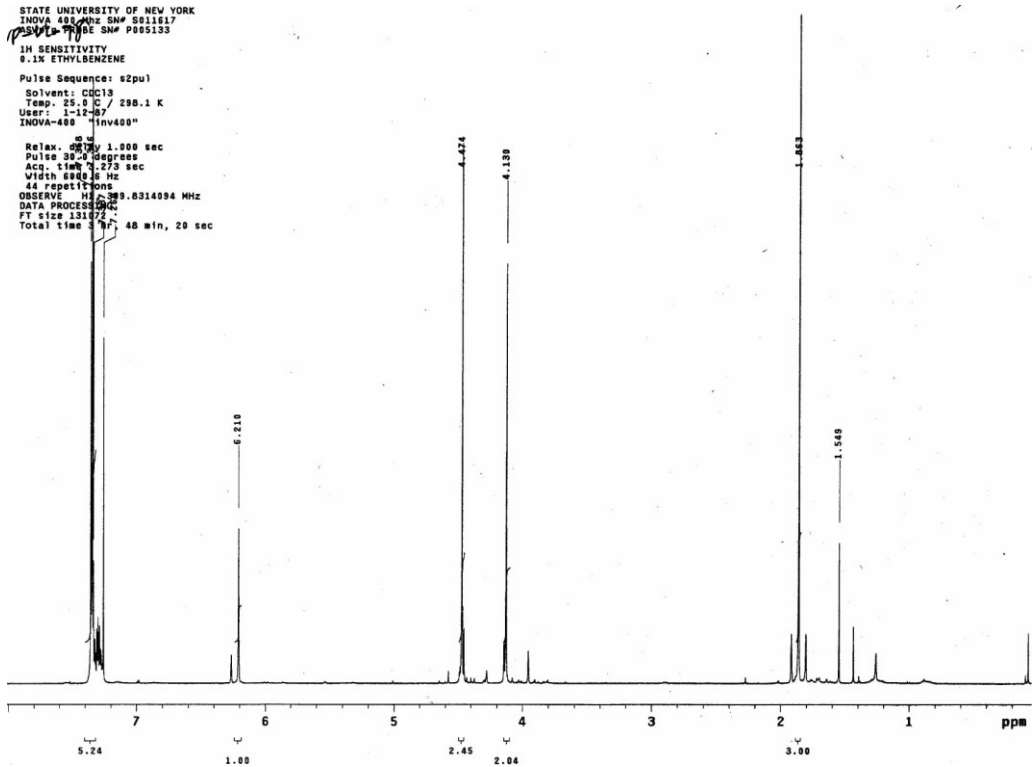


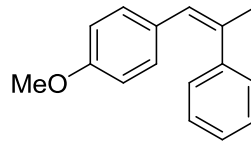
238



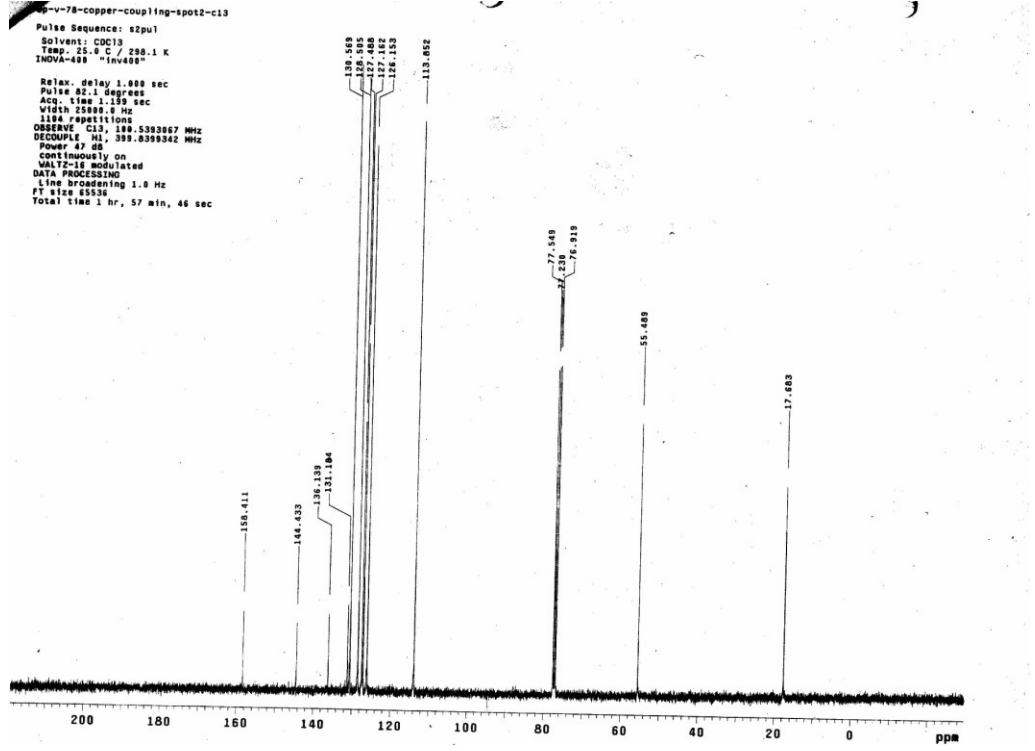
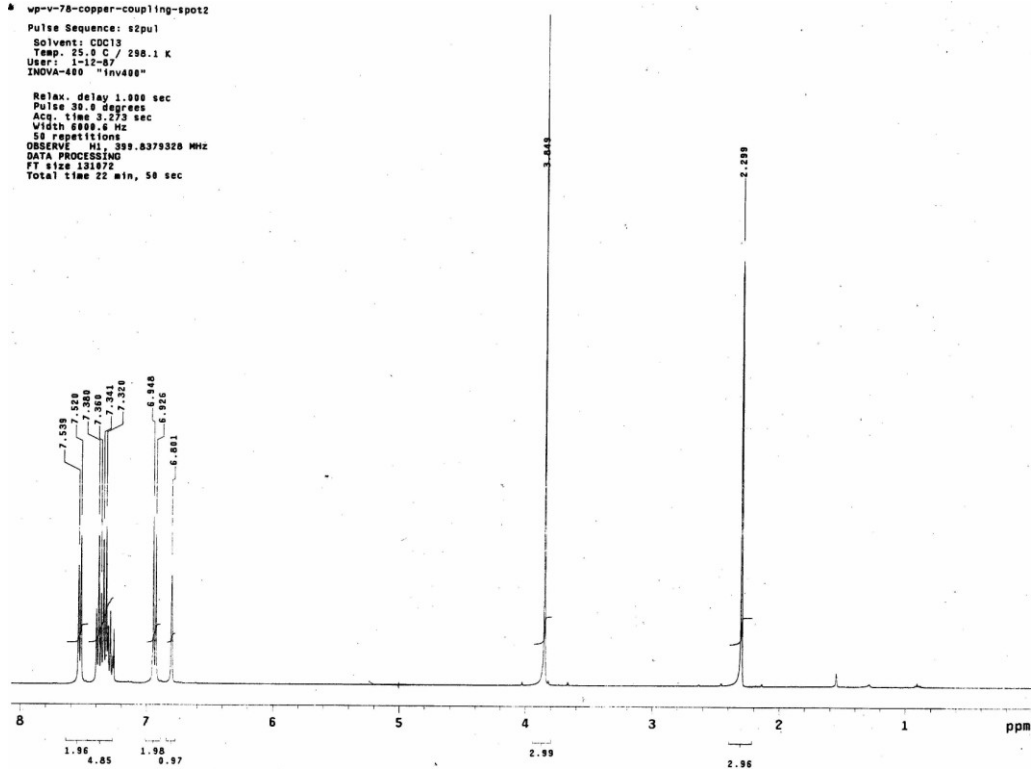


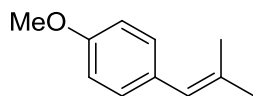
239



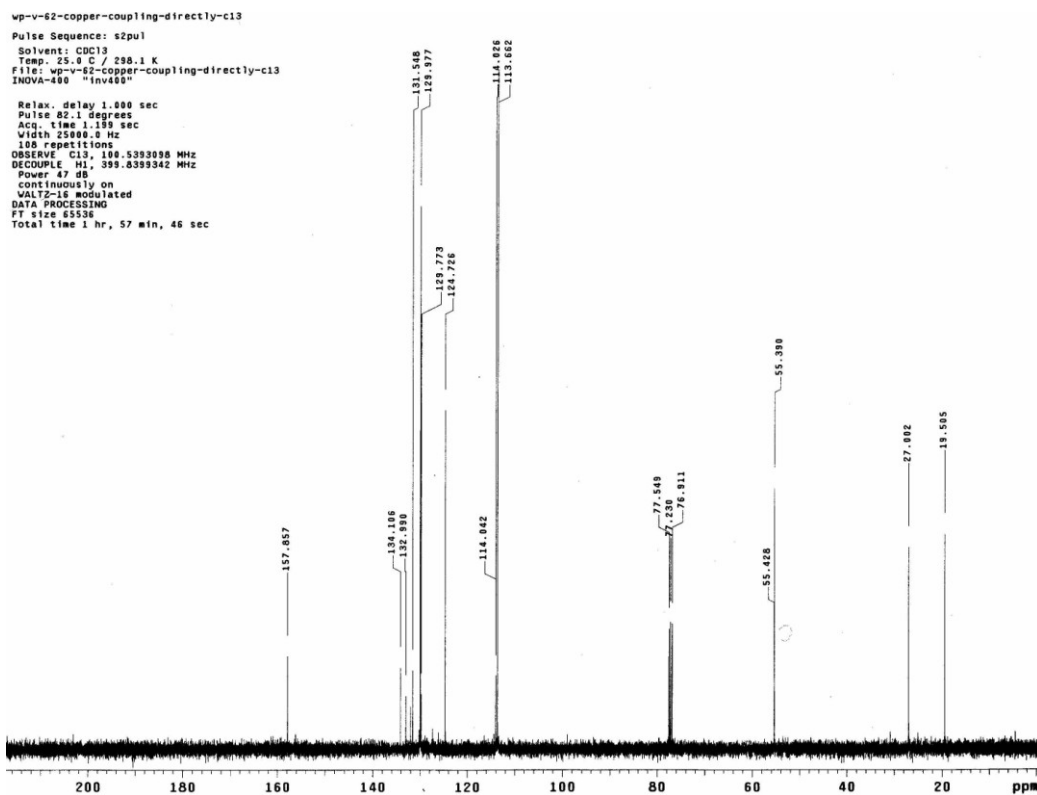
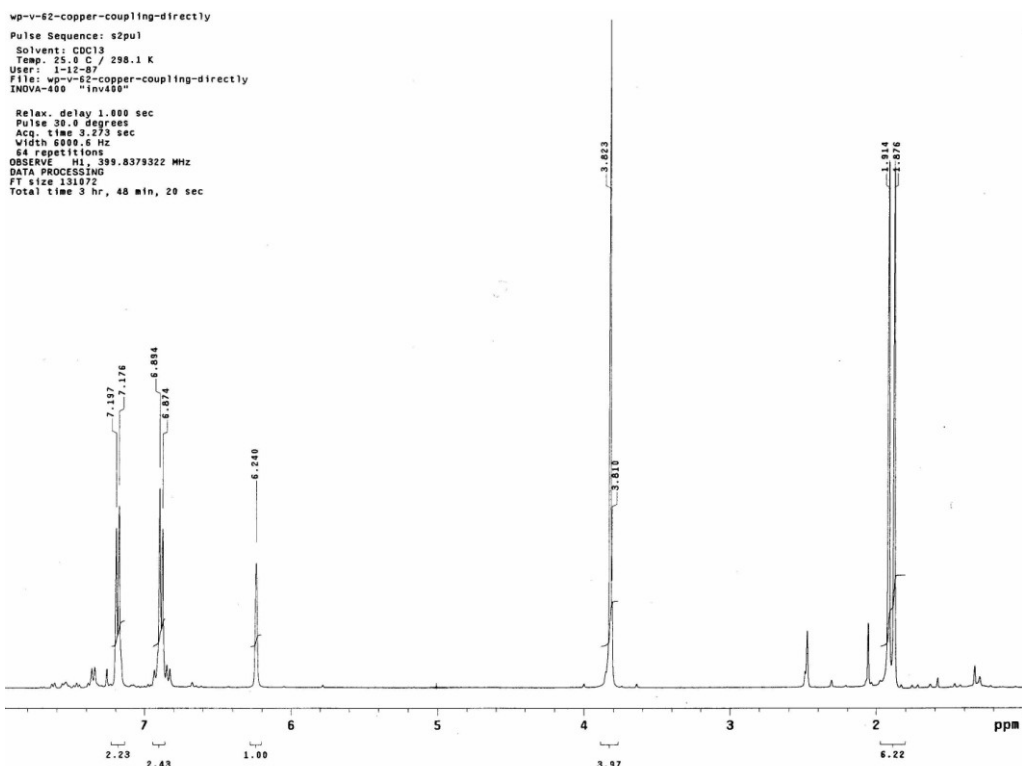


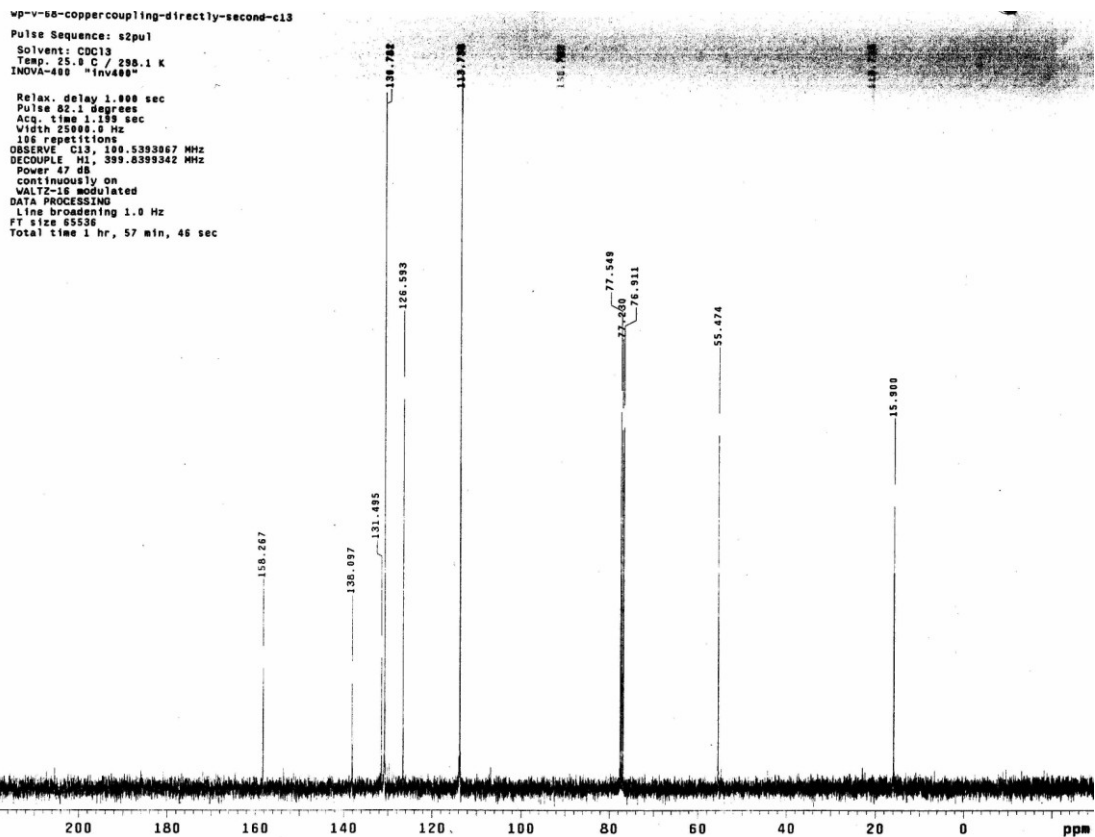
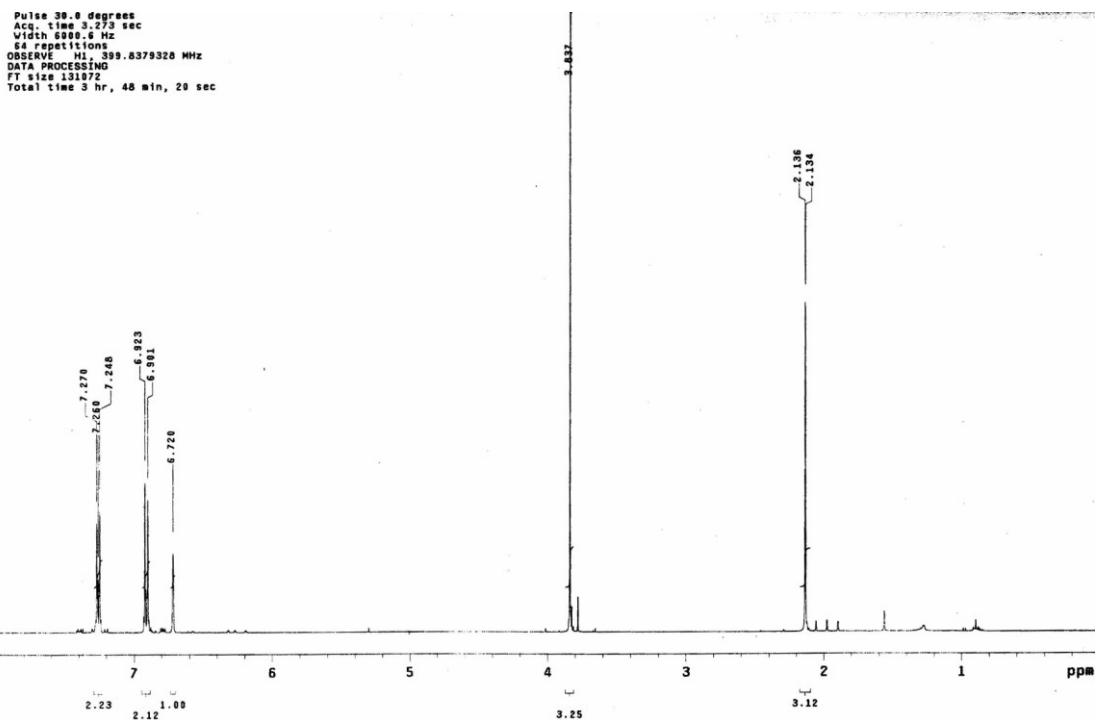
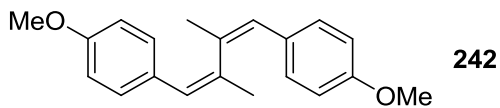
240

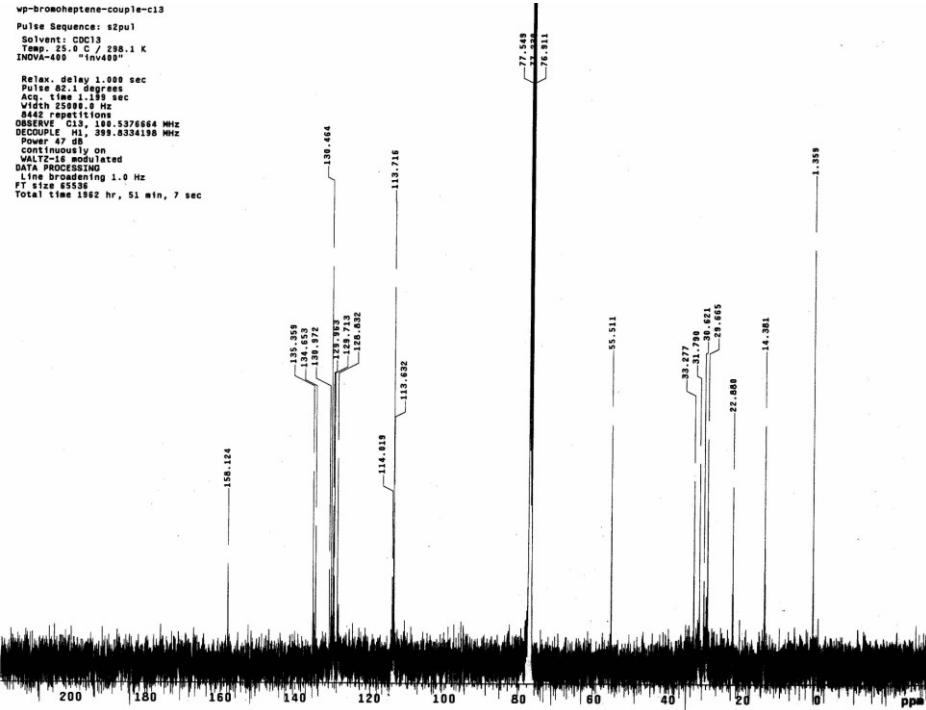
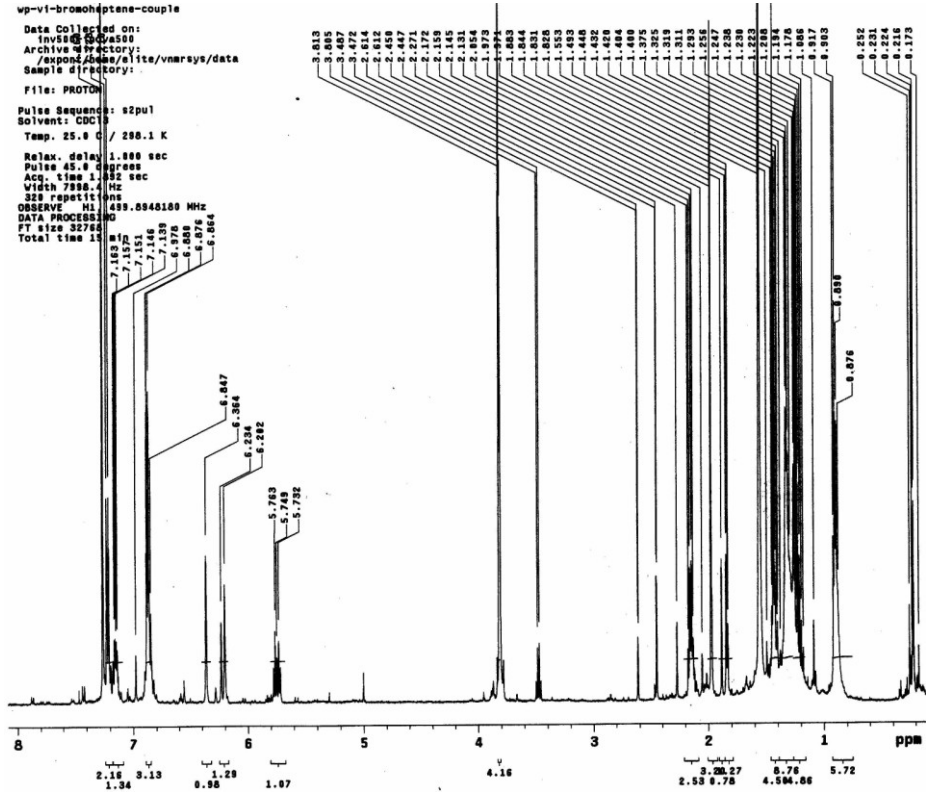
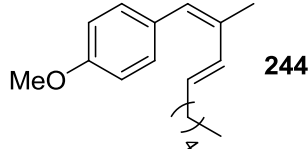


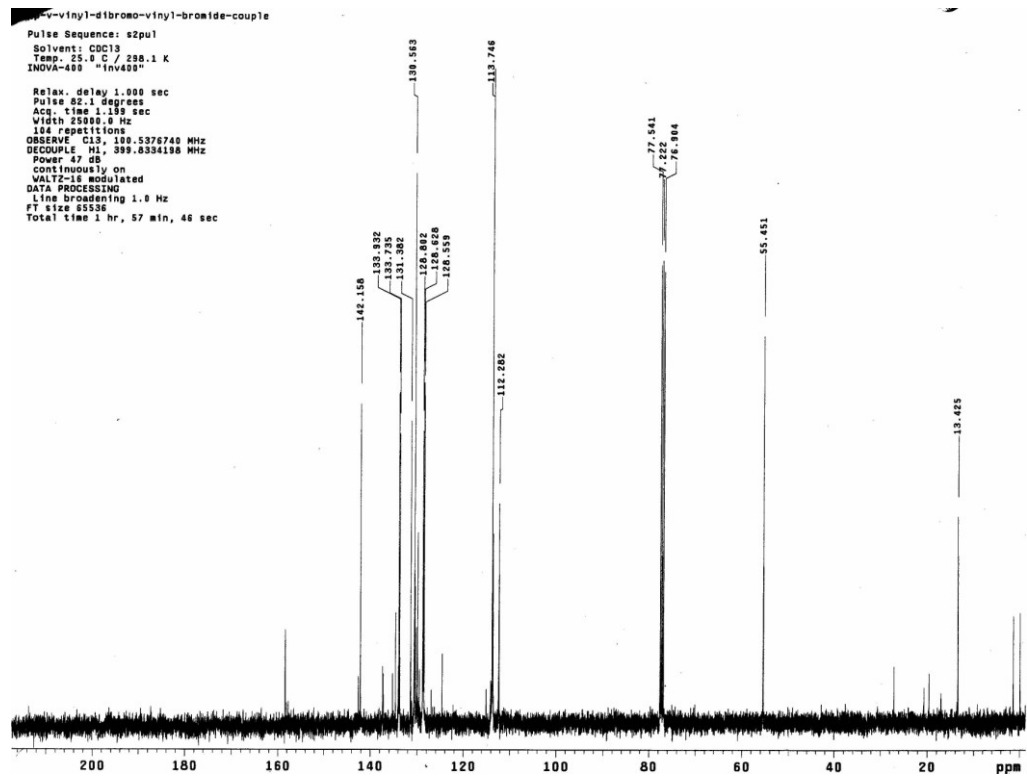
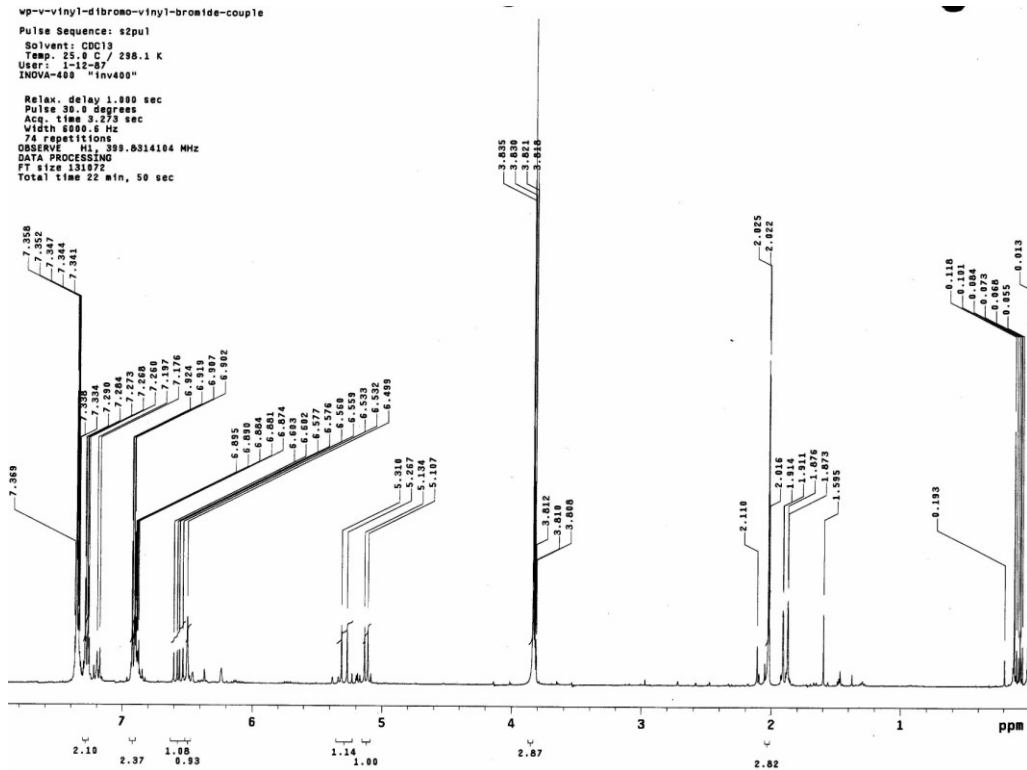
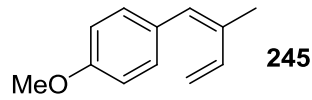


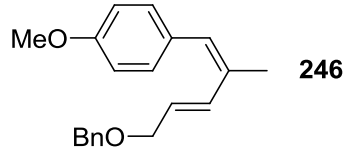
241



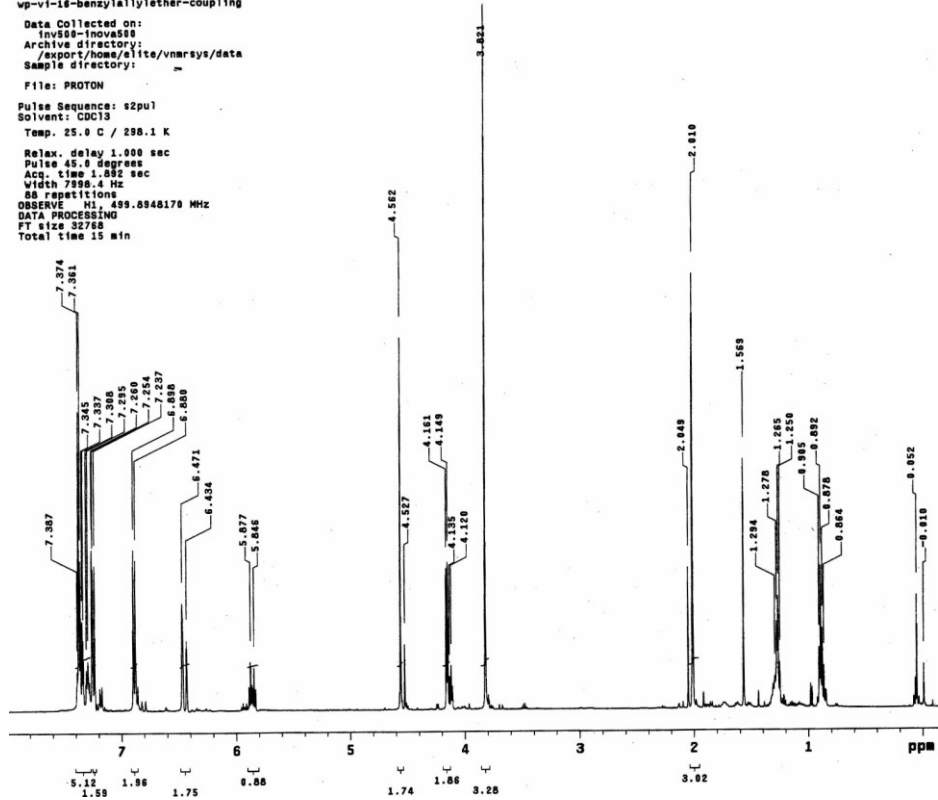




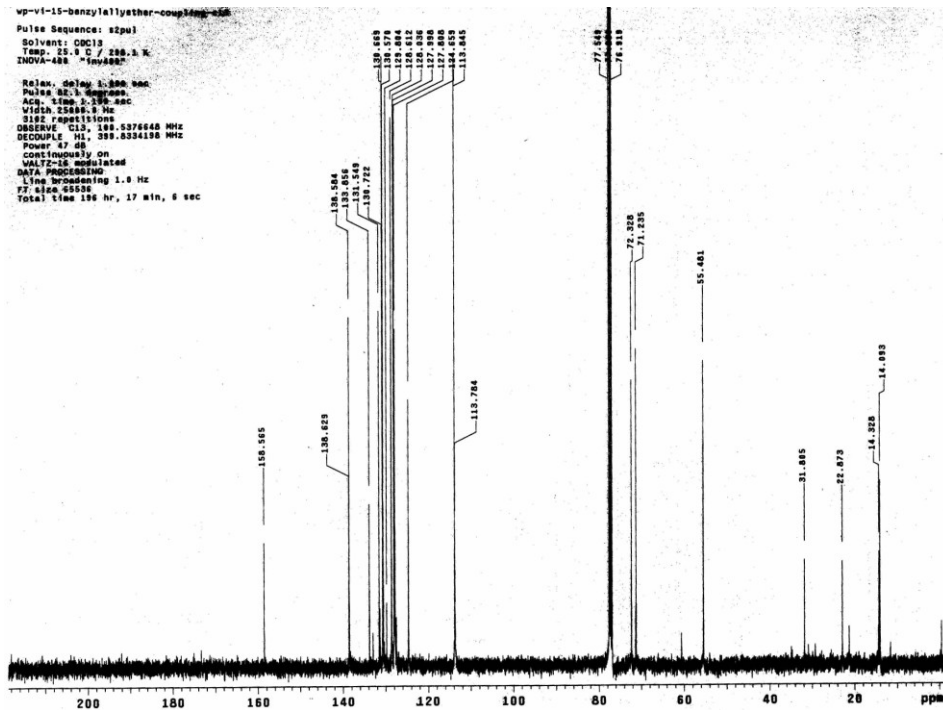


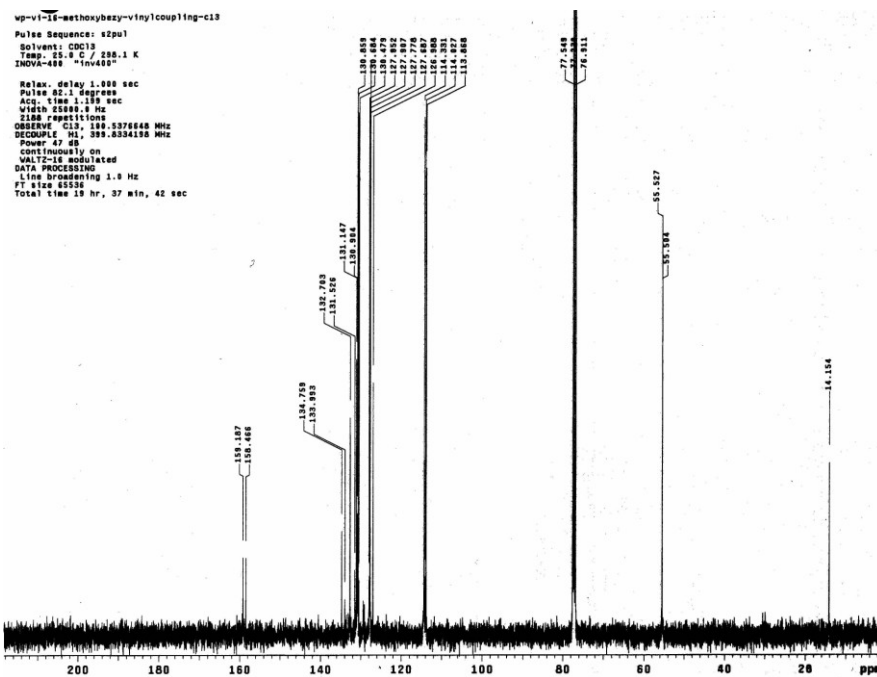
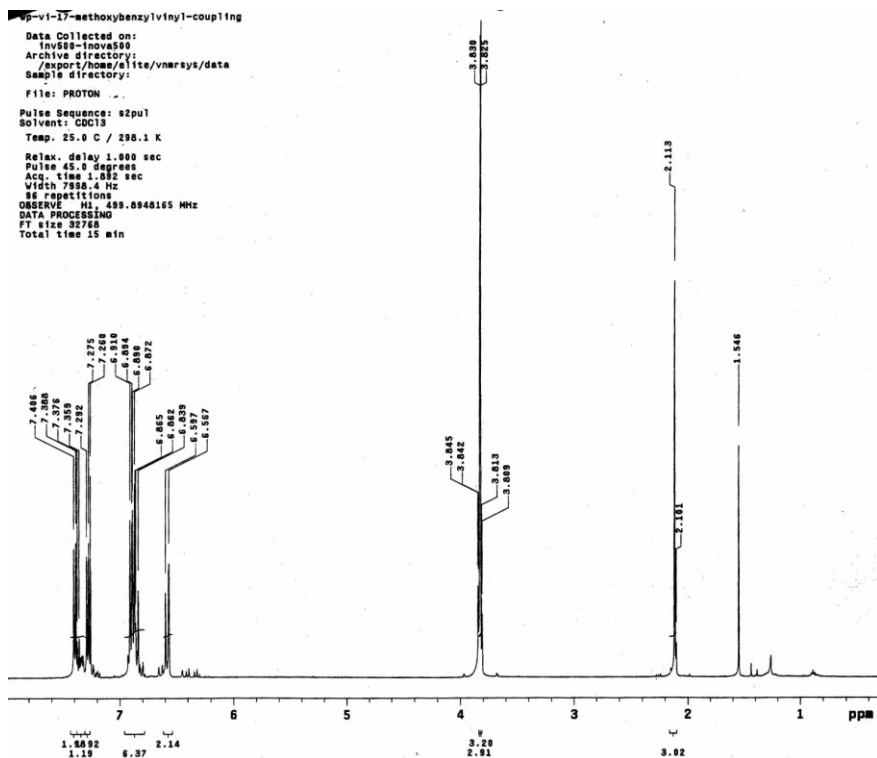
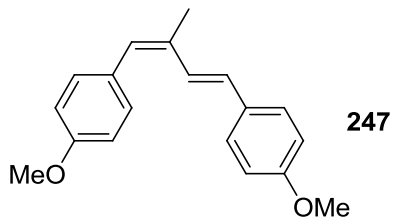


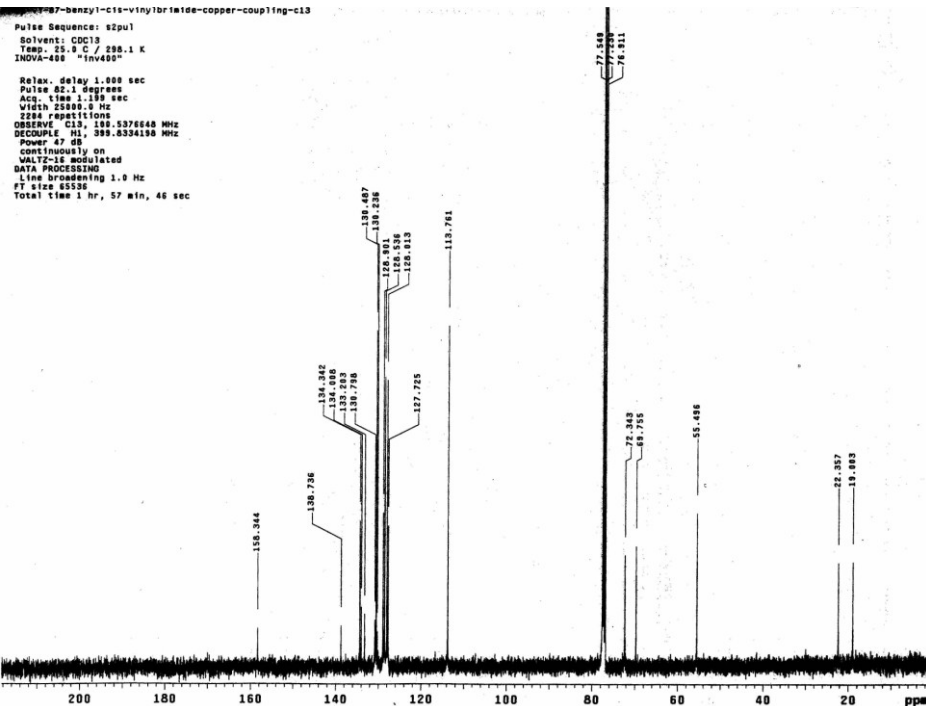
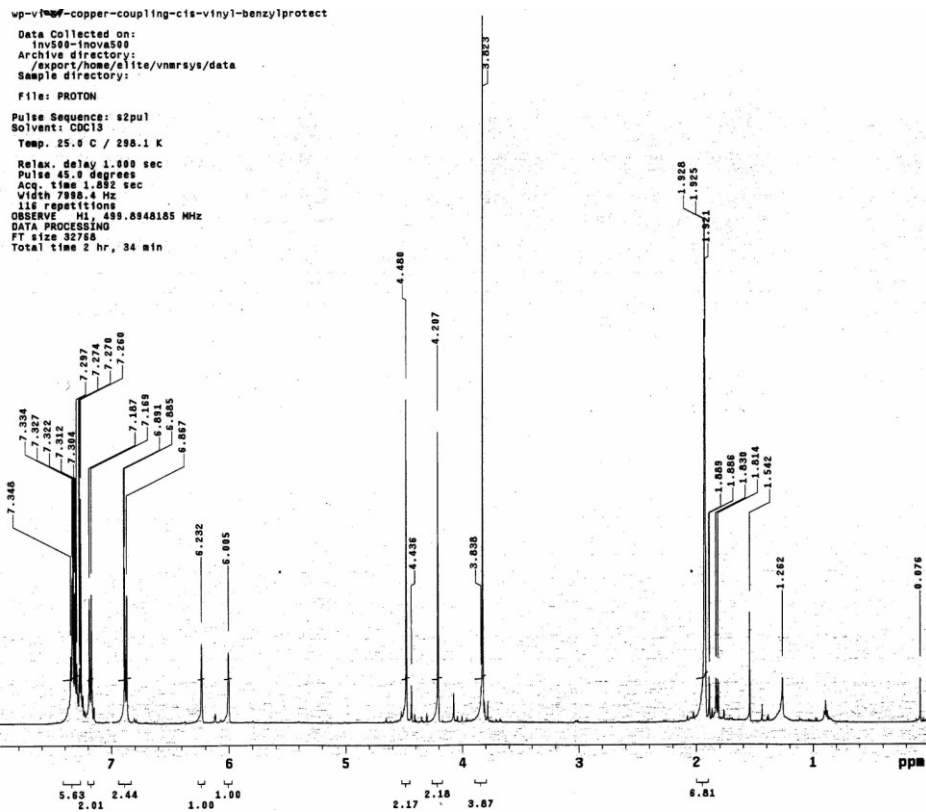
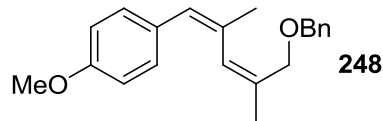
vp-vi-16-benzylallylether-coupling
 Data Collected on:
 inv508-1nova500
 Archive directory:
 /export/home/ajlita/vmarsys/data
 Sample directory:
 =
 File: PROTON
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 1.892 sec
 Width 7398.4 Hz
 86 repetitions
 OBSERVE HI, 499.8948170 MHz
 DATA PROCESSING
 FT size 32768
 Total time 15 min

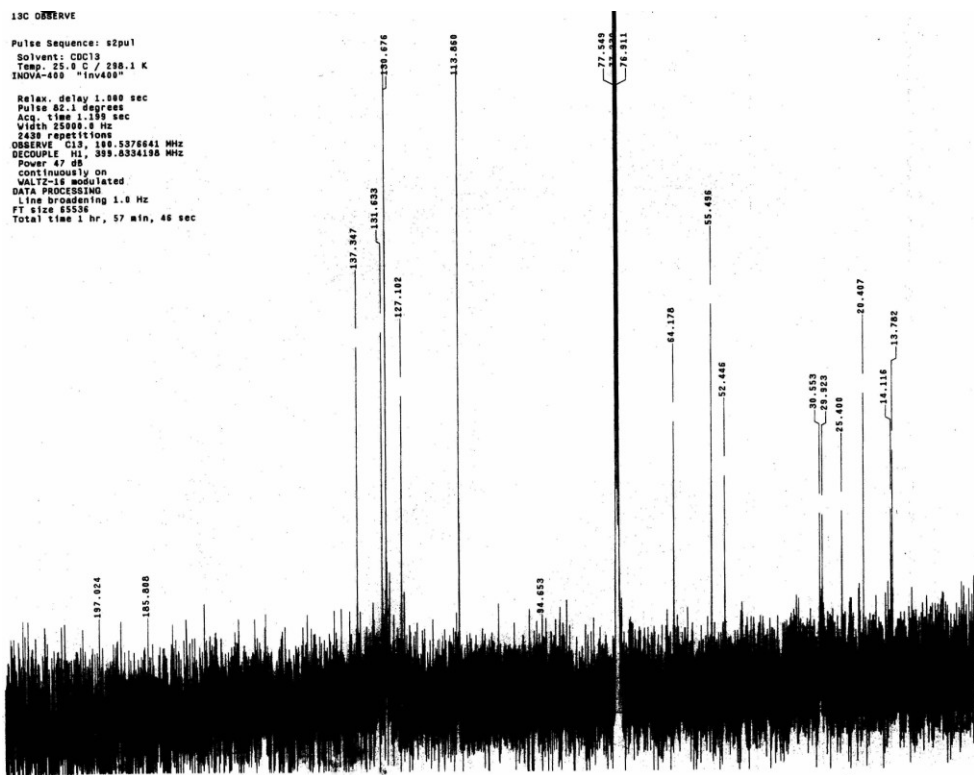
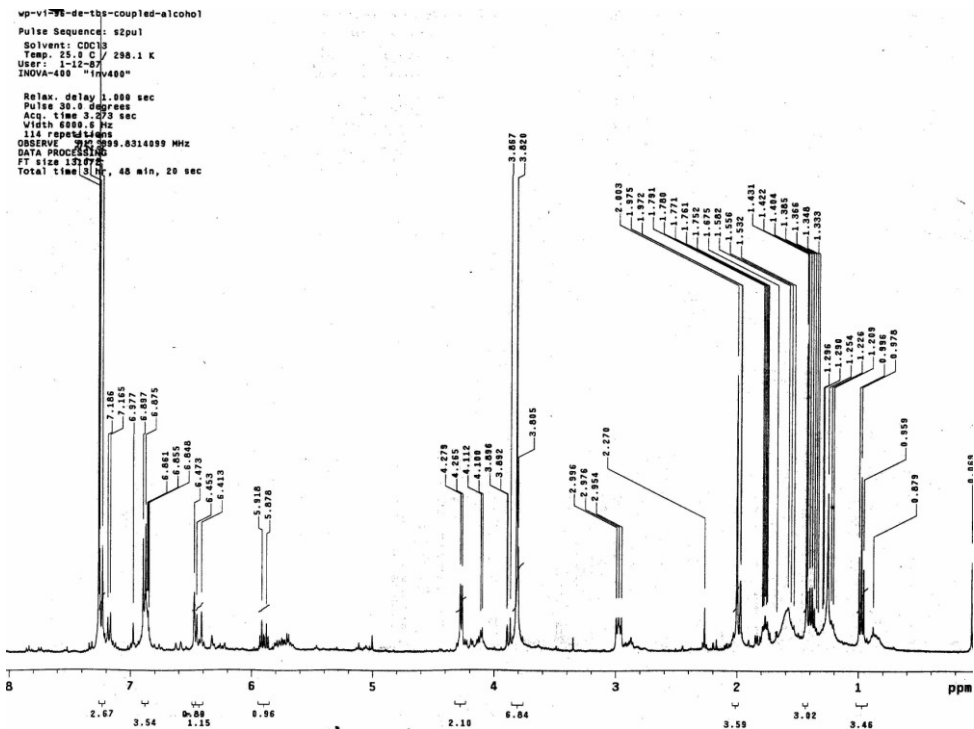
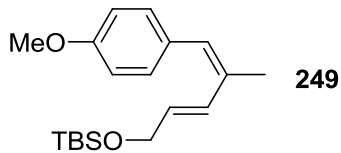


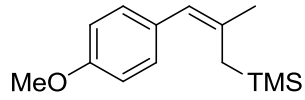
vp-vi-15-benzylallylether-coupling-55
 Pulse Sequence: s2pu1
 Solvent: CDCl3
 Temp: 25.0 C / 298.2 K
 INOVA-400 "1nova400"
 Relax. delay 1.000 sec
 Pulse 45.0 degrees
 Acq. time 2.390 sec
 Width 25888.0 Hz
 5182 repetitions
 OBSERVE HI, 148.5376648 MHz
 DECOUPLE HI, 399.8354198 MHz
 Power 47.00
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 196 hr, 17 min, 0 sec





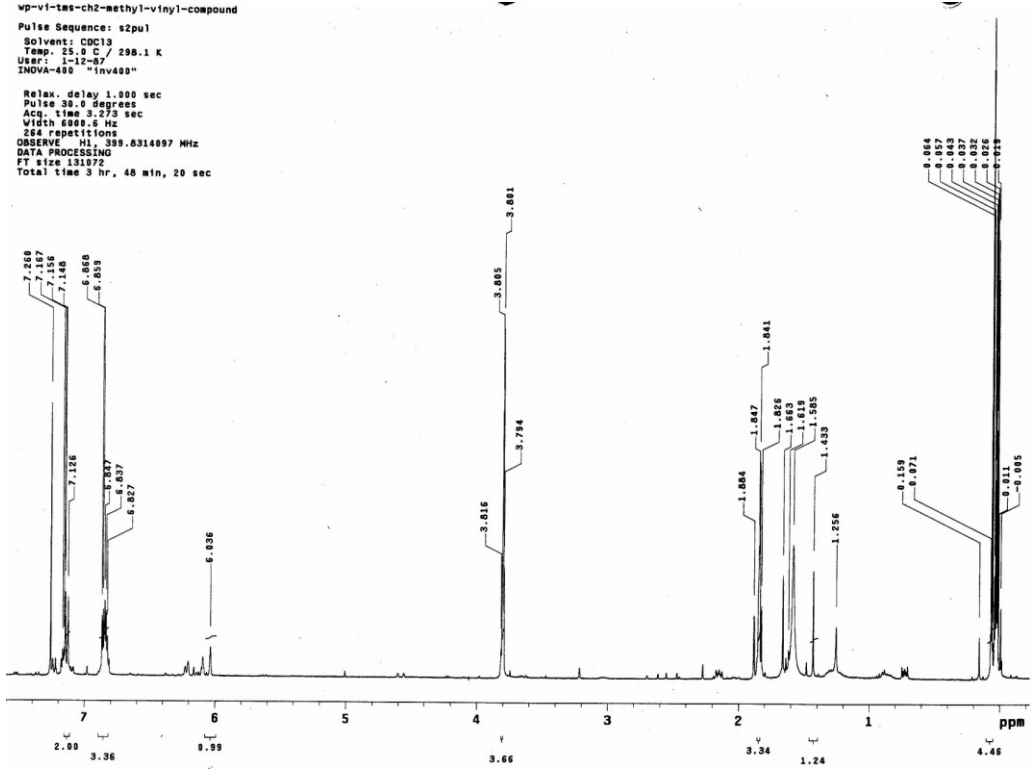




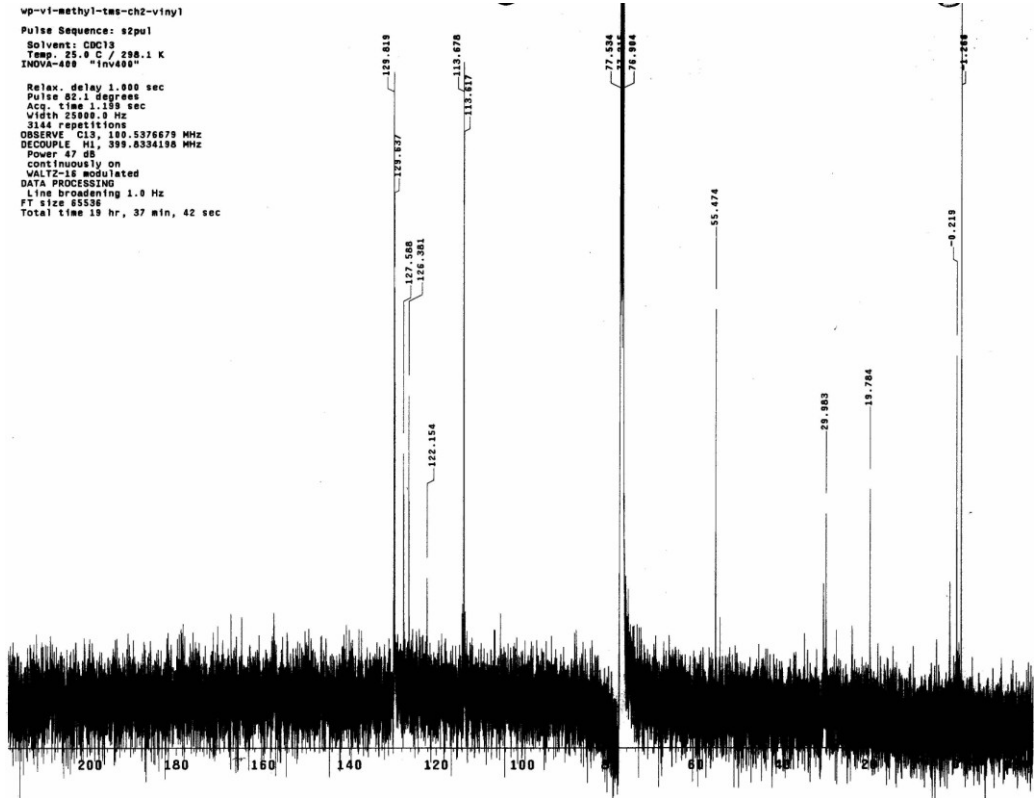


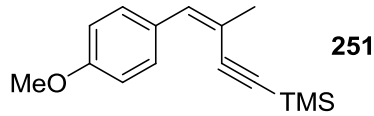
250

wp-vi-tas-ch2-methyl-vinyl-compound
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 User: 1-12-87
 INOVA-400 "Inv400"
 Relax. delay 1.000 sec
 Pulse 30.0 degrees
 Acq. time 3.273 sec
 Width 6000.0 Hz
 284 repetitions
 OBSERVE H1, 399.6314097 MHz
 DATA PROCESSING
 FT size 131072
 Total time 3 hr, 48 min, 20 sec

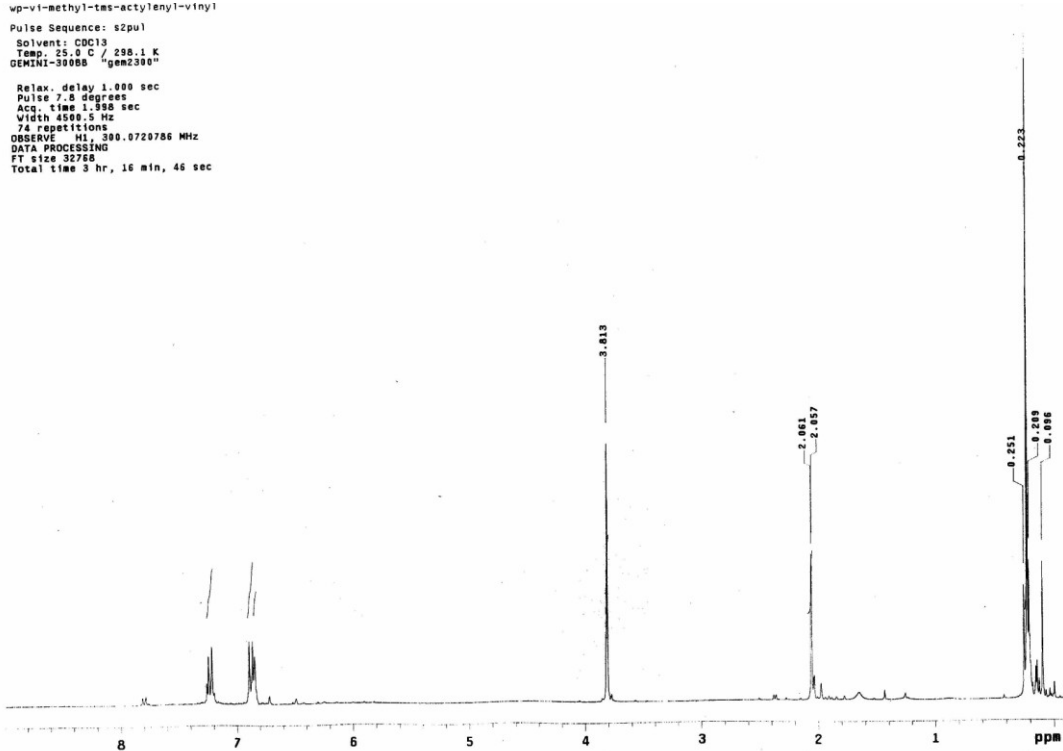


wp-vi-methyl-tas-ch2-vinyl
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp. 25.0 C / 298.1 K
 User: 1-12-87
 INOVA-400 "Inv400"
 Relax. delay 1.000 sec
 Pulse 62.1 degrees
 Acq. time 1.139 sec
 Width 25000.0 Hz
 3144 repetitions
 OBSERVE C13, 100.5376679 MHz
 DECOUPLE H1, 399.6334198 MHz
 Power 47 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 45536
 Total time 19 hr, 37 min, 42 sec





wp-vi-methyl-tms-actylenyl-vinyl
Pulse Sequence: e2pul
Solvent: CDCl3
Temp: 25.0 C / 298.1 K
GEMINI-300BB "gem2300"
Relax. delay 1.000 sec
Pulse 7.0 degrees
Acq. time 1.398 sec
Width 4500.5 Hz
74 repetitions
OBSERVE H1, 390.0720786 MHz
DATA PROCESSING
FT size 32768
Total time 3 hr, 16 min, 46 sec

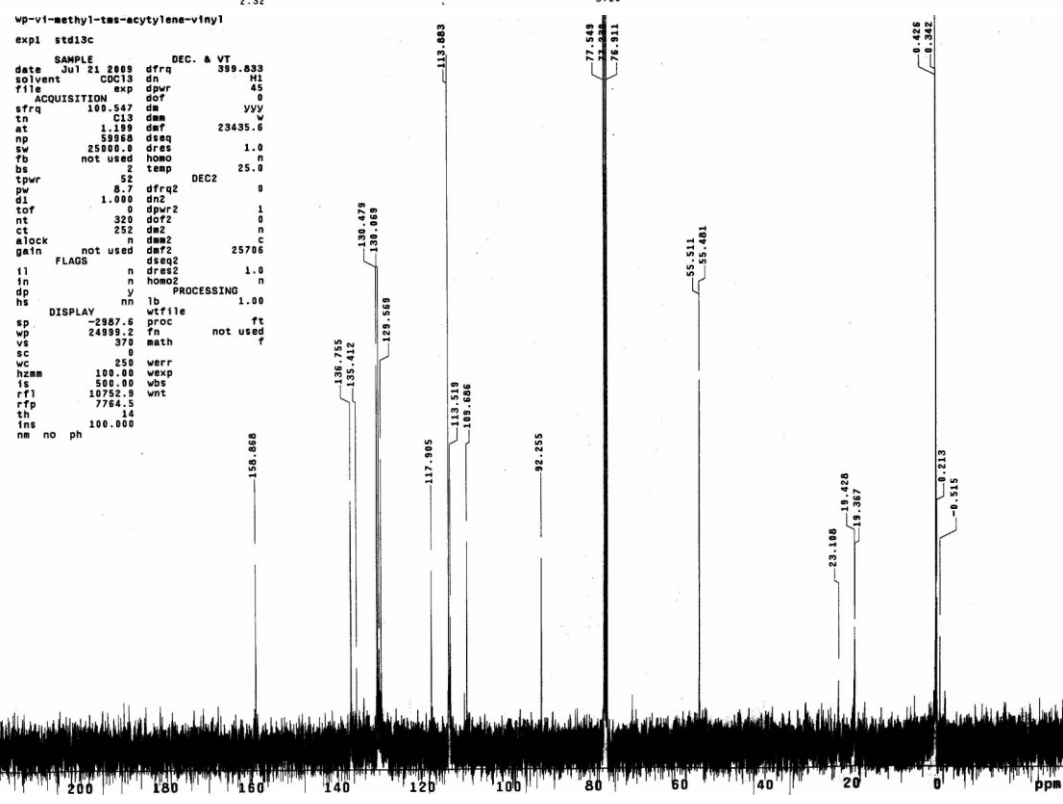


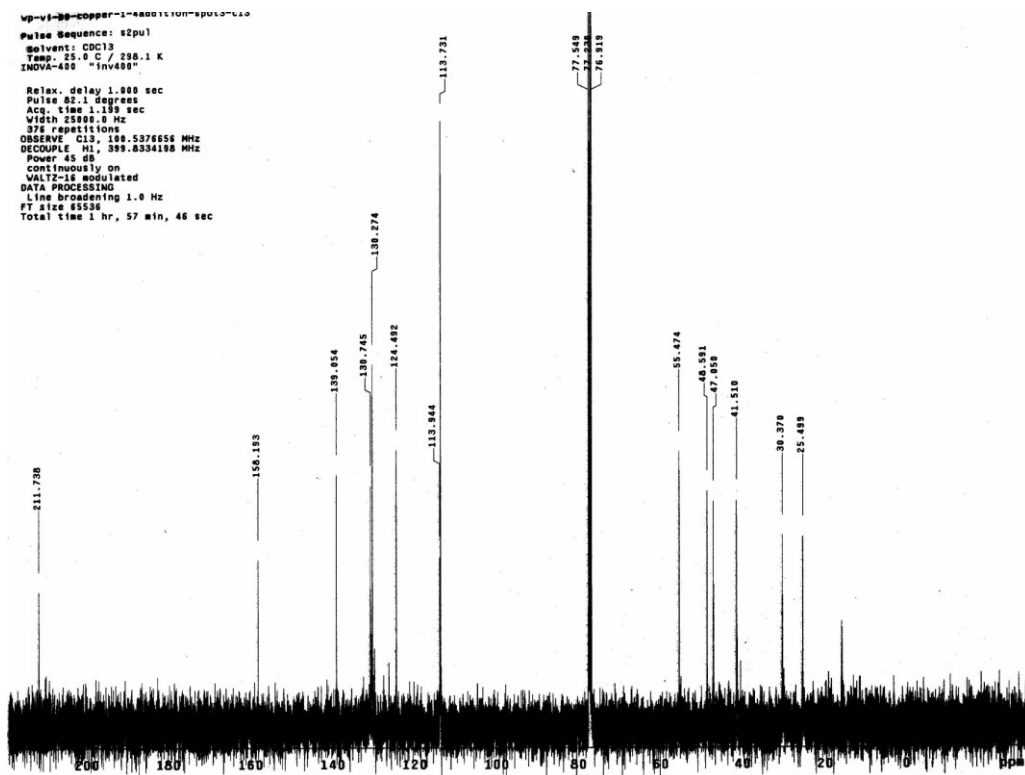
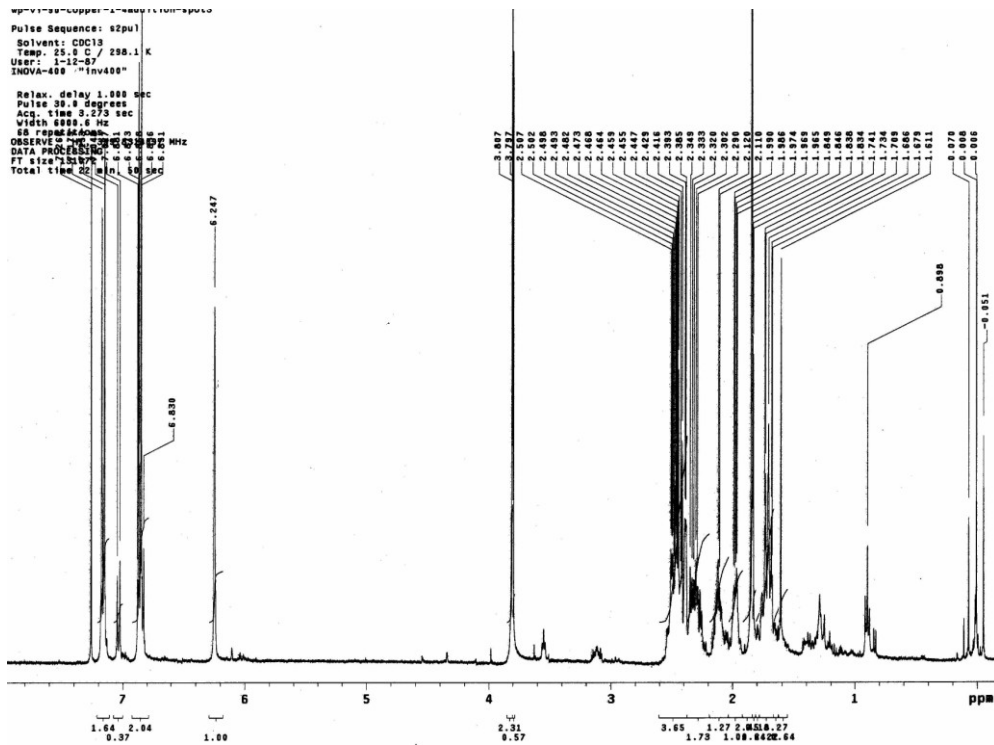
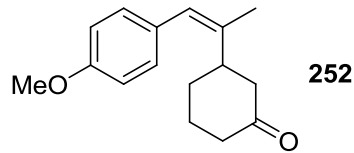
wp-vi-methyl-tms-actylene-vinyl
expl std13c

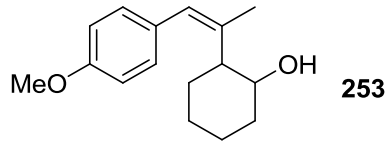
```

SAMPLE      date Jul 21 2889  dfrq  DEC. & VT  399.833
solvent     CDCl3      dn      M1
file        ACQUISITION  exp  dprw  45
            ACQUISITION  dof  0
sfrq       100.547   dm      Yyy
tn          213      dm      w
at          1.199   dm7     23435.6
np          59566   dseq    1.0
sw          25001.0 dres    25.0
fb          not used homo
bs          2       temp
spwr        52      DEC2
pw          8.7     dfrq2  0
sl          1.000   dn2
tof         0       dprw2  1
nt          320    dof2    0
ct          252    do2     n
alock      not used dma2    c
gain       not used dm7     25706
            FLAGS    n dres2  1.0
            in      n homo2  n
            gp      y  PROCESSING
            hs      nn  lb      1.00
            DISPLAY  vnfile
sp         -2987.6  proc   ft
wp         24999.2  fn    not used
vs         373    math   r
sc         0
wc         250    werr
hznm      100.00 wesp
ls         500.00 wbs
rfl       10782.0 wnt
rtp       7764.5
th        14
ins       100.000
nm no ph

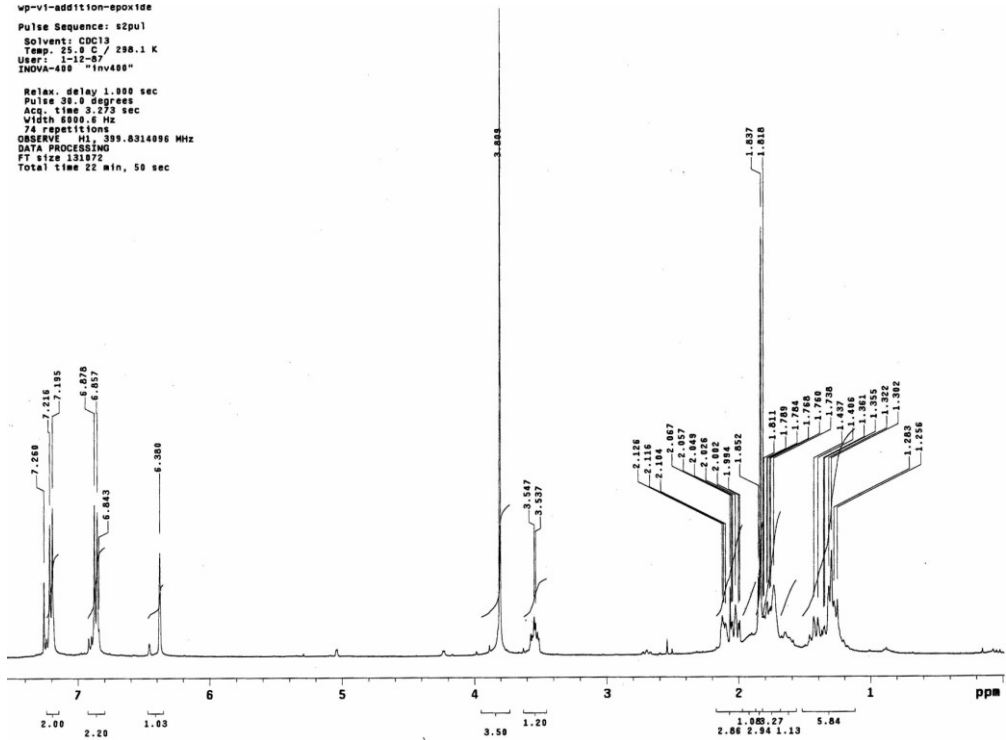
```







wp-vi-addition-epoxide
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: j-12-87
 INOVA-400 "Inv400"
 Relax. delay 1.000 sec
 Pulse 30.0 degree
 Acq. time 3.273 sec
 Width 6000.6 Hz
 74 repetitions
 OBSERVE H1, 399.8314096 MHz
 DATA PROCESSING
 FT size 131872
 Total time 22 min, 58 sec



wp-vi-epoxide-opening
 Pulse Sequence: s2pul
 Solvent: CDCl3
 Temp: 25.0 C / 298.1 K
 User: j-12-87
 INOVA-400 "Inv400"
 Relax. delay 1.000 sec
 Pulse 82.1 degree
 Acq. time 1.139 sec
 Width 25000.9 Hz
 638 repetitions
 OBSERVE C13, 100.5376664 MHz
 DECOUPLE H1, 399.8334188 MHz
 Power 47 dB
 continuously on
 VALTZ-18 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 1 hr, 57 min, 46 sec

